



Universiteit  
Leiden  
The Netherlands

## **FTIR spectroscopy shows weak symmetric hydrogen bonding of the Q<sub>B</sub> carbonyl groups in Rhodobacter sphaeroides R26 reaction centres**

Brudler, R.; Groot, H.J.M. de; Liemt, W.B.S.; Gast, P.; Hoff, A.J.; Lugtenburg, J.; Gerwert, K.

### **Citation**

Brudler, R., Groot, H. J. M. de, Liemt, W. B. S., Gast, P., Hoff, A. J., Lugtenburg, J., & Gerwert, K. (1995). FTIR spectroscopy shows weak symmetric hydrogen bonding of the Q<sub>B</sub> carbonyl groups in Rhodobacter sphaeroides R26 reaction centres. *Febs Letters*, 370(1-2), 88-92. doi:10.1016/0014-5793(95)00805-J

Version: Publisher's Version

License: [Licensed under Article 25fa Copyright Act/Law \(Amendment Taverne\)](#)

Downloaded from: <https://hdl.handle.net/1887/3239393>

**Note:** To cite this publication please use the final published version (if applicable).

# FTIR spectroscopy shows weak symmetric hydrogen bonding of the $Q_B$ carbonyl groups in *Rhodobacter sphaeroides* R26 reaction centres

R. Brudler<sup>a</sup>, H.J.M. de Groot<sup>b</sup>, W.B.S. van Liemt<sup>b</sup>, P. Gast<sup>c</sup>, A.J. Hoff<sup>c</sup>, J. Lugtenburg<sup>b</sup>, K. Gerwert<sup>a,\*</sup>

<sup>a</sup>Lehrstuhl für Biophysik, Ruhr-Universität Bochum, Postfach 102148, 44780 Bochum, Germany

<sup>b</sup>Department of Chemistry, Gorlaeus Laboratories, Leiden University, P.O. Box 9502, 2300 RA Leiden, The Netherlands

<sup>c</sup>Department of Biophysics, Huygens Laboratory, Leiden University, P.O. Box 9504, 2300 RA Leiden, The Netherlands

Received 22 May 1995; revised version received 6 July 1995

**Abstract** The absorption frequencies of the C = O and C = C (neutral state) and of the C $\cdots$ O (semiquinone state) stretching vibrations of  $Q_B$  have been assigned by FTIR spectroscopy, using native and site-specifically 1-, 2-, 3- and 4-<sup>13</sup>C-labelled ubiquinone-10 ( $UQ_{10}$ ) reconstituted at the  $Q_B$  binding site of *Rhodobacter sphaeroides* R26 reaction centres. Besides the main C = O band at 1641 cm<sup>-1</sup>, two smaller bands are observed at 1664 and 1651 cm<sup>-1</sup>. The smaller bands at 1664 and 1651 cm<sup>-1</sup> agree in frequencies with the 1- and 4-C = O vibrations of unbound  $UQ_{10}$ , showing that a minor fraction is loosely and symmetrically bound to the protein. The larger band at 1641 cm<sup>-1</sup> indicates symmetric H-bonding of the 1- and 4-C = O groups for the larger fraction of  $UQ_{10}$  but much weaker interaction as for the 4-C = O group of  $Q_A$ . The FTIR experiments show that different C = O protein interactions contribute to the factors determining the different functions of  $UQ_{10}$  at the  $Q_A$  and the  $Q_B$  binding sites.

**Key words:** Bacterial reaction centre; Ubiquinone; Fourier transform infrared spectroscopy; isotopic labelling; Photosynthesis

## 1. Introduction

The photosynthetic reaction centre (RC) of the purple non-sulfur bacterium *Rhodobacter (Rb.) sphaeroides* is a transmembrane pigment protein complex whose structure has been determined with up to 2.65 Å resolution [1–3]. Upon light excitation, an electron is transferred from the primary donor P (bacteriochlorophyll *a*-dimer) at the periplasmic side via a monomeric bacteriochlorophyll *a* and a bacteriopheophytin *a* molecule to the primary quinone  $Q_A$  and finally to the secondary quinone  $Q_B$  at the cytoplasmic side of the membrane [4]. Although  $Q_A$  and  $Q_B$  both are ubiquinone-10 ( $UQ_{10}$ ) molecules their functions are different.  $Q_A$  is tightly bound to the RC. It accepts only one electron and transfers it to  $Q_B$  in about 200 s;  $Q_B$  is less tightly bound to the protein. It accepts two electrons and two protons and is finally released from the RC as  $Q_BH_2$ .

In order to elucidate the protein-cofactor interactions which determine the different functions of  $UQ_{10}$  at their binding sites,

FTIR difference spectroscopy has been applied [5,6]. By use of  $UQ_{10}$ , <sup>13</sup>C-labelled at the ring positions 1, 2, 3 and 4, respectively, the 1- and 4-C = O and 2/3-C = C stretching vibrations of  $UQ_{10}$  have been specifically assigned in the  $Q_A^-$ - $Q_A$  difference spectra [7]. The mode dominated by the 4-C = O vibration is drastically downshifted in the RC as compared to unbound  $UQ_{10}$ , indicating unusually strong hydrogen bonding to His M219. In contrast, the 1-C = O group is only weakly bound to the protein. In the charge-separated state, the asymmetric binding is largely maintained. These results are fully supported by similar FTIR studies [8], by EPR [9] and NMR spectroscopy [10]. In the present contribution the interaction of  $Q_B$  with its protein environment is investigated using site specifically 1-, 2-, 3- and 4-<sup>13</sup>C-labelled  $UQ_{10}$ . The results and their implications for the different functions of  $UQ_{10}$  at the  $Q_A$  and the  $Q_B$  site will be discussed.

## 2. Materials and methods

RC protein was purified from *Rb. sphaeroides* strain R26 [11].  $UQ_{10}$ , selectively <sup>13</sup>C-labelled at positions 1, 2, 3 and 4 was synthesized as described in [12].  $Q_B$  was selectively removed by incubating RCs with OD<sub>800</sub> = 2 in 2% LDAO, 1 mM *o*-phenanthroline, 10 mM Tris pH 8 at 25–26°C for 6 h, followed by extensive washing on a DEAE Sephacel column [13]. The  $Q_A$  and  $Q_B$  content was determined by measuring the amplitude of the fast (100 ms) and slow (1 s) decay component of the photobleaching of P<sup>+</sup> at 865 nm after a light flash [7]. After the selective quinone-removal, the  $Q_B$  content was less than 10% while  $Q_A$  was present in more than 95% of the RCs.  $Q_B$  reconstitution with native and selectively <sup>13</sup>C-labelled  $UQ_{10}$  was done according to [13], except that the stock solution of  $UQ_{10}$  was in 1% Triton X-100. The  $Q_B$  content after reconstitution was better than 85% (except 70% for 1-<sup>13</sup>C- $UQ_{10}$ ).

Sample preparation for the IR measurements was performed according to [7]. 45 μl of 40 μM RCs, dissolved in 10 mM Tris/HCl, 1 mM EDTA, 0.025 (w/v) LDAO, pH 8, were pipetted on a CaF<sub>2</sub> window, approximately ten-fold concentrated under a gentle stream of nitrogen and mixed with 5 μl of 10 mM sodium ascorbate, 20 mM diamidodurene (DAD) dissolved in the same buffer as the RCs. After further drying to a final volume of ~1 μl, the sample was sealed with another CaF<sub>2</sub> window and thermostabilized at 295 K in the FTIR apparatus.

IR spectra of free ubiquinones and  $Q_B^-$ - $Q_B$  difference spectra were recorded as reported [7], except that the actinic light intensity allowing the saturation of  $Q_B^-$  was about ten-fold lower than that for the saturation of  $Q_A^-$  [14]. Spectral resolution was 4 cm<sup>-1</sup>.

Double difference spectra were computed as described [7]. The difference spectra with unlabelled and <sup>13</sup>C-labelled  $Q_B$  were normalized on the 1800–1700 cm<sup>-1</sup> region that was unaffected by the labelling.

## 3. Results

In Fig. 1a the  $Q_B^-$ - $Q_B$  difference spectrum of *Rb. sphaeroides* R26 RCs reconstituted with unlabelled  $UQ_{10}$  is shown. The

\*Corresponding author. Fax: (49) (234) 7094238.

**Abbreviations:** RC, reaction centre; Rb., *Rhodobacter*; P, primary electron donor;  $Q_A$ , primary acceptor quinone;  $Q_B$ , secondary acceptor quinone; UQ, ubiquinone; ENDOR, electron-nuclear double resonance; EPR, electron paramagnetic resonance; FTIR, Fourier transform infra-red; NMR, nuclear magnetic resonance.

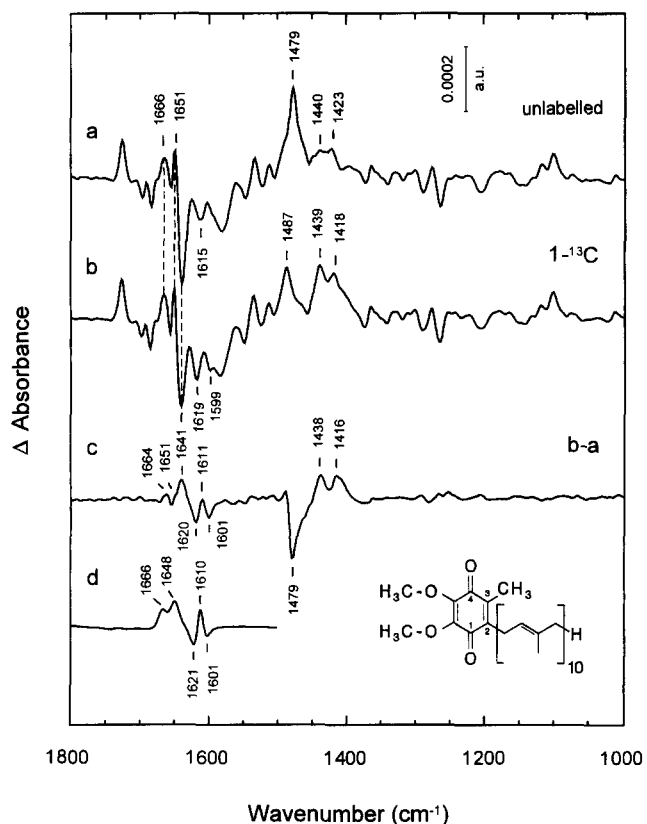


Fig. 1.  $Q_B^- - Q_B$  difference spectra of *Rb. sphaeroides* R26 RCs reconstituted with (a) unlabelled and (b)  $1-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$  at the  $Q_B$  site. (c) double difference spectrum (b - a). (d) difference spectrum between unbound unlabelled and  $1-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$ . The difference spectrum between unbound unlabelled and  $4-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$  is identical. Inset: structural formula of ubiquinone-10.

difference between the charge-separated and the ground state absorption selectively represents the light-induced absorption changes of the RCs. Positive bands belong to the charge-separated state, negative signals to the ground state.

The  $Q_B^- - Q_B$  difference spectrum (Fig. 1a) agrees remarkably well with the one published in [14]. It was shown that other species like  $Q_B^{2-}$ ,  $Q_B\text{H}$  and  $Q_B\text{H}_2$  do not significantly contribute to the difference spectra [14]. Only minor deviations in band positions and intensities within the experimental error are observed between the  $Q_B^- - Q_B$  difference spectra shown here and in [14], which may be caused by different detergent and water contents of the samples and different measuring temperatures (283 K instead of 295 K). Bauscher et al. [15] have also presented a  $Q_B^- - Q_B$  difference spectrum, obtained by the subtraction of an electrochemically generated  $\text{P}^+ - \text{P}$  difference spectrum from a light-induced  $\text{P}^+ Q_B^- - \text{P} Q_B$  difference spectrum. In this approach the much larger  $\text{P}^+ - \text{P}$  bands have to be subtracted and only the minor  $Q_B^- - Q_B$  bands should result. The experimental error of this method and consequently the deviations to the directly measured  $Q_B^- - Q_B$  difference spectra shown here and in [14] are larger.

In Fig. 1b the  $Q_B^- - Q_B$  difference spectrum of RCs containing  $1-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$  at the  $Q_B$  site is displayed. The isotope labelling induces a frequency shift of the labelled group absorption to lower wavenumbers and thereby allows the specific assignment of the quinone vibrations. The C = O and C = C

stretching vibrations of neutral  $\text{UQ}_{10}$  are expected to absorb between  $1670\text{--}1570\text{ cm}^{-1}$  [7]. In this spectral region the intensity of the strong negative band at  $1641\text{ cm}^{-1}$  decreases and two additional negative bands appear at  $1619$  and  $1599\text{ cm}^{-1}$  (compare Fig. 1a and 1b). In addition, the positive bands at  $1666$  and  $1651\text{ cm}^{-1}$  slightly gain intensity (Fig. 1b), indicating shifts of underlying negative bands. The C=O and C=C stretching vibrations of the semiquinone are found between  $1500$  and  $1400\text{ cm}^{-1}$  [7]. In Fig. 1b the strong band of unlabelled  $\text{UQ}_{10}$  at  $1479\text{ cm}^{-1}$  (Fig. 1a) is shifted and new bands appear at  $1439$  and  $1418\text{ cm}^{-1}$ .

In order to visualize more clearly the deviations between the difference spectra, they have been subtracted as described [7]. The resulting double difference spectrum is shown in Fig. 1c (= Fig. 1b - 1a). In the double difference spectra, the bands of the C = O and C = C stretching vibrations of neutral  $\text{UQ}_{10}$  ( $1670\text{--}1570\text{ cm}^{-1}$ ) are positive for unlabelled  $\text{UQ}_{10}$  while the corresponding shifted bands of labelled  $\text{UQ}_{10}$  are negative. Inversely, the signals of the C=O and C=C stretching vibrations of the semiquinone ( $1500\text{--}1400\text{ cm}^{-1}$ ) are negative for unlabelled  $\text{UQ}_{10}$  and the shifted bands of labelled  $\text{UQ}_{10}$  are positive. In Fig. 1c the bands of unlabelled  $\text{UQ}_{10}$  are seen at  $1664$ ,  $1651$ ,  $1641$ ,  $1611$  and  $1479\text{ cm}^{-1}$ . The positive band at  $1664\text{ cm}^{-1}$  and the shoulder at  $1651\text{ cm}^{-1}$  are small compared to the band at  $1641\text{ cm}^{-1}$  but have been reproduced in every measurement. The shifted bands of  $1-^{13}\text{C}$ - $\text{UQ}_{10}$  can be seen at  $1620$ ,  $1601$ ,  $1438$  and  $1416\text{ cm}^{-1}$ . For comparison, in Fig. 1d the difference spectrum between unbound unlabelled and  $1-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$  is depicted in the region of the C = O and C = C stretching vibrations. For the spectra of unbound unlabelled and  $1$ -,  $2$ -,  $3$ - and  $4-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$  see [7]. The difference spectrum shown in Fig. 1d is identical to the difference spectrum between unbound unlabelled and  $4-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$  [7]. As in the double difference spectra, the positive bands belong to unlabelled  $\text{UQ}_{10}$  while the negative signals represent the shifted bands of labelled  $\text{UQ}_{10}$ . Two bands at  $1666$  and  $1648\text{ cm}^{-1}$  are shifted to  $1621\text{ cm}^{-1}$  for  $1$ - and  $4-^{13}\text{C}$ -labelled  $\text{UQ}_{10}$  and are dominated by the  $1$ - and  $4\text{-C}=\text{O}$  stretching vibrations [7]. A distinction between the vibrations of the  $1$ - and  $4\text{-C}=\text{O}$  group is not possible due to mixing of the  $1$ - and  $4\text{-C}=\text{O}$  vibrations (both contribute to the bands at  $1666$  and  $1648\text{ cm}^{-1}$ ). A band at  $1610\text{ cm}^{-1}$  is shifted to  $1601\text{ cm}^{-1}$  (Fig. 1d) and is dominated by the  $2/3\text{-C}=\text{C}$  stretching vibration [7]. As discussed in [7], isotope labelling of C = O groups shifts also bands of C = C vibrations and vice versa due to extensive coupling of the C = O and C = C vibrations of  $\text{UQ}_{10}$ . As for unbound  $\text{UQ}_{10}$ , the two bands at  $1664$  and  $1651\text{ cm}^{-1}$  seem to be shifted to  $1620\text{ cm}^{-1}$  for RC bound  $\text{UQ}_{10}$  (Fig. 1c). They are therefore assigned to C = O stretching vibrations of  $\text{UQ}_{10}$  at the  $Q_B$  site. The largest band in the carbonyl region appears at  $1641\text{ cm}^{-1}$  (Fig. 1c). It is not directly seen to which position this band is shifted. The maximal shift for a C = O vibration due to  $^{13}\text{C}$ -labelling is  $36\text{ cm}^{-1}$  [16]. It seems to be shifted underneath the band at  $1611\text{ cm}^{-1}$  (Fig. 1c). The difference band at  $1611/1601\text{ cm}^{-1}$  in Fig. 1c represents, as compared to Fig. 1d, the shift of the  $2/3\text{-C}=\text{C}$  vibration of  $Q_B$ . The intensity of the band at  $1611\text{ cm}^{-1}$  (Fig. 1c) is reduced as compared to the band at  $1610\text{ cm}^{-1}$  (Fig. 1d) in agreement with the assumption that the C = O vibration is shifted from  $1641\text{ cm}^{-1}$  underneath the band at  $1611\text{ cm}^{-1}$ . This assumption is furthermore supported by the observation of a clear C = C vibration at  $1611\text{ cm}^{-1}$  for  $2-^{13}\text{C}$ - and

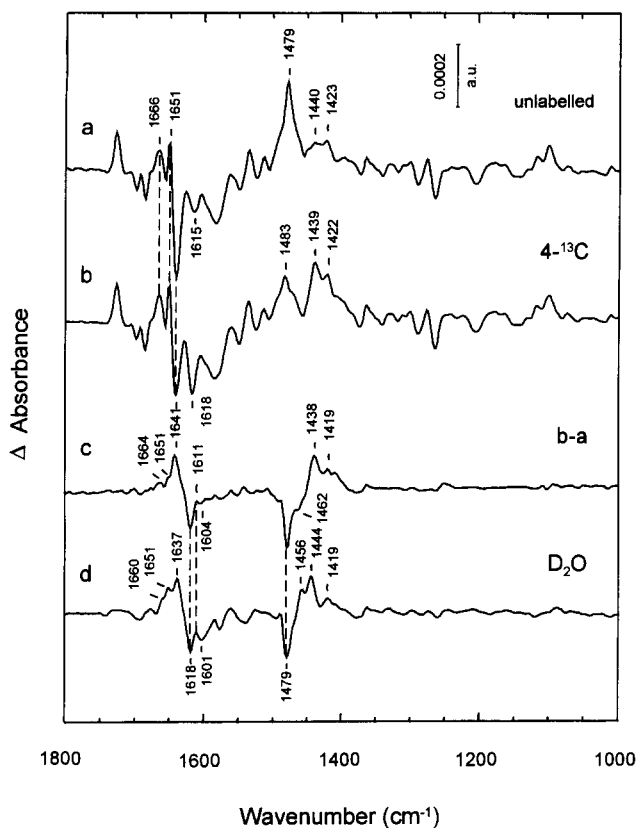


Fig. 2.  $Q_B^- - Q_B$  difference spectra of *Rb. sphaeroides* R26 RCs reconstituted with (a) unlabelled and (b)  $4\text{-}^{13}\text{C}$ -labelled  $UQ_{10}$  at the  $Q_B$  site. (c) double difference spectrum (b - a). (d) double difference spectrum as in Fig. 2c but measured in  $D_2O$  instead of  $H_2O$ .

$3\text{-}^{13}\text{C}$ -labelled  $UQ_{10}$  (Fig. 3d,e). In this case the  $C = O$  vibration at  $1643\text{ cm}^{-1}$  is less shifted to  $1621\text{ cm}^{-1}$  but not underneath the band at  $1611\text{ cm}^{-1}$ .

Fig. 2b shows the  $Q_B^- - Q_B$  difference spectrum of RCs with  $4\text{-}^{13}\text{C}$ - $UQ_{10}$  at the  $Q_B$  site. It is similar to the corresponding spectrum with  $1\text{-}^{13}\text{C}$ - $UQ_{10}$  at the  $Q_B$  site (Fig. 1b): the negative band at  $1641\text{ cm}^{-1}$  decreases and the positive bands at  $1666$  and  $1651\text{ cm}^{-1}$  slightly gain intensity. At  $1618\text{ cm}^{-1}$  a new negative band appears. The band at  $1479\text{ cm}^{-1}$  (Fig. 2a) is shifted and new bands appear at  $1439$  and  $1422\text{ cm}^{-1}$  (Fig. 2b). Again the labelling-induced band shifts can be more clearly seen in the double difference spectrum (Fig. 2c). The signals of unlabelled  $UQ_{10}$  (neutral state) appear at  $1664$ ,  $1651$  and  $1641\text{ cm}^{-1}$  as in the double difference spectrum with  $1\text{-}^{13}\text{C}$ -labelled  $UQ_{10}$  (Fig. 1c) and are correspondingly assigned to  $C = O$  stretching vibrations. As for unbound  $UQ_{10}$ , the 1- and 4- $C = O$  vibrations of  $Q_B$  cannot be distinguished. The shifted signal of the bands at  $1664$  and  $1651\text{ cm}^{-1}$  is seen at  $1618\text{ cm}^{-1}$  as a negative band (compare Fig. 2c with Fig. 1d). The difference band at  $1610/1601\text{ cm}^{-1}$  that is observed in the difference spectrum between unbound and 1- or  $4\text{-}^{13}\text{C}$ -labelled  $UQ_{10}$  (Fig. 1d) is missing in Fig. 2c. A  $^{13}\text{C}$ -induced shift of the  $C = O$  vibration from  $1641\text{ cm}^{-1}$  to around  $1610\text{ cm}^{-1}$  and of the  $C = C$  vibration from  $1610$  to  $1601\text{ cm}^{-1}$  seem to cancel each other more than in the case of  $1\text{-}^{13}\text{C}$ -labelling (compare Fig. 2c with Fig. 1c). In order to clarify, the samples have been measured in  $D_2O$  where the bands shift slightly different (Fig. 2d). The difference band at

$1611/1601\text{ cm}^{-1}$  is better resolved than in  $H_2O$ . In addition, the bands appearing at  $1664$ ,  $1651$ ,  $1641$  and  $1618\text{ cm}^{-1}$  in  $H_2O$  (Fig. 2c) are observed in  $D_2O$  as well at  $1660$ ,  $1651$ ,  $1637$  and  $1618\text{ cm}^{-1}$ , respectively (Fig. 2d). They are also found in the double difference spectrum between the difference spectra with  $1\text{-}^{13}\text{C}$ - $UQ_{10}$  and unlabelled  $UQ_{10}$  at the  $Q_B$  site, measured in  $D_2O$  (not shown). Their presence in  $D_2O$  excludes that they are caused by absorption of water. In  $D_2O$ , the O-H group is replaced by the O-D group and the corresponding bending vibration is shifted from  $1645$  to  $1215\text{ cm}^{-1}$ . For the semiquinone state the double difference spectra (Fig. 2c,d) show a band at  $1479\text{ cm}^{-1}$  of unlabelled  $UQ_{10}$ . Shifted bands of  $4\text{-}^{13}\text{C}$ - $UQ_{10}$  are found at  $1438$  and  $1419\text{ cm}^{-1}$  in  $H_2O$  (Fig. 2c) and at  $1456$ ,  $1444$  and  $1419\text{ cm}^{-1}$  in  $D_2O$  (Fig. 2d). The band at  $1456\text{ cm}^{-1}$  has a counterpart at  $1462\text{ cm}^{-1}$  in  $H_2O$  (Fig. 2c). It seems that the broad band at  $1479\text{ cm}^{-1}$  decreases the intensity of the band at  $1462\text{ cm}^{-1}$  in  $H_2O$  but has less effect on the band at  $1456\text{ cm}^{-1}$  in  $D_2O$ .

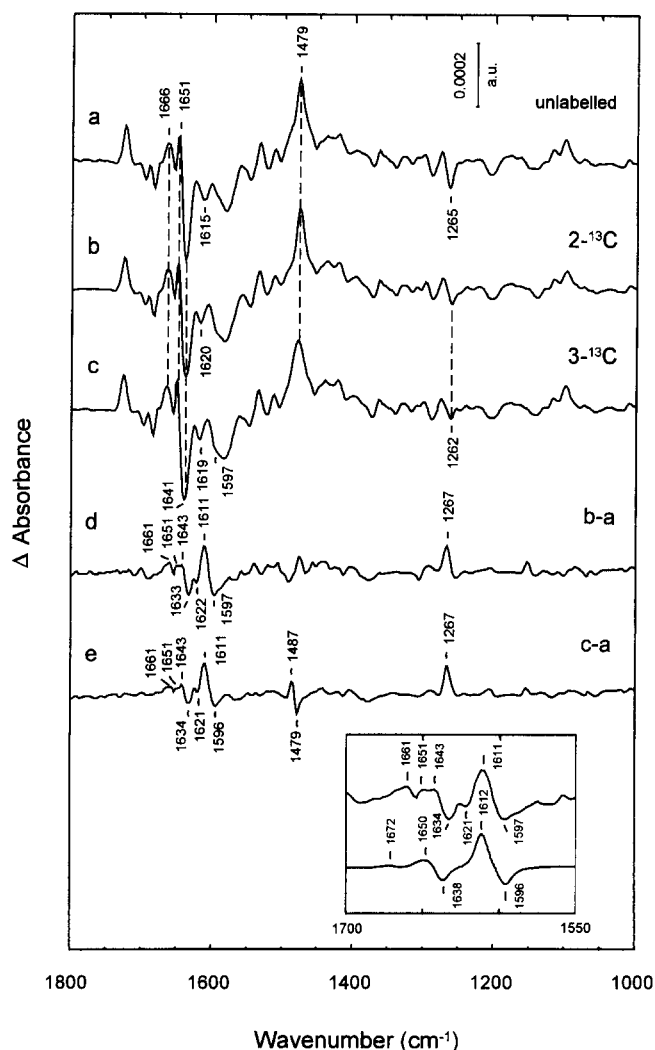


Fig. 3.  $Q_B^- - Q_B$  difference spectra of *Rb. sphaeroides* R26 RCs reconstituted with (a) unlabelled, (b)  $2\text{-}^{13}\text{C}$ -labelled and (c)  $3\text{-}^{13}\text{C}$ -labelled  $UQ_{10}$  at the  $Q_B$  site. (d) double difference spectrum (b - a). (e) double difference spectrum (c - a). Inset: double difference spectrum of Fig. 3d (upper trace) in comparison with the difference spectrum between unbound unlabelled and  $2\text{-}^{13}\text{C}$ -labelled  $UQ_{10}$  (lower trace). It is identical to the difference spectrum between unbound unlabelled and  $3\text{-}^{13}\text{C}$ -labelled  $UQ_{10}$  (not shown).

In Fig. 3 the  $Q_B^- - Q_B$  difference spectra of RCs reconstituted with 2- $^{13}C$ - (Fig. 3b) and 3- $^{13}C$ -UQ $_{10}$  (Fig. 3c) at the  $Q_B$  site are displayed. They are very similar. The corresponding double difference spectra agree therefore nicely and show small positive bands at 1661, 1651 and 1643  $cm^{-1}$  and a stronger band at 1611  $cm^{-1}$  (Fig. 3d,e). Negative bands are found at 1633/1634, 1622/1621 and 1597/1596  $cm^{-1}$  (Fig. 3d,e). An expanded view of Fig. 3d is shown in the inset of Fig. 3 in the upper trace. Interestingly, no band shifts are found for the semiquinone vibrations (1500–1400  $cm^{-1}$ ; Fig. 3d,e). For 3- $^{13}C$ -labelled UQ $_{10}$  (Fig. 3e) a positive band at 1487  $cm^{-1}$  is shifted to 1479  $cm^{-1}$ . It cannot be assigned to semiquinone vibrations since a semiquinone band should be negative and the shifted signal positive. The inset of Fig. 3 shows the difference spectrum between protein unbound unlabelled and 2- $^{13}C$ -labelled UQ $_{10}$  in the lower trace. It is identical to the difference spectrum between unbound unlabelled and 3- $^{13}C$ -labelled UQ $_{10}$  [7]. The strongest band of this difference spectrum at 1612  $cm^{-1}$  is shifted to 1596  $cm^{-1}$  and is assigned to the C = C stretching vibration [7]. These bands have counterparts in the double difference spectra at 1611 and 1597  $cm^{-1}$  (inset of Fig. 3, upper trace). The band at 1611  $cm^{-1}$  is therefore assigned to the C = C stretching vibration of UQ $_{10}$  at the  $Q_B$  site. Due to extensive coupling of the C = C and C = O vibrations  $^{13}C$ -labelling at the 2/3-C = C positions shifts also the C = O vibrations. The difference spectrum in Fig. 3 (inset, lower trace) shows also shifts of C = O vibrations from 1672 and 1650  $cm^{-1}$  to 1638  $cm^{-1}$ . These bands have counterparts in the double difference spectra at 1661, 1651 and 1634  $cm^{-1}$  (inset, upper trace) and therefore are assigned to C = O stretching vibrations. As for 1- and 4- $^{13}C$ -labelled UQ $_{10}$  (Figs. 1c and 2c) an additional band appears in the region of the C = O stretching vibrations at 1643  $cm^{-1}$  (inset, upper trace) but is less pronounced as compared to 1- and 4- $^{13}C$ -labelled UQ $_{10}$  (Figs. 1c and 2c). It seems to be shifted to 1621  $cm^{-1}$  and is assigned to a C = O stretching vibration of  $Q_B$  as discussed above.

#### 4. Discussion

The isotope shifts and the assignments of the dominant vibrations for UQ $_{10}$  bound at the  $Q_B$  site are summarized in Table 1. As for unbound UQ $_{10}$  the C = O and C = C stretching vibrations are strongly coupled but the dominant vibrations can be assigned.

Surprisingly, three carbonyl vibrations are observed for

UQ $_{10}$  at the  $Q_B$  site (Table 1) instead of two as for unbound UQ $_{10}$ , indicating heterogeneous binding to the protein. The two minor bands at 1664 and 1651  $cm^{-1}$  agree with the frequencies of unbound UQ $_{10}$ . The band at 1641  $cm^{-1}$  is slightly downshifted and indicates a stronger interaction of the 1- and 4-carbonyl groups with the  $Q_B$  site. Comparison of the integral intensity of the carbonyl bands results in roughly 25 % for the fraction absorbing at 1664 and 1651  $cm^{-1}$  and roughly 75% for the fraction absorbing at 1641  $cm^{-1}$ . Both fractions represent reacting UQ $_{10}$  because only bands of functionally active groups appear in the difference spectra. It is not clear, whether both fractions have physiological relevance.

The absorption frequencies of the two bands at 1664 and 1651  $cm^{-1}$  (Table 1) are almost unaltered as compared to unbound UQ $_{10}$  (1666 and 1648  $cm^{-1}$ ). They can be attributed to a fraction of  $Q_B$  with the carbonyl groups not specifically hydrogen bonded to the protein in the ground state. The C = O vibration of the other  $Q_B$  fraction, absorbing at 1641  $cm^{-1}$ , indicates symmetric hydrogen bonding to the protein in the ground state. In contrast, the FTIR difference spectra of  $Q_A^- - Q_A$  have revealed an asymmetry between the two carbonyls of  $Q_A$ . The 1-C = O group (absorbing at 1660  $cm^{-1}$ ) is only very weakly bound, whereas the 4-C = O group (absorbing at 1601  $cm^{-1}$ ) is much stronger hydrogen-bonded to the protein than the carbonyl groups of  $Q_B$ . The strong hydrogen bonding binds  $Q_A$  tightly to the protein and governs the fast electron transfer via His M219 to  $Q_B$  [7,8,17].

In agreement with the heterogeneous binding of UQ $_{10}$  at the  $Q_B$  site, seen in FTIR experiments, also in X-ray crystallography experiments different positions of  $Q_B$  have been observed. Eight possible binding positions have been reported for UQ $_1$  (ubiquinone with one isoprene unit) at the  $Q_B$  site of *Rhodospseudomonas viridis* RCs [18]. For *Rb. sphaeroides* RCs two different binding positions of  $Q_B$  are proposed. Ermler et al. [3] have positioned the  $Q_B$  molecule at the more hydrophobic entrance of the  $Q_B$  binding pocket, whereas Allen et al. [19] and El-Kabbani et al. [20] have located it deeper in the protein. Consequently, different hydrogen bond donors to the carbonyl groups of  $Q_B$  have been suggested. In [19] and [20] the 1-C = O group is in hydrogen bonding distance to Ser L223, the 4-C = O group to His L190. In [3] only the 1-C = O group could form a hydrogen bond with the peptide nitrogen of Ile L224. The differently bound fractions of UQ $_{10}$  at the  $Q_B$  site seen in the IR experiments may explain the high temperature factor for the  $Q_B$  site [3] and the different positions for UQ $_{10}$  at the  $Q_B$  site

Table 1

Absorption frequencies ( $cm^{-1}$ ) and assignments of UQ $_{10}$  IR bands at the  $Q_B$  site of *Rb. sphaeroides* R26 RCs in the neutral (above) and in the semiquinone state (below)  
The frequency shifts of the labels are indicated in brackets.

$Q_B$	1- $^{13}C$	4- $^{13}C$	2- $^{13}C$	3- $^{13}C$	Assignment of dominant vibration
1641	~ 1610 (- 31)	~ 1610 (- 31)	1622 (- 19)	1621 (- 20)	1/4-C = O
1664	1620	1618	1633	1634	1/4-C = O
1651	(- 44/- 31)	(- 46/- 33)	(- 31/- 18)	(- 30/- 17)	
1611	1601 (- 10)	1604 (- 7)	1587 (- 14)	1596 (- 15)	2/3-C = C
$Q_B^-$	1- $^{13}C$	4- $^{13}C$	2- $^{13}C$	3- $^{13}C$	Assignment of dominant vibration
1479	1438; 1416 (- 41/- 63)	1438; 1419 (- 41/- 60)	-	-	1/4-C...O

of *Rb. sphaeroides* RCs [3,19,20]. Furthermore, Giangiaco and Dutton [21] have shown that the  $Q_B$  site is not highly specific for the native  $UQ_{10}$  and functions as well with various substituted quinones, including quinones with the ortho-carbonyl configuration.

In the semiquinone spectral region (1500–1400  $\text{cm}^{-1}$ ) [22] the strong positive band at 1479  $\text{cm}^{-1}$  is shifted by the 1- and 4- $^{13}\text{C}$ -labels to 1438 and 1416/1419  $\text{cm}^{-1}$ . The band at 1479  $\text{cm}^{-1}$  is therefore assigned to the 1- and 4-C $\cdots$ O stretching vibration of  $Q_B^-$  (Table 1). Similar band shifts have been observed for uniformly  $^{13}\text{C}$ -labelled  $UQ_{10}$ , dissolved in  $\text{CH}_2\text{Cl}_2$ . A semiquinone band at 1483  $\text{cm}^{-1}$  is shifted to 1442 and 1412  $\text{cm}^{-1}$  and has been assigned to the C $\cdots$ O stretching vibrations [23]. Since no band shifts of the band at 1479  $\text{cm}^{-1}$  are seen in the spectra of 2- $^{13}\text{C}$ - and 3- $^{13}\text{C}$ -labelled  $UQ_{10}$  bound to the protein (Fig. 3d, 3e), the band is assigned to mixed 1- and 4-C $\cdots$ O vibrations without C $\cdots$ C contributions. The close similarity of the 1- and 4- $^{13}\text{C}$ -induced shifts of the band at 1479  $\text{cm}^{-1}$  (Fig. 1b,2b and 1c,2c) indicates that the carbonyl groups of  $Q_B$  are nearly equivalent and equally hydrogen bonded to the protein. This observation is in agreement with results of ENDOR measurements [24,25]. In contrast, at the  $Q_A$  site the asymmetry of the carbonyl groups is maintained as well in the semiquinone state [9,7]. The frequency of the 1- and 4-C $\cdots$ O vibrations of  $Q_B^-$  at 1479  $\text{cm}^{-1}$  indicates weaker hydrogen bonding than for the 4-C $\cdots$ O group of  $Q_A^-$ , which absorbs at 1466  $\text{cm}^{-1}$ . The weak symmetric binding in the charge separated state indicates an equally delocalized  $\pi$  electron distribution and thereby seems to support the stabilization of the negative charge on  $Q_B^-$  until a second electron is accepted.

In conclusion, the  $Q_A^- - Q_A$  and the  $Q_B^- - Q_B$  FTIR difference spectra demonstrate two strongly differing binding sites, which tune the different functions of the same prosthetic group by specific protein interactions.

**Acknowledgements:** We thank A.H.M. de Wit and S.J. Jansen for culturing the cells and isolating the RCs, and W.F. Steggerda and R. Esmeijer for  $^{13}\text{C}$ -labelling of  $UQ_{10}$ . P.G. is a recipient of a Royal Netherlands Academy of Arts and Sciences (KNAW) career fellowship. This work was supported by the Foundation for Chemical Research (SON), financed by the Netherlands Organization of Pure and Applied Research (NWO). K.G. acknowledges the DFG for financial support (Ge-599/7-1).

## References

- [1] Feher, G., Allen, J.P., Okamura, M.Y. and Rees, D.C. (1989) *Nature* 339, 11–116.
- [2] Chang, C.-H., El-Kabbani, O., Tiede, D., Norris, J. and Schiffer, M. (1991) *Biochemistry* 30, 5352–5360.
- [3] Ermler, U., Fritsch, G., Buchanan, S. and Michel, H. (1994) *Structure* 2, 925–936.
- [4] Norris, J.R. (1992) *Israel J. Chem.* 32, 418–422.
- [5] Gerwert, K. (1993) *Curr. Opin. Struct. Biol.* 3, 769–773.
- [6] Mäntele, W. (1993) in *The Photosynthetic Reaction Center* (Deisenhofer, J., Norris, J., eds.), pp. 239–283, Academic Press, New York, Vol. 2.
- [7] Brudler, R., de Groot, H.J.M., van Liemt, W.B.S., Steggerda, W.F., Esmeijer, R., Gast, P., Hoff, A.J., Lugtenburg, J. and Gerwert, K. (1994) *EMBO J.* 13, 5523–5530.
- [8] Breton, J., Boullais, C., Burie, J.-R., Nabedryk, E. and Mioskowski, C. (1994) *Biochemistry* 33, 14378–14386.
- [9] van den Brink, J.S., Spoyalov, A.P., Gast, P., van Liemt, W.B.S., Raap, J., Lugtenburg, J. and Hoff, A.J. (1994) *FEBS Lett.* 353, 273–276.
- [10] van Liemt, W.B.S., Boender, G.J., Gast, P., Hoff, A.J., Lugtenburg, J. and de Groot, H.J.M. (1995) *Biochemistry*, in press.
- [11] Feher, G. and Okamura, M.Y. (1978) in *The Photosynthetic Bacteria* (Clayton, R.K., Sistrom, W.R., eds.), pp. 349–386, Plenum Press, New York.
- [12] van Liemt, W.B.S., Steggerda, W.F., Esmeijer, R. and Lugtenburg, J. (1994) *Recl. Trav. Chim. Pays-Bas* 113, 153–161.
- [13] Okamura, M.Y., Isaacson, R.A. and Feher, G. (1975) *Proc. Natl. Acad. Sci. USA* 72, 3491–3495.
- [14] Breton, J., Berthomieu, C., Thibodeau, D.L. and Nabedryk, E. (1991) *FEBS Lett.* 288, 109–113.
- [15] Bauscher, M., Leonhard, M., Moss, D.A. and Mäntele, W. (1993) *Biochim. Biophys. Acta* 1183, 59–71.
- [16] Engelhard, M., Gerwert, K., Hess, B., Kreutz, W. and Siebert, F. (1985) *Biochemistry* 24, 400–407.
- [17] Bosch, M.K., Gast, P., Hoff, A.J., Spoyalov, A.P. and Tsvetkov, Yu.D. (1995) *Chem. Phys. Lett.*, in press.
- [18] Michel, H., Epp, O. and Deisenhofer, J. (1986) *EMBO J.* 5, 2445–2451.
- [19] Allen, J.P., Feher, G., Yeates, T.O., Komiya, H. and Rees, D.C. (1988) *Proc. Natl. Acad. Sci. USA* 85, 8487–8491.
- [20] El-Kabbani, O., Chang, C.-H., Tiede, D., Norris, J. and Schiffer, M. (1991) *Biochemistry* 30, 5361–5369.
- [21] Giangiaco, K. and Dutton, P.L. (1989) *Proc. Natl. Acad. Sci. USA* 86, 2658–2662.
- [22] Buchanan, S., Michel, H. and Gerwert, K. (1992) *Biochemistry* 31, 1314–1322.
- [23] Bauscher, M. and Mäntele, W. (1992) *J. Phys. Chem.* 96, 11101–11108.
- [24] Feher, G., Isaacson, R.A., Okamura, M.Y. and Lubitz, W. (1985) in *Antennas and Reaction Centers of Photosynthetic Bacteria* (Michel-Beyerle, M.E., ed.), pp. 174–189, Springer, Berlin.
- [25] Lubitz, W., Abresch, E.C., Debus, R.J., Isaacson, R.A., Okamura, M.Y. and Feher, G. (1985) *Biochim. Biophys. Acta* 808, 464–469.