

# The metallophilic interaction between cyclometalated complexes: photobiological applications

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# **APPENDIX I SUPPORTING INFORMATION FOR CHAPTER 2**



Scheme AI.1 (a) Toluene, BINAP, Pd(dba)<sub>2</sub>, KOt-Bu, 85 °C, N<sub>2</sub>, 3 days; (b) CH<sub>3</sub>COOH, N<sub>2</sub>, 135 °C, 12 h.



**Figure AI.1** IR spectra of ligands **HL**<sup>1</sup> (a), **HL**<sup>2</sup> (b), palladium complexes **PdL**<sup>1</sup> (c), **PdL**<sup>2</sup> (d), and the reference complex [Fe(bbpya)(NCS)<sub>2</sub>] (e).

Complex	PdL <sup>2</sup>
Crystal data	
Chemical formula	$C_{21}H_{14}N_4Pd$
$M_{ m r}$	428.76
Crystal system, space	Monoclinic, $P2_1/n$
group	
Temperature (K)	110
<i>a</i> , <i>b</i> , <i>c</i> (Å)	10.3734 (3), 7.2412 (2), 20.6547 (6)
β (°)	100.020 (3)
$V(Å^3)$	1527.83 (8)
Ζ	4
Radiation type	Μο Κα
$\mu (mm^{-1})$	1.23
Crystal size (mm)	0.23  imes 0.11  imes 0.04
Data callection	
Data conection	SumarNaus Dual Cu at zone Atlas
A has mation as mastion	Supernova, Dual, Cu al zero, Allas
Absorption correction	Gaussian Cruchlic DBO 117120200 (Disclus Outerd Diffusction 2017)
	CrysAlls PRO 1.1/1.39.29c (Rigaku Oxford Diffraction, 2017)
	Numerical absorption correction based on gaussian integration over a
	multifaceted crystal model Empirical absorption correction using
	spherical harmonics, implemented in SCALE3 ABSPACK scaling
<b>— —</b>	algorithm.
$T_{\min}, T_{\max}$	0.625, 1.000
No. of measured,	20278, 3517, 3063
independent and	
observed $[I > 2\sigma(I)]$	
reflections	
R <sub>int</sub>	0.036
$(\sin \theta / \lambda)_{max} (A^{-1})$	0.650
Refinement	
$R[F^2 > 2\sigma(F^2)].$	0.025, 0.060, 1.07
$wR(F^2)$ , S	
No. of reflections	3517
No. of parameters	320
No. of restraints	581
H-atom treatment	H-atom parameters constrained
$\Lambda_{0max}$ $\Lambda_{0min}$ (e Å <sup>-3</sup> )	0.88 -0.79

Table AI.1 Crystallographic Data for  $PdL^2$ 



**Figure AI.2** Time-evolution of the absorption spectra of  $PdL^1$  (left) and  $PdL^2$  (right) in PBS: DMSO (50 µm, 1:1) solution at 310 K for 24 hours. The spectra were measured every 30 min.



**Figure AI.3** Evolution of the absorption spectrum of complexes  $PdL^1$  (a) and  $PdL^2$  (b) when incubated in Opti-MEM containing 2.5% FCS at 310 K for 24 hours. The spectra were measured every 30 min. Inset: time evolution of the absorbance at 450 nm. The color of spectra change from black (0 h) to red (24 h).

Table AI.2 Octanol-water partition coefficients (log Pow) of the two palladium complexes.

Complex	log Pow	
PdL <sup>1</sup>	-0.64	
PdL <sup>2</sup>	0.046	



**Figure AI.4** Dose-response curves for A431 cells incubated with palladium complexes and irradiated 5 min with blue light (blue data points), or left in the dark (black data points). Photocytoxicity assay outline: cells seeded at  $5 \times 10^3$  cells/well at t = 0 h; treated with **PdL**<sup>1</sup> or **PdL**<sup>2</sup> at t = 24 h; irradiated at t = 48 h, blue light (455 nm, 5 min, 10.5 mW cm<sup>-2</sup>, 3.2 J cm<sup>-2</sup>); SRB assay performed at t = 96 h. Incubation conditions: 37 °C, 21% O<sub>2</sub>, 7% CO<sub>2</sub>.

Complex	$\lambda_{abs}$ , nm ( $\epsilon \ge 10^3 \text{ M}^{-1}$	λem	$\phi_p{}^b$	φa <sup>b</sup>	Lifetime (ns) <sup>a,c</sup>	
	<b>cm</b> <sup>-1</sup> ) <sup>a</sup>	(nm) <sup>b</sup>			$ au_1$	$ au_2$
PdL <sup>1</sup>	251 (26.8), 286 (17.1), 422 (4.3)	539	0.0017	0.89	0.271±0.002 (97%)	6.4±0.3 (3%)
PdL <sup>2</sup>	283 (16.7), 347 (11.7)	604	0.00084	0.38	0.333±0.005 (96%)	5.3±0.5 (4%)

Table AI.3 Photophysical data for PdL<sup>1</sup> and PdL<sup>2</sup>

<sup>a</sup> Measurements were carried out in methanol. <sup>b</sup> Measurements were carried out at 450 nm excitation wavelength and in a solution of deuterated methanol in air atmosphere according to literature. The absorption of complexes at 450 nm were set below 0.1 to avoid the generation of excimer. A solution of  $[Ru(bpy)_3]Cl_2$  (Tris(2,2'-bipyridyl)dichlororuthenium(II)) in deuterated methanol (photoluminescence quantum yield  $\varphi_P = 0.015$ , singlet oxygen quantum yield  $\varphi_A = 0.73$ ) was used as reference. <sup>c</sup> Excitation source 375 nm. Biexponential model:  $y = y_0 + A_1 \exp(-(x-x_0)/\tau_1) + A_2 \exp(-(x-x_0)/\tau_2)$ 



**Figure AI.5** The phosphorescence lifetime spectra and fit curve of palladium complexes in methanol. Fit equation:  $y = y_0 + A_1 \exp(-(x-x_0)/\tau_1) + A_2 \exp(-(x-x_0)/\tau_2)$ .

Table AI.4 HOMO and LUMO energies, Energy gap ( $\Delta E$ ) of PdL<sup>1</sup> and PdL<sup>2</sup>

Complex	LUMO/eV	HOMO/eV	ΔE/eV
$PdL^{\overline{1}}$	-2.048	-5.258	3.21
PdL <sup>2</sup>	-2.114	-5.644	3.53



**Figure AI.6** TDDFT-calculated spectra for **PdL**<sup>1</sup>, **PdL**<sup>2</sup> (left) and [**PdHL**<sup>1</sup>]<sup>+</sup>, [**PdHL**<sup>2</sup>]<sup>+</sup> (right) at the TZP/COSMO level in methanol. The theoretical curves were calculated and plotted in ADF with Gaussian Fixed Oscillator Strengths. Scaling factor = 1.0, peak width = 30 nm.

#### **APPENDIX II SUPPORTING INFORMATION FOR CHAPTER 3**



Scheme AII.1 Synthesis of ligands  $MeL^1$ - $MeL^3$  and of their palladium complexes. Reaction condition: (a) CH<sub>3</sub>I, KO*t*-Bu, DMF, room temperature, 24 h; (b) palladium(II) acetate, CH<sub>3</sub>COOH, N<sub>2</sub>, 135 °C, 24 h; (c) Palladium(II) acetate, MeOH, 65 °C, 24 h.



**Figure AII.1** The aromatic region of the <sup>1</sup>H NMR spectrum of complexes [1]<sup>+</sup>-[3]<sup>2+</sup> at low (2 mg/mL) and high (7 mg/mL) concentration. Solvent: CD<sub>3</sub>OD.



**Figure AII.2** DFT calculation of HOMOs (bottom) and LUMOs (top) orbitals of  $[1]^+-[3]^{2+}$ ; occupied orbitals (HOMO) have red and blue lobes, and unoccupied orbitals (LUMO) brown and cyan lobes. Element color code: grey = C; orange = Pd; blue = N; white = H. Level of theory: ADF/DFT/PBE0/TZP/COSMO(water).

Complex	НОМО	/eV	LUMO/eV	ΔE/eV
[1]+	-6.24		-2.41	3.83
[2]+	-6.58		-2.46	4.11
<b>[3]</b> <sup>2+</sup>	-6.95		-2.82	4.13

Table AII.2 HOMO and LUMO energies, energy gap ( $\Delta E$ ) of complexes [1]<sup>+</sup>-[3]<sup>2+</sup>.



**Figure AII.3** TDDFT-calculated spectra for palladium complexes [1]<sup>+</sup>-[3]<sup>2+</sup>. Level of theory: ADF/TDDFT/PBE0/TZP/COSMO (water).



**Figure AII.4** The phosphorescence lifetime spectra and fit curve of complexes  $[1]^+-[3]^{2+}$  in water under air atmosphere at room temperature. Fit equation:  $y = y_0 + A_1 \exp(-(x-x_0)/\tau_1)$  ([1]<sup>+</sup> and [2]<sup>+</sup>,  $y = y_0 + A_1 \exp(-(x-x_0)/\tau_1) + A_2 \exp(-(x-x_0)/\tau_2)$  ([3]<sup>2+</sup>). Excitation wavelength: 375 nm. The data were analysized via OriginPro 9.1.

comp	olex	H <sub>2</sub> O	PBS	Opti-MEM FCS	with	Opti-MEM FCS	without	Opti-MEM BSA	with
Solut	ion only	54	40	894		107		8257	
<b>[1]</b> +	5 μΜ	37	65	2268		1936			
[T]	50 µM	146	187	19144		18970		14105	
[2]+	5 μΜ	30	58	2775		5657			
	50 µM	67	61	22311		25377		13359	
Г <b>2</b> 12+	5 μΜ	56	60	927		136			
[3]-	50 µM	79	21	930		101		8056	

**Table AII.5** The derived countrate (kcps) values of complexes in different solutions and different concentrations.



Figure AII.5 Size distributions according to DLS of solution of [1]OAc-[3]OAc at 5 or 50  $\mu$ M in different solvents.



**Figure AII.6** DLS size distribution (left and middle) and derived count rate (right) of **[1]OAc** (50 μM) in Opti-MEM complete medium at different pH.



**Figure AII.7** Time evolution of the absorbance spectra of solutions of  $[1]^+$ - $[3]^{2+}$  in H<sub>2</sub>O (a), PBS (b), cell medium with FCS (c), cell medium without FCS (d), H<sub>2</sub>O with GSH (200  $\mu$ M, e) and H<sub>2</sub>O with ascorbic acid (200  $\mu$ M, f) at 310 K. Concentration 50  $\mu$ M. The color change of the spectra indicates the time, with black corresponding to the first curve (0 h) and red to the last one (24 h).



**Figure AII.8** TEM images of samples prepared from evaporated MilliQ water solutions of [1]OAc, [2]OAc, and [3](OAc)<sub>2</sub> (50 μM). Scale bar: 2 μm (top) and 200 nm (bottom).



Figure AII.9 The Cryo-TEM images of complexes  $[1]^+$ - $[2]^+$ (50  $\mu$ M) in the Opti-MEM medium with or without FCS.



**Figure AII.10** Structures of the DFT-optimized dimers  $\{[1]^+\}_2$  (left) and  $\{[2]^+\}_2$  (right).



Figure AII.11 The absorbance (a) and emission spectra (b) of [1]OAc (50  $\mu$ M) in water and Opti-MEM complete medium.



**Figure AII.12** The absorbance of ABMDMA Opti-MEM complete solution (100  $\mu$ M) in the absence or presence of [Ru(bpy)<sub>3</sub>]Cl<sub>2</sub> (50  $\mu$ M) under dark or blue light (450 nm) irradiation. (b) Absorbance time evolution and linear fit curve at 378 nm of ABMDMA Opti-MEM complete solution (100  $\mu$ M) in the absence or presence of [**1**]OAc (50  $\mu$ M), [Ru(bpy)<sub>3</sub>]Cl<sub>2</sub> (50  $\mu$ M) under blue light irradiation. The baseline for these spectra was a solution of [**1**]OAc (50  $\mu$ M) or [Ru(bpy)<sub>3</sub>]Cl<sub>2</sub> (50  $\mu$ M) in Opti-MEM medium without ABMDMA.



Figure AII.13 Dose-response curves for A549 and A431 cancer cells incubated with complexes [1]OAc-[3](OAc)<sub>2</sub> and cisplatin, either in the dark (black data points) or upon blue light irradiation (5 minutes, 5.66 mW cm<sup>-2</sup>, 1.7 J cm<sup>-2</sup>, blue data points), under normoxic condition (37 °C atmosphere, 21%  $O_2$  and 7.0% CO<sub>2</sub>).



**Figure AII.14** Dose-response curves for A549 and A431cancer cells incubated with complexes **[1]OAc-[3](OAc)**<sub>2</sub>, cisplatin, 5-ALA and Rose bengal either in the dark (black data points) or upon blue light irradiation (455 nm, 8 min, 3.54 mW cm<sup>-2</sup>, 1.7 J cm<sup>-2</sup>, blue data points) under hypoxic condition (37 °C atmosphere, 1% O<sub>2</sub> and 7.0% CO<sub>2</sub>).



**Figure AII.15** Graphical representation of the EC<sub>50</sub> values of complexes [1]OAc-[3](OAc)<sub>2</sub> in A549 and A431 in the dark or upon blue light irradiation, in normoxic vs. hypoxic conditions. (irradiation condition: normoxic 455 nm, 5 minutes, 5.66 mW cm<sup>-2</sup>, 1.7 J cm<sup>-2</sup>, hypoxic 455 nm, 8 min, 3.54 mW cm<sup>-2</sup>, 1.7 J cm<sup>-2</sup>).

Complex	Treatmen t (ng Pd)	Metal uptake (ng Pd/millio n cells)	Metal uptake efficienc y (%)	Fractions	Metal distributio n (ng Pd/million cells)	Relative metal distributio n (%)
[1]OAc	212.84	19±7	5.2	cytosol membranes nucleus cytoskeleto n	0.7±0.1 0.23±0.03 0.23±0.03 17±3	3.5 1.1 1.1 94.3
[2]OAc	212.84	14±4	9.8	cytosol membranes nucleus cytoskeleto n	$0.45\pm0.05$ $0.68\pm0.09$ $0.34\pm0.04$ $12\pm2$	3.3 5.1 2.5 89.1
[3](OAc) 2	212.84	1.7±0.2	0.8	cytosol membranes nucleus cytoskeleto n	0.42±0.09 0.28±0.06 0 1.1±0.2	23.3 15.6 0 61.1

**Table AII.7** Palladium cellular uptake according to ICP-MS analysis in the different fractions of A549 cells treated with **[1]OAc-[3](OAc)**<sub>2</sub> (1  $\mu$ M) in the dark after 24 h.



**FITC-Annexin V** 

**Figure AII.16** Annexin V/propidium iodide double staining FACS data for A549 cells after treatment with cisplatin (15  $\mu$ M) and complexes **[1]OAc-[3](OAc)**<sub>2</sub> (15  $\mu$ M) in the dark or upon blue light irradiation (455 nm, 5 minutes, 5.66 mW cm<sup>-2</sup>, 1.7 J cm<sup>-2</sup>).



Scheme AII.2 The sketch of *in vivo* experiments for complexes [1]<sup>+</sup> and [2]<sup>+</sup>.



Figure AII.17. Bodyweight of mice treated with [1]OAc and [2]OAc and control groups.

# **APPENDIX III SUPPORTING INFORMATION FOR CHAPTER 4**



**Figure AIII.1** <sup>1</sup>H NMR of complexes **[PtMeL<sup>2</sup>]OAc** and **[PtMeL<sup>2</sup>]OAc** at low (2 mg/mL MeOD) and high (7 mg/mL MeOD) concentration.



Figure AIII.2 (a) DFT calculation of HOMOs (bottom) and LUMOs (top) orbitals of  $[1]^+$  and  $[2]^+$  in monomer and dimer states. Occupied orbitals (HOMO) have red and blue lobes, and unoccupied orbitals (LUMO) brown and cyan lobes. Element color code: blue = N, grey = C, white = Pt and H.

Complex	Energy (nm)	Energy (eV)	Oscillator strength (f)	Orbital transition contribution
	370 3522	3 3177	0 3260	65.1% HOMO-1→LUMO
[1]+	570.3522	5.5477	0.3200	26.0% HOMO→LUMO+1
	441.1625	2.8104	0.2223	96.8% HOMO→LUMO
[2]+	275 1118	2 2052	0 4247	61.0%% HOMO-1→LUMO
	575.1118	5.5055	0.4247	35.7%% HOMO→LUMO
	518.4094	2.3916	0.0188	78.3% HOMO→LUMO
1[1],12+	471.8107	2.6278	0.0996	93.8% HOMO→LUMO
[1]2	151 6116	2 7272	0.0202	15.3% HOMO-2→LUMO
	434.0110	2.1213	0.0203	11.9% HOMO-1→LUMO+1
	473.1412	2.6204	0.0163	83.3% HOMO→LUMO
[ <b>2</b> ]2  <sup>2+</sup>	447.1158	2.7730	0.0697	93.5%HOMO→LUMO
	413.8552	2.9958	0.0506	91.8% HOMO-1→LUMO

**Table AIII.1** The lowest ( $\lambda$ >350 nm) and most intense (f > 0.01) TDDFT singlet-singlet transitions calculation information of platinum complexes in monomer and dimer states.



**Figure AIII.3** Singlet oxygen generation of aggregates of **[1]OAc-[2]OAc** and reference  $[Ru(bpy)_3]Cl_2$  in Opti-MEM complete medium. (a-c) The absorbance change of ABMDMA (100  $\mu$ M) in Opti-MEM complete in presence of **[1]OAc-[2]OAc** or reference  $[Ru(bpy)_3]Cl_2$  (50  $\mu$ M) upon blue light irradiation (d) Evolution of the absorbance at 378 nm vs. irradiation time of ABMDMA (100  $\mu$ M) in Opti-MEM complete medium in the absence or presence of **[1]OAc-[2]OAc** or reference  $[Ru(bpy)_3]Cl_2$  (50  $\mu$ M) under blue light irradiation. Irradiation conditions: 298 K, 450 nm, 5.23 mW cm<sup>-2</sup>, 80 s.



**Figure AIII.4** The phosphorescence lifetime spectra and fitting curves for platinum complexes [1]OAc and [2]OAc in aerated methanol, water and Opti-MEM. Fit equation:  $y = y_0 + A_1 \exp(-(x-x_0)/\tau_1) + A_2 \exp(-(x-x_0)/\tau_2)$ .



**Figure AIII.5** The solid emission spectra of complexes **[1]OAc** and **[2]OAc** (excitation 450 nm, laser intensity 50 mW cm<sup>-2</sup>, Edmund optics high-pass filter was used to filter the spectra below 500 nm).

Appendix III



**Figure AIII.6** Time evolution for 24 h of the absorbance spectra of solutions of **[1]OAc** and **[2]OAc** (50  $\mu$ M) in PBS (a), OptiMEM with (b) or without (c) FCS medium (2.5 % v/v). Interval: 15 min, from black (0 h) to red (24 h).



**Figure AIII.7** Size distribution observed by Dynamic Light Scattering (DLS) for **[1]OAc** and **[2]OAc** (50  $\mu$ M) dissolved in different buffers. PBS = phosphate buffer saline; BSA = bovine serum albumin (50 g/L in Opti-MEM medium); GB = globulin (30 g/L in Opti-MEM medium).



Figure AIII.8 TEM images of [1]OAc and [2]OAc (50  $\mu$ M) in Opti-MEM with FCS (2.5 % v/v).



**Figure AIII.9** Selected timepoints from 16-h time-lapse confocal fluorescence live-cell imaging of **[1]OAc** uptake by A549 cells. The complex was added in the dish 15 min before initiation of the measurement (t=0). The top row shows overlayed bright field and **[1]OAc** fluorescence images. The bottom row shows **[1]OAc** fluorescence images. Red arrows indicate **[1]OAc** uptake by the cell. Brightfield and fluorescence (561 nm excitation) images were acquired every 50 s. Time is indicated in hours:minutes:seconds format. The selected time points are extracted from <u>Video S1</u>.



Figure AIII.10 Confocal images of A549 cells treated with or without complexes [1]OAc and [2]OAc (red, 650-750 nm,  $\lambda_{ex} = 552$  nm), and staining with Hoechst 33342 (blue, 420-480 nm,  $\lambda_{ex} = 405$  nm). Scale bar: 20 µm. complex concentration 5 µM, Pearson coefficients -1.



**Figure AIII.11** Confocal microscopy overlapping images and pixel intensity curves for A549 cells co-treated with complexes **[1]OAc** (red, 5  $\mu$ M ) and (green) dyes Endoplasmic reticulum Green Fluorescent Protein (GFP), Mitochondria GFP, MitoTracker Green FM, LysoTracker<sup>TM</sup> Green DND-26, HCS LipidTOX<sup>TM</sup> Green Neutral Lipid Stain and BODIPY<sup>TM</sup> FL C5-Ceramide complexed to BSA Golgi. Green channel ( $\lambda_{ex} = 488$  nm,  $\lambda_{em} = 500-530$  nm). Red channel ( $\lambda_{ex} = 552$  nm,  $\lambda_{em} = 650-750$  nm).



**Figure AIII.12** Confocal microscopy overlapping images and pixel intensity curves for A549 cells co-treated with complex **[2]OAc** (red, 5  $\mu$ M ) and (green) dyes Endoplasmic reticulum Green Fluorescent Protein (GFP), Mitochondria GFP, MitoTracker Green FM, LysoTracker<sup>TM</sup> Green DND-26, HCS LipidTOX<sup>TM</sup> Green Neutral Lipid Stain and BODIPY<sup>TM</sup> FL C5-Ceramide complexed to BSA Golgi. Green channel ( $\lambda_{ex} = 488$  nm,  $\lambda_{em} = 500 - 530$  nm). Red channel ( $\lambda_{ex} = 552$  nm,  $\lambda_{em} = 650 - 750$  nm).

**Table AIII.2** Pearson's correlation coefficients measured for co-localization studies of complexes **[1]OAc** or **[2]OAc** and different organelle-staining dyes in confocal cell imaging shown in Figure S15 and S16.

Dro	Complex	
Dye	[1]OAc	[2]OAc
Endoplasmic reticulum GFP	-0.1 to 0.2	-0.1 to 0.2
Mitochondria GFP	-0.2 to 0.4	0.3 to 0.6
MitoTracker Green FM	0.2 to 0.4	0.3 to 0.5
LysoTracker <sup>™</sup> Green DND-26	-0.3 to 0.1	0.0 to 0.3
HCS LipidTOX <sup>™</sup> Green Neutral Lipid Stain	0.0 to 0.1	-0.1 to 0.1
BODIPY <sup>TM</sup> FL C5-Ceramide complexed to BSA golgi Dye	0.5 to 0.7	-0.4 to 0.6

**Table AIII.3** Platinum cellular uptake according to ICP-MS analysis in A549 cells treated with [1]OAc and [2]OAc (5  $\mu$ M, 1 mL) in combination with different inhibition conditions after 2 h.

Complex	inhibitor	uptake (ng Pd/million cells)	Inhibition percent <sup>a</sup>
	control	111±25	0%
[1]OAc	dynasore	84±7	24%
	4 °C	52±2	53%
	control	159±5	0%
	dynasore	58±7	64%
	4 °C	66±8	58%

<sup>a</sup> Inhibition percent = (uptake<sub>control</sub>-uptake<sub>inhibitor</sub>)/uptake<sub>control</sub>  $\times$  100%.



Figure AIII.13 Electron Microscopy imaging of control A549 cells without platinum treatment (scale bar 1  $\mu$ M).

## **APPENDIX IV SUPPORTING INFORMATION FOR CHAPTER 5**

Table AIV.1 Selected bond distances (Å) and angels (degree) in the crystal structure of PdL.

Distance (Å)		Angel (°)	
Pd-N1	2.144(3)	C11-Pd1-C17	92.09(12)
Pd-C11	1.969(3)	C11-Pd1-N1	80.17(11)
Pd-C17	1.972(3)	C17-Pd1-N1	171.99(11)
Pd-N3	2.163(3)	C17-Pd1-N3	80.24(11)°
Pd-Pd	3.518	N1-Pd1-N3	107.42(10)

Table AIV.2 TDDFT singlet-singlet transitions calculation information of PdL in monomeric or dimeric state.

State	Energy	Energy	<b>Oscillator strength</b>	Orbital transition
	( <b>nm</b> )	(eV)	( <b>f</b> )	contribution
Monomer	383.0	4.0697	0.1262	HOMO→LUMO 89.9%
	335.2	3.6989	0.3642	HOMO→LUMO+1 84.7%
	304.65	3.2369	0.7603	HOMO-1→LUMO 70.9%
Dimer	540.1232	2.2955	0.0043	HOMO→LUMO 100%
	501.6673	2.4714	0.0484	HOMO→LUMO 96%
	450.6614	2.7512	0.0168	HOMO→LUMO 47.3%
				HOMO-1→LUMO 36.9%
	400.7565	3.0938	0.0819	HOMO→LUMO+1 86.4%

Table AIV.3 The photophysical properties of PdL in monomeric state.

Solvent	$\lambda_{abs}$ , nm ( $\epsilon \ge 10^3 \text{ M}^{-1} \text{ cm}^{-1}$ ) <sup>a</sup>	$\lambda_{em} (nm)^{a,b}$	lifetime (ns) <sup>a,d</sup>	φ <sub>p</sub> <sup>c</sup>	φ <sub>Δ</sub> <sup>c</sup>
DMSO	343 (25.8), 405 (5.2), 481 (3.7)	564	$0.406\pm0.004$	0 0008	0.00
THF	347 (22.5), 410 (4.3), 480 (2.9)	540	$0.432 \pm 0.005$	0.0008	0.09

<sup>a</sup> measurement were carried out in aerated DMSO or THF <sup>b</sup> excitation and concentration: 419 nm, 100 μM.

<sup>c</sup> measurement was carried out in MeOD.

<sup>d</sup> excitation source : 340 nm. Monoexponential model:  $y = y_0 + A_1 \exp[-(x-x_0)\tau_1]$ 



**Figure AIV.1** (a) DLS derived count rate in the DMSO/H<sub>2</sub>O or THF/H<sub>2</sub>O system of **PdL** (100  $\mu$ M) after 30 min self-assembly; (b) Size distribution of the DLS analysis in the DMSO/H<sub>2</sub>O or THF/H<sub>2</sub>O system of **PdL** (100  $\mu$ M) after 30 min self-assembly; TEM images of samples prepared from the DMSO/H<sub>2</sub>O (c) or THF/H<sub>2</sub>O (d) system of **PdL** (100  $\mu$ M) after 30 min self-assembly. Inset picture scale bar: 500 nm.



**Figure AIV.2** (a) The molar absorption coefficient (black solid line) and emission spectra of **PdL** in tetrahydrofuran (THF) solution at different concentrations (blue dash line 10  $\mu$ M; black dash line 100  $\mu$ M, red dash line 1000  $\mu$ M). (b) Time evolution of the absorption spectra of H<sub>2</sub>O/THF solution (100  $\mu$ M, 9:1, v/v) of **PdL** at 298 K for 30 min (30 s interval, the color of

spectra change from black (0 min) to red (30 min); the blue line is the absorbance spectra of **PdL** (100  $\mu$ M) in pure THF). Inset: time evolution of the absorption at 350 nm (black square), 480 nm (red dot), 504 nm (green triangle) of the solution. (d) Emission spectra of **PdL** (100  $\mu$ M) in pure THF (fw (V<sub>water</sub>/V<sub>total</sub>) = 0.0) and water/THF mixture (9:1, v/v, fw = 0.9); excitation 419 nm. (d) Emission spectra of **PdL** (20  $\mu$ M) in different THF/water ratio (from v/v = 10/0 to 1/9, excitation 450 nm).



**Figure AIV.3** DLS scattering-derived count rate of **PdL** in Opti-MEM complete medium at 0 and 30 min under room temperature.



**Figure AIV.4** Time evolution of the absorption spectrum of an Opti-MEM complete solution of 9,10-anthracenediyl-bis(methylene)-dimalonic acid (ABMDMA, 100  $\mu$ M) in the absence or presence of **PdL** (25  $\mu$ M) or rose Bengal, under green light irradiation (515 nm).



Figure AIV.5 The emission spectra of dihydroethidium (DHE) solution (DMSO or Opti-MEM complete) in the absence or presence of PdL (25  $\mu$ M) under green light irradiation (520 nm) or in the dark, over 60 s.



**Figure AIV.6** Dose-response curves for 2D-monolayer (a) or 3D-spheroid (b) for different human cancer cell lines incubated with **PdL**, either in the dark (black data points) or upon green

light irradiation (green data points) under normoxic-2D (520 nm, 20 min, 10.92 mW/cm<sup>2</sup>, 13 J/cm<sup>2</sup>), hypoxic-2D (520 nm, 32 min, 6.90 mW/cm<sup>2</sup>, 13 J/cm<sup>2</sup>), normoxia-3D spheroid condition (520 nm, 32 min, 6.90 mW/cm<sup>2</sup>, 13 J/cm<sup>2</sup>), or hypoxia-3D spheroid condition (520 nm, 55 min, 3.99 mW/cm<sup>2</sup>, 13 J/cm<sup>2</sup>).



**Figure AIV.7** Annexin V/Propidium iodide double staining FACS data for A375 cells after treatment with cisplatin (7.5  $\mu$ M) or **PdL** (0.5  $\mu$ M or 2  $\mu$ M) in the dark or upon green light irradiation (normoxic 520 nm, 20 min, 10.9 mW/cm<sup>2</sup>, 13 J/cm<sup>2</sup>) after 2 (a), 4 (b) and 24 h (c)



**Figure AIV.8** Bright filed images (left) and diameter (right,  $\mu$ m) for A549 (a) and A375 (b) 3D tumor spheroids kept in the dark (black bars) or irradiated with green light (green bars, 520 nm, 13 J/cm<sup>2</sup>). Scar bar 500  $\mu$ m.



Figure AIV.9 The H&E staining of different mice organs after treatment with vehicle control or PdL, and either without or with green light irradiation (100 mW/cm<sup>2</sup>, 10 min, 60 J/cm<sup>2</sup>). Scale bar 200  $\mu$ m.



**Figure AIV.10** EM images showing the morphology of nanoparticles found in the blood of mice 12 h after intravenous tail injection of **PdL** (middle and right images), or in an untreated control mice (left image). Injection dose: 2.1  $\mu$ mol/kg, 0.9 mg/kg, 420  $\mu$ M, 100  $\mu$ L saline.

## **APPENDIX V SUPPORTING INFORMATION FOR CHAPTER 6**

[1]Cl		[2]AuCl4	
Au-C28	2.030(7)	Au-N3	1.995(18)
Au-N3	2.046(6)	Au-N6	2.00(2)
Au-N4	2.050(6)	Au-N4	2.002(16)
Au-C1	2.060(5)	Au-N1	2.081(18)
C28-Au-N3	166.6(6)	N3-Au-N6	167.2(13)
C28-Au-N4	92.7(3)	N3-Au-N4	81.4(6)
N3A Au-N4A	81.5(3)	N6-Au-N4	93.2(9)
C28-Au-C1	96.5(3)	N3-Au-N1	88.8(9)
N3-Au-C1	91.3(3)	N6-Au-N1	99.0(7)
N4-Au-C1	166.9(4)	N4-Au-N1	163.0(13)

Table AV1. The selected bond distances (Å) and angels (°) of [1]Cl and [2]AuCl4.



**Figure AV1.** The crystal packing structure of [1]Cl with the intermolecular  $\pi$ - $\pi$  stacking distance. The H and counterions were omitted for clarity.

Table AV2. The TDDFT singlet-singlet transitions calculation information of gold complexes.

Complex	Energy (nm)	Energy (eV)	Oscillator strength (f)	Orbital transition contribution
[1]+	451	2.7451	0.2282	97.4% HOMO→LUMO
	370	3.3471	0.1146	95.6% HOMO-1→LUMO
	694	1.7863	0.0074	97.8% HOMO→LUMO
[2]+	531	2.3334	0.0066	98.1% HOMO-1→LUMO
	469	2.6413	0.1915	95.9% HOMO→LUMO+1



Figure AV2. The TDDFT absorbance peaks and spectra of [1]<sup>+</sup> and [2]<sup>+</sup>.



**Figure AV3.** Time evolution of the absorbance spectrum of [1]Cl and [2]Cl (50  $\mu$ M) in PBS solution (100  $\mu$ M) for 24 hours at 37 °C; measurement interval 15 min, color changes from black (0 s) to red (24 h).



**Figure AV4.** <sup>1</sup>H NMR of the precipitate formed after 24 h of incubation of compound [2]Cl (3.0 mM) and GSH (6.0 mM) at a ratio of 1:2 (top) and spectra of H<sub>2</sub>biqbpy2 (bottom) added as a reference for peak identification. All spectra were measured in DMSO-d<sub>6</sub> at 25 °C.



Figure AV5. Dose-response curves for four cancer cells and one healthy skin cells incubated with [1]Cl under normoxic or hypoxic conditions, in the 2D-monolayer or 3D-spheroid model.



**Figure AV6.** Dose-response curves for four cancer cells and one healthy skin cells incubated with **[2]Cl** under normoxic or hypoxic conditions, in the 2D-monolayer or 3D-spheroid model.



**Figure AV7.** Dose-response curves for four cancer cells and one healthy skin cells incubated with cisplatin under normoxic or hypoxic conditions, in the 2D-monolayer or 3D-spheroid model.



**Figure AV8.** Bright-field imaging of 3D tumor spheroids treated with gold complexes or vehicle control (DMSO); scale bar is 500  $\mu$ m. The cells were seeded at t=0 (500 cells per well), treated at t=4 days, measured (diameter) at t=7 days.