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Tyrosine Protonation Changes in Bacteriorhodopsin

A FOURIER TRANSFORM INFRARED STUDY OF BR548 AND ITS PRIMARY PHOTOPRODUCT*

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Paul D. Roepe‡§, Patrick L. Ahl‡¶, Judith Herzfeld∥, Johan Lugtenburg**, and Kenneth J. Rothschild‡‡‡

From the ‡Departments of Physics and Physiology, and the Program in Cellular Biophysics, Boston University, Boston, Massachusetts 02215, the ||Department of Chemistry, Brandeis University, Waltham, Massachusetts 02154, and the **Department of Chemistry, Leiden University, 2300 RA Leiden, The Netherlands

The structural alterations which occur in bacteriorhodopsin (bR) during dark adaptation (BR₅₇₀→BR₅₄₈) and the primary phototransition of the dark photocycle $(BR_{548} \rightarrow K_{610}^D)$ have been investigated by Fourier transform infrared and UV difference spectroscopy. Possible contributions of tyrosine to the Fourier transform infrared difference spectra of these transitions were assigned by incorporating ring per-deuterated tyrosine into bR. Based on these data and UV difference measurements, we conclude that a stable tyrosinate exists in BR₅₇₀ at physiological temperature and that it protonates during formation of BR₅₄₈. A tyrosinate protonation has also been observed at low temperature during the primary phototransition of BR₅₇₀ to the redshifted photoproduct K_{630} (1). However, we now find that no tyrosine protonation change occurs during the primary phototransition of BR₅₄₈ to the red-shifted intermediate K_{610}^D . Through analysis of bR containing isotopically labeled retinals, it was also determined that the chromophore of K_{610}^D exits in a 13-trans,15cis configuration. On the basis of this evidence and previous studies on the structure of the chromophore in BR₅₇₀, BR₅₄₈, and K₆₃₀, it appears that only the 13trans, 15-trans configuration of the protonated chromophore leads to a stable tyrosinate group. It is proposed that a tyrosinate residue is stabilized due to its interaction with the Schiff base positive charge in the BR₅₇₀ chromophore. Isomerization of the chromophore about either the C₁₃=C₁₄ or C=N bond disrupts this interaction causing a protonation of the tyrosinate.

Bacteriorhodopsin (bR),¹ the 26-kDa retinal-containing protein of the purple membrane of Halobacteria halobium,

functions as a light-driven transmembrane proton pump (2). Photon absorption by the light-adapted state of purple membrane (BR $_{670}$) results in the initiation of a photochemical cycle which is characterized by several distinct intermediates that have different visible absorption maxima. The formation of the early photointermediate K_{630} involves a C_{13} = C_{14} trans to cis isomerization of the retinylidene chromophore (3). Protein changes during K_{630} formation have been detected by FTIR and UV spectroscopy. These include a tyrosinate protonation (1, 4, 5), tryptophan perturbation (1), and the perturbation of at least one carboxyl group (6, 7).

In the dark, at least 50% of the light-adapted BR_{570} species converts to a blue-shifted form designated BR548. Recent resonance Raman and NMR studies indicate that in contrast to the 13-trans, 15-trans configuration of the retinylidene chromophore of BR₅₇₀, the BR₅₄₈ chromophore exists in a 13cis,15-cis configuration (8, 9). The photocycle of BR₅₄₈, as shown in Fig. 1 (10-12), also exhibits an early red-shifted intermediate, K₆₁₀.² However, the "dark photocycle" which occurs upon light absorption by BR548 does not result in the formation of a blue-shifted "M"-like intermediate or result in proton pumping as found for the photocycle of the lightadapted BR₅₇₀ species. In general, a comparison of the molecular changes that accompany the $BR_{548} \rightarrow K_{610}^D$ and $BR_{570} \rightarrow$ K₆₃₀ transitions may help explain why both transitions produce a red-shifted intermediate yet only one leads to proton pumping.

In this work, we have used FTIR spectroscopy in combination with isotopic labeling of both the retinal chromophore and bR tyrosine residues in order to study the molecular changes occurring in these moieties during both dark adaptation (the BR₅₇₀ \rightarrow BR₅₄₈ transition) and during the primary phototransition of the dark photocycle (the BR₅₄₈ \rightarrow K^D₅₁₀ transition). In agreement with earlier reports (4, 5), it is found that dark adaptation involves the protonation of a tyrosinate group. In contrast, a tyrosine protonation is not observed during the BR₅₄₈ \rightarrow K^D₅₁₀ transition. We have also determined that the retinylidene chromophore in K^D₆₁₀ exists in a 13-trans, 15-cis configuration. On the basis of this evidence it appears

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[§] Currently a Jane Coffin Childs postdoctoral fellow for medical research. Current address: Dept. of Biochemistry, Roche Institute of Molecular Biology, Nutley, NJ 07110.

[¶] Current address: Bio/molecular Engineering Branch, Naval Research Laboratory, Washington, D. C. 20375.

^{‡‡} To whom correspondence should be addressed: 590 Commonwealth Ave., Dept. of Physics, Boston University, Boston, MA 02215.

¹ The abbreviations used are: bR, bacteriorhodopsin; bR-[²H₄]Tyr, bR containing ring per-deuterated tyrosines; FTIR, Fourier transform infrared; HOOP, hydrogen-out-of-plane; RPSB, retinal-protonated Schiff base.

 $^{^2}$ The primary photoproduct of BR₅₄₈ formed at 77 K is referred to by Iwasa et~al.~(12) as batho-13-cis-bR in analogy to the K₆₃₀ photoproduct of BR₅₇₀, batho-trans-bR. The first photoproduct detected by the flash kinetic absorption studies of Dencher et~al.~(10) at $-90~{\rm ^{\circ}C}$ has been designated C* and the decay product at this temperature, C₆₁₀. C refers to the "cis" photocycle of BR₅₄₈ in contrast to the "trans" photocycle of BR₅₇₀. In this paper, we use the designation K^D₆₁₀ (i.e. the K photoproduct of the dark photocycle) in order to stress the analogy to the K photoproduct of BR₅₇₀. This name also avoids potential confusion which might arise from designations which imply a cis or trans structure of the intermediate, which is under investigation in these studies.

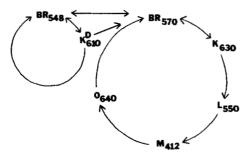


Fig. 1. A simplified representation of cyclic BR_{548} and BR_{670} photochemistry. The subscripts denote the visible absorption maxima of the intermediates. At room temperatures, K forms in less than 6 ps, L in 2 μ s, M in 40 μ s, and BR_{570} reforms in ~10 ms. Precise rate constants for the dark cycle have not been determined.

that the existence of a tyrosinate in bR depends on the existence of a 13-trans,15-trans retinal configuration and that it is stabilized by interaction with the positive charge on the Schiff base.

EXPERIMENTAL PROCEDURES

Materials—The production of bR containing L ring per-deuterated tyrosines (bR-[2H4]Tyr) has been described previously (1, 13, 14). Purple membrane sheets containing either normal bR or bR-[2H4] Tyr were isolated by the method of Oesterhelt and Stoeckenius (15). White membrane containing bacterio-opsin was isolated from a retinal-deficient strain of *H. halobium* (JW5) as reported previously (6). Regeneration of bacterio-opsin with $^{13}C_{14}$ - $^{13}C_{16}$ - or ^{2}H - C_{15} -labeled retinals (16) was done in dim red light at room temperature and was monitored with a Cary 219 UV-visible spectrophotometer. Membrane fragments were washed in distilled water and deposited on AgCl windows by the isopotential spin-dry method (17) or by slow air drying of a concentrated suspension. Prior to loading the sample in a sealed cell, purple membrane films were rehydrated by direct exposure to liquid H₂O or ²H₂O for 2-5 min, and the excess liquid was shaken off. The amount of H₂O or ²H₂O in the film was monitored by comparing the integrated intensities of the bands at 3300 or 2500 cm⁻¹ to that of the amide I at 1660 cm⁻¹. Ratios of 6:1 or higher were taken as indicative of adequate hydration.

FTIR Difference Spectroscopy—BR₅₇₀—BR₅₄₈ difference spectra were computed by subtracting an FTIR spectrum of a hydrated purple membrane film recorded at 310 K while the film was illuminated by 550-nm light (550-nm broad band filter, Ditric Optics, Hudson, MA) for 15 min from a spectrum taken at the same temperature while the same film was in the dark for 15 min. As many as 25 of these differences were averaged together. By comparing the intensities of peaks in successive individual differences, it was found that dark adaptation of the film was complete before the end of a 15-min dark scan at 310 K.

BR₅₇₀→K₆₃₀ FTIR difference spectra were obtained as reported previously (1, 6, 14) from completely light-adapted films at 81 K, and an analogous procedure was used to obtain spectra at 81 K from a dark-adapted film. The procedure used to compute the pure BR₅₄₈— K_{610}^{D} difference spectrum is described under "Results." Comparison of individual differences revealed that all the reactions observed were completely reversible under the illuminations used and that the intermediates observed were thermally stable at 81 K. Light adaptation of films was accomplished by illuminating the sample with 550nm light for 15 min at 310 K. In order to avoid trapping undesired photointermediates, this illumination was shut off at 273 K while the film was being cooled rapidly (~2 K/s) to 81 K. All measurements were made on a Nicolet MX-1 spectrometer interfaced to a Nicolet 1200S data analysis terminal, and experimental variables were controlled using a Forth program developed in our laboratory. Each FTIR scan was 15 min long and resulted in the acquisition of 480 interferograms which were multiplied by a Happ-Genzel instrument function before Fourier transformation. Interferometer mirror velocity was 0.85 s/cm, and final effective spectral resolution was 2 cm⁻¹

UV-visible Difference Spectroscopy—The UV-visible measurements were made as previously described (1, 13, 14) on fully hydrated purple membrane films. These films were formed by slow air drying of a concentrated purple membrane suspension on a quartz substrate which was then hydrated and sealed in a specially constructed cell.

The spectra were recorded at 1-nm resolution with a Cary 219 spectrophotometer equipped with a Janis (Waltham, MA) liquid nitrogen cryostat and interfaced to an Apple IIe computer. Room temperature light adaptation was done with a 600-watt tungsten lamp in combination with a Schott OG475 glass filter (Duryea, PA) and two heat filters (Edmund Scientific, Camden, NJ). Difference spectra were recorded at 80 K from either light-adapted or dark-adapted fully humidified films. Samples were illuminated using a 1-kilowatt Hg-Xe lamp in combination with a monochromator and either a 540- or 640-nm narrow band interference filter (Ditric Optics, Hudson, MA).

RESULTS

Dark Adaptation

Figs. 2 and 3 compare FTIR difference spectra obtained for dark adaptation of normal bR (solid) and bR-[²H₄]Tyr (dashed). These spectra represent the BR₅70→BR₅48 reaction. Several small but reproducible changes are caused by the isotopic label. In the 1450 to 1550 cm⁻¹ region (Fig. 3A), two small positive peaks, one centered at about 1510 cm⁻¹ and the other at 1453 cm⁻¹ (cf. Fig. 3A), are reduced in the bR-[²H₄] Tyr difference spectrum, possibly shifting to near 1470 and 1415 cm⁻¹ due to isotope labeling. Consistent with this interpretation, a comparison of the IR spectra of unlabeled tyrosine and [²H₄]Tyr reveals that tyrosine modes at 1514 cm⁻¹ (aromatic ring C=C stretching vibration(s)) and 1456 cm⁻¹ shift due to ring per-deutero labeling (1, 13, 18) in a very similar manner to that seen in the difference spectra of bR and bR-[²H₄]Tyr.

In the 1290–1210 cm⁻¹ region (Fig. 3B) tyrosine labeling causes a disappearance of the negative peak at 1276 cm⁻¹ concomitant with an increase in negative intensity between 1249 and 1230 cm⁻¹. This effect is consistent with the ~25 cm⁻¹ downshift of the phenolate C–O⁻ stretching mode near 1270 cm⁻¹ in tyrosinate due to ring per-deuteration (1, 18). An increase in positive intensity in the bR-[2 H₄]Tyr difference spectrum seen near 1212 cm⁻¹ may also be due to the downshift of a tyrosine C–OH stretch band centered at 1234 cm⁻¹ in unlabeled bR.

The Primary Photoproduct of BR₅₄₈

The Configuration of the K_{610}^D Chromophore—Fully darkadapted films contain as much as 67% BR₅₄₈ (19). In order to obtain an FTIR difference spectrum of the primary photoreaction of BR₅₄₈, we subtracted a difference spectrum for the primary photoreaction of a fully light-adapted film (BR₅₇₀-> K₆₃₀) (Fig. 4, solid) from that for a completely dark-adapted film obtained under identical conditions. Scaling during the subtraction was done interactively, so that the ethylenic stretching modes of the BR₅₇₀ and K₆₃₀ chromophores at 1528 and 1515 cm⁻¹, respectively, vanished. This procedure led to the disappearance of most other BR₅₇₀ and K₆₃₀ chromophore peaks, and it enhanced negative peaks associated with BR548 chromophore vibrations (20, 21) including the Schiff base C=N stretch at 1636 cm⁻¹, the ethylenic C=C stretch at 1536 cm⁻¹, the NH rock at 1346, and C-C stretches at 1234 and 1185 cm⁻¹. Peaks in the resonance Raman spectrum of BR₅₄₈ at 1202 and 1167 cm⁻¹ assigned to C-C stretches (21) are not observed, most likely because they are cancelled by positive peaks due to K_{610}^D and 1205 and 1177 cm⁻¹, respectively. The positive ethylenic line at 1521 cm⁻¹ falls close to the

The positive ethylenic line at 1521 cm⁻¹ falls close to the frequencies expected for a species (K_{510}^D) with an absorption maximum near 610 nm, based on the linear correlation between the λ_{max} and the C=C stretching frequencies of bR photocycle intermediates (22). Other chromophore lines at 1315, 1205, 1177, and 1004 cm⁻¹ can tentatively be assigned to the 12C-H in-plane rock, the 14-15 and 10-11 C-C stretches, and the 19CH₃ rock, respectively, based on previous

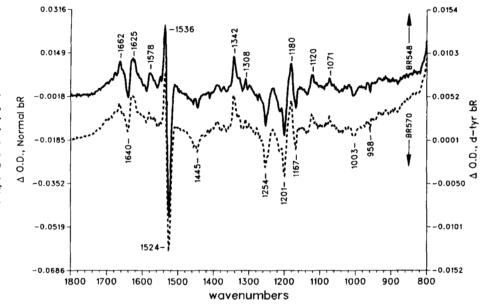


FIG. 2. Comparison of FTIR difference spectra of the light (BR₅₇₀, negative peaks) to dark (BR₅₄₈, positive peaks) transition at 310 K for normal bR (solid line) and bR containing ring per-deuterated tyrosines (dashed line) in the 1800-800 cm⁻¹ region. Spectra are at 2 cm⁻¹ resolution and are unsmoothed. Scaling of the spectra is done by matching the intensities of the chromophore peaks.

work with all-trans and 13-cis isotopically labeled retinals (20, 21, 23) and the data presented below. The protonated Schiff base C=N stretch of K_{610}^D is surmised to fall near 1622 cm⁻¹ because of the effects in the 1660–1550 cm⁻¹ region produced by $^2\text{H-C}_{15}$ and $^2\text{H-N}$ labeling (data not shown). This frequency is very close to that expected for a 610-nm absorbing species from a linear plot of the C_{15} =N stretch versus the visible λ_{max} of photocycle intermediates (6).

In contrast to the intense positive hydrogen-out-of-plane (HOOP) modes observed in the $1000-800 \text{ cm}^{-1}$ region for the primary photointermediates of rhodopsin (24) and light-adapted bacteriorhodopsin (3) (cf. Fig. 4, top) by both resonance Raman and FTIR spectroscopy (25–27) weak HOOPS are observed for K_{610}^D (cf. Fig. 4, bottom). The intensification of HOOP modes has been attributed to significant single bond twisting of the chromophore. In view of the similarity between the resonance Raman and FTIR difference spectra for the $BR_{570} \rightarrow K_{630}$ transition (27), the absence of intense HOOP modes in the $BR_{548} \rightarrow K_{610}^{B1}$ difference spectrum may indicate a more relaxed chromophore for K_{610}^{B1} . However, other possible mechanisms for a decrease in the IR activity of these modes cannot be completely excluded at this time.

In order to deduce the configuration of the K^D₆₁₀ chromophore, spectra of this transition were obtained for bR containing isotopically labeled retinals. Fig. 5 presents expansions of the $BR_{548} \rightarrow K_{610}^{D}$ spectrum in the fingerprint region for normal bR in H2O and 2H2O (A and B), bR regenerated with ${}^{13}C_{14}$ - ${}^{13}C_{15}$ retinal in H_2O and ${}^{2}H_2O$ (C and D), and bR regenerated with ²H-C₁₅ retinal in H₂O (E). It is known, through comparison with model retinal-protonated Schiff base (RPSB) compounds, that certain 2H-C15 effects on the fingerprint region of bR spectra are highly dependent on the configuration of the chromophore $C_{13}=C_{14}$ bond (3). Also, the magnitude of the upshift of the C₁₄-C₁₅ single bond stretch upon ²H/H exchange at the Schiff base nitrogen is known to be highly dependent on the C₁₅=N bond conformation (8). Thus, analysis of the spectra from the above samples can be used to determine the retinal configuration for K_{610}^D .

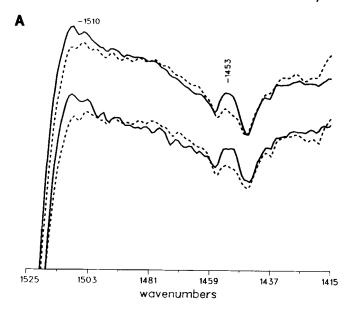
We first note that the prominent positive peak at $1205~\rm cm^{-1}$ can be assigned to the C_{14} – C_{15} stretch of K_{01}^{D} on the basis of its shift upon $^{13}C_{14}$ – $^{13}C_{15}$ labeling (cf. Fig. 5C). As seen in Fig. 6, this peak for a 610-nm absorbing species falls on an empirical straight line correlation between the C_{14} – C_{15} stretching

frequency and the λ_{max} of 13-trans intermediates and the 13trans RPSB. In contrast, all of the points which correspond to intermediates known to be 13-cis (as well as the point corresponding to the 13-cis RPSB) do not fall near this line or exhibit a straight line correlation. The latter may be due to uncertainty in assignment of the C14-C15 stretch. For example, the shoulder near 1185 cm⁻¹ in K₆₃₀ also contains significant C₁₄-C₁₅ character (28); thus this point may actually fall further from the all-trans line. Uncertainty also exists about the assignment of the position of the L_{550} C_{14} – C_{15} stretch mode, which has also been assigned near 1155 cm⁻¹ (28) and attributed to a cis conformation about the C14-C15 bond. Since the 13-trans straight line correlation is based on only a few data points, the conclusion that the K_{610}^D species contains a 13-trans chromophore is not compelling. However, additional considerations discussed below also support this conclusion.

Table I summarizes the effects $^2\text{H-C}_{15}$ substitution of the chromophore has on the K_{610}^D spectrum and compares them to the effects previously seen in spectra of the 13-cis and 13-trans RPSB model compounds (3). It is clear from examination of this table that the $^2\text{H-C}_{15}$ substitution effects, which are diagnostic for the C_{13} = C_{14} configuration, are more 13-trans-like than 13-cis-like. In particular, the K_{610}^B HOOP mode alterations upon $^2\text{H-C}_{15}$ substitution include a decrease in intensity of the weak line at 877 cm $^{-1}$. This is similar to the disappearance of the 870 cm $^{-1}$ line in the trans model compound. The major HOOP effects seen in the cis model compound include the apparent shift of a line at 972 cm $^{-1}$ to 975 cm $^{-1}$, as well as the dramatic appearance of an intense new line at 988 cm $^{-1}$. None of these effects are seen in the K_{610}^D spectrum.

In the cis model compound spectrum, a line at $1174~\rm cm^{-1}$ shifts to $1168~\rm cm^{-1}$; however, in the K_{610}^{D} spectrum the line at $1177~\rm cm^{-1}$ does not shift upon $^2H-C_{15}$ substitution (cf. Fig. 5, A and E). Furthermore, in the cis model compound, a line at $1206~\rm cm^{-1}$ does not shift upon $^2H-C_{15}$ substitution (3), but the $1205~\rm cm^{-1}$ lines does shift in the K_{610}^{D} spectrum (cf. Fig. 5E). Also, the decrease at $1277~\rm cm^{-1}$ concomitant with the appearance of a small peak at $1262~\rm cm^{-1}$ in the $^2H-C_{15}~K_{610}^{D}$ spectrum (cf. Fig. 5E) is quite similar to effects at $1279~\rm and~1271~\rm cm^{-1}$ in the trans model compound. Finally, other effects such as the decrease at $1412~\rm cm^{-1}$ and the increase at $1600~\rm cm^{-1}$ are much more similar to the effects seen in the 13-trans RPSB than the 13-cis RPSB.

While complete agreement between the K_{610}^D spectrum and



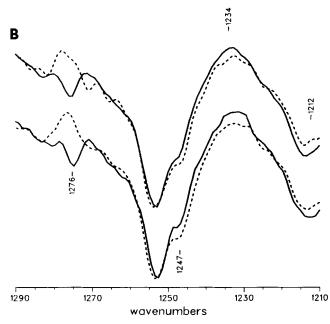


FIG. 3. A, expanded overlay of the spectra in Fig. 1 in the 1525–1415 cm⁻¹ region. Two additional spectra of two other samples are also shown to demonstrate reproducibility. B, same as in A but in the 1290–1210 cm⁻¹ region.

the 13-trans model compound is not observed, the environment surrounding the chromophore in bR as well as the configuration of the retinal C=N bond (see below) most likely influence the electronic character of many bonds, thus affecting the frequencies and coupling of many modes. However, the generally good agreement between the K_{610}^D and 13-trans RPSB model compound $^2\text{H-C}_{15}$ effects (as well as the disagreement between the K_{610}^D and 13-cis model compound $^2\text{H-C}_{15}$ spectra) and the empirical correlation seen in Fig. 6 support but do not unequivocally establish a 13-trans assignment for K_{610}^D .

Assignment of the C_{14} – C_{15} stretch and observation of its behavior upon N-²H substitution can also be used to determine the C=N bond configuration (8). When the C_{15} =N bond is cis, a significant upshift in the C_{14} – C_{15} single bond stretch of the chromophore is observed upon deutero substitution at the Schiff base nitrogen. Conversely, when the C_{15} =N bond is trans, shifts of less than 5 cm⁻¹ are seen because of de-

creased coupling between the individual modes. As noted, comparison of spectra in Fig. 5, A and C, indicates the $K_{610}^{\rm E}$ C_{14} – C_{15} stretch is localized to a large degree in a band at 1205 cm⁻¹, as this line is seen to shift down due to di-¹³C labeling, very likely to near 1195 cm⁻¹. (A similar 10 cm⁻¹ shift has previously been observed in model compounds, cf. Ref. 21.) Comparison of spectra in Fig. 5, A and B, indicates the 1205 line has upshifted to 1205 cm⁻¹ upon deutero substitution at the Schiff base nitrogen, this line (1234 cm⁻¹) and then appears to downshift to 1224 cm⁻¹ for the di-¹³C sample in 2 H₂O (spectrum D). The magnitude of this downshift is exactly that of the 1205 cm⁻¹ line upon di-¹³C labeling. Since the C_{14} – C_{15} stretch upshifts 30 cm⁻¹ upon 2 H–N substitution, we conclude the C_{15} =N bond is cis in K_{610}^{B} .

Protein Groups Involved in the Formation of K_{510}^D —A notable feature of the BR_{548} — K_{610}^D difference spectrum is the absence of peaks near 853, 842, and 833 cm⁻¹ due to the Fermi resonance doublet of tyrosine (1) (cf. Fig. 7A, top dashed line) which have previously been associated with the protonation of a tyrosinate group in the BR_{570} — K_{630} transition (cf. Fig. 7, solid lines). These peaks disappear upon introduction of perdeuterated tyrosine into bR, as seen in Fig. 7B (bottom dashed line). The Fermi doublet peaks are those tyrosine vibrations most sensitive to protonation changes. Their absence in the BR_{548} — K_{610}^B difference spectrum indicates no tyrosine group undergoes a protonation change during this reaction.

Additional protein change(s) involving one or more peptide amide groups is indicated by the prominent negative peak at $1660~\rm cm^{-1}$ and a smaller $1668~\rm cm^{-1}$ band (cf. Fig. 4). The major amide I band of bR is observed at $1660~\rm cm^{-1}$, and thus the peak in the difference spectrum may indicate a perturbation in amide structure during this transition. A peak at $1662~\rm cm^{-1}$ is observed upon formation of BR₅₄₈ from BR₅₇₀ (c.f. Fig. 2) and may reflect a change in the same peptide groups. Since the C=N stretch of the BR₅₄₈ retinal Schiff base has previously been assigned at $1636~\rm cm^{-1}$ (20, 21), we can exclude the possibility that the negative $1660~\rm cm^{-1}$ peak is due to this mode.

We also note that small negative and positive peaks at 1740 and 1730 cm⁻¹ due to the carboxyl COOH stretches of aspartic acid or glutamic acid, seen in $BR_{570} \rightarrow K_{630}$ difference spectra (6, 7), are also seen in the $BR_{548} \rightarrow K_{610}^D$ difference spectrum (cf. Fig. 4). These peaks are due either to a hydrogen bond perturbation of a carboxyl group or the simultaneous protonation/deprotonation of two carboxyls.

UV-visible Difference Spectra

Fig. 8A and the expansion in Fig. 8B show the UV-visible difference spectrum for light adaptation (the BR₅₄₈ → BR₅₇₀ transition). The film was kept in total darkness for over 2 h before being illuminated with yellow light (cf. "Experimental Procedures") to produce the light-adapted BR₅₇₀ state. Unlike the UV differences previously measured (1, 13) for the BR₅₇₀ \rightarrow K₆₃₀ and M₄₁₂ transitions, the dominant feature in the UV region for light adaptation is a positive peak at 275 nm. Retinal model compound studies indicate that this peak is not due to the chromophore (30) but is consistent with a tryptophan absorption.3 Two positive peaks are also observed at 244 and 304 nm which are consistent with the deprotonation of a tyrosine during the BR₅₄₈ → BR₅₇₀ transition. However, the intensity of the 243-nm peak relative to the 304-nm peak is smaller than expected on the basis of model compound tyrosine deprotonation difference spectra (1). This may be due to cancellation near 243 nm of retinal contributions at

³ P. L. Ahl and K. J. Rothschild, unpublished data.

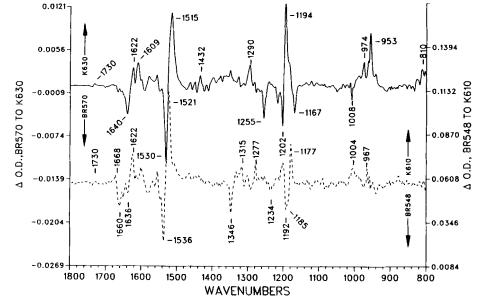


FIG. 4. Comparison of the FTIR difference spectra of the $BR_{570} \rightarrow K_{630}$ (solid) and $BR_{548} \rightarrow K_{610}^{p}$ (dashed) transitions in the 1800–800 cm⁻¹ region. The $BR_{548} \rightarrow K_{610}^{p}$ spectrum is computed by subtracting the $BR_{570} \rightarrow K_{630}$ spectrum from a $BR \rightarrow K$ spectrum obtained with a completely dark-adapted film (see "Experimental Procedures").

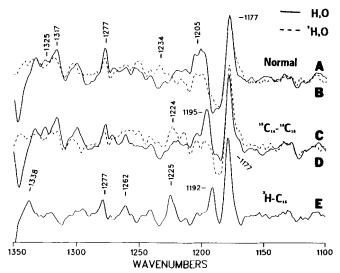


FIG. 5. FTIR difference spectra of the BR₅₄₈ \rightarrow K⁰₁₀ transition in the fingerprint region for bR in H₂O (A, solid line), and in 2 H₂O (B, dashed line), bR regenerated with 13 C₁₄ $^{-13}$ C₁₅ retinal in H₂O (C, solid line), and in 2 H₂O (D, dashed line), and bR regenerated with 2 H 2 H 2 C₁₅ retinal in H₂O (E). All spectra were obtained as described under "Experimental Procedures."

the same wavelength which were found in difference spectra for conversion of 13-cis to all-trans protonated Schiff base retinals (30).

In Fig. 8, C and D, we compare the UV-visible difference spectra obtained under identical conditions for the primary phototransition of light-adapted (top) and dark-adapted (bot-tom) humidified films. In the visible region, we find a blue shift of the absorption maxima for the negative and positive peaks of dark-adapted relative to light-adapted films due to the occurrence in the dark-adapted films of the BR₅₄₈ \rightarrow K $_{610}^{D}$ transition. In the case of the UV region (Fig. 8D), except for the "filling in" of the negative K $_{630}$ difference peak at 270 nm observed in the light-adapted difference spectra, no new major UV difference peaks are observed in the dark-adapted case. We do find, however, an apparent reduction in the doublet at 290 and 294 nm. Furthermore, interactive subtraction (not shown) of the light-adapted and dark-adapted difference spec-

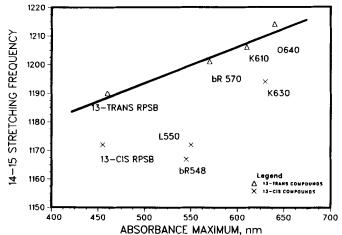


FIG. 6. Plot of the $C_{14}-C_{15}$ stretching frequency versus λ_{max} of 13-trans retinal compounds. Also shown are points which correspond to known 13-cis compounds. Plotted is a C_{14} - C^{15} assignment for the 13-trans intermediate O_{640} at 1214 cm⁻¹. Although originally $C_{14}-C_{15}$ stretch character was assigned at 1198 for O_{640} (8), data in Ref. 34 are consistent both with $C_{14}-C_{15}$ character in a band at 1214 cm⁻¹ and a 13-trans,15-trans configuration for O_{640} as originally proposed (34). $C_{14}-C_{15}$ stretching character in more than one peak can occur due to mixing between individual C-C stretches (8, 20, 21, 23, 34). Data is from Refs. 20, 21, and 34 and the present work. It is seen that $C_{14}-C_{15}$ stretches of 13-cis compounds (i.e. K_{630} , $C_{14}-C_{15}$ stretch at 1193 cm⁻¹) do not fall close to this straight line correlation.

tra in Fig. 8C so that the visible region resembled previously reported $BR_{548} \rightarrow K_{510}^{D}$ visible difference spectra (12) also revealed that the peaks near 300 and 243 nm are reduced or absent. In a previous study (1) of the $BR_{570} \rightarrow K_{630}$ transition, the negative peak (cf. expansion in Fig. 8D, top) at 244 nm was assigned to a tyrosine protonation and the positive doublet at 290 and 294 nm to a tryptophan hydrogen bonding change on the basis of UV absorption measurements of model compounds and their perturbation difference spectra (29–31). Thus, the present results support our conclusion based on FTIR spectroscopy that a tyrosinate protonation does not occur during the primary phototransition in the dark-adapted

Tabulated are the observed shifts and intensity changes of chromophore vibrations for $K_{0,0}^0$, the 13-trans RPSB, and the 13-tis RPSB upon deutero substitution at C_{15} . All entries are in cm⁻¹. \downarrow or \uparrow indicates the peak decreases or increases significantly in intensity, respectively, upon the isotopic substitution. Model compound data is from Ref. 3.

13-cis RPSB		K_{610}^D		13-trans RPSB	
15H	15°H	15H	15 ² H	15H	15°H
		877	875↓	870	
		957	957		
972	975 988	964	964	972	972
				1161	1164
1174	1168	1177	1177		
		1197	1192	1197	1202
1206	$1206 \\ 1232$	1205		1239	
		1277	1277↓ 1262	1279	1279↓ 1271 (sh)
1332		1323			
			1338		
1400		1412	1412↓↓	1410	
		1463	1463↓		1440
		1452			
				1525	
1572	1572	1522 1600 (sh)	1508 1600↑	1565	1561 1605

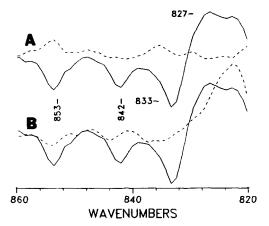


FIG. 7. A, comparison of a BR₅₇₀ \rightarrow K₆₃₀ difference spectrum (solid line) and a BR₅₄₈ \rightarrow K⁰₅₁₀ spectrum (dashed line) in the 860-820 cm⁻¹ region. B, comparison of bR (solid line) and bR-[²H₄]Tyr (dashed line) BR₅₇₀ \rightarrow K₆₃₀ difference spectra in the Fermi resonance region showing the disappearance in the labeled spectrum of tyrosina and tyrosinate peaks which indicate a tyrosinate protonation during K₆₃₀ formation (1). The same peaks eliminated by [²H₄]Tyr labeling are also seen to be absent in the BR₅₄₈ \rightarrow K⁰₅₁₀ spectrum.

photocycle. Furthermore, if a tryptophan perturbation occurs for $BR_{548} \rightarrow K_{610}^D$ it is relatively smaller than in the $BR_{570} \rightarrow K_{630}$ case.

DISCUSSION

The present results on dark adaptation and the primary photoreaction of BR_{548} add to the conclusions already reached on the basis of FTIR and UV difference spectroscopy pertaining to tyrosine protonation changes occurring during the bR photocycle $(1,\ 4,\ 5,\ 13,\ 14)$. The major conclusions of this study are summarized below.

(i) At Physiological Temperature (310 K) a Tyrosinate Group in BR_{570} Protonates during the $BR_{570} \rightarrow BR_{548}$ Transition—Features in both the FTIR and UV difference spectrum of dark adaptation support this conclusion. Furthermore, the similarity between the frequencies and intensities of tyrosine and tyrosinate peaks seen in the $BR_{570} \rightarrow K_{630}$ FTIR difference spectrum at 81 K and the $BR_{570} \rightarrow BR_{548}$ FTIR difference spectrum at 310 K indicates that one specific tyrosinate protonates in both transitions. The existence of a tyrosinate group in BR_{570} at physiological temperature has also been deduced recently from both time-resolved FTIR (32) and FTIR-attenuated total reflection difference spectroscopy (33) of the $BR_{570} \rightarrow M_{412}$ transition.

(ii) The $BR_{548} \rightarrow K_{610}^{D}$ Transition Does Not Involve a Protonation Change of Tyrosine—This conclusion is based primarily on the absence of peaks, particularly in the 900–800 cm⁻¹ region of the FTIR difference spectrum, associated with tyrosine protonation changes at various stages of the photocycle (1, 13, 14, 18). In addition, tyrosinate protonation peaks near 240 and 300 nm are not apparent in the UV difference spectrum for this transition.

(iii) Carboxyl Group(s) of Glutamic Acid and/or Aspartic Acid Are Perturbed in the $BR_{570} \rightarrow K_{630}$ and $BR_{548} \rightarrow K_{610}^D$ Transitions but Not in the $BR_{570} \rightarrow BR_{548}$ Transition—This conclusion is based on the appearance of almost identical peaks in the COOH stretching region for the primary phototransition of BR_{548} and BR_{570} . Thus, it is likely that one or more aspartic acid and/or glutamic acid groups are perturbed in the same way, despite the fact that the chromophore configurational changes during the two transitions are quite different.

(iv) The Conformation of K_{610}^D Is 13-trans,15-cis—This conclusion is based on the isotopic induced shifts for peaks assigned to the K_{610}^D retinylidene chromophore. In particular, 2 H-C₁₅ retinal substitution causes shifts similar to those seen for C₁₃=C₁₄ trans model compounds. Furthermore, the frequency of the C₁₄-C₁₅ single bond stretch supports a 13-trans assignment. In the case of the C₁₅=N bond, the large upshift of the C₁₄-C₁₅ stretch upon N deutero substitution indicates this bond is cis, as no large upshift of the C₁₄-C₁₅ line is expected when the C₁₅=N bond is trans (8).

(v) Intense HOOP Lines are Absent in Spectra of the K_{610}^D Chromophore—In contrast to the BR₅₇₀→K₆₃₀ transition and other primary retinal pigment phototransitions studied by FTIR difference spectroscopy (including the rhodopsin to bathorhodopsin transition (25)), very weak peaks in the HOOP mode region are found in the BR₅₄₈→K^D₆₁₀ difference spectrum. In resonance Raman studies, intensified HOOP modes have been associated with conformationally distorted forms of the chromophore which contain significant single bond twists (3, 24, 34). Such peaks have been observed in the primary photoproducts of BR₅₇₀ (3) and dry BR₅₃₀ (35) as well as in the O_{640} intermediate of the bR photocycle (34). Thus, the K_{610}^D chromophore may be in a more relaxed conformation, relative to K₆₃₀. However, since HOOP modes are expected to be infrared active and normally produce intense absorption lines in spectra of retinal model compounds (36), it is also possible that the absence of intense peaks in the 1000-800 cm⁻¹ region of the BR₅₄₈ \rightarrow K^D₆₁₀ spectrum indicates a cancellation of HOOP intensity, which would occur if the HOOP modes for BR₅₄₈ and K_{610}^D were very similar in frequency and intensity.

The above conclusions are represented diagramatically in Fig. 9. Several features are worth noting with regard to possible mechanisms for proton pumping and wavelength control of chromophore absorbance.

First, it is found that BR₅₇₀ is the only one of the four states

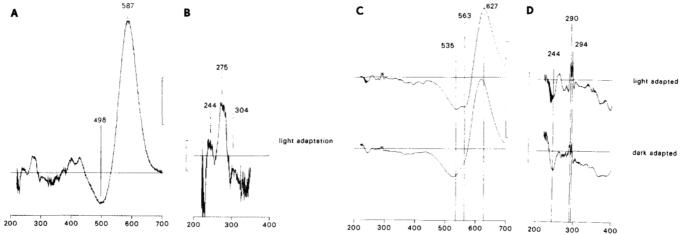


FIG. 8. The UV-visible difference spectrum for light adaptation at room temperature in the 700–200-nm region (A) and in an expanded view of the 400–200-nm UV region (B). The spectrum shown is the average of two differences recorded from a humidified purple membrane film. Absorbance scale markers are 0.002 (B) and 0.01 absorbance unit (A). C, comparison of the UV-visible difference spectra in the 700–200-nm region; D, expanded view of the 400–200-nm UV region for the primary photoproduct of light-adapted (top) and dark-adapted (bottom) bR films at 80 K. Scale markers are 0.01 (D) and 0.05 absorbance unit (C) for both top and bottom spectra (cf. "Experimental Procedures" for further details).

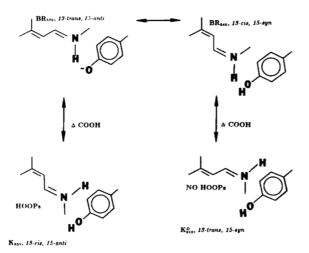


FIG. 9. A summary of observed chromophore isomerizations and protein changes during the early photochemistry of BR₅₇₀ and BR₅₄₈. The positioning of the tyrosine group is *not* meant to imply it is the sole counterion to the Schiff base. Note that 15-anti and 15-syn are equivalent to 15-trans and 15-cis, respectively.

which possesses a tyrosinate ion. Since the existence of a tyrosinate would require stabilization by one or more positive groups, direct interaction of the tyrosinate with the Schiff base proton as previously proposed (1) is quite plausible. If this is the case, the tyrosinate must be positioned such that the 13-trans, 15-trans configuration confers a more energetically favorable interaction than any of the other three configurations. However, it is possible that additional groups in bR interact electrostatically with the Schiff base.

Examination of proposed two-dimensional folding patterns for bR (38, 39) reveals that at least one tyrosine (Tyr-185) is in a position to interact with the C=N group of the chromophore. Indeed, recent FTIR studies on a bR mutant that has Tyr-185 replaced by a phenylalanine indicate that vibrations in the BR₅₇₀ \rightarrow K₆₃₀ difference spectrum attributable to the BR₅₇₀ tyrosinate are absent in the site-specific mutant (40). Thus, it is likely that it is this residue which gives rise to the tyrosinate to tyrosine signals in both the primary phototransition of light-adapted bR as well as during dark adaptation.

Finally, determination of the chromophore configuration for K_{610}^D reveals a consistent pattern for wavelength regulation in BR_{570} , BR_{548} , K_{630} and K_{610}^D , whose chromophores have all four possible combinations of isomerization about the C_{13} = C_{14} and C=N bonds. As seen in Fig. 9, the 13-trans,15-trans (BR₅₇₀) and 13-cis,15-cis (BR₅₄₈) states result in a similar orientation of the Schiff base proton relative to the long axis of the retinal chromophore (the axis drawn along the polyene chain). The absorption maxima of these two intermediates are relatively close, being separated by only 20 nm. Similarly, the 13-cis, 15-trans (K_{630}) and 13-trans, 15-cis (K_{610}^D) forms, which have significantly red-shifted λ_{max} , also have a similar N-H bond dipole orientation (relative to the long axis). A recent polarized FTIR study has determined that the average angle of the long axis of the chromophore remains fixed during the primary photochemistry of bR (37); thus, the major factor which appears to be associated with the λ_{max} of these intermediates is the orientation of the N-H bond dipole and not the configuration about either the $C_{13}=C_{14}$ or C=N bonds.

In conclusion, these FTIR and UV difference measurements of the BR₅₄₈ and K^D₆₁₀ intermediates of bacteriorhodopsin provide new information about structural changes occurring during the photochemistry of bR. A correlation appears to exist between the configuration of the retinylidene chromophore, particularly near the 13=14 and C=N double bonds. and the state of protonation of a tyrosine group. The most consistent picture is that the positive charge of the Schiff base stabilizes a tyrosinate group, most likely Tyr-185. The movement of this charge during dark adaptation or during the primary phototransition of the light-adapted photocycle destabilizes this tyrosinate causing its protonation. In the case of the primary phototransition for the dark-adapted photocycle, a protonation signal is not observed since the tyrosinate group has already become protonated during the formation of BR₅₄₈.

Additional progress will be possible by studying the effects of site-specific mutagenesis on the UV-visible (41) and FTIR difference spectra of the various bR transitions. Such studies are presently underway to determine the origin of the tyrosine signals detected during the FTIR difference spectra of the light to dark adaptation reaction and the carboxyl signals

detected during primary phototransition of both the light and dark photocycles.

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