

Accelerating the photocatalytic water splitting in catalyst-dye complexes Shao, Y.

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Accelerating the Photocatalytic Water Splitting in Catalyst–Dye Complexes

Yang Shao 邵洋

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Accelerating the Photocatalytic Water Splitting in Catalyst-Dye Complexes

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ter verkrijging van de graad van Doctor aan de Universiteit Leiden, op gezag van Rector Magnificus prof.dr.ir. H. Bijl, volgens besluit van het College voor Promoties te verdedigen op woensdag 24 februari 2021 klokke 13:45 uur

door

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geboren te Shandong, China in 1990

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For my parents, my wife, and my son

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List of Abbreviations

ADF	Amsterdam Density Functional
AIMD	Ab Initio Molecular Dynamics
APT	Concerted Atom-Proton Transfer
BO	Born-Oppenheimer approximation
BOMD	Born-Oppenheimer Molecular Dynamics
bpy	2,2'-bipyridine
СВ	Conduction Band
CFF	Consistent Force Field
CHARMM	Chemistry at HARvard Macromolecular Mechanics
COSMO	Conductor-like Screening Model
CPMD	Car-Parrinello Molecular Dynamics
су	<i>p</i> -cymene
DCACP	Dispersion-Correcting Atom-Centered Potential
DFT	Density Functional Theory
DFT-MD	DFT-based Car-Parrinello Molecular Dynamics
DS-PEC	Dye-Sensitized Photoelectrochemical Cell
DSSC	Dye-sensitized Solar Cells
EPT	Concerted Electron-Proton Transfer
ET	Electron Transfer
FMD	Free Molecular Dynamics
FS	Final State
GEA	Gradient Expansion Approximation
GGA	Generalized Gradient Approximation
GTH	Goedecker-Teter-Hutter
HEC	Hydrogen-Evolving Catalyst
НОМО	Highest Occupied Molecular Orbital
IEM	Ion Exchange Membrane
IS	Initial State
I2M	Oxo-oxo Coupling
KS	Kohn-Sham
LDA	Local Density Approximation
LUMO	Lowest Unoccupied Molecular Orbital
MD	Molecular dynamics
NCAP	Nonadiabatic Conversion by Adiabatic Passage
NDI	2,6-diethoxy-1,4,5,8-diimide-naphthalene
NPT	Isothermal-isobaric Ensemble

NVT	Canonical Ensemble
OEC	Oxygen Evolving Center
OPBE	OPTX-Perdew-Burke-Ernzerhof
OPTX	Handy's Optimized Exchange
PBC	Periodic Boundary Conditions
PBE	Perdew-Burke-Ernzerhof
PCET	Proton-Coupled Electron Transfer
PEC	Photoelectrochemical Cell
PEM	Proton Exchange Membrane
PSI	PhotoSystem I
PSII	PhotoSystem II
PV-E	PV-Electrolysis
РТ	Proton Transfer
PV	Photovoltaics
SOMO	Singly Occupied Molecular Orbital
TD-DFT	Time-Dependent Density Functional Theory
TIP3P	Transferable Intermolecular Potential with 3 Points
TS	Transition State
TZP	Triple-Zeta Polarized Basis Set
VDOS	Vibrational Density of States
VMD	Visual Molecular Dynamics
WNA	Water Nucleophilic Attack
WOC	Water Oxidation Catalyst

List of Symbols

pre-exponential frequency factor
C–N bond length
C–N bond length of the initial intermediate
C–N bond length of the final intermediate
time-averaged C–N bond length
Euler's number
overpotential
ground state energy
exchange-correlation functional
total bonding energy
energy difference between molecular orbitals
energy difference between intermediates
excitation energy around the transition state
oscillator strength
radial distribution function
activation free energy barrier
thermodynamic driving force
free energy change
calculated free energy change
experimentally measured free energy change
Planck constant
classical Coulomb interaction
reaction rate
Boltzmann constant
Kohn-Sham orbital
coordination number
O…O distance
Universal gas constant
electron density
total spin angular momentum
spin multiplicity
thermodynamic temperature
kinetic energy
standard deviation
dihedral angle
dihedral angle of the initial intermediate

$\theta_{\text{fin}} < \theta >$	dihedral angle of the final intermediate time-averaged dihedral angle
λ	constraint force
<λ>	time-averaged constraint force
<λ> _r	running average of constraint force
μ	fictitious mass of the electronic degrees of freedom
$\Lambda_{ m ij}$	Lagrange multipliers
$V_{ m ee}[ho]$	electron-electron interaction
$V_{ m ext}[ho]$	nucleus-electron interaction
$v_{\rm ext}(r)$	external potential
ω	vibrational frequency