1	Microplastics in freshwater and terrestrial environments: evaluating the current understanding to
2	identify the knowledge gaps and future research priorities
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15	Keywords:
16	Plastic pollution, nanoplastics, litter, rivers, soil, hazard.

18 Abstract

19 Plastic debris is an environmentally persistent and complex contaminant of increasing concern. 20 Understanding the sources, abundance and composition of microplastics present in the environment 21 is a huge challenge due to the fact that hundreds of millions of tonnes of plastic material is 22 manufactured for societal use annually, some of which is released to the environment. The majority 23 of microplastics research to date has focussed on the marine environment. Although freshwater and terrestrial environments are recognised as origins and transport pathways of plastics to the oceans, 24 25 there is still a comparative lack of knowledge about these environmental compartments. It is highly 26 likely that microplastics will accumulate within continental environments, especially in areas of high 27 anthropogenic influence such as agricultural or urban areas. This review critically evaluates the current 28 literature on the presence, behaviour and fate of microplastics in freshwater and terrestrial 29 environments and, where appropriate, also draws on relevant studies from other fields including 30 nanotechnology, agriculture and waste management. Furthermore, we evaluate the relevant 31 biological and chemical information from the substantial body of marine microplastic literature, 32 determining the applicability and comparability of this data to freshwater and terrestrial systems. 33 With the evidence presented, the authors have set out the current state of the knowledge, and 34 identified the key gaps. These include the volume and composition of microplastics entering the 35 environment, behaviour and fate of microplastics under a variety of environmental conditions and 36 how characteristics of microplastics influence their toxicity. Given the technical challenges 37 surrounding microplastics research, it is especially important that future studies develop standardised techniques to allow for comparability of data. The identification of these research needs will help 38 39 inform the design of future studies, to determine both the extent and potential ecological impacts of 40 microplastic pollution in freshwater and terrestrial environments.

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44 Introduction

45 Research on microplastics as an environmental contaminant is rapidly advancing. Although marine microplastics research remains at the forefront, in recent years researchers recognising the 46 47 comparative lack of studies on microplastics in freshwater environments have begun to address this 48 field as a matter of priority, quantifying microplastics in lake and river systems and assessing exposure 49 to, and uptake by, organisms (Dris et al., 2015b; Wagner et al., 2014). Despite the knowledge that 50 microplastics (and indeed plastics of all sizes) are also widespread within terrestrial environments as 51 a result of human activities, there is a dearth of studies that have quantified microplastics in terrestrial 52 environments. In fact, much of the existing information about the environmental presence of 53 microplastics considers terrestrial and freshwater environments only as sources and transport 54 pathways of microplastics to the oceans. However, given that the majority of all plastics will be used 55 and disposed of on land, both terrestrial and adjacent freshwater environments will themselves be 56 subject to extensive pollution by plastics of all sizes, based on large amounts of anthropogenic litter 57 from both point (e.g. wastewater treatment discharge, sewage sludge application) and diffuse (e.g. 58 general littering) sources. As such it is highly likely that soils will act as long term sinks for microplastic 59 debris (Rillig, 2012; Zubris and Richards, 2005). Hence it is important to understand release rates, fate 60 and transport of microplastics entering terrestrial systems as well as freshwater systems in order to 61 allow for the assessment of hazards and risks posed by microplastics, and indeed plastics in general, 62 to ecosystems.

The aim of this review is to synthesise available information relevant to understanding microplastics behaviour, fate and ecological effects within freshwater environments and soils. The review draws primarily on the published literature available from freshwater and the relatively few terrestrial microplastic studies published to date, setting out the key factors that will influence microplastic distribution, fate and exposure. One important consideration is that the processes governing distribution and exposure to plastics are not necessarily exclusive to a specific environmental 'compartment' (e.g. plastics within a shallow freshwater system may be exposed to

similar levels of UV radiation as a particle in coastal marine systems) and plastics can be transported between compartments (e.g. from land to rivers and the sea, and from rivers and sea to land during flooding, storm events or tidal surges). Therefore it is not realistic to consider such studies in isolation from the body of marine work. Thus, where appropriate, we also include key studies from the extensive body of marine literature that will inform knowledge of the processes likely to occur in freshwaters and soils.

76 Microplastics as a term has quite a broad definition and can refer to a wide range of polymers, 77 particle sizes and densities (see section 2). In this review we will predominantly focus on microplastics 78 defined as being any polymer within the size range 1 µm to 5 mm as this is the size range which has 79 been the major focus of reported microplastics research to date. Where information is available, we 80 have in places included relevant information from reported studies for nanoplastics (< 100nm) as 81 contaminants that are also likely to occur in soils and water. For the purposes of this review, 82 microplastics and nanoplastics have been defined as per the study in which they were used/discussed 83 and parallels drawn between the two where appropriate. However, we do not intend to carry out a 84 complete review of nanoplastics or compare them with other nanomaterials as this topic has been 85 has been previously addressed (Syberg et al., 2015). Finally in places throughout the text, we also use 86 the term "plastics" to refer to plastics as a whole class (macro-, micro- and nano-sized plastics). This is 87 in order to capture the relevant influence of processes such as wind or water flow, exposure to UV, 88 temperature fluctuations and associations with organic matter that can, alone or together, commonly 89 affect the fate and behaviour with different sized plastic materials. The reality is that there are likely to be significant similarities between the effects and behaviours of plastics of different size 90 91 classifications, for example when comparing 'large nanoplastics' to 'small microplastics'. As the size 92 and state of plastics within the environment can change with time, we believe it is necessary to include 93 information that extends beyond plastics in the micron size range to fully understand the drivers of 94 microplastic and indeed all plastic transport, fate and resulting bioavailability.

95 Available information on plastic usage and presence on land is used in order to make informed 96 estimations about the likely presence and effects of microplastics within terrestrial environments. 97 This includes considering relevant data on plastic sources and transport through different 98 environmental compartments, and therefore the organisms that may encounter and be affected by 99 these plastics. We evaluate the available literature on ecological effects of microplastics to freshwater 100 species (using both studies with freshwater species and any studies in comparable marine species) 101 that can be directly related to organisms occupying the same ecological niche within aquatic and 102 terrestrial environments. Finally, we review chemical associations and plasticiser leaching, including 103 examples from microplastics and also large plastic products ('macroplastics') that may have 104 implications for the toxicity of microplastics within freshwater and terrestrial environments. If we are 105 to fully understand or predict the effects of microplastic pollution within the environment as a whole, 106 a multidisciplinary approach will be needed to integrate knowledge on presence and behaviour of 107 plastic waste, particles and associated chemical pollution in the environment. Our review sets out to 108 reflect this by drawing together knowledge from all relevant fields including waste management, 109 nanotechnology, agriculture and toxicology. By using all available knowledge we are able to establish 110 how previous studies can inform our knowledge of presence and effects of microplastics in terrestrial 111 and freshwater environments and, thus, make recommendations for further research.

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113 2. Plastic as an environmental contaminant

114 **2.1.** Plastic pollution in the environment

In 2014, annual plastic production exceeded 311 million tonnes, an increase of nearly 84 million tonnes since 2004 (PlasticsEurope, 2015; Thompson et al., 2005). By 2050 it is estimated that this may increase to a colossal 33 billion tonnes (Rochman et al., 2013a). Of anthropogenic waste materials released to the environment, plastic can constitute up to 54% by mass (Hoellein et al., 2014). Established widespread uses of plastic include packaging materials (39.5% total plastic production),

120 building materials (20.1%), automotive components (8.6%), electronic appliances (5.7%) and 121 agricultural materials (3.4%), with the remainder including products such as household appliances and 122 sporting equipment (PlasticsEurope, 2015). There are approximately 30,000 different polymer 123 materials registered for use in the European Union. A 'polymer' is difficult to characterise as definitions 124 will vary between manufacturers, with much information commercially confidential. However the 125 European Commission report states that 84% of this 30,000 are represented by thermoplastics (Postle 126 et al., 2012). Although they share similar characteristics, each polymer has different physical 127 properties with respect to their plasticity and density. The density of the material in particular will be 128 important for determining environmental fate. For example, density will influence how particles 129 partition in the aquatic environment including whether they float on water surfaces or settle to 130 sediment and the ease with which they will be transported by wind action across land (Zylstra, 2013). 131 However, even when properties are known, it can be difficult to predict the fate of polymers. For 132 example, it has been observed that supposedly buoyant particles such as polyethylene and 133 polypropylene can be retained within sediments (Horton et al., 2016). This could be due to biofouling 134 or agglomeration with organic materials. These differences highlight polymers to be complex 135 environmental pollutants.

136 For many plastic products their useful lifetimes are often relatively short. This is especially the 137 case for single-use packaging materials. However, the qualities which make plastic a good material for 138 consumer products: waterproof, durable and resistant to wear and biodegradation, can also make 139 plastic extremely persistent (Barnes et al., 2009; Imhof et al., 2012). Many commonly-used polymers 140 are extremely resistant to biodegradation, for example polyethylene and polystyrene (Gautam et al., 141 2007). Common characteristics of plastics that can impede biodegradation are high molecular weight, 142 hydrophobicity and cross-linked chemical structure (Gautam et al., 2007; Shah et al., 2008). There is evidence that biodegradation of polymers by some organisms can occur, for example bacteria, fungi 143 144 and mealworms (due to gut bacteria) (Gu, 2003; Yang et al., 2015a, b). However, when biodegradation 145 does occur, it is reliant on exposure of polymers to these and other specific degrading organisms that 146 have the ability to degrade these specific polymers - conditions that may not necessarily be 147 encountered in the environment. Indeed it has been proposed that no polymers can be efficiently 148 biodegraded in landfill sites (Shah et al., 2008). Therefore, apart from incineration, it is understood 149 that the vast majority of plastic ever made is still present in the environment in some form (Barnes et 150 al., 2009; Thompson et al., 2005). It is this persistence that makes plastic pervasive as an 151 environmental pollutant and is a main driver underpinning current concerns about the possible 152 ecological impacts of the growing burden of plastic materials present in ecosystems. Plastic litter is present in terrestrial, freshwater, estuarine, coastal and marine environments, particularly in 153 154 urbanised regions (Cole et al., 2011; Free et al., 2014; Zylstra, 2013). Plastics have been observed even 155 in remote areas of the world including deep-sea sediments (Van Cauwenberghe et al., 2013; Woodall 156 et al., 2014), submarine canyons (Pham et al., 2014) and encapsulated in Arctic sea ice (Obbard et al., 157 2014), far from any potential land-based source. It has even been observed in some locations that 158 plastic debris can fuse together, becoming associated with volcanic rocks, sediment and organic 159 materials forming 'plastiglomerates', solid rock-like substances, that have the potential to become 160 preserved in the fossil record. As human influence begins to dominate even the most fundamental 161 processes on earth, the potential for this evidence of human impact to last far into geological records 162 has prompted the suggestion that we are moving into a new geological epoch from the Holocene to 163 the 'Anthropocene' (Corcoran et al., 2014).

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166 **2.2.** Microplastics: a brief background

Plastic debris is broadly classified by size: mega-debris (> 100 mm), macro-debris (> 20 mm), meso-debris (20-5 mm) and micro-debris (< 5 mm) (Barnes et al., 2009). Although microscale plastic particles were first observed in the marine environment in the early 1970s (Buchanan, 1971; Carpenter and Smith, 1972), it was not until 2004 that the term "microplastic" became commonly used as the result of a study by Thompson et al. (2004). Microplastics are now commonly defined as particles with the largest dimension smaller than 5 mm, although no lower size limit has been specifically defined (Arthur and Baker, 2009; Duis and Coors, 2016; Faure et al., 2012). It is understood that plastic particles in the environment will continue to degrade and become steadily smaller, eventually forming 'nanoplastics' (Koelmans et al., 2015; Mattsson et al., 2015). Microplastics in environmental samples can currently be detected down to a size of 1 μ m, however few environmental studies identify particles <50 μ m due to methodological limitations (Hidalgo-Ruz et al., 2012; Imhof et al., 2016).

178 Microplastics fall within two categories: primary and secondary. Primary microplastics are 179 specifically manufactured in the micrometre size range, for example those used in industrial abrasives 180 for sandblasting, either acrylic or polyester beads (von Moos et al., 2012; Zitko and Hanlon, 1991), 181 plastic pre-production pellets ('nurdles') or in personal care products such as exfoliating agents in 182 creams and cleansers containing polyethylene 'microbeads' (Napper et al., 2015). Primary microplastic particles are likely to be washed down industrial or domestic drainage systems and into wastewater 183 184 treatment streams (Fendall and Sewell, 2009; Lechner and Ramler, 2015). Despite the capability of 185 some sewage treatment works to remove up to 99.9% microplastic particles from wastewater 186 (dependent on the processes employed by the treatment plant), the sheer number of particles 187 entering the system may still allow a significant number to bypass filtration systems and be released 188 into the freshwater environment with effluent (Carr et al., 2016; Murphy et al., 2016).

189 Secondary microplastics are formed as a result of meso and macroplastic litter fragmentation. 190 Plastics are susceptible to the effects of UV radiation and high temperatures which can cause chemical 191 changes making plastics brittle and thus more susceptible to fragmentation (Andrady, 2011; Barnes et 192 al., 2009; Hidalgo-Ruz et al., 2012; Ivar do Sul and Costa, 2014; Rillig, 2012; Shah et al., 2008). 193 Fragmentation increases surface area and number of particles per unit of mass. Both exposure to 194 sunlight and wave action are primary causes of fragmentation in marine waters. On land, especially at 195 the soil surface, fragmentation of plastics is thought to occur readily as a result of direct exposure to 196 UV radiation from sunlight, aided also by temperature fluctuations which will generally be greater

197 than those in sea water (Andrady, 2011). Similarly, exposure to UV may be higher in small shallow 198 aquatic systems such as ponds and rivers than in large lakes or the open ocean. However many 199 freshwater environments may lack the fragmentation potential that is offered by turbulence and wave 200 action in coastal waters, especially in rocky tidal areas (Barnes et al., 2009). An additional source of 201 secondary microplastics is derived from synthetic fabrics, which can shed up to 1900 fibres per 202 garment during washing (Browne et al., 2011). Although microfibres are secondary particles they will 203 be released to the environment along with primary microplastics through wastewater effluents and 204 sludge application. Hence in this respect the fate and transport of these fibres may be more closely 205 aligned with that of primary microplastics, based on similar release routes.

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3. Sources, environmental presence and transport of microplastics

208 **3.1. Sources of microplastics to freshwater and terrestrial environments**

209 A significant direct input of primary microplastics to terrestrial environments has been 210 identified as being through the application of sewage sludge containing synthetic fibres or sedimented 211 microplastics from personal care or household products to land (Habib et al., 1996; Zubris and 212 Richards, 2005). Polymers used in synthetic textiles include polyester and nylon, while polyethylene 213 or polypropylene are commonly used as microbeads or glitter in cosmetics. As sewage treatment 214 works are efficient in removing the majority of microplastic particles from wastewater, many of the 215 particles that are removed will be retained within the sludge (Magnusson and Norén, 2014; Mintenig 216 et al., 2017). This suggests that the major routes of release for secondary microfibres and primary 217 microplastics are the same. In Europe it is common practice to compost and pasteurise sewage sludge 218 for use as agricultural fertiliser as well as dispose of large quantities of sludge produced by wastewater 219 treatment to land (DEFRA, 2012). Between four and five million tons dry weight of sewage sludge are 220 applied to arable land every year in the European Union (Cieślik et al., 2015; Willén et al., 2016), 221 although application rates are highly variable between countries (Nizzetto et al., 2016b). Despite

222 regulations on harmful substances within sludge applied to land, microplastics are not yet considered 223 by these and thus the mass of microplastics inadvertently applied to land annually may exceed 224 400,000 tonnes – higher than the mass currently estimated to be present in oceanic surface waters 225 worldwide (Nizzetto et al., 2016b). Zubris & Richards (2005) found that soils with a known history of 226 sewage sludge application contained significantly higher concentrations of synthetic microfibres than 227 soils which had not received sewage sludge. In some field sites, synthetic microfibres were found 15 228 years after the last sludge application (Zubris and Richards, 2005). This suggests that microplastics and 229 synthetic fibres are likely to accumulate in soils after repeated sludge applications.

230 Those particles that are not retained within the sewage sludge, or removed by skimming 231 during the treatment process, will enter the environment via effluent input to rivers. For primary 232 microplastics and secondary microfibres, effluent from sewage treatment is thought to be a major 233 source of microplastics to freshwater bodies. Synthetic microfibres have been identified by many studies as the most abundant microplastic particle type found throughout freshwater, terrestrial and 234 235 marine environments (Browne et al., 2011; Dubaish and Liebezeit, 2013; Free et al., 2014; Zubris and 236 Richards, 2005), with primary microbeads from personal care products also likely to be a significant 237 contributor to microplastic pollution (Castañeda et al., 2014; Murphy et al., 2016; Napper et al., 2015). 238 However, it must be noted that the sampling equipment and methodology will influence the size of 239 particles observed, and therefore may determine the dominant particle type observed. For example, 240 because fibres have at least one very small dimension, they may not always be retained on a mesh 241 even if the length of the fibre exceeds the mesh size. This variation in sampling methodology could 242 lead to fragments or pellets being erroneously identified as the most abundant particle type and may 243 make comparison of particle types and abundances between studies difficult (Dris et al., 2015b; Ivleva et al., 2016). 244

Due to the small size of primary microplastics they are unlikely to be removed by existing screening of debris, with coarse screens retaining particles >10 mm and even the finest screens retaining particles >1.5 mm (Fendall and Sewell, 2009). An important predictor of microplastic

248 partitioning in sewage treatment will be particle density, with dense particles settling to sludge and 249 buoyant particles floating in effluents (Fig. 1). The extent to which this occurs will also depend on a 250 number of relevant processes that may affect the characteristics of the microplastics. For example, 251 the aggregation of microplastic particles, either with themselves or more likely with other (organic) 252 particulate materials can increase size and density leading to an increase in sedimentation rate (Long 253 et al., 2015). The growth of bacterial biofilms on microplastic surface may again increase particle 254 weight and density, resulting in settling (Cozar et al., 2014; Kowalski et al., 2016; Moret-Ferguson et 255 al., 2010).

256 Figure 1 shows a schematic diagram of waste water treatment processes and how particle 257 partitioning is likely to occur through processing. Removal of coarse debris with physical screens, 258 primary settling lagoons and aerobic oxidation are common across many treatment plants, additional 259 settling lagoons and tertiary treatments may also be present. Plastic materials will generally not be 260 degraded at any point throughout the process and as a consequence, any plastic not removed for 261 disposal during the initial filtering steps will remain in the solids or the effluent after processing. Many 262 microplastics from sewage treatment works will therefore ultimately be directly released to the 263 environment in effluents or through sludge application to land. Other methods of sludge disposal 264 include landfilling, incineration and even in production of cement for use in construction. In these 265 cases, plastic particles are likely to be well-contained and so unlikely to leach into the surrounding 266 environment (Browne et al., 2011; Cieślik et al., 2015; Dubaish and Liebezeit, 2013; Rillig, 2012; Zubris 267 and Richards, 2005).

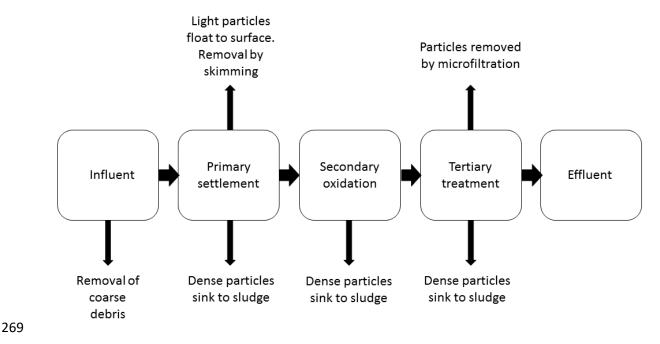


Figure 1. Schematic diagram of standard wastewater treatment processes and particle behaviour influenced by
density at each stage of treatment. Adapted from Baird and Cann (2012).

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273 A recent study observed microbeads originating from cosmetic products in wastewater 274 treatment influents and effluents at seven wastewater reclamation plants in California, in which waste waters were treated for reuse with tertiary treatment. The treatment processes at these plants 275 276 resulted in the complete removal of microparticles (45–400 μ m) from water outputs, as a result of 277 tertiary treatment including surface skimming, sludge settling and microfiltration processes (Carr et 278 al., 2016). After secondary treatment only (elimination microfiltration), effluents contained on 279 average one plastic particle per 1140 litres of effluent, compared to an estimated one particle per litre 280 in the influent (Carr et al., 2016). No fibres were found despite these being the most frequently 281 reported kind of microplastics found in environmental samples, however as previously highlighted, this may be a result of the sampling technique used. Murphy et al. (2016) similarly found that 282 microplastics were significantly reduced in effluent following a secondary treatment process. In this 283 284 study, plastic flakes and fibres were the two most abundant microplastic types (67.3% and 18.5%

285 respectively), with microbeads only contributing to 3% of total particles. For this mixture of materials, 286 average microplastic concentrations reduced from 15.7 particles litre⁻¹ (\pm 5.23) in sewage treatment influents to 0.25 particles litre⁻¹ (± 0.04) in final effluents, which represents a 98% reduction in 287 288 microplastic concentrations (Murphy et al., 2016). Other recent studies have reported similar high 289 removal rates: 95% (Talvitie et al., 2017), 97% (Mintenig et al., 2017) and 99% (Magnusson and Norén, 290 2014). Notably, these proportions of partitioning between solid waste and effluent are similar to 291 estimates that have been provided for nanomaterials: 90% removal of titanium (Ti) associated with 292 titanium dioxide (TiO₂) nanoparticles (Johnson et al., 2011), 96% removal of Ti (Westerhoff et al., 293 2011), 94% removal of surfactant-coated silicon dioxide (SiO_2) nanoparticles (Jarvie et al., 2009). This 294 suggests that similar processes may affect the fate of microplastics as they do poorly soluble and 295 potentially inert nanomaterials such as gold and titanium dioxide during waste water treatment (e.g. 296 heteroaggregation), and highlights the importance of interdisciplinary research for understanding the 297 fates and behaviours of microplastics and nanoparticles and the parallels that can be drawn between 298 them (Bouwmeester et al., 2015; Syberg et al., 2015). Despite the significant removal of particles from 299 treated wastewater, given the large volumes passing through wastewater treatment plants the 300 remaining 5%, or less, of the microplastics that are not filtered out will likely represent a large number 301 and mass entering the freshwater environment in effluent (Murphy et al., 2016; Ziajahromi et al., 302 2016). It is also important to note that these results are based on efficient current-generation 303 wastewater treatment processes that may not be widely available or utilised worldwide. In many 304 countries, untreated sewage is input directly to watercourses without treatment (Duis and Coors, 305 2016; Hammer et al., 2012). Where the most modern facilities are not available, these estimates could 306 fall short by up to 100-fold in places.

307 Sources of secondary microplastics derived from plastic litter are both numerous and diverse, 308 ranging from releases during municipal solid waste collection, processing and land-filling, release from 309 transportation and disposal systems to individuals creating litter either accidentally or intentionally 310 (Fig. 2). This includes large plastic items and sanitary waste input to rivers via combined sewage 311 overflows (CSOs). Runoff via drainage ditches from agricultural land, or storm drains from roads 312 containing plastics such as tyre wear particles, vehicle-derived debris or fragments of road-marking 313 paints is another significant source of riverine microplastic loads (Browne et al., 2010; Eriksen et al., 314 2013; Galgani et al., 2015; Horton et al., 2016; Tibbetts, 2015). Additionally, wind action may also 315 transport lighter plastic items into water bodies or across land (Zylstra, 2013) and there is evidence to 316 suggest that anthropogenic fibres can be transported and deposited by atmospheric fallout. This 317 appears to be especially significant in urban areas, with deposition increasing during periods of rain 318 (Dris et al., 2016). Although the fibres found in atmospheric studies were not exclusively synthetic 319 (<33% fibres were pure polymers), with an estimated deposition of between 3-10 tonnes of fibres 320 deposited annually in an area approximately 2500 km² (based on the Paris region), this may therefore 321 still represent a significant pathway of microplastics from consumer products to the environment (Dris 322 et al., 2017; Dris et al., 2016). Airborne particles are determined to originate from a variety of sources 323 including construction materials, artificial turf and household dust (Magnusson et al., 2016).

324 Another direct source of secondary microplastics to land is the use and fragmentation of 325 agricultural plastics. For example, plastic mulches and polytunnels are used to control temperature 326 and moisture, and retard weed growth in agricultural and horticultural applications (Kasirajan and 327 Ngouajio, 2012; Kyrikou and Briassoulis, 2007; Rillig, 2012; Steinmetz et al., 2016). Polymer seed 328 coatings can also be used to control germination (Clayton et al., 2004). These may consist of various 329 polymers and often contain incorporated pesticides and fertilisers. Commonly used polymers for seed 330 coatings are non-biodegradable and therefore following germination, will remain in the soil (Schultz 331 et al., 2014; Turnblad and Chen, 1998). Additional products used in agriculture include bale twines and 332 wraps, containers, packaging and netting, all of which have the potential for dispersal within the 333 environment (Scarascia-Mugnozza et al., 2012). Exposure of these materials to sunlight and high 334 temperatures may lead to their relatively rapid fragmentation after which they are difficult to 335 completely remove from soils. Dense polymers are more likely to remain in soil and ultimately to be 336 transported into deeper soil layers, whereas lighter polymers will be more likely to be transported by

wind and water action either to other terrestrial locations or to surface waters. To our knowledge, to date there are no studies which quantify microplastic presence at terrestrial field sites. Based on the above evidence, however, it is highly likely that microplastics will be present within terrestrial environments and, if investigated in detail, may be found to be as equally pervasive as they are in freshwater and marine environments (Nizzetto et al., 2016a).

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3 3.2. Presence of microplastics in the freshwater environment

Studies of microplastics in freshwater environments are rapidly advancing, with microplastic particles found across a range of freshwater environments worldwide, including lakes and rivers. Area of water surface, depth, wind, currents and density of particles are all factors determining transport and fate of particles within these aquatic systems (Eriksen et al., 2014; Eriksen et al., 2013; Fischer et al., 2016; Free et al., 2014). Given the lack of terrestrial studies to date, it is necessary to use our knowledge of microplastics in the freshwater environment, notably sediments, to infer the presence and behaviour of microplastics in soils and to inform future sampling efforts.

351 A study carried out on lake beaches by Imhof et al (2013) measured microplastics found in 352 sediments of two beaches on the north and south shores of Lake Garda (Italy). Particle numbers 353 between these sites were significantly different, with these differences attributed to the prevailing 354 southerly wind direction transporting plastics either directly or by surface water movement to the 355 opposite shore (Imhof et al., 2013). The number of local sources, together with factors including water 356 surface area, depth, wind, currents and density of particles are all factors determining transport and 357 fate of particles within these aquatic systems and can lead to large variation, even within a relatively 358 small area (Castañeda et al., 2014; Eriksen et al., 2014; Eriksen et al., 2013; Fischer et al., 2016; Free 359 et al., 2014). Another significant factor influencing particle presence and abundance is urbanisation of 360 the area surrounding and influencing the waterbody. Eriksen et al. (2013) conducted a study in the 361 Great Lakes (USA) and found that downstream of highly populated Detroit and Cleveland metropolitan

362 areas, particle concentrations ranged from 280,947-466,305 particles km⁻². In Lake Huron, where the 363 shorelines are less influenced by the presence of major urban centres, particle concentrations 364 estimated from sampling were generally orders of magnitude lower, ranging from 456-6541 particles km⁻², with one trawl finding no particles (Eriksen et al., 2013). A similar study of the remote lake 365 366 Hovsgol (Mongolia) also found microplastics present in all samples at concentrations comparable to 367 those found in the Great Lakes (Table 1). Although the area surrounding Lake Hovsgol has a low 368 population density, poor local waste management and inputs of wastewater are blamed for the 369 presence of microplastic particles in the lake (Free et al., 2014). Additionally, the smaller volume of 370 Lake Hovsgol, compared to the Great Lakes of the USA, may be an important reason for microplastic 371 concentrations being comparable between these two studies.

372 Urbanisation has also been observed to be a significant factor influencing presence of 373 microplastics in riverine environments, with plastics being introduced from a variety of sources 374 including effluent, road runoff, littering and atmospheric deposition (discussed further in Section 3.1). 375 Mani et al. (2015) and Yonkos et al. (2014) are among those who have found microplastics in higher 376 abundances at sites in close proximity to urban areas than at more remote sites. However, although 377 particle numbers are regularly found to be high near urban areas, this is not the only factor influencing presence of microplastic particles. For example, Horton et al. (2016), in addition to finding high 378 379 numbers of particles downstream of urban discharge points, also found particles in rural areas where 380 few human-associated inputs would be expected.

Given the growing need to make comparative assessments in order to identify regional, national and global trends in microplastic distribution, it would be desirable to be able to collate the available data to conduct meta-analyses. However, a major challenge to this is that no standard protocol for collecting particles from environmental samples exists, with different authors using different approaches. While many studies use broadly similar techniques to extract microplastics from environmental samples, including size fractionation, digestion of organic matter and density

387 separation, the specific parameters of methods differ between studies regarding volume of sample 388 studied, upper and lower particle size limits, density separation media and particle identification 389 criteria (Besley et al., 2016; Hidalgo-Ruz et al., 2012). Given that many methods currently rely on visual 390 identification, there are also many opportunities for the introduction of sampling error, bias or 391 omission of particles of certain size or density, leading many results to be qualitative rather than 392 quantitative (Ivleva et al., 2016). Although many studies have established 'standard methods' for 393 particle extraction in an effort to introduce consistency across studies, these methods are in fact quite 394 disparate. Moreover, studies are still identifying new and reportedly more effective criteria. Thus no 395 standardised methods have yet been agreed (Hidalgo-Ruz et al., 2012; Syberg et al., 2015). An 396 additional issue is the use of non-standard units of measurement for reporting microplastic 397 concentrations. In order to compare studies where units are not consistent, units must be transformed 398 to units per volume, either as particles per litre of sampled water or as particles per kilogram of 399 sediment (see Table 1). It is therefore of utmost importance that authors detail results in all units, or 400 provide sufficient detail on the sampling methodology to do so (Phuong et al., 2016; Van 401 Cauwenberghe et al., 2015). These differences between studies highlight the need for continued 402 efforts to standardise methods for microplastic extraction and quantification, as has been recognised 403 in environmental nanomaterial research (Delay et al., 2010).

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405 Table 1. Summary of selected freshwater microplastic environmental sampling studies, covering a range of 406 freshwater environments (water, plus benthic and shore sediments of lakes and rivers). Selected studies were 407 those which quantified specifically microplastics and provided sufficient methodological detail to allow for 408 conversion of units, to standardise by volume or mass for comparability.. Converted units for water and 409 sediment were calculated by multiplying area sampled by sampling depth to estimate total volume, then 410 converting this volume into litres or kg (dry weight). For sediment this calculation is based on typical dry 411 sediment bulk density of 1.3 g cm⁻³ (Sekellick et al., 2013) Conversion was not required where the study already 412 reports results as particles L⁻¹ or kg⁻¹. For details of additional freshwater studies, refer to (Dris et al., 2015b).

Water body type	Sample type	Sample location and description	Study findings (reported units)	Study findings (converted units)	Study
Lake	Water	Great Lakes (USA) 16 cm sampling depth	Average particle concentration 43,000 km ⁻²	Average 0.00027 particles L ⁻¹	Eriksen et al. (2013)
Lake	Water	Lake Hovsgol (Mongolia), sampling depth 16 cm	Average particle concentration 20,264 km ⁻²	Average 0.00012 particles L ⁻¹	Free et al. (2014)
Lake	Benthic sediment	Lake Ontario (Canada) sampling depth 8 cm	26 particles in 42.2 g (station 403) 9 particles in 103.2 g (station 208)	616.1 particles kg ⁻¹ (station 403) 87 particles kg ⁻¹ (station 208)	Corcoran et al. (2015)
Lake	Shore sediment	Lake Garda (Italy), sampling depth 5 cm	Average particle abundance 1108 and 108 m ⁻² (north and south shores respectively)	Average 17 particles kg ⁻¹ (north) 1.7 particles kg ⁻¹ (south)	Imhof et al. (2013)
Lake	Shore sediment	Lake Garda (Italy), sampling depth 5 cm	Average particle abundance 75 m ⁻²	Average 1.2 particles kg ⁻¹	lmhof et al. (2016)
Lake	Shore sediment	Various lakes (Switzerland), sampling depth 5 cm	Average particle abundance 1300 m ⁻²	Average 20 particles kg ⁻¹	Faure et al. (2015)
Lake	Water and shore sediment	Lake Chiusi and Lake Bolsena (Italy)	Average particle abundance 234 kg ⁻¹ , 3.02 m ⁻³ surface water (Chiusi) Average particle abundance 112 kg ⁻¹ , 2.51 m ⁻³ surface water (Bolsena)	Average 0.03 particles L ⁻¹ surface water (Chiusi) Average 0.025 particles L ⁻¹ surface water (Bolsena)	Fischer et al. (2016)
Lake	Water and benthic sediment	Taihu Lake (China)	Particle abundance range: 3.4 – 25.8 L ⁻¹ surface water 11 – 234.6 kg ⁻¹ benthic sediment	-	Su et al. (2016)
Lake	Benthic and shore sediments	Lake Ontario (Canada)	Average particle abundance 980 kg ⁻¹ lake benthic 140 kg ⁻¹ lake beach	-	Ballent et al. (2016)
River	Water	Great Lakes tributaries (USA)	Particle abundance range: 0.05 – 32 m ⁻³	0.00005 – 0.032 particles L ⁻¹	Baldwin et al. (2016)
River	Water	River Seine, urban area (Paris, France)	Average particle abundance 30 m ⁻³ (plankton trawl) Average particle abundance 0.35 m ⁻³ (manta trawl)	Average 0.03 particles L ⁻¹ Average 0.00035 particles L ⁻¹	Dris et al. (2015a)
River	Water	Various rivers (Switzerland)	Average particle abundance 7 m ⁻³	Average particles 0.007 L ⁻¹	Faure et al. (2015)
River	Water	River Danube (Austria)	Average particle abundance 316.8 m ⁻³	Average 0.32 particles L ⁻¹	Lechner et al. (2014)
River	Water	River Rhine (various) sampling depth 18 cm	Average particle abundance 892,777 km ⁻²	Average particles 0.005 L ⁻¹	Mani et al. (2015)

River	Water	Nine different rivers, Chicago area (USA)	Average particle abundance 2.4 m ⁻³ , upstream sewage treatment works (STW) Average particle abundance 5.7 m ⁻³ , downstream STW	Average particles 0.002 L ⁻¹ Average particles 0.006 L ⁻¹	McCormick et al. (2014)
River	Water	Rivers: Papatsco Corsica Rhode Magothy Sampling depth 15 cm	Average particle abundance 155,374 km ⁻² 40,852 km ⁻² 67,469 km ⁻² 112,590 km ⁻²	Average particles 0.001 L ⁻¹ 0.00027 L ⁻¹ 0.00045 L ⁻¹ 0.00075 L ⁻¹	Yonkos et al. (2014)
River	Shore sediment	Rivers Rhine and Main (Germany)	Particle abundance range: 228 - 3763 kg ⁻¹	-	Klein et al. (2015)
River	Benthic sediment	Lake Ontario tributaries (Canada)	Average particle abundance 610 kg ⁻¹	-	Ballent et al. (2016)
River	Benthic sediment	St Lawrence river sediments, sampling depth 10-15 cm (Canada).	Average particle abundance 13,759 m ⁻²	Average approx. 70.6- 105.8 particles kg ⁻¹ (depending on depth sampled)	Castañeda et al. (2014)
River	Benthic sediment	River Thames Basin (UK), sampling depth approx. 10cm	Average particle abundance range: 185 kg ⁻¹ to 660 kg ⁻¹ depending on site.	-	Horton et al. (2016)
River	Benthic sediment	Beijiang River (China)	Particle abundance range: 178 - 554 particles kg ⁻¹	-	Wang et al. (2016)

413

The numbers of particles reported in marine and freshwater surface waters are extremely 414 415 variable. Concentrations of microplastics in marine surface waters have been reported from 0.0005 416 particles L^{-1} (Carson et al., 2013) (calculated as per Table 1) to 16 particles L^{-1} (Song et al., 2014) with 417 a range of intermediate concentrations reported (Lusher et al., 2014; Zhao et al., 2014). Studies of 418 freshwater surface samples generally show concentrations comparable to the lower end of the 419 reported marine surface concentrations such as those seen by Carson et al. (2013) (see Table 1). Dris 420 et al. (2015a) highlight the consequence of using different mesh sizes when determining the number 421 of particles observed. When sampling with a plankton net (80 µm mesh), up to 100-fold more particles 422 can be collected compared to use of a manta net (330 µm mesh). This effect of mesh size is an 423 important consideration when comparing surface water studies, as differences in sampling method 424 and equipment may lead to inconsistencies that prohibit the comparability of datasets (Cole et al., 425 2011). However, despite this variation, it remains possible that freshwater concentrations comparable 426 to the higher marine concentrations will be found, likely within urban areas.

427 Studies in river sediments consistently report abundances of microplastics in the tens to hundreds of particles kg⁻¹ (Table 1), values that are broadly comparable to those reported in marine 428 429 sediment studies. For example, Dekiff et al. (2014) and Nor and Obbard (2014) reported marine 430 microplastic concentrations in the range from individual particles to tens of particles per kilogram of 431 dry sediment, consistent with a study of the sediments of the St Lawrence River (Castañeda et al., 432 2014). Hundreds of particles per kilogram of dry sediment were reported by Horton et al. (2016) in UK 433 river sediments, values also reflected by Laglbauer et al. (2014) in coastal sediments in Slovenia. At 434 the highest concentrations, thousands of particles kg⁻¹ of dry sediment have been reported in river sediments in Germany (Klein et al., 2015), comparable to the 2000-8000 particles kg⁻¹ reported by 435 436 Mathalon and Hill (2014) in coastal sediments in Canada.

437 Efforts in colloid science and nanotoxicology have shown the value of working towards standard methods for key measurements of colloid and nanomaterial characteristics, such as size, 438 stability and surface properties (Hassellov et al., 2008; Montes-Burgos et al., 2009). Similar efforts 439 440 seem warranted in the microplastic community with respect to environmental sampling and 441 qualification. Currently in the field of microplastics research, there are two widely accepted methods 442 of polymer identification – Fourier transform infra-red (FTIR) spectroscopy and Raman spectroscopy, although both have drawbacks. Alternative identification methods such as differential scanning 443 444 calorimetery (DSC) and thermo-gravimetric analysis (TGA) have been tested but not been widely 445 applied (Dumichen et al., 2015). Of the sampling configurations available for FTIR, there are two that 446 are most common: attenuated total reflectance (ATR) and or transmission (or absorbance). ATR is not 447 effective for analysing very small particles due to the fact that the sample needs to be large enough 448 to cover an 'ATR window' in order for a satisfactory spectrum to be obtained (typically > 1 mm). 449 Additionally, while in transmission mode refractive or scattering artefacts can occur, most notably for 450 particles with irregular surfaces (Harrison et al., 2012). Raman spectroscopy can be overridden by 451 fluorescence from some polymer particles, while other interferences may occur if particles are dirty 452 or contain larger amounts of filler, such as dyes or plasticisers (Löder and Gerdts 2015). These

limitations reduce the possibility of determining probable sources, fate and potential short and longterm environmental impacts of these microplastics as well as advising policy makers on how to regulate microplastic pollutants. It could be that in order to effectively identify environmental polymers, a combined and complementary approach is required, for example using both spectroscopy and thermal analysis (Gigault et al., 2016; Majewsky et al., 2016; Sgier et al., 2016). It will be important to use the experience of working with microplastics in aquatic environments, especially sediments, to inform methods for terrestrial studies.

460

461 **3.3. Transport of microplastics within the environment**

462 Estimating the quantity of plastic litter which is released to the environment is difficult due 463 to a lack of data and international variations between plastic waste generation and disposal. These disparities arise as a result of international differences in societal attitudes, education and 464 465 investment in waste management infrastructure. For example, in China in 2010, 76% of plastic waste 466 (8.82 million metric tonnes) was considered to be mismanaged, compared with 2% (0.28 million 467 metric tonnes) in the United States (Jambeck et al., 2015). Mismanaged waste accounts for plastic 468 released to land by littering and wind-blown debris. The best available estimates for managed and 469 mismanaged plastic waste worldwide are from Jambeck et al. (2015), who modelled how much 470 plastic waste was emitted globally to the oceans from land-based sources during 2010. Our 471 estimates presented in Table 2 focus on Europe and assume that the proportion of waste that is 472 mismanaged in the European Union (EU) is equivalent to that of the United States (2%). This is a 473 reasonable assumption based on similarities in national income and development of waste 474 management infrastructure, evidenced by the application of EU wide policies governing waste management, such as the 1999 EC landfill directive (1999/31/EC) (European Council, 1999). Based on 475 476 this assumption we estimate how much of this mismanaged waste, plus the additional source of

477 microplastics from sewage sludge application, is likely to remain on land annually within Europe

478 (Table 2).

479

480 Table 2. Waste management data and estimates of plastic waste released to terrestrial and freshwater 481 (continental) environments, based on figures for the European Union. Rows highlighted in grey are those directly 482 related to plastic within continental environments. [#]Values for specific waste management practises do not 483 account for mismanaged waste. *Managed and mismanaged waste figures are calculated based on the 484 proportion of waste categorised as managed or mismanaged in the United States: 2% (Jambeck et al., 2015). 485 [¥]Values are calculated based on mismanaged waste to include plastics within sewage sludge, minus plastic that 486 is transported to the oceans. Some sources, such as atmospheric fallout have not been considered due to the 487 limited data available. ¹PlasticsEurope (2015) ²Jambeck et al. (2015) ³Nizzetto et al. (2016b)

Plastic handling/disposal	Plastic million metric tonnes/year
Plastic production (EU total, 2014) ¹	59
Plastic waste (EU total, 2014) ¹	25.8
Managed plastic waste (-2% mismanaged waste)*	25.28
Landfill (EU total) ^{1¤}	8
Recycling (EU total) ^{1^µ}	7.6
Energy recovery (EU total) ^{1^µ}	10.2
Mismanaged plastic waste (2% of plastic waste in the EU)*	0.52
Plastic in sewage sludge (EU total) ³	0.063 - 0.43
Ocean input (EU total) ²	0.04 - 0.11
Total mismanaged plastic waste remaining in continental environments (EU) ¥	0.47 - 0.91

488

Plastic materials used in consumer, domestic and agricultural products in Europe amounted
to 59 million metric tonnes in 2014 (PlasticsEurope, 2015). Mismanaged plastic waste within the EU is
calculated at 520,000 metric tonnes (plastic waste – managed waste). In addition to this, it is estimated
that between 63,000 and 430,000 metric tonnes of microplastics in sewage sludge are deposited on

493 land annually (Nizzetto et al., 2016b). As a result we calculate that in the EU between 473,000 and 494 910,000 metric tonnes of plastic waste is released and retained annually within continental 495 environments, between 4 and 23 times the amount estimated to be released to oceans (Table 2). With 496 the current lack of data on microplastics in soils, it is not possible to distinguish between particles that 497 are retained within terrestrial environments and those retained within freshwater systems. As plastic 498 production and thus environmental deposition increases, this will also result in greater accumulation, 499 and larger amounts being ultimately transferred to the marine environment. However, for a 500 considerable time into the future it remains likely that the amount of plastic deposited and retained 501 within continental environments will exceed that entering the oceans. It is important to note that the 502 study by Jambeck et al. (2015) considers all waste within the US to be well-managed, with the 503 exception of litter (2% of all waste). However, it is possible that some fraction of the waste that is 504 considered to be well-managed could enter the environment during waste processing (e.g. as wind-505 blown debris or mechanical or human error). Therefore it remains plausible that the figures for 506 mismanaged waste may be higher than the stated value. When it is also considered that there may be 507 additional pathways of release that are poorly known, such as atmospheric deposition, then it may be 508 the case that the calculations presented here may be an underestimation of plastic releases.

509 Freshwater and soil systems are subject to both point and diffuse inputs of plastics and so 510 great research effort is warranted to understand transport, exposure and ecological effects of 511 microplastics in these systems. This knowledge will also inform our understanding of rivers and 512 freshwater bodies as transport pathways for plastics from land to oceans (Jambeck et al., 2015; 513 Lechner et al., 2014; Rillig, 2012). It has been estimated that between 70-80% of marine plastics are 514 transported to the sea through the conduits provided by rivers (Bowmer and Kershaw, 2010). 515 Recognising this need, freshwater environments have received more attention than terrestrial 516 environments thus far as they are seen as a direct link between land-based plastic waste and the open 517 oceans, as well as interest in the toxicological impact of microplastics on freshwater ecosystems (see

Table 1). Studies of microplastics in soil ecosystems are, however, notably lacking (Huerta Lwanga etal., 2016; Zubris and Richards, 2005).

520 Figure 2 shows a conceptual diagram of the main flows of microplastics within and between 521 three environmental compartments: terrestrial, freshwater and marine. A key concept of the diagram 522 is partitioning of plastic particles between aquatic and terrestrial environments, highlighting that 523 plastic debris will not only be transported by rivers from land to sea, but that even once in the aquatic 524 environment, may also return to land during high tide or flooding events (Fig. 2). The extent of overall 525 deposition, retention and transport of microplastics will depend on many factors including human 526 behaviours, such as littering or recycling, particle characteristics such as density, shape and size, 527 weather, including wind, rainfall and flooding, and environmental topography and hydrology. This 528 variation can make predicting the spread of litter difficult (Zylstra, 2013). Transport of plastic particles 529 within river systems will be largely affected by the same factors affecting sediment transport, such as 530 hydrological characteristics and environmental conditions (Nizzetto et al., 2016a). Conditions such as 531 low flows and change in river depth or velocity (for example, on a bend) may lead to deposition of 532 particulate matter, whereas high velocity flood conditions and erosion could lead to mobilisation of 533 previously sedimented particles, in addition to the introduction of particles via runoff (Milliman et al., 534 1985; Naden et al., 2016; Walling, 2009). Surrounding land-use can also affect the dynamics of 535 sediment and particulate transport within a river due to erosion, use of soils, irrigation and runoff 536 (Chakrapani, 2005). Plastic residing in river systems may also be subject to in-situ degradation, either 537 by photodegradation or mechanical fragmentation (Williams and Simmons, 1999).

To date only scant attention has been paid to investigating sources, fate and transport of microplastics in terrestrial environments. However it not unreasonable to suggest that microplastics are widely present across land. Litter has been widely reported as a common observation, with many studies commenting on land based (macro)plastic debris (Derraik, 2002; Hoellein et al., 2014; Jambeck et al., 2015; Townsend and Barker, 2014; Williams and Simmons, 1999; Zylstra, 2013).

543

544

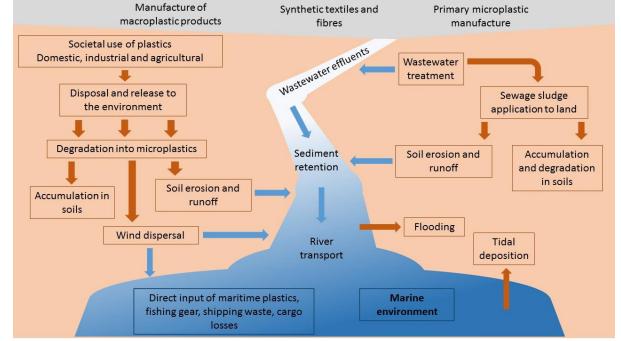


Figure 2. Conceptual diagram of microplastic sources and flows throughout and between anthropogenic,
terrestrial, freshwater and marine environmental compartments.

547

548 4. Microplastics as an environmental hazard

549 4.1. Ecological impacts of microplastics

550 **4.1.1. Organism interaction and ingestion of microplastics**

Based on the evidence of widespread presence of plastics, it is highly likely that organisms in 551 552 terrestrial and freshwater ecosystems will encounter microplastic particles. Depending on the particle 553 size and the physiological and behavioural traits of the organism, there is an opportunity for the 554 ingestion of these particles by invertebrates and vertebrates. Indeed such consumption has been 555 widely observed in many marine species. Although plastic is largely excreted following ingestion, there is evidence to suggest that microplastics can be retained in the gut over timescales beyond those 556 557 expected for other ingested matter (Browne et al., 2008). Further, there is evidence that particles may even cross the gut wall and be translocated to other body tissues, with unknown consequences 558

559 (Browne et al., 2008; Farrell and Nelson, 2013; von Moos et al., 2012). Given the similarity of some 560 phyla that are commonly found in freshwater and marine ecosystems (e.g. nematodes, annelids, 561 molluscs, arthropods) and indeed in soils, similar findings of ingestion in species in these ecosystems 562 are almost inevitable. Since many of these species, likely to take up microplastics, are important to 563 ecosystems (Lavelle, 1997; Sampedro et al., 2006) ecosystem processes such as decomposition and 564 nutrient cycling may be affected by microplastic exposure. Further, there is the potential for food web 565 effects either through effects on keystone species or possibly through the trophic transfer of 566 microplastics themselves.

567 Research to date, predominantly on marine species, has shown the ingestion of microplastic 568 particles in a wide range of species at many organisational levels and with different feeding strategies, 569 including detritivores, filter feeders and predators. In addition to accumulation of particles in 570 organisms at lower trophic levels (Browne et al., 2008), there is also evidence for the trophic transfer 571 of microplastic particles between marine species, especially bivalves and crustaceans (Farrell and 572 Nelson, 2013; Van Cauwenberghe and Janssen, 2014; Watts et al., 2014). This is also likely to occur in 573 terrestrial ecosystems in a similar manner to that of observed trophic transfer and accumulation of 574 gold nanoparticles between earthworms Eisenia fetida and bullfrogs Rana catesbeina (Unrine et al., 575 2012). Gold nanoparticles are comparable to (nano)plastic particles in that are they are similarly 576 poorly soluble (Bouwmeester et al., 2015). There is also evidence that exposure to inert anthropogenic 577 particles can cause physical damage to body tissues (Lahive et al., 2014; Van Der Ploeg et al., 2013).

As far as we are aware, to date only three terrestrial species, the earthworms *Lumbricus terrestris* (Huerta Lwanga et al., 2016) and *Eisenia andrei* (Rodriguez-Seijo et al., 2017) and the nematode *Caenorhabditis elegans* (Kiyama et al., 2012), have been studied in the literature exposed to microplastic particles under laboratory conditions and with ingestion being observed. Among freshwater organisms, the filter feeder *Daphnia magna* has been observed to ingest microplastics (Besseling et al., 2014; Casado et al., 2013; Rehse et al., 2016). Synthetic fibres have also been observed in the digestive systems of freshwater fish collected from the wild, indicating consumption

585 either directly or in association with consumed prey items (Sanchez et al., 2014). Through such 586 consumption, mobile organisms such as fish, mammals and birds may also contribute to the dispersal 587 of microplastics over long distances following the ingestion and subsequent egestion of consumed 588 microplastics (Eerkes-Medrano et al., 2015). A major factor that is known to influence particle 589 ingestion by organisms is particle to mouth size ratio, with smaller particles having greater potential 590 to be ingested by a greater range of organisms. If ingested by lower tropic level organisms, this may support further transfer and accumulation along food chains (Cole et al., 2013; Farrell and Nelson, 591 592 2013; Setälä et al., 2014).

593

594 4.1.2. Observed toxicological effects of microplastics

595 Ingestion of microplastic particles by marine invertebrates has been linked with a wide range 596 of sub-lethal effects including reduced reproduction, reduced growth of individuals and reduced 597 fitness. These are generally the result of the physical effects of ingested microplastics including 598 internal damage such as lacerations, inflammatory responses and plastic particles replacing digestible food, causing individuals to reduce feeding hence resulting in lower energy intake, although effects 599 600 vary between species and plastic types (Moore, 2008; von Moos et al., 2012; Wright et al., 2013a; 601 Wright et al., 2013b). While there are fewer studies conducted to date with soil and freshwater 602 species, the studies that have been conducted generally confirm the potential for microplastics to 603 have detrimental effects on the physiology of species across many ecological niches.

In a recent soil study, Huerta Lwanga et al. (2016) observed mortality in *Lumbricus terrestris* earthworms exposed to polyethylene particles; mortality was increased by 8% at a concentration of 450 g kg⁻¹ polyethylene (in overlying leaf litter) and 25% mortality at 600 g kg⁻¹. Reduced growth and negative effects on burrow construction were also observed. As the concentrations of plastic litter micro-fragments found on soil surfaces are currently unknown, it is difficult to place the concentrations that are used in this study within the range of possible microplastic concentrations

610 that may occurs in soils. The exposure concentrations would certainly seem high compared to 611 expected microplastic levels resulting from diffuse pollution. However, it remains possible that they 612 may be consistent with exposure around some point sources, especially following *in situ* degradation. 613 This finding that annelid worms can be affected by microplastics is consistent with a number of studies 614 conducted for marine species. For example, in a study of Arenicola marina exposed to uPVC 615 (unplasticised PVC) particles experienced weight loss and reduced lipid reserves were observed. A uPVC treatment of 10 g kg⁻¹ dry sediment reduced energy reserves by 30% while at a uPVC 616 617 concentration of 50 g kg⁻¹ dry sediment, energy reserves were reduced by 50%. This effect overall 618 suggests that exposure to UPVC causes metabolic stress to marine benthic sediment worms (Wright 619 et al., 2013a). Due to the close relatedness of worm species in terms of morphology and how they 620 feed in sediment it is likely that similar effects would be observed in freshwater and terrestrial worm 621 species (Rillig, 2012). In the marine copepod, Tigriopus japonicas, Lee et al. (2013) found that although 622 acute exposure (96 hours) to three different particle sizes (0.05, 0.5 and 6 μ m) of polystyrene 623 microbeads, had no impact on the survival rate of adults, in a two generation chronic exposure 624 experiment mortality was observed at concentrations above 12.5 μ g ml⁻¹, with the second generation 625 observed to be much more sensitive than the first generation, especially when exposed to the nano-626 scale particles (0.05 μ m). Larger particles in contrast (6 μ m) had no effect on survival even over two 627 generations, although fecundity was affected at concentrations above 25 µg ml⁻¹. Although the species 628 of copepod used in this study were marine, they are directly comparable to freshwater copepod 629 species and other planktonic filter feeding organisms like Daphnia sp. This implies that toxic effects of 630 microplastics may be size-dependent either as a result of particle ability to permeate body tissues or 631 to cause greater inflammatory response. Studies conducted with nanoplastics also highlight possible 632 size dependent influences on toxicity for both acute survival effects (Besseling et al., 2014; Nasser and 633 Lynch, 2016) and different reproductive effects observed in response to smaller particle fractions (Lee 634 et al., 2013).

635 It is also important to consider how alteration of particle characteristics over different 636 environmental timescales may affect toxicity. Exposure to artificially aged (nano)polystyrene has been 637 found to cause mortality, growth and reproduction effects to the standard test species Daphnia 638 magna over a 21 day period, whereas pristine nano-polystyrene particles caused no significant effects 639 on mortality. Mixtures of nano-polystyrene and fish kairomones (known to cause stress in D. magna) 640 produced an additive effect on body size and reproductive endpoints, indicating that exposure to 641 plastic particles can exacerbate existing environmental stress responses (Besseling et al., 2014). Many 642 studies investigating the toxicological impacts of microplastics have used virgin plastic particles. 643 However, if aged and contaminated, particles can have the potential for greater chemical transfer 644 than virgin particles (see section 4.2.2.). This use of pristine particles could thus lead to a potential 645 underestimation of the toxicological impacts of microplastic exposure under more realistic 646 environmental exposure scenarios. Recently the nanotoxicology research community have recognised 647 the need to conduct experiments with environmentally 'aged' nanomaterial forms (Christian et al., 648 2008; Judy et al., 2015; Lahive et al., 2017). Common nanomaterial transformations, such as hetero-649 and homo-aggregation, changes in surface charge and in particular the development of a surface 650 'corona' of associated macromolecules and chemicals may all occur for both nanoparticles and 651 microplastics (Syberg et al., 2015). Hence future studies with these 'aged' particle forms may be 652 needed to more accurately identify the possible effects of anthropogenic materials in real 653 environments (Schultz et al., 2015).

When considering microplastics and chemical co-transport, principles used in mixture toxicology may be useful to assess these multifaceted stresses in the environment. Given that most environmental microplastic studies quantify microplastics by number of particles rather than by weight (as is more common for bioassays), and none to our knowledge have yet detected nanoplastics in environmental samples, it is not yet possible to determine whether the concentrations used in these studies are environmentally relevant. This is a similarly common criticism of microplastic studies in that the concentrations of particles used are likely not environmentally realistic. Even though the 661 relationship between environmental concentrations and those used in toxicity bioassays is not fully 662 established, it is likely that the concentrations used in laboratory tests are comparable to only the 663 highest levels of environmental contamination. However, it is still valuable to understand the potential 664 ecological implications of microplastic pollution at these high concentrations as a contribution to 665 understanding of hazard and developing risk assessments. Further, given that environmental 666 concentrations of microplastics are likely to increase with input and fragmentation of plastics already 667 present in the environment, the future presence of higher concentrations can be expected (Phuong 668 et al., 2016).

669

670 **4.2. Microplastics as a chemical hazard**

671 **4.2.1.** Leaching of plasticiser chemicals in freshwater and terrestrial environments

672 Plastic materials often contain a wide range of plasticiser chemicals to give them specific physical properties such as elasticity, rigidity, UV stability, flame retardants and colourings (Browne et 673 674 al., 2013; Lithner et al., 2009; Moore, 2008; Teuten et al., 2009). Many of the chemicals associated 675 with plastics have been identified as either toxic or endocrine disruptors including bisphenol-A, 676 phthalates such as di-n-butyl phthalate and di-(2-ethylhexyl) phthalate, polybrominated diphenyl 677 ethers (PBDEs) and metals used as colourings (Hua et al., 2005; Kim et al., 2006; Lithner et al., 2009; Oehlmann et al., 2009; Rochman et al., 2013b; Teuten et al., 2009). Additive chemicals like these are 678 679 weakly bound, or not bound at all to the polymer molecule and as such these chemicals will leach out 680 of the plastic over time. Such releases can be facilitated in environments where particle dispersal is 681 limited and where plastics will experience UV degradation and high temperatures (Andrady, 2011). 682 The locations where microplastics may accumulate in soil and surface waters are therefore likely to 683 be subject to the possible release of these chemicals from plastics and their subsequent transfer to 684 water, sediment and organisms. Lithner et al. (2009) showed that different plastic items can leach 685 toxic chemicals into water that can cause varying effects on Daphnia magna. Different items made of the same polymer may have varying toxicity effects following leaching, based on the type and amount
of plasticisers added during manufacture. This demonstrates that plastic materials can act as a source
of complex leachate mixtures to the environment.

689 As a major environmental sink for all types of plastic waste, landfill material and the leachates 690 arising from landfill sites are highly likely to contain high concentrations of plasticiser chemicals (do 691 Nascimento Filho et al., 2003; Slack et al., 2005; Yamamoto et al., 2001). Within a landfill site chemical 692 conditions change over time with regards to temperature fluctuation, oxygen presence, acid/alkaline 693 conditions and dissolved organic carbon all of which have the potential to change plasticiser leaching 694 (Teuten et al., 2009; Xu et al., 2011). Large scale chemical monitoring studies have identified the 695 presence of phthalate esters (plasticiser chemicals) in a wide range of agricultural and peri-urban soils 696 in various regions of China. Zeng et al. (2008) analysed soil samples from a range of field sites around 697 Guangzhou city, China. The study identified 16 phthalate compounds with concentrations for 698 individual phthalate found ranged from 0.195–33.5 mg kg⁻¹ dry weight soil. The highest concentration 699 of phthalates were found in an agricultural soil, in close proximity to a water course into which 700 wastewater was discharged from nearby industrial activities including manufacture and disposal of 701 plastics and this was identified as the key source of phthalates in soil. Similarly Kong et al. (2012) 702 analysed soil samples from farmland finding concentrations of phthalates ranging from 0.05–10.4 mg 703 kg⁻¹ dry weight. The highest concentrations were found in vegetable plots close to domestic rubbish 704 sites, from which phthalates could be expected to leach. High concentrations were found at sites close 705 to busy roads and at wasteland sites where plastic debris abundance was high. Further to these 706 studies, Wang et al. (2013) sampled soils used for vegetable production near Nanjing (east China). Measured concentrations of phthalates ranged between 0.15–9.68 mg kg⁻¹ dry weight; the highest 707 708 concentrations were found at sites where plastic mulches and polytunnels were in use. Proximity to 709 municipal solid waste sites and application of sewage sludge were also identified as major sources of 710 phthalates, indicating leaching of plasticiser chemicals from plastic particles deposited on land. Taken 711 together, the results suggest that plastic materials release chemicals to soil via a number of the

pathways and are a potential source of plasticisers to soils. This may have significant implications for
terrestrial locations where microplastic concentrations are high, although further studies are needed
to confirm this early evidence.

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6 **4.2.2.** Microplastic associations with organic pollutants

717 Microplastics themselves are widely understood to bind to a range of different hydrophobic 718 organic chemicals (HOCs) within the environment, such as organochlorine pesticides, PAHs, PCBs, 719 PBDEs, dioxins and metals (Besseling et al., 2013; Mato et al., 2001; Rochman et al., 2013c). This may 720 be especially significant in continental freshwater and terrestrial environments, where concentrations 721 of these chemicals are expected to be higher than in marine systems, due to proximity to the use of 722 these chemicals (Dris et al., 2015b). HOCs are recognised as having high lipophilicity (i.e. high 723 octanol/water partition coefficient, Kow), determining whether a chemical will dissolve in water and 724 remain in solution). Chemicals with such a high K_{ow} will typically have a strong affinity for adsorption 725 to organic and particulate matter within water, soil and sediment. These same characteristics, in 726 addition to factors including hydrophobicity of polymer, large or abraded surface properties and 727 biofouling, mean that HOCs also have the potential for sorption to plastic materials (Karapanagioti and 728 Klontza, 2008; Teuten et al., 2007). Microplastics and representative chemicals from many POP classes 729 may become associated in waste streams (e.g. sewage effluent and sludge, landfill waste and leachate) 730 or in anthropogenically influenced environments. Hence, the interactions between microplastics and 731 organic pollutants are particularly pertinent in freshwaters inland, especially those in close proximity 732 to industrialised and populated areas with a high discharge of industrial and domestic wastewater, 733 where small dispersal areas can lead to high pollutant concentrations (Eerkes-Medrano et al., 2015; 734 Free et al., 2014). This will be especially relevant in agricultural areas where plastic products are used 735 in close proximity or in association with the application of hydrophobic chemicals such as some 736 pesticides.

737 Changes to environmental conditions will influence equilibrium dynamics between chemicals 738 and plastics, impacting on chemical accumulation and bioavailability (Bakir et al., 2016; Bakir et al., 739 2014; Karapanagioti and Klontza, 2008; Koelmans et al., 2016). Additionally, particle size and texture 740 will affect the capacity of microplastics to either adsorb or leach contaminants and indeed plasticiser 741 additives. The greater surface area per unit of mass as particles decrease in size increases the potential 742 for surface chemical interactions and thus binding with hydrophobic chemicals. Physically weathered 743 particles are expected to have a larger surface area as a result of cracking and abrasion which increases 744 overall surface area (Ivar do Sul and Costa, 2014; Teuten et al., 2009). Such environmentally-induced 745 changes may be particularly relevant for terrestrial microplastics, which may be exposed to high levels 746 of UV radiation and wind. The ecological impacts of plastic-chemical associations are difficult to predict due to the many interactions between polymers, plastic additives, adsorbent characteristics 747 748 and environmental conditions which will impact on bioavailability (Bakir et al., 2014; Koelmans et al., 749 2016; Velzeboer et al., 2014).

750

751 5. Future research recommendations

752 As this review highlights, the largest gaps in current knowledge are in our understanding of 753 microplastic pollution in terrestrial ecosystems, especially environmental concentrations, sources and 754 ecological impacts. In freshwater systems, knowledge of concentrations of microplastics is rapidly 755 growing. However, in most instances this knowledge has yet to be related to ecological effects. Due 756 to the lack of quantitative data, it is difficult to assess quantitatively the exact nature of the 757 microplastic hazard in these systems and how the consequences of microplastic presence in these 758 ecosystems will manifest themselves. Indeed this is true of microplastics research as a whole, where 759 the long term implications of microplastics are still unclear compared to better-studied chemical 760 pollutants.

761 There is a large degree of uncertainty around the volume, composition and diversity of 762 microplastic particles entering the environment. Information on the scale of production is available 763 as is some data on plastic entry into major waste management systems, however current release rates 764 from these streams either by deliberate or accidental release of refuse or wind action is not quantified. 765 This route from accidental release and littering is, hence, one of the greatest uncertainties for emission 766 predictions. This review highlights the complex challenge of understanding the dynamics and impacts 767 of microplastics as an environmental pollutant, especially understanding microplastics in a freshwater 768 and terrestrial context, but also demonstrates how information from marine studies can be used to 769 infer or predict what may occur in these less studied systems. In a similar way, nanomaterial research 770 can also provide insights into particulate behaviour and fate.

771 To progress the field of research, it is of utmost importance in the first place to define 772 'microplastics' clearly as an environmental contaminant, and thereafter to develop standardised 773 methods for collecting, processing and analysing environmental samples. Such standardisation has 774 the potential to reduce ambiguity and thus allow direct comparison between studies with a view to 775 understanding sources and transport pathways. Spectroscopy methods have already been used to 776 identify freshwater and terrestrial nanoparticles and the continued development of such methods, as 777 well as alternatives such as differential scanning calorimetery (DSC) and thermo-gravimetric analysis 778 (TGA), is important to provide additional information on the polymers present in terrestrial and 779 freshwater ecosystems.

While an ideal scenario would be to reduce the amount of plastic entering the environment, the challenges of reduction from changes in manufacturer and consumer behaviour mean that releases can be expected to continue for some time. Given the volume of plastic currently present in the environment, and the likely increase of microplastics due to fragmentation, it therefore remains important to understand the potential effects of this ever-accumulating pollution (Nizzetto et al., 2016a; Phuong et al., 2016).

Based on the evidence presented in this review, it is clear that our understanding of microplastics in the environment is rapidly developing. However, there are still fundamental gaps in the knowledge and many questions still remain. In summary, the most important questions remaining are:

- What is the current extent of microplastic pollution in terrestrial environments, and how does
 this compare to known contamination in aquatic environments? Which polymers are most
 abundant and does this vary across habitats and regions?
- 793 2) To what extent do environmental conditions and properties of different plastic materials
 794 affect microplastic behaviour and bioavailability under the conditions that are found in
 795 freshwater and terrestrial environments?
- Are adverse effects primarily due to physical impacts of the particle itself, chemical toxicity or
 mixture effects, and does this vary between polymers and species? Are there parallels that
 can be drawn with what is known concerning mechanisms of action for some nanoparticles?
- 4) What are the likely ecological implications of plastics under realistic exposure conditions (i.e.
- 800 microplastics of the type and concentrations likely to be encountered by organisms)?
- 801

802 6. Conclusions

803 The available literature reporting information on plastic use and release suggests that primary 804 and certainly secondary microplastics are likely to be found ubiquitously across terrestrial and 805 freshwater environmental compartments due to their proximity to most point and diffuse sources. 806 Both primary and secondary microplastics entering the environment will persist and continue to 807 fragment to smaller particles. These smaller fragments are likely to pose a greater risk to organism health due to their increased likelihood of uptake, increased surface area for interactions with 808 809 chemicals and greater number of particles per unit of bulk mass (Jeong et al., 2016; Lee et al., 2013). 810 The focus on nanoparticle hazards has recently generated a greater understanding of the behaviour

of particulate pollutants, as well as methods for their detection and hazard assessment. Clear parallels exist from this work to future studies with nanoparticles, with collaboration between the disciplines likely to improve understanding (Bouwmeester et al., 2015; Syberg et al., 2015). This takes the more environmentally relevant approach that it is necessary to understand the fate, behaviour and impacts of microplastics as an environmental pollutant and, therefore, their potential implications for keys ecosystem components and processes.

817 As microplastics can act as both a direct (particulate) hazard and an indirect (chemical) hazard, 818 unravelling ecological effects may call for the application of approaches for mixture toxicity may be 819 beneficial for the analysis of combined plastic-chemical effects. Despite land being the least studied 820 environmental compartment, many of the ecological risks of microplastics identified in aquatic species 821 will also apply to terrestrial ecosystems due to the many ecological and taxonomic parallels that exist 822 between resident species. Studies on the dynamic interactions between plastic particles, plasticiser 823 additives and environmental contaminants is also a field that needs to be expanded to understand 824 how organic chemical partition coefficients to plastics are altered in the presence of sediment and 825 soil. Studies of chemical dynamics within the gut of organisms are also needed in order to better 826 understand the processes that govern bioaccumulation of plasticisers and co-transported chemicals. 827 Ultimately, studies are needed to link the finding in the field studies to laboratory results to better 828 understand both environmentally relevant scenarios of real-world risks posed by microplastics and 829 the underlying mechanisms.

830

831 Acknowledgements

We would like to thank Professor Tamara Galloway, Dr. Martina Vijver and Professor Peter van
Bodegom for their support during the preparation of this review. This work was funded by the UK
Natural Environment Research Council through National Capability funding of the Centre for Ecology

- and Hydrology Pollution and Environmental Risk Program. AW received support from the NERC
- 836 GW4+ DTP Grant Reference NE/L002434/1.

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