



Universiteit
Leiden
The Netherlands

Testing electronic friction models: Vibrational De-excitation in scattering of H₂ and D₂ from Cu(111)

Spiering, P.; Meyer, J.

Citation

Spiering, P., & Meyer, J. (2018). Testing electronic friction models: Vibrational De-excitation in scattering of H₂ and D₂ from Cu(111). *Journal Of Physical Chemistry Letters*, 9(7), 1803-1808. Retrieved from <https://hdl.handle.net/1887/61995>

Version: Not Applicable (or Unknown)

License:

Downloaded from: <https://hdl.handle.net/1887/61995>

Note: To cite this publication please use the final published version (if applicable).

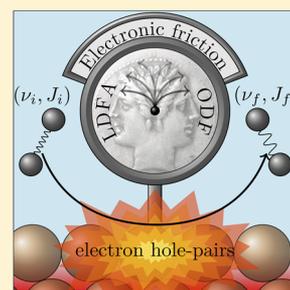
Testing Electronic Friction Models: Vibrational De-excitation in Scattering of H₂ and D₂ from Cu(111)

Paul Spiering^{1b} and Jörg Meyer^{*1b}

Leiden Institute of Chemistry, Gorlaeus Laboratories, Leiden University, P.O. Box 9502, 2300 RA Leiden, The Netherlands

Supporting Information

ABSTRACT: At present, molecular dynamics with electronic friction (MDEF) is the workhorse model to go beyond the Born–Oppenheimer approximation in modeling dynamics of molecules at metal surfaces. Concomitant friction coefficients can be calculated with either the local density friction approximation (LDFA) or orbital-dependent friction (ODF), which, unlike LDFA, accounts for anisotropy while relying on other approximations. Due to the computational cost of ODF, extensive high-dimensional MDEF trajectory calculations of experimentally measurable observables have hitherto only been performed based on LDFA. We overcome this limitation with a continuous neural-network-based representation. In our first application to the scattering of vibrationally excited H₂ and D₂ from Cu(111), we predict up to three times higher vibrational de-excitation probabilities with ODF than with LDFA. These results indicate that anisotropic electronic friction can be important for specific molecular observables. Future experiments can test for this “fingerprint” of different approximations underlying state-of-the-art MDEF.



The motion of atomic and molecular adsorbates on metal surfaces underlies every elementary reaction step in heterogeneous catalysis. Due to the absence of an energy gap between valence and conduction band electrons, these motions can result in the excitation of electron–hole pairs (EHPs) and thus violate the Born–Oppenheimer approximation.^{1–3} A growing number of experiments points to the importance of this nonadiabatic energy loss channel.⁴ On the other hand, the development of suitable theoretical models to account for these nonadiabatic effects is still an ongoing process.^{5–9} For systems with weak nonadiabatic coupling, molecular dynamics with electronic friction (MDEF)¹⁰ is currently the most popular approach.¹¹ MDEF relies on a potential energy surface (PES), nowadays typically obtained from density functional theory (DFT),³ and accounts for the effects of the EHPs on the motion of the nuclei by electronic friction coefficients.¹⁰ One state-of-the-art technique for calculating these coefficients as functions of the adsorbate positions relies on mapping to an atom-in-jellium model for which only the surface electron density is considered (local density friction approximation, LDFA^{12,13}). Alternatively, the electronic states of the molecule–surface system can be taken into account (orbital-dependent friction, ODF^{10,14}). For the inelastic scattering of H atoms from Au(111), millions of MDEF trajectories based on a high-dimensional PES¹⁵ and LDFA have recently been demonstrated to yield accurate scattering probabilities in excellent agreement with experimental data.¹⁶

The situation is quite different for molecules. Due to its combination with the independent atom approximation, the LDFA completely neglects any molecular effects.¹² ODF on the other hand accounts for the anisotropic tensorial character of friction coefficients on corrugated metal surfaces and along adsorbate-internal bonds,^{17–19} which is why ODF has been argued to be “theoretically” more accurate for (diatomic)

molecules.²⁰ However, this discussion^{12,20,21} has still remained inconclusive because an evaluation of ODF comes at very high computational costs. Consequently, extensive MDEF trajectory calculations for molecules including all relevant degrees of freedom (DOF) can be easily performed with LDFA,^{11,12} whereas only two molecular DOF have so far been included for ODF.²² The very recent on-the-fly evaluation of ODF within ab initio molecular dynamics by Maurer et al.¹⁹ is an important step, but the less than 20 calculated trajectories make direct validation via molecular beam experiments impossible. Modeling the nonadiabatic contribution to vibrational lifetimes of molecules adsorbed on metal surfaces on the other hand does not require such extensive statistical averaging.^{23,24} The most recent implementations of LDFA and ODF both yield results that agree with experimental data within the error bars.^{25,26} Furthermore, Novko et al. have shown recently in this context²⁷ that the numerical evaluation of friction tensors within ODF^{19,26} effectively includes potentially spurious electronic memory effects with unclear consequences for MDEF.²⁸ Given this situation, theoretical understanding and modeling relying on MDEF faces an important question: Is the molecular anisotropy as described by ODF important for any observables that can be validated by high-precision molecular beam experiments like for atoms?¹⁶

In this work, we provide an answer to this question using H₂ and D₂ on Cu(111). For this system, weak nonadiabatic coupling and static surface approximation are well justified,^{29–31} and an accurate DFT-based PES relying on the semiempirically constructed “SRP” exchange–correlation func-

Received: December 1, 2017

Accepted: March 12, 2018

Published: March 12, 2018

tional is available.^{32,33} We construct a six-dimensional neural-network-based continuous representation of ODF that allows us to perform extensive MDEF trajectory calculations on equal footing with LDFA. While dissociative sticking probabilities are hardly affected in general and by the type of electronic friction coefficients used, we find vibrational de-excitation probabilities to be a “fingerprint” that can be used to distinguish and validate LDFA and ODF in future experiments.

Quasi-classical trajectory calculations³² within MDEF rely on a generalized Langevin equation¹⁰

$$m_i \frac{d^2 R_{i\alpha}}{dt^2} = -\frac{\partial V(\mathbf{R})}{\partial R_{i\alpha}} - \underbrace{\sum_{j=1}^N \sum_{\beta=1}^3 \eta_{i\alpha j\beta} \dot{R}_{j\beta}}_{F_{i\alpha}^{\text{fric}}(\mathbf{R})} + \mathcal{F}_{i\alpha}(t) \quad (1)$$

where i, j indicate atoms and α, β Cartesian coordinates. Atomic masses and positions are denoted by m_i and \mathbf{R}_i , respectively. For a H_2 or D_2 molecule on a static surface, the total number of moving atoms N is two, resulting in six DOF. In addition to the forces from the PES $-\frac{\partial V(\mathbf{R})}{\partial R_{i\alpha}}$, which yield the adiabatic dynamics, nonadiabatic effects on the nuclear dynamics originate from electronic friction forces $F_{i\alpha}^{\text{fric}}(\mathbf{R})$ and thermal white noise $\mathcal{F}_{i\alpha}(t)$, respectively. In this work, $V(\mathbf{R})$ is mainly taken to be the static surface PES based on the SRP48 exchange–correlation functional from ref 33, but we also compare with the PW91-PES from earlier work.³⁴ The friction forces are linear in nuclear velocities $\dot{R}_{j\beta}$ and are in general given by a symmetric 6×6 friction tensor $\eta_{i\alpha j\beta}(\mathbf{R})$, which consists of 21 independent elements each depending on six nuclear coordinates. These coordinates can be Cartesian $\mathbf{R} = (\mathbf{R}_1, \mathbf{R}_2)$ or expressed in the center-of-mass centered spherical coordinate system $\mathbf{R} = (X, Y, Z, d, \theta, \phi)$, which is commonly used for diatomics and described by Figure 1A.

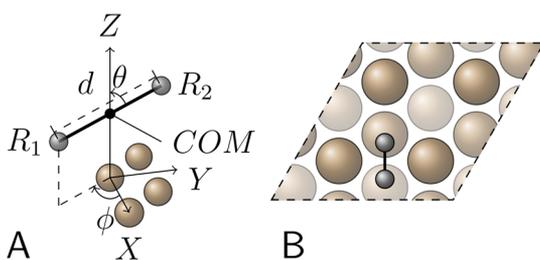


Figure 1. (A) Molecular coordinate system denoting the center of mass positions (X, Y, Z) , bond length d , as well as spherical orientation (θ, ϕ) . (B) Top view of a reference configuration with $X = \frac{1}{2}a$, $Y = 0$, and $\theta_0 = \phi_0 = 90^\circ$ from the minimum-energy reaction path for dissociative chemisorption over the bridge site,³⁴ where a denotes the surface lattice constant. Cu atoms in the first, second, and third layer are depicted by increasing transparency. Note that $X, Y, Z = 0$ corresponds to the position of a Cu atom in the surface plane (top site).

Within ODF, these 21 friction coefficients are obtained according to a Fermi golden rule-like expression resulting from time-dependent perturbation theory, which can be written in the quasi-static limit as^{10,24,26,35}

$$\eta_{i\alpha j\beta}^{\text{ODF}}(\mathbf{R}) = 2\pi\hbar \sum_{kab} g_{kab}^{i\alpha}(\mathbf{R})^* \cdot g_{kab}^{j\beta}(\mathbf{R}) \delta(\epsilon_{ka} - \epsilon_F) \delta(\epsilon_{kb} - \epsilon_F) \quad (2)$$

The electron–phonon matrix elements $g_{kab}^{i\alpha}$ describe the nonadiabatic coupling between two electronic states of the molecule at the metal surface with band indices a and b at wave vector \mathbf{k} due to the motion of (adsorbate) atom i along direction α . In general, the ODF tensor can have different diagonal elements even for the same atom. This anisotropy yields very different friction forces when the atoms move (with the same velocity) in different directions. Its generally nonzero off-diagonal elements are responsible for coupling the motion in different directions and between both atoms in a way that is not accounted for by the PES. In particular, this can lead to a strong damping of the molecular stretch vibration of a diatomic molecule and thus a pronounced molecular anisotropy.^{19,20,22}

In order to use the so-calculated $\eta_{i\alpha j\beta}^{\text{ODF}}(\mathbf{R})$ in MDEF trajectory calculations of generic experimentally measurable observables, a continuous representation of this 6×6 tensor is required that can be evaluated at low computational cost. We have designed such a representation based on a symmetry-adapted neural network fit that is briefly described in the Supporting Information and will be discussed extensively in a forthcoming publication.

Within LDFA, friction coefficients for hydrogen atoms $\eta^{\text{H}}(\rho)$ are obtained from a spherical atom-in-jellium model with background density ρ , which is solved via DFT at the level of the local density³⁶ or generalized gradient approximation.³⁷ Mapping of the actual surface problem is accomplished by taking the electron density of the bare surface (without the molecule) at each atom’s position $\rho(\mathbf{R}_i)$ as the background density of the jellium.¹² This independent-atom approximation (IAA) results in electronic friction coefficients that are isotropic for each atom and depends on its own three coordinates alone. In Cartesian coordinates, only diagonal elements of the friction tensor are nonzero

$$\eta_{i\alpha j\beta}^{\text{LDFA}}(\mathbf{R}) = \eta^{\text{H}}(\rho(\mathbf{R}_i)) \delta_{\alpha\beta} \delta_{ij} \quad (3)$$

A continuous representation of $\eta_{i\alpha j\beta}^{\text{LDFA}}(\mathbf{R})$ for extensive MDEF trajectory calculations can be easily constructed.^{12,38}

Going beyond the IAA within LDFA is possible, for example, by determining the background electron density using an atoms-in-molecules technique (LDFA-AIM).²⁵ However, this approach does not lift the isotropy and, as detailed in the Supporting Information, cannot be applied to H_2 and D_2 molecules. The other way round, isotropic friction can be constructed from ODF by neglecting the coupling between different directions and atoms plus averaging the remaining (generally anisotropic) friction over different directions

$$\eta_{i\alpha j\beta}^{\text{ODF-iso}}(\mathbf{R}) = \frac{1}{3} \sum_{\gamma} \eta_{i\gamma j\gamma}^{\text{ODF}}(\mathbf{R}) \delta_{ij} \delta_{\alpha\beta} \quad (4)$$

This ODF-iso allows one to disentangle the influence of anisotropy from the very different electronic structure inherent to ODF and LDFA.

Figure 2A–C shows η_{Zd} , η_{db} , and η_{ZZ} , respectively, as obtained from eqs 2–4 along the minimum-energy reaction path for dissociative chemisorption over the bridge site, as depicted in Figures 2D and 1B. We focus here on these three particular friction coefficients in order to compare with the earlier two-dimensional ODF calculations.^{20,22} The agreement is quite good except for some differences close to the transition state for η_{ZZ}^{ODF} . As the molecule approaches the surface, each model yields increasing friction for the six diagonal elements of the friction tensor, and the absolute values of the off-diagonal

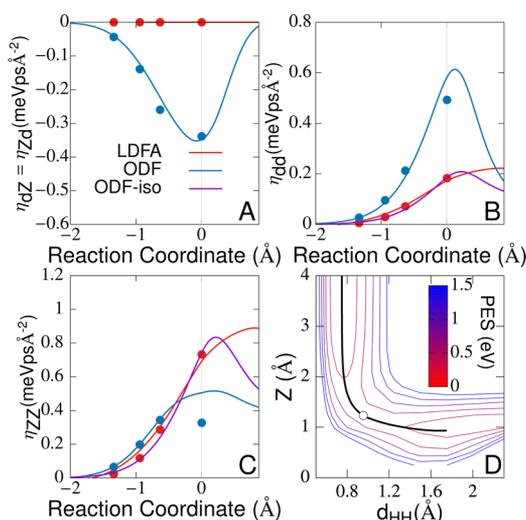


Figure 2. (A–C) η_{dZ} , η_{dd} , and η_{ZZ} in the molecular coordinate system (see Figure 1A), respectively, along the minimum-energy reaction path for dissociative chemisorption over the bridge site (see Figure 1B) as depicted in (D) together with the corresponding two-dimensional PES cut. The blue, red, and purple lines indicate the continuous representation from this work for ODF, LDFA, and ODF-iso, respectively, as obtained from eqs 2–4. Blue (red) dots show the ODF (LDFA) results from previous work of Luntz et al.,^{20,22} and the reaction coordinate is defined in the same way as in that work. The barrier and thus the transition state for dissociation are located at the vertical gray line (i.e. 0 Å) in (A–C) and indicated by the empty circle in (D). Negative numbers up to the transition state denote the approach from the gas phase (i.e., decreasing heights Z above the surface).

elements increase likewise in the case of ODF. Furthermore, ODF directly reflects the strong rearrangement of (Kohn–Sham) orbitals when approaching the dissociation barrier by significantly higher friction along the molecular bond (and thus reaction) coordinate, resulting in $\eta_{dd}^{\text{ODF}} \approx 3\eta_{dd}^{\text{LDFA}}$ at the transition state, in agreement with earlier work.²⁰ For the observables calculated below, friction beyond the dissociation barrier is not relevant. Quite remarkably, $\eta_{dd}^{\text{ODF-iso}}$ ($\eta_{ZZ}^{\text{ODF-iso}}$) and η_{dd}^{LDFA} (η_{ZZ}^{LDFA}) are almost identical up to the transition state and thus much more alike than what has originally been found for the diffusion of H atoms on Pd(100).^{17,39}

In order to study the effect of the different friction models on actual experimental observables, we perform MDEF calculations according to the quasi-classical trajectory method.²⁸ In view of the short interaction time of the molecules with the Cu(111) surface during all simulated trajectories, we neglect the fluctuating forces in eq 1.

Figure 3A,B shows the results for the dissociative chemisorption probability S_0 for both H₂ and D₂ molecular beams based on the SRP48-PES, respectively. Due to the construction of the latter,^{32,33} already the adiabatic calculations yield good agreement with the experimental data.^{40,41} Inclusion of electronic friction reduces S_0 , leading to even better agreement with the experimental data, in particular, at high incidence energies. The reduction is strongest for ODF and weaker for LDFA and ODF-iso, which are very similar to each other. It can be rationalized by the differences of the friction models for the friction η_{dd} along the reaction coordinate close to the dissociation barrier (see Figure 2B). This effect of η_{dd}^{ODF} on S_0 for H₂ and D₂ on Cu(111) has not been reported for two-dimensional ODF calculations.²² Consequently, a high-dimen-

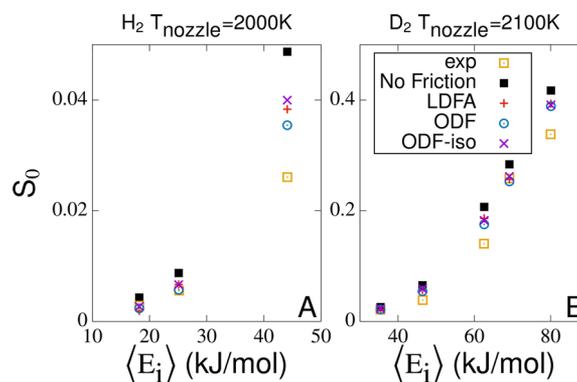


Figure 3. Calculated reaction probabilities S_0 based on the SRP48-PES for dissociative chemisorption of (A) H₂ and (B) D₂ molecular beams as a function of average normal incidence energy $\langle E_i \rangle$ for the indicated nozzle temperatures T_{nozzle} in comparison to experimental data (brown unfilled squares) from refs 40 and 41, respectively. The calculations are adiabatic (filled black squares) or employ the LDFA (red pluses), ODF (blue circles), and ODF-iso (purple crosses) models for the electronic friction coefficients.

sional treatment of ODF in MDEF, on an equal footing with LDFA,²¹ is important. However, the overall small effect of electronic friction on S_0 makes this not an optimal observable for experimental validation of the different friction models.

Instead, we have identified vibrational de-excitation probabilities $P_{\text{transition}}$ to yield a clearly distinguishable difference between LDFA and ODF. On the basis of 50000 MDEF trajectories, we calculate $P_{\text{transition}}$ as a function of incidence energy E_i from the scattered trajectories by a conventional binning procedure. The concomitant average gain in translational energy $\langle \Delta E_{\text{trans}} \rangle$ is calculated from the final center-of-mass velocities. As detailed in the Supporting Information, the error bars reflect the error due to statistical sampling of the initial conditions. Only by employing our continuous representation to compute a large amount of trajectories were we able to reduce these errors so that the different electronic friction models can be distinguished. We focus the discussion on de-excitation from vibrational state $\nu = 2$, $J = 1$ (2) to $\nu = 1$, $J = 1$ (2) for H₂ (D₂), respectively, as shown in Figure 4A,C (B,D). Unlike for other vibrational transitions,⁴² for this transition, we obtain results that are not only qualitatively but even almost quantitatively identical to corresponding results obtained with the PW91-PES (see the Supporting Information).

At low incidence energies, with increasing E_i , more and more molecules come close enough to the surface so that the curvature of the PES and electronic friction lead to an increase of $P_{\text{transition}}$. Both effects are additive and result in de-excitation probabilities that are up to 6 (2) times larger for H₂ (Figure 4C) and up to 3 (2) times larger for D₂ with ODF (LDFA), respectively (Figure 4D). At high incidence energies, the dissociation channel (see Figure 3A,B) becomes more effective, which is why $P_{\text{transition}}$ decreases again in all cases. For the adiabatic simulations on the static surface, $\langle \Delta E_{\text{trans}} \rangle$ is equal to the rovibrational energy loss of one vibrational quantum and thus by about $\sqrt{2}$ larger for H₂ than that for D₂ (Figure 4A,B). Electronic friction reduces the energy gain. The reduction is almost twice as large for ODF compared to that for LDFA. The fact that it does not very strongly depend on E_i for the energy range considered here suggests that it is dominated by η_{dd} and thus directly reflects the differences observed along the

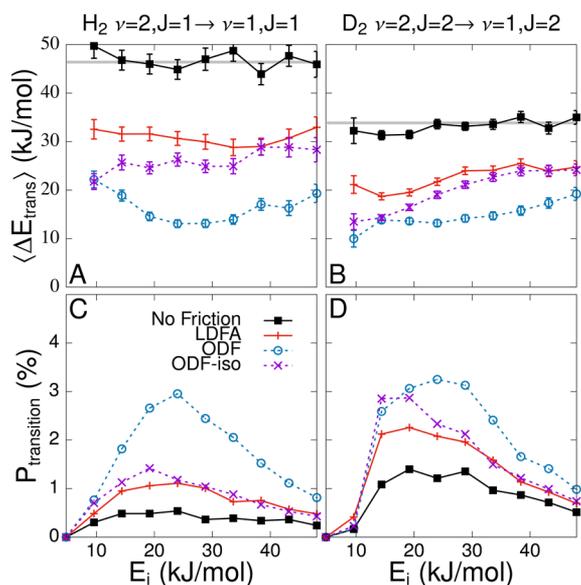


Figure 4. Vibrational de-excitation probabilities $P_{\text{transition}}$ (lower row) and concomitant average gain in translational energy $\langle \Delta E_{\text{trans}} \rangle$ (upper row) as a function of normal incidence energy E_i for state-to-state scattering using the SRP48-PES. Panels (A,C) [(B,D)] are for the transition from the rovibrational state $\nu = 2, J = 1$ [2] to $\nu = 1, J = 1$ [2] for H₂ [D₂]. Shown are results from adiabatic calculations (filled black squares), as well as those including electronic friction according to the LDFA (red pluses), ODF (blue circles), or ODF-iso (purple crosses) models. The error bars indicate the error due to statistical sampling as described in detail in the Supporting Information.

minimum-energy path depicted in Figure 2A–D. Consequently, when comparing MDEF with other nonadiabatic models^{43–45} for vibrational de-excitation, our results suggest that it is crucial to also take into account whether the friction coefficients include any molecular anisotropy. Unfortunately, because molecular beam experiments for this system have hitherto focused on rovibrational excitation rather than de-excitation,^{46–48} experimental verification of this effect is still pending.

Although ODF-iso inherits the spurious memory effects as well as going beyond the independent atom approximation from ODF, quite remarkably, for $E_i > 15$ (20) kJ/mol for H₂ (D₂), we obtain results with ODF-iso that are almost identical to those with LDFA. That means that (at least in this energy range) these do not affect the dynamics and the molecular anisotropy is the most important difference. For lower incidence energies, scattering over the top site has been found to dominate vibrational de-excitation from $\nu = 1$ in adiabatic calculations.⁴² Indeed, for top sites, $\eta_{ad}^{\text{ODF-iso}}$ is rather different from η_{ad}^{LDFA} so that the difference in electronic structure inherent to LDFA and ODF also becomes visible in the dynamics in this case. If the molecular anisotropy could be experimentally validated, it would greatly encourage future theoretical work to develop extensions to LDFA that might be able to (at least approximately) account for it.

In summary, we have obtained different observables for H₂ and D₂ on Cu(111) from extensive MDEF trajectory calculations for the first time using full-dimensional friction tensors based on both LDFA and ODF. The molecular anisotropy as described by ODF and absent from LDFA leads to strongly enhanced friction for motion along the molecular axis when the molecules are close to the surface. The

dissociative sticking probability is almost negligibly reduced compared to adiabatic simulations. The effect is slightly stronger for ODF compared to LDFA and improves the agreement with experimental data in both cases. For the state-to-state scattering of vibrationally excited molecules (from $\nu = 2, J = 1$ (2) to $\nu = 1, J = 1$ (2) for H₂ (D₂)), we predict up to six (two) times larger vibrational de-excitation probabilities with ODF (LDFA) compared to adiabatic simulations. Remarkably, isotropicalization of ODF yields results almost identical to LDFA for incidence energies larger than 15 (20) kJ/mol for H₂ (D₂). The predicted differences between the vibrational de-excitation probabilities are a “fingerprint” of the molecular anisotropy as described by ODF. Recently suggested techniques to prepare H₂ molecular beams with $1 \leq \nu \leq 4$ ⁴⁹ should allow testing for this fingerprint. This would provide unprecedented insight into the accuracy of state-of-the-art electronic friction models for molecules and allow analysis of the importance of concomitant approximations.

COMPUTATIONAL DETAILS

In eq 2, we calculate the electron–phonon matrix elements $g_{kab}^{ja} = \langle \phi_{ka} | \frac{\partial v_{\text{KS}}}{\partial \mathbf{R}_{ia}} | \phi_{kb} \rangle$ from the change of the Kohn–Sham potential $\frac{\partial v_{\text{KS}}}{\partial \mathbf{R}_{ia}}$ with respect to nuclear coordinate $\mathbf{R}_{i\omega}$ which is obtained from density functional perturbation theory (DFPT)⁵⁰ employing the PW91⁵¹ exchange–correlation functional as implemented in the QUANTUM ESPRESSO package.⁵² Surfaces are modeled by 2×2 Cu(111) slabs with four layers and 10 Å of vacuum. A plane-wave cutoff energy of 816 eV is used, together with ONCV pseudopotentials⁵³ from the SG15⁵⁴ library and an 18×18 k-point grid. These settings reproduce the PW91-PES from ref 34 up to a few meV. They also enable an accurate evaluation of the sum over electronic states in eq 2 at the Fermi level using an equivalent Gaussian envelope technique to broaden the δ -function with a width of 0.6 eV as suggested in ref 26. We note that this implies the possible presence of spurious electronic memory effects, as argued in ref 27.

The neural network fits for the 21 independent elements of $\eta_{iaj\beta}^{\text{ODF}}(\mathbf{R})$ are based on ~ 30000 ODF coefficients obtained from DFT calculations on the same seven lateral sites that have been used to construct the SRP48-PES.³³ For LDFA, we extract the background electron density $\rho(\mathbf{R}_i)$ from a DFT calculation with the same computational setup as described above. Employing the functional form for $\eta^{\text{H}}(\rho)$ suggested in ref 38, we then construct three-dimensional neural network interpolations for $\eta^{\text{H}}(\rho(\mathbf{R}_i))$ based on the symmetry-adapted coordinates^{55,56} in order to obtain a continuous representation of $\eta_{iaj\beta}^{\text{LDFA}}(\mathbf{R})$.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jpcllett.7b03182.

LDFA-AIM for H₂ and D₂, continuous representation of 6×6 friction tensors, errors due to statistical sampling, and vibrational de-excitation for PW91-PES (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: j.meyer@chem.leidenuniv.nl. Phone: +31 (0)71 527 5569.

ORCID 

Paul Spiering: 0000-0001-7692-7273

Jörg Meyer: 0000-0003-0146-730X

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

J.M. gratefully acknowledges financial support from The Netherlands Organisation for Scientific Research (NWO) under Vidi Grant No. 723.014.009. We would like to thank Geert-Jan Kroes and Reinhard Maurer for very insightful and stimulating discussions.

REFERENCES

- (1) Wodtke, A. M.; Tully, J. C.; Auerbach, D. J. Electronically Non-Adiabatic Interactions of Molecules at Metal Surfaces: Can We Trust the Born-Oppenheimer Approximation for Surface Chemistry? *Int. Rev. Phys. Chem.* **2004**, *23*, 513–539.
- (2) Kroes, G.-J. Frontiers in Surface Scattering Simulations. *Science* **2008**, *321*, 794–797.
- (3) Golibrzuch, K.; Bartels, N.; Auerbach, D. J.; Wodtke, A. M. The Dynamics of Molecular Interactions and Chemical Reactions at Metal Surfaces: Testing the Foundations of Theory. *Annu. Rev. Phys. Chem.* **2015**, *66*, 399–425.
- (4) Wodtke, A. M. Electronically Non-Adiabatic Influences in Surface Chemistry and Dynamics. *Chem. Soc. Rev.* **2016**, *45*, 3641–3657.
- (5) Shenvi, N.; Roy, S.; Tully, J. C. Nonadiabatic Dynamics at Metal Surfaces: Independent-Electron Surface Hopping. *J. Chem. Phys.* **2009**, *130*, 174107.
- (6) Meyer, J.; Reuter, K. Electron-Hole Pairs during the Adsorption Dynamics of O₂ on Pd(100): Exciting or Not? *New J. Phys.* **2011**, *13*, 085010.
- (7) Ryabinkin, I. G.; Izmaylov, A. F. Mixed Quantum-Classical Dynamics Using Collective Electronic Variables: A Better Alternative to Electronic Friction Theories. *J. Phys. Chem. Lett.* **2017**, *8*, 440–444.
- (8) Dou, W.; Miao, G.; Subotnik, J. E. Born-Oppenheimer Dynamics, Electronic Friction, and the Inclusion of Electron-Electron Interactions. *Phys. Rev. Lett.* **2017**, *119*, 046001.
- (9) Rittmeyer, S. P.; Meyer, J.; Reuter, K. Nonadiabatic Vibrational Damping of Molecular Adsorbates: Insights into Electronic Friction and the Role of Electronic Coherence. *Phys. Rev. Lett.* **2017**, *119*, 176808.
- (10) Head-Gordon, M.; Tully, J. C. Molecular Dynamics with Electronic Frictions. *J. Chem. Phys.* **1995**, *103*, 10137–10145.
- (11) Alducin, M.; Díez Muiño, R.; Juaristi, J. I. Non-Adiabatic Effects in Elementary Reaction Processes at Metal Surfaces. *Prog. Surf. Sci.* **2017**, *92*, 317–340.
- (12) Juaristi, J. I.; Alducin, M.; Díez Muiño, R.; Busnengo, H. F.; Salin, A. Role of Electron-Hole Pair Excitations in the Dissociative Adsorption of Diatomic Molecules on Metal Surfaces. *Phys. Rev. Lett.* **2008**, *100*, 116102.
- (13) Novko, D.; Blanco-Rey, M.; Alducin, M.; Juaristi, J. I. Surface Electron Density Models for Accurate *Ab Initio* Molecular Dynamics with Electronic Friction. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2016**, *93*, 245435.
- (14) Hellsing, B.; Persson, M. Electronic Damping of Atomic and Molecular Vibrations at Metal Surfaces. *Phys. Scr.* **1984**, *29*, 360–371.
- (15) Janke, S. M.; Auerbach, D. J.; Wodtke, A. M.; Kandratsenka, A. An Accurate Full-Dimensional Potential Energy Surface for H-Au(111): Importance of Nonadiabatic Electronic Excitation in Energy Transfer and Adsorption. *J. Chem. Phys.* **2015**, *143*, 124708.
- (16) Bünermann, O.; Jiang, H.; Dorenkamp, Y.; Kandratsenka, A.; Janke, S. M.; Auerbach, D. J.; Wodtke, A. M. Electron-Hole Pair Excitation Determines the Mechanism of Hydrogen Atom Adsorption. *Science* **2015**, *350*, 1346–1349.
- (17) Askerka, M.; Maurer, R. J.; Batista, V. S.; Tully, J. C. Role of Tensorial Electronic Friction in Energy Transfer at Metal Surfaces. *Phys. Rev. Lett.* **2016**, *116*, 217601.
- (18) Inoue, K.-i.; Watanabe, K.; Sugimoto, T.; Matsumoto, Y.; Yasuike, T. Disentangling Multidimensional Nonequilibrium Dynamics of Adsorbates: CO Desorption from Cu(100). *Phys. Rev. Lett.* **2016**, *117*, 186101.
- (19) Maurer, R. J.; Jiang, B.; Guo, H.; Tully, J. C. Mode Specific Electronic Friction in Dissociative Chemisorption on Metal Surfaces: H₂ on Ag(111). *Phys. Rev. Lett.* **2017**, *118*, 256001.
- (20) Luntz, A. C.; Makkonen, I.; Persson, M.; Holloway, S.; Bird, D. M.; Mizielinski, M. S. Comment on “Role of Electron-Hole Pair Excitations in the Dissociative Adsorption of Diatomic Molecules on Metal Surfaces. *Phys. Rev. Lett.* **2009**, *102*, 109601.
- (21) Juaristi, J. I.; Alducin, M.; Muiño, R. D.; Busnengo, H. F.; Salin, A. Juaristi et al. Reply. *Phys. Rev. Lett.* **2009**, *102*, 109602.
- (22) Luntz, A. C.; Persson, M. How Adiabatic is Activated Adsorption/Associative Desorption? *J. Chem. Phys.* **2005**, *123*, 074704.
- (23) Krishna, V.; Tully, J. C. Vibrational Lifetimes of Molecular Adsorbates on Metal Surfaces. *J. Chem. Phys.* **2006**, *125*, 054706.
- (24) Forsblom, M.; Persson, M. Vibrational Lifetimes of Cyanide and Carbon Monoxide on Noble and Transition Metal Surfaces. *J. Chem. Phys.* **2007**, *127*, 154303.
- (25) Rittmeyer, S. P.; Meyer, J.; Juaristi, J. I.; Reuter, K. Electronic Friction-Based Vibrational Lifetimes of Molecular Adsorbates: Beyond the Independent-Atom Approximation. *Phys. Rev. Lett.* **2015**, *115*, 046102.
- (26) Maurer, R. J.; Askerka, M.; Batista, V. S.; Tully, J. C. *Ab Initio* Tensorial Electronic Friction for Molecules on Metal Surfaces: Nonadiabatic Vibrational Relaxation. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2016**, *94*, 115432.
- (27) Novko, D.; Alducin, M.; Blanco-Rey, M.; Juaristi, J. I. Effects of Electronic Relaxation Processes on Vibrational Linewidths of Adsorbates on Surfaces: The Case of CO/Cu(100). *Phys. Rev. B: Condens. Matter Mater. Phys.* **2016**, *94*, 224306.
- (28) Kroes, G.-J.; Juaristi, J. I.; Alducin, M. Vibrational Excitation of H₂ Scattering from Cu(111): Effects of Surface Temperature and of Allowing Energy Exchange with the Surface. *J. Phys. Chem. C* **2017**, *121*, 13617–13633.
- (29) Diaz, C.; Pijper, E.; Olsen, R. A.; Busnengo, H. F.; Auerbach, D. J.; Kroes, G.-J. Chemically Accurate Simulation of a Prototypical Surface Reaction: H₂ Dissociation on Cu(111). *Science* **2009**, *326*, 832–834.
- (30) Nattino, F.; Genova, A.; Guijt, M.; Muzas, A. S.; Diaz, C.; Auerbach, D. J.; Kroes, G.-J. Dissociation and Recombination of D₂ on Cu(111): *Ab Initio* Molecular Dynamics Calculations and Improved Analysis of Desorption Experiments. *J. Chem. Phys.* **2014**, *141*, 124705.
- (31) Kroes, G.-J.; Diaz, C. Quantum and Classical Dynamics of Reactive Scattering of H₂ from Metal Surfaces. *Chem. Soc. Rev.* **2016**, *45*, 3658–3700.
- (32) Diaz, C.; Olsen, R. A.; Auerbach, D. J.; Kroes, G.-J. Six-Dimensional Dynamics Study of Reactive and Non Reactive Scattering of H₂ from Cu(111) Using a Chemically Accurate Potential Energy Surface. *Phys. Chem. Chem. Phys.* **2010**, *12*, 6499–6519.
- (33) Mondal, A.; Wijzenbroek, M.; Bonfanti, M.; Diaz, C.; Kroes, G.-J. Thermal Lattice Expansion Effect on Reactive Scattering of H₂ from Cu(111) at T_s = 925 K. *J. Phys. Chem. A* **2013**, *117*, 8770–8781.
- (34) Diaz, C.; Olsen, R. A.; Busnengo, H. F.; Kroes, G.-J. Dynamics on Six-Dimensional Potential Energy Surfaces for H₂/Cu(111): Corrugation Reducing Procedure versus Modified Shepard Interpolation Method and PW91 versus RPBE. *J. Phys. Chem. C* **2010**, *114*, 11192–11201.
- (35) Butler, W. H.; Pinski, F. J.; Allen, P. B. Phonon Linewidths and Electron-Phonon Interaction in Nb. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1979**, *19*, 3708–3721.
- (36) Puska, M. J.; Nieminen, R. M. Atoms Embedded in an Electron Gas: Phase Shifts and Cross Sections. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1983**, *27*, 6121.

- (37) Gerrits, N.; Juaristi, J. I.; Meyer, J. Atoms in jellium revisited - Implications for the Local Density Friction Approximation. In preparation.
- (38) Saalfrank, P.; Juaristi, J. I.; Alducin, M.; Blanco-Rey, M.; Muiño, R. D. Vibrational Lifetimes of Hydrogen on Lead Films: An Ab Initio Molecular Dynamics with Electronic Friction (AIMDEF) Study. *J. Chem. Phys.* **2014**, *141*, 234702.
- (39) Askerka, M.; Maurer, R. J.; Batista, V. S.; Tully, J. C. Erratum: Role of Tensorial Electronic Friction in Energy Transfer at Metal Surfaces [Phys. Rev. Lett. 116, 217601 (2016)]. *Phys. Rev. Lett.* **2017**, *119*, 069901.
- (40) Rettner, C. T.; Michelsen, H. A.; Auerbach, D. J. Quantum-State-Specific Dynamics of the Dissociative Adsorption and Associative Desorption of H₂ at a Cu(111) Surface. *J. Chem. Phys.* **1995**, *102*, 4625–4641.
- (41) Michelsen, H. A.; Rettner, C. T.; Auerbach, D. J.; Zare, R. N. Effect of Rotation on the Translational and Vibrational Energy Dependence of the Dissociative Adsorption of D₂ on Cu(111). *J. Chem. Phys.* **1993**, *98*, 8294–8307.
- (42) Muzas, A. S.; Juaristi, J. I.; Alducin, M.; Díez Muiño, R.; Kroes, G.-J.; Diaz, C. Vibrational Deexcitation and Rotational Excitation of H₂ and D₂ Scattered from Cu(111): Adiabatic versus Non-Adiabatic Dynamics. *J. Chem. Phys.* **2012**, *137*, 064707.
- (43) Luntz, A. C.; Persson, M.; Sitz, G. O. Theoretical Evidence for Nonadiabatic Vibrational Deexcitation in H₂(D₂) State-to-State Scattering from Cu(100). *J. Chem. Phys.* **2006**, *124*, 091101.
- (44) Golibrzuch, K.; Shirhatti, P. R.; Rahinov, I.; Kandratsenka, A.; Auerbach, D. J.; Wodtke, A. M.; Bartels, C. The Importance of Accurate Adiabatic Interaction Potentials for the Correct Description of Electronically Nonadiabatic Vibrational Energy Transfer: A combined Experimental and Theoretical Study of NO($v = 3$) Collisions with a Au(111) Surface. *J. Chem. Phys.* **2014**, *140*, 044701.
- (45) Krüger, B. C.; Bartels, N.; Bartels, C.; Kandratsenka, A.; Tully, J. C.; Wodtke, A. M.; Schäfer, T. NO Vibrational Energy Transfer on a Metal Surface: Still a Challenge to First-Principles Theory. *J. Phys. Chem. C* **2015**, *119*, 3268–3272.
- (46) Hodgson, A.; Moryl, J.; Traversaro, P.; Zhao, H. Energy transfer and vibrational effects in the dissociation and scattering of D₂ from Cu(111). *Nature* **1992**, *356*, 501.
- (47) Rettner, C. T.; Auerbach, D. J.; Michelsen, H. A. Observation of Direct Vibrational Excitation in Collisions of H₂ and D₂ with a Cu(111) Surface. *Phys. Rev. Lett.* **1992**, *68*, 2547–2550.
- (48) Hodgson, A.; Samson, P.; Wight, A.; Cottrell, C. Rotational Excitation and Vibrational Relaxation of H₂ ($v = 1, J = 0$) Scattered from Cu(111). *Phys. Rev. Lett.* **1997**, *78*, 963.
- (49) Perreault, W. E.; Mukherjee, N.; Zare, R. N. Preparation of a Selected High Vibrational Energy Level of Isolated Molecules. *J. Chem. Phys.* **2016**, *145*, 154203.
- (50) Baroni, S.; de Gironcoli, S.; Dal Corso, A.; Giannozzi, P. Phonons and Related Crystal Properties from Density-Functional Perturbation Theory. *Rev. Mod. Phys.* **2001**, *73*, 515–562.
- (51) Perdew, J. P.; Chevary, J. A.; Vosko, S. H.; Jackson, K. A.; Pederson, M. R.; Singh, D. J.; Fiolhais, C. Atoms, Molecules, Solids, and Surfaces: Applications of the Generalized Gradient Approximation for Exchange and Correlation. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1992**, *46*, 6671–6687.
- (52) Giannozzi, P.; Baroni, S.; Bonini, N.; Calandra, M.; Car, R.; Cavazzoni, C.; Ceresoli, D.; Chiarotti, G. L.; Cococcioni, M.; Dabo, L.; et al. QUANTUM ESPRESSO: A Modular and Open-Source Software Project for Quantum Simulations of Materials. *J. Phys.: Condens. Matter* **2009**, *21*, 395502.
- (53) Hamann, D. R. Optimized Norm-Conserving Vanderbilt Pseudopotentials. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2013**, *88*, 085117.
- (54) Schlipf, M.; Gygi, F. Optimization Algorithm for the Generation of ONCV Pseudopotentials. *Comput. Phys. Commun.* **2015**, *196*, 36–44.
- (55) Meyer, J. Ab Initio Modeling of Energy Dissipation during Chemical Reactions at Transition Metal Surfaces. Ph.D. thesis, Freie Universität Berlin, Germany, 2012.
- (56) Goikoetxea, I.; Beltrán, J.; Meyer, J.; Juaristi, J. I.; Alducin, M.; Reuter, K. Non-Adiabatic Effects during the Dissociative Adsorption of O₂ at Ag(111)? A First-Principles Divide and Conquer Study. *New J. Phys.* **2012**, *14*, 013050.