

Synthesis & biological applications of glycosylated iminosugars Duivenvoorden, B.A.

Citation

Duivenvoorden, B. A. (2011, December 15). *Synthesis & biological applications of glycosylated iminosugars*. Retrieved from https://hdl.handle.net/1887/18246

Note: To cite this publication please use the final published version (if applicable).

2

A Preparative Synthesis of Human Chitinase Fluorogenic Substrate*[⋆]*

2.1 Introduction

Chitinases are a class of enzymes capable of cleaving natural chitin (a linear polymer of *β*-1,4-linked-*N*-acetylglucosamine) and a wide variety of artificial chitinlike substrates. The existence of endogenous chitinases in mammals was discovered at the end of the last century.¹ First to be identified was chitotriosidase (CHIT1), a human chitinase that is strongly expressed and secreted by lipid-laden tissue macrophages that are found in patients suffering from Gaucher disease. $2-4$ Gaucher disease is a rare lysosomal storage disease in which the influx/efflux balance of glucosylceramide (GC) is disturbed by the inefficient hydrolysis, of GC, by mutant *β*-glucocerebrosidase (GBA1). A second mammalian chitinase, named AMCase (Acidic Mammalian Chitinase), was identified some years later^{5,6} and its role in the etiology of asthma has been proposed. ⁷

The CHIT1 activity in plasma of Gaucher patients correlates to the progression of the disease and the effect of therapeutic intervention. 1,8 Chitotriosidase activity thus is an ideal marker by which Gaucher patients are identified and by which their susceptibility towards therapeutic agents is monitored. ⁹ Currently two

[⋆] Duivenvoorden, B. A.; Dinkelaar, J.; Wennekes, T.; Overkleeft, H. S.; Boot, R. G.; Aerts, J. M. F. G.; Codée, J. D. C.; van der Marel, G. A. *Eur. J. Org. Chem.*, **2010**, *13*, 2565-2570.

therapies for the treatment of Gaucher patients are applied, namely enzyme replacement therapy (ERT) and substrate reduction therapy (SRT) (see also chapter 1). 10–15 Both therapies are expensive and monitoring their effect (i.e. optimal dosage and treatment regimen) through measuring of serum CHIT1 activity has a considerable clinical value.

In the years immediately following the discovery of CHIT1 activity as marker for Gaucher, umbelliferyl chitobioside **126** (Figure 2.1) was used as a fluorogenic substrate in biological assays to give a fluorescent read-out.¹ However, it was found that human CHIT1 possesses intrinsic transglycosylase activity, 16 in that higher oligomers are formed through hydrolysis of **126** followed by coupling to the 4-position of the non-reducing end carbohydrate of another substrate. This side reaction complicates interpretation of the kinetics of the enzyme-mediated generation of the fluorescent 4-methylumbelliferyl anion (4-MU). To circumvent the possibility of CHIT1 mediated transglycosylation a modified fluorogenic substrate **125** (Figure 2.1) was synthesized, in which the 4'-OH is removed. ¹⁶ This modification gave a superior CHIT1 and AMCase substrate as compared to **126**. ⁸ Given the growing interest to monitor CHIT1 activity and given the present interest of AM-Case in relation to asthma, umbelliferone 4'-deoxychitobioside **125** has become a very desired fluorogenic substrate.

Figure 2.1: Umbelliferyl chitobioside fluorogenic substrates **126** and **125**.

In the original paper ¹⁶ the synthesis of **125** started from chitobiose, a disaccharide, forming the target compound **125** in nine consecutive synthetic steps. Although sufficiently effective for the preparation of several milligrams, the route falls short when aiming for larger quantities. The nine-step sequence is quite inefficient (3% overall yield) and furthermore the starting disaccharide (chitobiose) is rather expensive. This chapter describes a more efficient and reliable route towards the superior substrate **125**.

2.2 Results and Discussion

4-Methylumbelliferyl deoxychitobioside **125** was synthesized as outlined in Figure 2.2. The linkage between the chitobiose and the 4-methylumbelliferyl fluorophore could best be achieved via a S*^N* 2-displacement of the anomeric *α*-chloride in **127** by the umbelliferyl phenolate anion. This procedure was selected as aromatic hydroxyl functions are substantially less nucleophilic than the corresponding aliphatic hydroxyls. In addition, Lewis acid mediated glycosylation of phenols can give rise to an unwanted Fries rearrangement. ¹⁷ For the construction of the chitobiose-core a thiophenyl glycoside building block was used because the anomeric thiophenyl group can be introduced early in the synthesis, is stable to the reaction conditions employed and can be selectively activated with a variety of soft electrophiles to provide glycosylating species. Furthermore thiophenyl glycosides are shelf stable and often crystalline, which for the large-scale preparation of building blocks is a valuable asset. To maximize the efficiency in the construction of 4'-deoxychitobiosyl umbelliferone **125** a route was designed, in which a single thioglycoside (**130**) serves as an advanced precursor for both the non-reducing and reducing end glucosamine building blocks. To protect the glucosamine nitrogen function the phthaloyl group was selected because it is cheap, robust under both basic and acidic conditions, and can be readily introduced on glucosamine on a large scale using well-established chemistry. $18,19$ The phthalimide group reliably provides anchimeric assistance in the coupling of the two glucosamines to give the 1,2-trans glycosidic bond and does not give rise to oxazoline side products. Benzyl ethers will mask all hydroxyls during the assembly of the chitobiose disaccharide.

Figure 2.2: Outline for the large-scale synthesis of umbelliferyl chitobioside fluorogenic substrate **125**.

Scheme 2.1 depicts the synthesis of 4'-deoxychitobiosyl umbelliferone which started with the synthesis of thioglycoside **130**. Thioglucosamine **131** was obtained from D-glucosamine in 40% yield over 8 steps on a 147 g scale. 18,19 Only a single chromatographic purification was required in this sequence of reactions. Reductive opening of the benzylidene acetal in the next step was affected by treat-

ment of **131** with TFA and TES to selectively provide key thioglycoside **130** in 85% yield. ²⁰ The formation of the regioisomeric C-4 benzyl ether was not observed. To provide the non-reducing end glucosamine building block, alcohol **130** was subjected to a Barton-McCombie deoxygenation by treatment with NaH and CS_2 followed by MeI to provide the methyl dithiocarbonate. Radical fragmentation using Bu $_3$ SnH and AIBN as initiator in refluxing toluene then led to the deoxygenated glucosamine 128 in 87% yield.²¹ For the construction of the reducing end glucosamine building block **129**, partly protected thioglycoside **130** was condensed with benzyl alcohol employing *N*-iodosuccinimide (NIS) and a catalytic amount of TMSOTf as activator cocktail. ²² The use of a large excess of BnOH (5 equivalents), and the high nucleophilicity of this alcohol as compared to the glucosamine 4-OH, completely prevented self-condensation of **130**. Glucosamine **129** was obtained in 75% yield. In the ensuing NIS/TMSOTf mediated glycosylation deoxy glucosamine **128** and benzyl glucosamine **129** were reacted in a 1 : 1 ratio to provide chitiobioside derivative **132** in excellent yield.

To introduce the umbelliferyl chromophore, disaccharide **132** was transformed into the disaccharide *α*-chloride **127**. To this end, both *N*-phthaloyl groups in **132** were removed by transamidation with ethylenediamine in refluxing *n*-butanol. Subsequent acetylation of the resulting free amines then provided the crystalline dimer **133**. Removal of all benzyl groups from this disaccharide proved to be more troublesome than expected due to the low solubility of the partly debenzylated-*N*-acetyl chitobioside intermediates. The best results were obtained when disaccharide **133** was treated under 5 bar hydrogen pressure using 5 mol% of Pearlman's catalyst in a 1:1 THF/MeOH solvent mixture in the presence of 5 equivalents of AcOH. The fully deprotected 4'-deoxychitobioside was then acetylated to give the penta-*O*-acetate **134** in 65% yield as an amorphous white solid over two steps.

The final stages of the synthesis followed procedures slightly adapted from literature. ²³ Chlorination of the reducing end glucosamine derivative required careful tuning of the reaction conditions. The anomeric acetate **134** was treated with dry HCl in a mixture of AcOH and Ac₂O at 5℃ for 42 hours to afford 4'-deoxychitobiosyl chloride **127** in 74%. Shorter reaction times led to incomplete chlorination and higher reaction temperatures gave interglycosidic bond cleavage. Previously it has been reported that the anomeric chlorination of chitobiosyl acetate can be readily accomplished at room temperature. ²³ Presumably the absence of the hydroxyl function on the 4-position of the non-reducing end GlcNAc in **134** makes the glycosidic linkage more labile towards acidic cleavage. Introduction of the 4 methylumbelliferyl chromophore was accomplished by S_N 2 displacement of the anomeric α -chloride by the tetrabutyl ammonium salt of 4-methylumbelliferone, generated under phase transfer conditions (PTC) .²⁴ The protected umbelliferyl derivative was obtained in 62% as a white amorphous solid. Saponification of the acetyl esters with NaOMe and HPLC purification completed the synthesis of target compound **125**, yielding 227 mg (28%) of product.

Scheme 2.1: Large-scale synthesis of umbelliferyl chitobioside fluorogenic substrate **125**.

Reagents and conditions: a) DCM, BnOH, NIS, 0°C , TMSOTf (75%) b) (1) THF, imidazole, CS₂, 0°C, NaH, 1h, then rT, MeI (93%); (2) Tol, B $\rm n_{3}SnH$, AIBN, Δ (87%); c) DCM, NIS, 0°C , TMSOTf (86%); d) (1) $nBuOH$, ethylenediamine, Δ ; (2) MeOH, Ac₂O, Et₃N (82% over two steps); e) (1) THF, MeOH, AcOH, Pd(OH)₂, H₂; (2) pyr., Ac₂O (65% over two steps); f) AcOH, Ac₂O, HCl, 0°C to 5°C (74%); g) $\mathrm{CHCl}_3, \mathrm{H}_2\mathrm{O}, \mathrm{NaHCO}_3,$ umbelliferone sodium salt, TBAHS (62%); h) MeOH, NaOMe (28% after HPLC purification).

2.3 Conclusion

This chapter describes an efficient, reliable and scalable route for the synthesis of 4'-deoxychitobiosyl umbelliferone **125**. The synthesis is based on the use of a partially protected thiophenyl glucosamine derivative **130** as main building block, which is readily transformed into both the reducing and non-reducing end building blocks for the construction of 4'-deoxychitobiose core. This carbohydrate core is then converted to an α -chloride donor, which was then coupled, under PTC, with 4-methylumbelliferone salt to yield, after deprotection, target compound **125**.

2.4 Experimental section

All reagent were of commercial grade and used as received (Acros, Fluka, Merck, Schleicher & Schuell) unless stated otherwise. Diethyl ether (Et, O) , light petroleum ether $(PE 40-60)$, en toluene (Tol) were purchased from Riedel-de Haën. Dichloromethane (DCM), *N*,*N*dimethylformamide (DMF), methanol (MeOH), pyridine (pyr) and tetrahydrofuran (THF)

were obtained from Biosolve. THF was distilled over LiAlH $_4$ before use. Dichloromethane was boiled under reflux over $\mathrm{P}_2\mathrm{O}_5$ for 2 h and distilled prior to use. Molecular sieves 3Å were flame dried under vacuum before use. All reactions sensitive to moisture or oxygen were performed under an inert atmosphere of argon unless stated otherwise. Solvents used for flash chromatography were of pro analysis quality. Flash chromatography was performed on Screening Devices silica gel 60 (0.004 - 0.063 mm). TLC-analysis was conducted on DC-alufolien (Merck, Kieselgel60, F245) with detection by UV-absorption (254 nm) for UV-active compounds and by spraying with 20% $\rm H_2SO_4$ in ethanol or with a solution of (NH₄)₆Mo₇O₂₄ · 4 H₂O 25 g/L, (NH₄)₄Ce(SO₄)₄ · 2 H₂O 10 g/L, 10% H₂SO₄ in H₂O followed by charring at ∼150 ◦C. ¹H and ¹³C NMR spectra were recorded on a Bruker DMX-400 (400/100 MHz), a Bruker AV 400 (400/100 MHz), a Bruker AV 500 (500/125 MHz) or a Bruker DMX-600 (600/150 MHz) spectrometer. Chemical shifts (*δ*) are given in ppm relative to the chloroform residual solvent peak or tetramethylsilane as internal standard. Coupling constants are given in Hz. All given 13 C spectra are proton decoupled. High resolution mass spectra were recorded on a LTQ-Orbitrap (Thermo Finnigan) Mass spectrometer. LC/MS analysis was performed on a Jasco HPLC-system (detection simultaneous at 214 nm and 245 nm) equipped with an analytical Alltima C_{18} column (Alltech, 4.6 mmD x 50 mmL, 3μ particle size) in combination with buffers A: H₂O, B: MeCN and C: 0.5% aq. TFA and coupled to a Perkin Almer Sciex API 165 mass spectrometer. Optical rotations were measured on a Propol automatic polarimeter. IR spectra were recorded on a Shimadzu FTIR-8300 and are reported in $\rm cm^{-1}$.

Phenyl 3,6-di-*O***-benzyl-2,4-di-deoxy-2-phthalimido-1-thio-***β***-D-glucopyranoside (128).**

$$
\begin{array}{ccc}\n\text{BnO} & \begin{matrix}\n\end{matrix} & \text{SPh} \\
\text{BnO} & \begin{matrix}\n\end{matrix} & \text{SPh} \\
\text{NPhth}\n\end{array}
$$

Glycoside **130** (25.4 g, 43.8 mmol) was coevaperated thrice with dioxane, then taken up in THF (220 mL). Imidazole (0.298 g, 4.38 mmol) and CS_2 (7.9 mL, 131 mmol) were added after which the mixture was cooled to 0◦C. NaH (2.63 g, 60% dispersion in oil, 65.7 mmol) was added and

the reaction was kept at 0°C for one hour then allowed to warm to rT. At rT MeI (4.82 mL, 77.5 mmol) was added. After 30 min. the mixture was quenched by addition of AcOH and subsequently diluted with EtOAc (250 mL). The mixture was then washed with NaHCO $_{\rm 3}$. The layers were separated and the organic layer was dried over MgSO $_4$ and concentrated *in vacuo*. Purification by column chromatography (Tol-EA 100-0→95-5) yielded the thiocarbamate intermediate as the yellow oil (27.4 g, 93%). The thiocarbonate (27.4 g, 40.8 mmol) was coevaporated three times with toluene, dissolved in toluene (800 mL) and degassed with sonication under argon flow for 5 min. $\,\mathrm{Bu}_3\mathrm{SnH}$ (16.4 mL, 61.2 mmol) and AIBN (0.33 g, 2.04 mmol) were added and the mixture was warmed to $120°C$. After 1h when TLC analysis showed complete consumption of starting material the reaction was cooled to rT and concentrated *in vacuo*. The residue was taken up in ACN and washed twice with hexane, the ACN layer was concentrated *in vacuo*. Column chromatography (EtOAc/PE 30%) afforded **128** as an oil (20.1 g, 87%). TLC: EtOAc/PE 45%; ¹H NMR (400 MHz, CDCl₃) δ 7.66-7.83 (m, 4H, H_{arom}), 7.28-7.39 (m, 7H, H_{arom}), 7.16-7.17 (m, 3H, H_{arom}), 6.98-7.02 (m, 5H, H_{arom}), 5.57 (d, 1H, $J = 10.4$ Hz, H-1), 4.55-4.57 (m, 3H, CH₂ Bn), 4.19-4.34 (m, 3H, H-2, H-3, CH₂ Bn), 3.84 (m, 1H, H-5), 3.65-3.69 (m, 1H, H-6), 3.54-3.58 (m, 1H, C-6), 2.31 (dd, 1H, *J* = 3.6 Hz, 12.8 Hz), 1.59 (q, 1H, *J* = 11.6 Hz, H-4); ¹³C NMR (100 MHz, CDCl₃) δ 167.7, 168.0 (C=O Phth), 137.7, 138.0 (C_q), 133.7 (CH_{arom}), 132.5 (C_q), 132.0 (CH_{arom}) , 131.5 (C_q) , 127.3-128.6 (CH_{arom}) , 123.3, 123.1, 83.6 $(C-1)$, 75.4 $(C-5)$, 73.5 $(C-3)$,

73.3 (CH₂ Bn), 72.3 (C-6), 70.6 (CH₂ Bn), 55.5 (C-2)34.0 (C-4); HRMS: $C_{34}H_{31}NO_5S + Na^+$ requires 588.18151, found 588.18115.

Benzyl 3,6-di-*O***-benzyl-2-deoxy-2-phthalimido-***β***-D-glucopyranoside (129).**

BnO HO O NPhth -OBr BnO

Glycoside **130** (23.6 g, 40.5 mmol) was coevaperated thrice with toluene. DCM (810 mL), BnOH (21 mL, 202 mmol) and NIS (10.9 g, 48.6 mmol) were added. The mixture was stirred over activated 3Å molsieves for 30 min. After cooling to 0◦C a catalytic amount of TM-

SOTf (0.81 mL, 4.5 mmol) was added. After 1h the mixture was allowed to warm to rT when TLC analyses showed complete consumption of starting material the reaction was quenched by addition of $Et₃N$ (5.6 mL, 40.5 mmol). The reaction mixture was diluted with DCM and washed with Na $_{2} \mathrm{S}_{2} \mathrm{O}_{3}$. The water layer was extracted twice with DCM, the collected organic layers were dried over MgSO₄ and concentrated *in vacuo*. Purification by column chromatography (EtOAc/PE 20%) yielded **129** as a colorless oil (17.6 g, 75%). TLC: EtOAc/PE 50%; ¹H NMR (400 MHz, CDCl₃) δ, 7.76 (bs, 1H, H_{arom}), 7.62-7.63 (m, 2H, Harom), 7.53 (bs, 1H, Harom), 7.28-7.36 (m, 5H, Harom), 7.02-7.07 (m, 7H, Harom), 6.89-6.93 (m, 3H, H_{arom}), 5.15-5.17 (m, 1H, H-1), 4.78 (d, 1H, *J* = 12.4 Hz, CH₂ Bn), 4.74 (d, 1H, *J* = 12.4 Hz, CH₂ Bn), 4.64 (d, 1H, *J* = 12.0 Hz, CH₂ Bn), 4.58 (d, 1H, *J* = 12.0 Hz, CH₂ Bn), 4.51 (d, 1H, *J* = 12.4 Hz, CH₂ Bn), 4.47 (d, 1H, *J* = 12.4 Hz, CH₂ Bn), 4.23-4.25 (m, 2H, H-2, H-3), 3.82 (m, 3H, H-4, H-6, H-6), 3.62-3.65 (m, 1H, H-5),3.20 (d, 1H, *J* = 2.4 Hz, OH); ¹³C NMR (100 MHz, CDCl₃) δ 168.0 (C=O Phth), 168.1 (C=O Phth), 136.9, 137.6, 138.0 (C_q), 131.4 (Cq), 133.5 (CHarom), 127.2-128.3 (CHarom), 97.2 (C-1), 123 (CHarom), 78.4 (C-3), 74.1 $\left(\text{CH}_{2} \text{ Bn} \right)$, 73.8 (C-4), 73.7 (C-5), 73.5 (CH₂ Bn), 70.6 (CH₂ Bn), 70.3 (C-6), 55.3 (C-2); HRMS: $C_{35}H_{33}NO_7 + Na^+$ requires 602.21492, found 602.21471.

Phenyl 3,6-di-*O***-benzyl-2-deoxy-2-phthalimido-4-O-(3,6-di-***O***-benzyl-2,4-di-deoxy-2 phthalimido-***β***-D-glucopyranosyl)-***β***-D-glucopyranoside (132)**.

OBn ene. DCM (350 mL) and NIS (9.56 g, 42.5 mmol) were added A mixture of donor **128** (20.1 g, 35.4 mmol) and acceptor **129** (20.6 g, 35.4 mmol) were coevaperated three times with toluand the mixture was stirred over activated 3Å molsieves for 30

min. The mixture was cooled to $0°C$ before a catalytic amount of TMSOTf (0.32 mL, 1.77 mmol) was added. After TLC analysis showed complete consumption of starting material (3 h) at 0°C, the reaction was quenched with Et₃N (5.0 mL, 35 mmol). The reaction mixture was diluted with DCM and washed with Na $_2\rm S_2\rm O_3$. The water layer was extracted twice with DCM, the collected organic layers were dried over MgSO₄ and concentrated *in vacuo*. Purification by column chromatography (EtOAc/PE 30%) yielded **132** as a colorless oil (31.7 g, 86%). TLC : EtOAc/PE 30%; ¹H NMR (400 MHz, CDCl₃) $\delta = 7.88$ -7.89 (m, 2H, H_{arom}), 7.67-7.71 (m, 4H, H_{arom}), 7.58-7.59 (m, 2H, H_{arom}), 7.20-7.37 (m, 10 H, H_{arom}), 6.96-7.02 (m, 12H, Harom), 6.82 (bs, 3H, Harom), 5.29 (d, 1H, *J* = 8.0 Hz, H-1'), 4.98 (d, 1H, *J* = 6.4 Hz, H-1), 4.84 (d, 1H, $J = 12.4$ Hz, CH₂ Bn), 4.68 (d, 1H, $J = 12.4$ Hz, CH₂ Bn), 4.44-4.58 (m, 6H, CH₂ Bn), 4.11-4.39 (m, 7H), 3.34-3.58 (m, 6H), 2.28 (dd, 1H, *J* = 4.8 Hz, 12.8 Hz, H-4'), 1.52 (q, 1H, *J* = 12.0 Hz, H-4'); ¹³C NMR (100 MHz, CDCl₃) δ 167.5-168.1 (C=O Phth), 136.9-138.6 (C_q) , 133.4-133.7 (CH_{arom}), 131.5 (C_q) , 126.7-128.3 (CH_{arom}), 122.9-123.5 (CH_{arom}), 97.2 (C-1'), 97.0 (C-1), 76.5, 75.5, 74.5, 74.0, 73.2, 72.5, 72.4, 71.9, 71.1, 70.6, 70.3, 68.1, 57.7 (C-2'), 55.6 (C-2), 34.2 (C-4'); HRMS: $C_{63}H_{58}N_2O_{12} + Na^+$ requires 1057.38820, found 1057.38876.

Phenyl 3,6-di-*O***-benzyl-2-deoxy-2-acetamido-4-***O***-(3,6-di-***O***-benzyl-2,4-di-deoxy-2-acetamido-***β***-D-glucopyranosyl)-***β***-D-glucopyranoside (133).**

OBn ture was refluxed overnight and subsequently concentrated *in* Disaccharide **132** (31.7 g, 30.6 mmol) was dissolved in *n*-BuOH (275 mL) and ethylene diamine (30 mL). This mix*vacuo*. The reaction was then coevaporated thrice with toluene and taken up in MeOH (300 mL). At 0℃ Ac₂O (30 mL, 300

mmol) and Et₂N (8.5 mL, 61.2 mmol) were added and the mixture was allowed to warm to rT. The resulting mixture was concentrated *in vacuo* and taken up in CHCl₃ and washed with $H₂O$. The collected organic layer was stirred over activated carbon and filtered over hyflo-gel concentrated *in vacuo*. Crystallization PE-EA yielded **133** (26.6 g, 82%) as slightly yellow crystals. TLC: EtOAc/PE 25%; ¹H NMR (400 MHz, CDCl₃ / CD₃OD, 1/1): *δ* 7.21-7.35 $(m, 25H, H_{arom})$, 4.86 (d, 1H, J = 12.4 Hz, CH₂ Bn), 4.75 (d, 1H, J = 11.6 Hz, CH₂ Bn), 4.54 4.69 (m, 5H, CH² Bn, H1'), 4.38-4.49 (m, 5H, CH² Bn, H1), 4.12 (t, 1H, *J* = 6.8 Hz), 3.97 (t, 1H, *J* = 6.4 Hz), 3.63-3.79 (m, 5H), 3.44-3.51 (m, 3H), 3.37-3.38 (m, 1H), 2.20 (dd, 1H, *J* = 4.8 Hz, 12.8 Hz, H-4'), 1.94 (s, 6H, CH₃ NHAc), 1.45 (q, 1H, *J* = 12.0 Hz, H-4'); ¹³C NMR (100 MHz, CDCl₃/CD₃OD, 1/1): δ 171.4 (C_q NHAc), 170.8, 137.1-138.3 (C_q), 126.8-128.0 (CHarom), 100.2 (C-1, C-1'), 99.6, 78.4, 75.4, 74.4, 74.1, 73.1, 73.0, 72.3, 71.8, 70.5, 70.2, 69.8, 68.9, 55.5 (C-2'), 51.8 (C-2), 33.1 (C-4'), 22.5 (CH₃ NHAc), 22.2; HRMS: C₅₁H₅₈N₂O₁₀ + Na⁺ requires 881.39837, found 881.39865.

1,3,6-Tri-*O***-acetyl-2-deoxy-2-acetamido-4-***O***-(3,6-di-***O***-acetyl-2,4-di-deoxy-2-acetamido-***β***-D-glucopyranosyl)-D-glucopyranoside (134).**

$$
\begin{picture}(120,15) \put(0,0){\line(1,0){100}} \put(15,0){\line(1,0){100}} \put(15,0){
$$

 α and $Pd(OH)_{2}$ (1 g, 20% on activated carbon, 1.33 mmol) Disaccharide **133** (22.9 g, 26.6 mmol) was disolved in THF (250 mL) then MeOH (250 mL), AcOH (9 mL, 106 mmol) were added. The mixture was shaken overnight on a Parr

apparatus[®] under 5 bar hydrogen pressure. The resulting mixture was filtered over Whatmann[®] filter paper, concentrated *in vacuo* and taken up in pyridine (180 mL). At 0°C, Ac $_2$ O (55 mL) was added, after 1h the mixture was allowed to warm to rT and stirred o.n.. The reaction was quenched by addition of MeOH at 0◦C then concentrated *in vacuo*. The residue was taken up in CHCl $_3$ and washed with 1M HCl:NaHCO $_3$ and brine. The organic layer was dried over MgSO₄ and concentrated *in vacuo*. Purification by column chromatography (MeOH/DCM 3%) yielded **134** as a white amorphous solid (10.6 g, 65%). TLC: MeOH/DCM 5%; ¹H NMR of α acetate (400 MHz, CD₃OD): δ 5.99 (d, 1H, $J = 3.6$ Hz, H-1), 5.24 (t, 1H, *J* = 10.0 Hz), 5.04 (dt, 1H, *J* = 5.2 Hz, 10.8 Hz), 4.56 (d, 1H, *J* = 8.0 Hz, H-1'), 4.44 (d, 1H, *J* = 12.0 Hz), 4.30 (dd, 1H, *J* = 3.6 Hz, 10.8 Hz), 4.23 (dd, 1H, *J* = 5.6 Hz, 11.6 Hz), 4.04-4.12 (m, 2H), 3.88 (t, 1H, *J* = 9.6 Hz), 3.98 (m, 1H), 3.79 (m, 1H), 3.61 $(t, 1H, J = 9.2 \text{ Hz})$, 1.86-2.14 (22H, CH₃ Ac, H-4'), 1.51 (q, 1H, $J = 11.6 \text{ Hz}$, H-4'); ¹³C NMR of *α*-acetate (100 MHz, CD₃OD): δ 171.9-172.4 (C=O Ac), 102.6 (C-1'), 91.5 (C-1), 76.9, 72.4, 72.0, 71.6, 70.7, 66.7 (C-6, C-6'), 63.5, 56.4 (C-2, C-2'), 52.2, 33.8 (C-4'), 20.8-23.0 (CH₃ Ac);HRMS: $C_{26}H_{38}N_2O_{15} + Na^+$ requires 641.21644, found 641.21643.

4-Methylumbelliferyl 1,3,6-tri-*O***-acetyl-2-deoxy-2-acetamido-4-***O***-(3,6-di-***O***-acetyl-2,4-di-deoxy-2-acetamido-***β***-D-glucopyranosyl)-***β***-D-glucopyranoside (135).**

Disaccharide **134** (1.61 g, 2.59 mmol) was dissolved in AcOH (13 mL) and Ac_2O (3.2 mL). At 0◦C dry HCl*^g* was bubbled through (liberated under Kipp conditions) for 3h. The reaction mixture was then placed at 5◦C for 42 h at which TLC analysis (DCM-acetone 60-40)

showed complete consumption of starting material. The reaction diluted with CHCl $_{\rm 3}$ (50 $\,$ mL, 0°C) and washed twice with H₂O (25 mL, 0°C) and twice with NaHCO₃ (25 mL, 0°C). The organic layer was dried over MgSO₄ and concentrated *in vacuo* yielding an amorphous solid **127** (1.14 g) and purity was evaluated by ¹H NMR (400 MHz, CDCl3): *δ* 1.55 $(q, 1H, J = 11.6 Hz, H-4)$, 1.86-2.14 (22H, CH₃ Ac, H-4'), 3.73-3.83 (m, 4H), 4.03 (dd, 1H, J = 4.0 Hz, 11.6 Hz), 4.20-4.25 (m, 2H), 4.36-4.54 (m, 3H), 4.48 (d, 1H, *J* = 8.0 Hz, H-1'), 5.02 (dt, 1H, *J* = 5.2 Hz, 11.2 Hz), 5.30 (t, 1H, *J* = 10.0 Hz), 5.94 (d, 2H, *J* = 8.0 Hz, NHAc), 5.96 (d, 2H, *J* = 8.0 Hz, NHAc), 6.12 (d, 1H, *J* = 3.6 Hz, H-1). The resulting solid was dissolved in CHCl₃ (76 mL) and added to a solution of H₂O (76 mL), NaHCO₃ (1.29 g, 15 mmol), 4methylumbelliferyl sodium salt²⁵ (1.9 g, 9.59 mmol) and TBAHS (1.3 g, 3.84 mmol). The biphasic mixture was stirred overnight under exclusion of light. The phases were separated and the organic layer was washed twice with NaHCO $_3$ (0.2 M) and twice with H $_2$ O. The organic layer was dried over MgSO₄ and concentrated *in vacuo*. Purification by column chromatography (MeOH/CHCl₃ 3%) yielded **135** (0.88 g, 62%) as a white amorphous solid. TLC: MeOH/DCM 5%; ¹H NMR (400 MHz, CDCl₃ / CD₃OD, 1/1): δ 7.44 (d, 1H, *J* = 10.4 Hz), 6.86 (m, 2H), 6.72 (d, 1H, *J* = 9.2 Hz, NHAc), 6.54 (d, 1H, *J* = 9.2 Hz, NHAc), 6.09 (s, 1H), 5.46-5.49 (m, 2H), 5.23-5.26 (m, 3H), 5.14 (t, 1H, *J* = 8.1 Hz), 4.04-4.29 (m, 7H), 2.38 (s, 3H, CH₃ 4-methylumbelliferyl), 1.86-2.14 (19H, CH₃ Ac, H-4'), 1.71 (q, 1H, *J* = 12.4 Hz, H-4'); ¹³C NMR of (100 MHz, CDCl₃/CD₃OD, 1/1): *δ* 171.9-172.4 (C_q Ac), 160.1 (C_q), 159.9, 154.4, 153.3, 125.5 (CH_{arom}), 114.7 (C_q), 133.3 (CH_{arom}), 112.4, 103.5 (CH_{arom}), 102.9 (C-1'), 98.7 (C-1), 72.1, 72.0, 70.0, 69.9, 68.5, 65.2 (C-6, C-6'), 62.0, 54.4 (C-2, C-2'), 54.2, 32.5 (C-4'), 20.5-23.2 (CH₃ Ac), 18.5 (CH₃ 4-methylumbelliferyl); HRMS: $\rm C_{34}H_{42}N_2O_{16}+Na^+$ requires 757.24265, found 757.24278.

4-Methylumbelliferyl 2-deoxy-2-acetamido-4-*O***-(2,4-di-deoxy-2-acetamido-***β***-Dglucopyranosyl)-***β***-D-glucopyranoside (125).**

To a suspension of **135** (0.878 g, 1.195 mmol) in MeOH (60 mL) was added NaOMe (44 *µ*L, 30 wt% in MeOH, 0.24 mmol). The reaction was stirred under exclusion of light. When LCMS (gradient 0 to 50% MeOH) showed complete conversion to the product, the mixture

was quenched with AcOH (70 μ L, 1.2 mmol). The reaction was diluted with H₂O (60 mL), the MeOH was evaporated *in vacuo* and the remaining H₂O was lyophilized. Purification by HPLC (gradient $H_2O-MeOH + 0.1\%$ TFA 80-20 \rightarrow 60-40) evaporation of MeOH and lyophilizing H2O yielded **125** (227 mg, 28%) as white fluffy solid. ¹H NMR (400 MHz, (D₆) DMSO): δ 7.90 (d, 1H, J = 8.8 Hz, NH), 7.67-7.71 (m, 2H, 4-methylumbelliferyl, NH), 7.02 (d, 1H, *J* = 1.6 Hz), 6.94 (d, 1H, *J* = 8.8 Hz, 4-methylumbelliferyl), 6.25 (s, 1H, 4 methylumbelliferyl), 5.17 (d, 1H, *J* = 8.4 Hz, H-1), 4.84-4.90 (m, 3H, OH), 4.69 (bs, 1H, OH), 4.30 (d, 1H, *J* = 8.4 Hz, H-1'), 3.78 (q, 1H, *J* = 9.2 Hz, C-2 or C-2'), 3.36-3.68 (m, 10 H), 2.39 (s, 3H, CH₃ 4-methylumbelliferyl), 1.84 (s, 4H, CH₃ NHAc, H-4'), 1.80 (s, 3H, CH₃ NHAc), 1.21 (q, 1H, $J = 11.6$ Hz, H-4'); ¹³C NMR (100 MHz, (D_6) DMSO): δ 169.2, 169.4 (C=O Ac), 160.1 (Cq), 159.9, 154.4, 153.3, 126.5 (CHarom), 114.3 (Cq), 113.5 (CHarom), 111.9 (CHarom), 103.2 (CHarom), 102.5 (C-1'), 98.3 (C-1), 80.9, 75.1, 72.9, 72.3, 68.2 (C-3, C-4, C-5, C-3', C-5'), 59.7, 63.5 (C-6 and C-6'), 54.4, 57.0 (C-2 and C-2'), 35.8 (C-4'), 23.1 (CH₃ NHAc), 23.0 (CH₃ NHAc), 18.1 (CH₃ 4-methylumbelliferyl); HRMS: $C_{26}H_{34}N_2O_{12} + Na^+$ requires 589.20040, found 589.20031.

References

- [1] Hollak, C. E. M.; Van Weely, S.; Van Oers, M. H. J.; Aerst, J. M. F. G. *J. Clin. Invest.* **1994**, *93*, 1288–1292.
- [2] Aerts, J. M. et al. *Acta Paediatr.* **2008**, *97*, 7–14.
- [3] Boven, L. A.; van Meurs, M.; Boot, R. G.; Mehta, A.; Boon, L.; Aerts, J. M.; Laman, J. D. *Am. J. Clin. Pathol.* **2004**, *122*, 359–369.
- [4] Bussink, A. P.; van Eijk, M.; Renkema, G. H.; Aerts, J. M.; Boot, R. G. *Int. Rev. Cytol.* **2006**, *252*, 71–128.
- [5] Boot, R. G.; Blommaart, E. F. C.; Swart, E.; Ghauharali-van der Vlugt, K.; Bijl, N.; Moe, C.; Place, A.; Aerts, J. M. F. G. *J. Biol. Chem.* **2001**, *276*, 6770–6778.
- [6] Bussink, A. P.; Vreede, J.; Aerts, J. M. F. G.; Boot, R. G. *FEBS Lett.* **2008**, *582*, 931–935.
- [7] Zhu, Z.; Zheng, T.; Homer, R. J.; Kim, Y. K.; Chen, N. Y.; Cohn, L.; Hamid, Q.; Elias, J. A. *Science* **2004**, *304*, 1678–1682.
- [8] Aerts, J. M. F. G.; Hollak, C. E. M. *Baillieres Clin. Haem.* **1997**, *10*, 691–709.
- [9] Boot, R. G.; van Breemen, M. J.; Wegdam, W.; Sprenger, R. R.; de Jong, S.; Speijer, D.; Hollak, C. E. M.; van Dussen, L.; Hoefsloot, H. C. J.; Smilde, A. K.; de Koster, C. G.; Vissers, J. P. C.; Aerts, J. M. F. G. *Expert Rev. Proteomics* **2009**, *6*, 411–419.
- [10] Barton, N. W.; Brady, R. O.; Dambrosia, J. M.; Dibisceglie, A. M.; Doppelt, S. H.; Hill, S. C.; Mankin, H. J.; Murray, G. J.; Parker, R. I.; Argoff, C. E.; Grewal, R. P.; Yu, K. T. *N. Engl. J. Med.* **1991**, *324*, 1464–1470.
- [11] de Fost, M.; Hollak, C. E. M.; Greener, J. E. M.; Aerts, J. M. F. G.; Maas, M.; Poll, L. W.; Wiersma, M. G.; Haussinger, D.; Brett, S.; Brill, N.; vom Dahl, S. *Blood* **2006**, *108*, 830–835.
- [12] Platt, F. M.; Jeyakumar, M.; Andersson, U.; Priestman, D. A.; Dwek, R. A.; Butters, T. D.; Cox, T. M.; Lachmann, R. H.; Hollak, C.; Aerts, J. M. F. G.; Hrebicek, M.; Moyses, C.; Gow, I.; Elstein, D.; Zimran, A. *J. Inherit. Metab. Dis.* **2001**, *24*, 275–290.
- [13] Elstein, D.; Dweck, A.; Attias, D.; Hadas-Halpern, I.; Zevin, S.; Altarescu, G.; Aerts, J. F. M. G.; van Weely, S.; Zimran, A. *Blood* **2007**, *110*, 2296–2301.
- [14] Cox, T.; Lachmann, R.; Hollak, C.; Aerts, J.; van Weely, S.; Hrebicek, M.; Platt, F.; Butters, T.; Dwek, R.; Moyses, C.; Gow, I.; Elstein, D.; Zimran, A. *Lancet* **2000**, *355*, 1481–1485.
- [15] Aerts, J. M. F. G.; Hollak, C. E. M.; Boot, R. G.; Groener, J. E. M.; Maas, M. *J. Inherit. Metab. Dis.* **2006**, *29*, 449–456.
- [16] Aguilera, B.; Ghauharali-van der Vlugt, K.; Helmond, M. T. J.; Out, J. M. M.; Donker-Koopman, W. E.; Groener, J. E. M.; Boot, R. G.; Renkema, G. H.; van der Marel, G. A.; van Boom, J. H.; Overkleeft, H. S.; Aerts, J. M. F. G. *J. Biol. Chem.* **2003**, *278*, 40911–40916.
- [17] Jacobsson, M.; Malmberg, J.; Ellervik, U. *Carbohydr. Res.* **2006**, *341*, 1266–1281.
- [18] Ogawa, T.; Nakabayashi, S.; Sasajima, K. *Carbohydr. Res.* **1981**, *95*, 308–312.
- [19] Medgyes, A.; Farkas, E.; Liptak, A.; Pozsgay, V. *Tetrahedron* **1997**, *53*, 4159–4178.
- [20] Pratt, M. R.; Bertozzi, C. R. *J. Am. Chem. Soc.* **2003**, *125*, 6149–6159.
- [21] Iacono, S.; Rasmussen, J. R.; Card, P. J.; Smart, B. E. *Org. Synth.* **1986**, *64*, 57–63.
- [22] Veeneman, G. H.; van Leeuwen, S. H.; van Boom, J. H. *Tetrahedron Lett.* **1990**, *31*, 1331–1334.
- [23] Osawa, T. *Carbohydr. Res.* **1966**, *1*, 435–443.
- [24] Roy, R.; Tropper, F. D.; Grandmaitre, C. *Can. J. Chem.* **1991**, *69*, 1462–1467.
- [25] Kostova, I. P.; Manolov, I. I.; Radulova, M. K. *Acta Pharm.* **2004**, *54*, 37–47.