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# Allosteric modulation and ligand binding kinetics at the $K_{\nu}11.1$ channel

The research described in this thesis was performed in Division of Medicinal Chemistry at the Leiden Academic Centre for Drug Research, Leiden University (Leiden, The Netherlands). Cover design: Shuo Mi and Zhiyi Yu Thesis layout: Zhiyi Yu Printed in the Netherlands ©Copyright, Zhiyi Yu, 2015. All right reserved. No part of this book may be reproduced in any form or by any means without permission of the author.

## Allosteric modulation and ligand binding kinetics at the $K_{\nu}11.1$ channel

#### **Proefschrift**

ter verkrijging van de graad van Doctor aan de Universiteit Leiden, op gezag van Rector Magnificus prof. mr. C.J.J.M. Stolker, volgens besluit van het College voor Promoties te verdedigen op dinsdag 20 oktober 2015 klokke 13:45 uur

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## Chapter 1

**General introduction** 

Ion channels are transmembrane proteins that create a gated and water-filled pore to allow the flow of ions down their electrochemical gradient between the intracellular and extracellular environment.<sup>1, 2</sup> They are selective for a wide range of cations and anions, such as Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, and Cl<sup>-</sup> ions. The distribution of ion channels has been found in cell membranes as well as the membranes of intracellular mitochondria, endoplasmic reticulum, and nucleus, which suggests their crucial physiological functions in a plethora of excitable and non-excitable tissues. Accordingly, ion channels are important targets for many drugs in current use. Indeed, application of ion channel modulators as medications was operative long before their existence had been demonstrated.<sup>3</sup> For instance, in the 19th century cocaine extraction from coca leaves gave rise to a pharmaceutical class of local anesthetics, which are selective sodium channel blockers.4 In addition to the widespread therapeutic properties, ion channels are also crucially related to human toxicology due to their involvement in human physiology. K., 11.1 (hERG) channel induced cardiotoxicity has become a thorny issue for the pharmaceutical industry in the past years. Blockade of the K\_11.1 channel leads to the prolongation of cardiac action potential duration (APD) and long QT syndrome, which in some individuals can degenerate into ventricular arrhythmias and sudden cardiac death.<sup>5</sup> It has been pointed out that 50% of drugs recalled by the FDA from 1995 to 2000 were also K<sub>2</sub>11.1 blockers. In this thesis, I, therefore, focus on parameters that might determine the K\_11.1 toxicity of compounds at the molecular level, as well as possible strategies for relieving the K\_11.1-related cardiac side effects of drugs. In order to provide some background for the research and concepts that are discussed in this thesis, a brief overview of ion channels has been outlined. Subsequently, the target receptor, the K<sub>v</sub>11.1 channel, is introduced, followed by a summary of objectives and contents of this thesis.

#### Ion channels

Several hundred genes in the human genome encode the pore-forming ion channels across plasma membranes. 1 According to the precise control of channel gating, ion channels are classified as ligand- and voltage-gated channels, of which the opening and closure are mediated by a specific ligand and membrane voltage, respectively.7

#### Ligand-gated ion channels

Ligand-gated ion channels (LGICs) are oligomeric protein assemblies that span the cell membrane and form both the binding site for an endogenous ligand and an ion-conducting pore, which converts a chemical signal into an ion flux through the membrane. 8, 9 Based on homology in their amino acid sequence and topology of their component subunits, LGICs are divided into three superfamilies, including P2X, ionotropic glutamate, and Cys-loop receptors. 10 They play an important role in fast synaptic transmission and in the modulation of cellular activity, in particular for basic brain functions like learning, memory and attention.

#### P2X receptors

P2X receptors are non-selective cationic channels gated by extracellular ATP, which are permeable to Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, and in some cases also to Cl<sup>-</sup> ions. <sup>11</sup> A P2X channel contains three subunits, and each subunit is composed of two transmembrane domains connected by a large, glycosylated extracellular loop, and intracellular C- and N-terminus (Figure 1A).9 At present, there are seven P2X receptor subtypes found in mammals, which are termed P2X<sub>1</sub> to P2X<sub>2</sub>. <sup>12</sup> These channels play paramount roles in various physiological processes such as nerve transmission, pain sensation, the response to inflammation, multiple facets of diabetes, and tumor cell growth, making them attractive drug targets in recent years. 12-14

#### **Ionotropic glutamate receptors**

Ionotropic glutamate receptors are cation-selective ion channels that open their ion-conducting pores in response to the binding of glutamate. 15 This channel is made up of four homologous subunits with each subunit containing three transmembrane segments, an intracellular C-terminus and extracellular N-terminus (**Figure 1B**). The ionotropic glutamate receptors are grouped into three subtypes according to their distinct responses to certain small molecule agonists: N-methyl-D-aspartate (NMDA), α-amino-3-hydroxy-5-methyl-4-isoxazolepropionic-acid (AMPA) and kainate (KA) receptors. 10, 15, 16 These distinct ion channels are implicated in almost all aspects of nervous system development and function, and their dysfunctions can be linked to epilepsy or chronic neurodegenerative conditions like schizophrenia, Alzheimer's and Parkinson's diseases, as well as stroke.15,16

#### **Cys-loop receptors**

The Cys-loop receptors, named from the conserved cysteine loop in their extracellular domain, constitute the largest subfamily amongst all LGICs.<sup>9, 17</sup> The best characterized members of the Cys-loop receptor superfamily are nicotinic acetylcholine receptors (nAChRs), serotonin type 3 receptors (5-HT<sub>3</sub>Rs), GABA receptors type A (GABA, Rs), glycine receptors (GlyRs) and zinc activated channels (ZACs). 18 Although these ion channels exhibit diverse kinetic and pharmacological profiles, they share a common structure of five identical subunits that are pseudo-symmetrically arranged to form a rosette with a central ion-conducting pore for either anion (GABA, Rs and GlyRs) or cation (nAChRs, 5-HT, Rs, and ZACs) flux.<sup>17, 19</sup> As shown in **Figure 1C**, each subunit of the channel is made up of an extensive N-terminus and a short C-terminus on the extracellular side, and four transmembrane segments.8 All these receptors are very important targets for a wide variety of drugs, and their dysfunctions are associated with different kinds of diseases. For example, nAChRs have been demonstrated to play many critical roles in brain and body functions, indicating their therapeutic modulation of Alzheimer's disease, epileptic disorders and chronic pain. 20-22 Likewise, GABA, Rs have been unraveled to regulate the controlling neuronal activity in distinct brain regions, and dysfunction of this ion channel involves a range of central nervous system diseases such as anxiety, insomnia, muscle spasms, Alzheimer's disease and schizophrenia.23,24

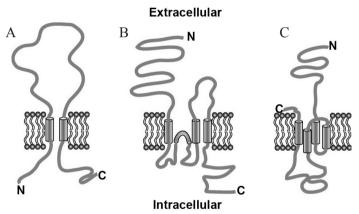


Figure 1. Schematic representation of a single subunit from the three superfamilies of ligand-gated ion channels across the membrane: (A) P2X receptors, (B) Ionotropic glutamate receptors, and (C) Cys-loop receptors.9

#### Voltage-gated ion channels

Voltage-gated ion channels (VGICs) are integral transmembrane proteins that conduct selected inorganic ions at high rates across the membrane in response to changes in transmembrane voltage.<sup>25</sup> The VGIC superfamily, with 143 members in the human genome, forms the third largest family of signal-transduction proteins, following the G protein-coupled receptors and protein kinases. <sup>26, 27</sup> As shown in Figure 2, ten families are included in this VGIC superfamily: voltage-gated sodium/calcium/potassium channels (Na,/Ca,/K,), calcium-activated potassium channels (K<sub>Ca</sub>), cyclic nucleotide-modulated (CNG) ion channels, hyperpolarization-activated cyclic nucleotide-modulated (HCN) channels, transient receptor potential (TRP) channels, two-pore channels (TPCs), inwardly rectifying potassium channels  $(K_{ir})$  and two-pore potassium channels  $(K_{2p})$ . <sup>10, 28</sup> All these channels are composed of a common cation-selective pore-forming module containing two transmembrane segments and an intervening pore loop. 10 Except for  $K_{ir}$  and  $K_{2P}$  channels, all other ion channels consist of four subunits symmetrically arranged around the central ion-conducting pore, and each subunit encompasses six transmembrane segments (S1-S6). 10, 26, 27 Amongst these segments, S1-S4 segments form the voltage-sensor domain with four to eight positively charged residues in the S4 segment conferring voltage dependence, whereas S5 and S6 make up the pore domain including the ionic pore with the selectivity filter. 1 Movement of the voltage sensors and pore domains constitutes three main conformational states of the VGICs: closed, open and inactivated states. On the other hand, K. and  $K_{\mbox{\tiny 3DD}}$  which are composed of four and three subunits respectively, are structurally the simplest VGICs with only pore-forming segments for each subunit. These two potassium channels are regulated by membrane lipids and intracellular 'ligands', as diverse as G proteins, Mg<sup>2+</sup> ions, polyamines and ATP. <sup>10</sup> Apart from the main subunits, VGICs also possess some smaller, auxiliary subunits to modify their functions, increasing the diversity and complexity of this superfamily. VGICs play a fundamental role in most aspects of cell physiology and underlie complex integrative processes like learning and memory in the brain and coordinated movements of the muscles, which makes them the molecular target for a number of therapeutic agents and also toxins.<sup>27</sup>

Potassium-selective channels with 78 members are the most diverse and widespread group of ion channels, whereas K, channels, in humans encoded by 40 genes and divided into 12 subfamilies (K<sub>v</sub>1-12), form the largest family among these potassium channels.<sup>29,30</sup> All K, channels have a high level of structural similarity, and play a pivotal role in a variety of cellular processes, such as functioning of excitable cells, cell apoptosis, growth and differentiation, the release of neurotransmitters and hormones, and maintenance of cardiac activity.<sup>31</sup> Therefore, K, channels constitute potential drug targets and provide tremendous opportunities for treating cancer, autoimmune diseases, metabolic, neurological and cardiovascular disorders. However, K, channels also present a challenge to the development of  $K_v$ -targeted medications, of which the  $K_v$ 11.1 channel is an explicit example with its promiscuity to drug binding, resulting in drug-induced ventricular arrhythmias.<sup>2</sup>

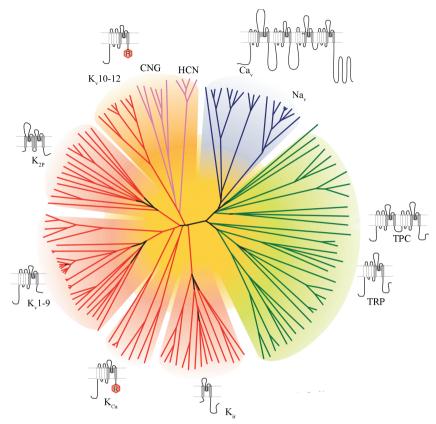
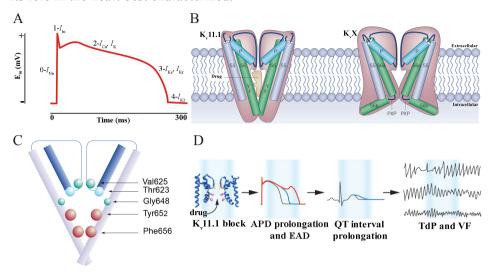


Figure 2. A phylogenetic tree of voltage-gated ion channels, and a schematic representation of the structure of each voltage-gated ion channel within this superfamily.<sup>28</sup>

#### The K<sub>v</sub>11.1 channel

#### Overview of the K\_11.1 channel

The K 11.1 channel, of which the pore-forming α-subunits are encoded by the human ether-à-go-go-related gene (hERG), carries a rapidly activating delayed rectifier potassium current  $(I_{Kr})$  underlying its essential role in cardiac action potential repolarization.<sup>32</sup> In addition, the K<sub>v</sub>11.1 channel has been found to be expressed in various brain regions, smooth muscle cells, endocrine cells, and a wide range of tumor cell lines, which implicates the channel's therapeutic opportunities beyond the cardiac system.<sup>32, 33</sup> However, due to its involvement in life-threatening cardiac arrhythmias, the terminology "hERG" or "K\_11.1" has become notorious in the drug discovery community. Hence, much attention has been paid to K<sub>v</sub>11.1-induced cardiotoxicity over the past decades, which makes its role in the heart best characterized.



**Figure 3.** (**A**) The ventricular action potential waveform indicating the major ionic currents that contribute to its morphology and duration ( $I_{\text{Na}}$ : inward sodium current,  $I_{\text{to}}$ : transient outward potassium current;  $I_{\text{Ca}}$ : L-type inward calcium current;  $I_{\text{K}}$ : outward repolarizing potassium current;  $I_{\text{Ks}}$ : slowly activating delayed rectifier potassium current;  $I_{\text{KI}}$ : inward rectifier potassium current). At activating delayed rectifier potassium current;  $I_{\text{KI}}$ : inward rectifier potassium current). The presence of a Pro-X-Pro (PXP) motif in the S6 helix of other  $K_{\text{v}}$  channels causes a kink and thereby reduces the size of the channel pore cavity. At activating delayed rectifier potassium current;  $I_{\text{KI}}$ : inward rectifier potassium current;  $I_{\text{KI}}$ : inward

The  $K_v$ 11.1 channel displays unusual gating properties that impact cardiac electrical function. Upon membrane depolarization the  $K_v$ 11.1 channel opens slowly but inactivates rapidly, while it recovers rapidly from inactivation but deactivates slowly during repolarization.<sup>5, 34</sup> Therefore, the  $K_v$ 11.1 channel conducts robust outward currents ( $I_{Kr}$ ) relatively late in the ventricular action potential, accelerating the phase 3 repolarization without significantly disturbing the preceding plateau phase 2 (**Figure 3A**). Nevertheless, the  $K_v$ 11.1 channel lacks a Pro-X-Pro (PXP) motif existing in other  $K_v$  channels (**Figure 3B**), resulting in a larger cavity that can accommodate a wide range of drugs.<sup>36</sup> Furthermore, the  $K_v$ 11.1 channel has multiple aromatic rings lining the pore (**Figure 3C**), which facilitates the binding of drugs containing aromatic rings at the channel.<sup>37</sup> Collectively, a number of drugs including cardiac and non-cardiac medications can bind to the  $K_v$ 11.1 channel. As depicted in **Figure 3D**, drug block of the  $K_v$ 11.1 channel at the molecular level induces APD prolongation, early afterdepolariza-

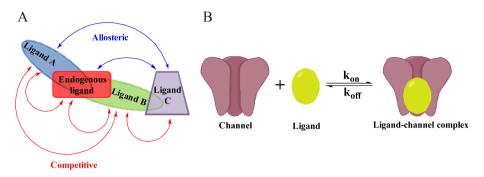
tions (EADs) and heterogeneity of repolarization.<sup>38</sup> This generates a lengthening of the QT interval in the electrocardiogram at the tissue level, which can further lead to life-threatening arrhythmias like Torsade de Pointes (TdP) and ventricular fibrillation (VF). Therefore, it has become obligatory routine practice for all pharmaceutical and biotechnology companies to screen for K\_11.1 liability of drug candidates in the early stage of drug development. More details on the structural features and physiological functions of K, 11.1 channel will be stated in the chapter 2.

#### Ligands for the K<sub>2</sub>11.1 channel

#### Blockers and modulatory activators

A large number of structurally diverse compounds have been shown to bind to the  $K_v 11.1$  channel. In most cases these drugs reduce the  $I_{Kr}$  by directly blocking the conduction pore or less frequently by interfering with channel trafficking to the surface membrane, and these drugs are referred to as K<sub>2</sub>11.1 blockers. <sup>32, 39, 40</sup> In addition, some blockers like APETx1 exert their inhibitory effects on the K\_11.1 current by modulating the gating properties of the channel. 41, 42 On the other hand, a small amount of compounds have also been characterized as K, 11.1 activators, which have the opposite activity of blockers and thus enhance the  $I_{K_T}$  current.<sup>39, 43</sup>

Most K<sub>v</sub>11.1 blockers share a common binding region within the aqueous inner cavity at the pore of the channel, where residues Tyr652 and Phe656 are found to be crucial for drug binding via hydrophobic and electrostatic interplays, including  $\pi$ - $\pi$  and  $\pi$ -cation interactions.<sup>37, 39</sup> Apart from the intracellular binding pocket, various scorpion, spider and sea anemone toxins inhibit the K\_11.1 channel from the extracellular side or by binding to the voltage-sensor domain. 42 With regard to the K 11.1 activators, different binding locations at the channel have been identified, depending on the mechanisms by which they potentiate the  $I_{Kr}$ . According to the primary action model, these K<sub>v</sub>11.1 activators are separated into types 1 and 2 classes.<sup>32, 39</sup> Type 1 K, 11.1 channel activators attenuate the inactivation and deactivation rates of the channel, whereas type 2 activators primarily impair the channel's inactivation rate. Distinct from each other as well as blockers, type 1 and type 2 K, 11.1 activators bind to a site between the pore and voltage sensor and a region that is closer to the selectivity filter, respectively. Additionally, several type 2 K<sub>v</sub>11.1 activators like NS1643 and PD307243 have been proposed to interact with the channel at the extracellular face. 43, 44 Taken together, multiple binding sites exist at the K\_11.1 channel, which illustrates the potential for different ligands to display either competitive or allosteric interactions depending on their binding sites relative to one another (**Figure 4A**).<sup>45</sup>



**Figure 4.** Simple schematic representations of the ligand-channel interaction: **(A)** Arrows demonstrate a competitive (endogenous ligand with ligands A and B, ligand B with ligands A and C) or allosteric interaction (ligand C with ligand A and endogenous ligand) of different ligands based on their binding sites relative to one another,<sup>45</sup> and **(B)** Single-step binding of a ligand to the K<sub>0</sub>11.1 channel.

#### **Binding kinetics of ligands**

The specific ligand-receptor interaction is the chemical basis of virtually all biological activities of compounds. 46 As shown in Figure 4B, binding of a K\_11.1 ligand to the channel can be considered as a single-step reaction in the simplest scenario. At first, the ligand binds to the K<sub>y</sub>11.1 channel with a second-order association rate (k<sub>on</sub>), and then detaches from the channel with a first-order dissociation rate  $(k_{\text{off}})$ . In a closed system, the extent of ligand-channel binding is commonly quantified in terms of the equilibrium dissociation constant  $(K_p)$  defined as the value of  $k_{off}/k_{on}$  or its (often inhibitory) affinity (IC<sub>50</sub> or  $K_i$ ). To date, more and more evidence from a wide range of receptors has addressed the concept that in vivo efficacy of a ligand is not only described by the in vitro measured equilibrium dissociation constant, but also depends on its association and, more critically, dissociation rates. 47-49 A ligand's residence time (RT), i.e. the reciprocal of its dissociation rate, may be at the basis of the ligand's duration of action and target selectivity. 48 As for the K\_11.1 channel, a 30-fold safety margin, quantified as the ratio of IC<sub>50</sub> and maximum plasma concentration (C<sub>max</sub>), has been proposed to define the cardiac safety of K\_11.1 blockers. 50 However, the binding of a K\_11.1 ligand to the channel remains a dynamic process in which binding (association) and unbinding (dissociation) events develop over time.<sup>51</sup> Moreover, the approach in which only the IC<sub>50</sub> value is considered, is rather arbitrary and may not satisfactorily account for the degree of ion channel inhibition reached in various conditions, leading to unsatisfactory quantitative predictions.<sup>51,52</sup> Importantly, K, 11.1 blockers with similar IC50 values but different binding kinetics at the channel have been reported to pose distinct proarrhythmic risks.<sup>52, 53</sup> Therefore, binding kinetics of ligand-K\_11.1 interactions constitute a critical component in understanding the mechanism of action, and thus should be incorporated within the present screening strategy to improve the risk profiles of drug candidates. 51, 52, 54

#### **Objectives and outline of this thesis**

Drug-induced arrhythmia due to blockade of the K<sub>v</sub>11.1 channel has emerged as an unanticipated side effect of many pharmacological agents and is a major obstacle for drug development. Although much attention has been paid to screening K 11.1 liability of drug candidates over the past decades, the current testing paradigm has important limitations and may have led to withdrawal of blockbusters on the present market or stopping the development of potentially valuable therapeutics. Therefore, investigating the mechanism of action of ligands at the K\_11.1 channel, where in particular the ligand binding kinetics may yield comprehensive understanding of K\_11.1-related cardiotoxicity, may further provide new strategies for circumventing this side effect in drug development. In addition, compounds that can relieve the unintended K 11.1 blockade of drugs, e.g., allosteric modulators, are of considerable interest because of their potential to prevent drug-induced arrhythmia via combination therapy.

In Chapter 2, the gating kinetics and ligand binding characteristics at the K<sub>2</sub>11.1 channel are extensively discussed, following the basic introduction of the channel biosynthesis, structure and function, and pharmacological target and anti-target applications in drug development.

At the start of this project, high-throughput assays had not been reported to characterize allosteric modulators of the K<sub>2</sub>11.1 channel, nor had ligand binding kinetics been determined at the channel. Thus, [3H]astemizole and [3H]dofetilide binding assays were developed and utilized to identify positive and negative allosteric modulators of the K<sub>u</sub>11.1 channel in **Chapter 3**.

Based on the work in chapter 3, a novel negative allosteric modulator (LUF7244) was identified in several different [3H]dofetilide binding assays in Chapter 4. LUF7244 was found to reduce the K<sub>v</sub>11.1 affinity of dofetilide, astemizole, sertindole and cisapride in a [3H]dofetilide competitive displacement assay, and more importantly, to prevent the proarrhythmic effects induced by astemizole, sertindole and cisapride in a newly validated neonatal rat ventricular myocyte model. This research raises the possibility to resume the clinical use of unintended K, 11.1 blockers via pharmacological combination therapy. Following this, in Chapter 5 the synthesis of a series of possible allosteric modulators and the evaluation of their allosteric effects on [3H]dofetilide binding at the channel are reported. Compared to LUF7244, several more potent negative allosteric modulators were obtained via small chemical modifications.

In Chapter 6, a novel [3H]dofetilide competition association assay was developed and validated to evaluate the kinetic parameters of fifteen prototypical K<sub>2</sub>11.1 blockers at the channel. Next to that, an immobilized artificial membrane column was applied to measure the membrane affinity of these blockers to elucidate the effect of membrane affinity of ligands on their affinity and kinetic parameters. In addition to varied affinity values and association rates of K\_11.1 ligands at the channel, Chapter 7 focuses on the discovery of blockers with diversified dissociation rates or RTs. Compounds with very short (< 1 min) and much longer RTs (> 100 min) were disclosed in this chapter, which enabled the construction of a "k<sub>on</sub>-k<sub>off</sub>-K<sub>D</sub>" kinetic map for all these compounds. Additionally, results from patch clamp studies implicated the importance of RTs in regulating the functional IC<sub>50</sub> values of K<sub>v</sub>11.1 blockers.

Overall, allosteric modulation and binding kinetics of ligands at the K\_11.1 channel have been thoroughly studied in this thesis. The general conclusions and future perspectives for this field of research have been summarized in Chapter 8. Hopefully, the findings in this thesis will add to the current understanding in ligand-K<sub>2</sub>11.1 interactions, and also provide effective opportunities for abrogating K, 11.1-induced cardiotoxicity.

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## Chapter 2

Kinetic studies of the K<sub>v</sub>11.1 (hERG) channel

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#### **Abstract**

The K<sub>v</sub>11.1 channel, encoded by the human ether-à-go-go-related gene (hERG), mediates a rapid delayed rectifying potassium current  $(I_{K_r})$  underlying repolarization of the cardiac action potential. This channel has been proven to be involved in a diversity of physiologic and pathological processes. "Loss-of-function" mutations in the gene or blockade of the channel by a wide range of prescription medications can prolong the cardiac action potential duration (APD), producing the long QT syndrome (LQTS) that is associated with a markedly increased risk of ventricular arrhythmias and sudden cardiac arrest. On the other hand, "gainof-function" mutations in the gene shorten the ventricular APD and cause the short QT syndrome (SQTS). Accordingly, the K 11.1 channel has elicited intense scientific interest in the past several decades. The counter-intuitive gating kinetics of the channel have been suggested to play a specific role in suppressing arrhythmias, and recent studies have pointed out the requirement of incorporating ligand binding kinetics with channel block affinity to refine the K, 11.1 liability of drug candidates. In this review, the biogenesis, structure, function, and pharmacological applications of the K<sub>v</sub>11.1 channel are briefly outlined first. Afterwards, the unusual gating kinetics of the channel together with the influences on its gating are summarized. Finally, binding and corresponding kinetics of ligands at the channel as well as assay conditions and techniques for measuring these parameters are discussed.

#### Introduction

Ion channels are membrane-spanning proteins that conduct ions across the cell and internal organelle membranes. They are involved in a tremendously diverse range of fundamental physiological processes, from rapid responses driven by the neuronal system or fast contractile forces produced by skeletal and smooth muscles to more protracted procedures such as cell growth, differentiation and migration. After G protein-coupled receptors (GPCRs), ion channels constitute the second largest target class for known drugs. Potassium-selective channels, represented by some 70 known loci in the mammalian genome, are the largest and most diverse group of ion channels. This superfamily of potassium channels is mainly divided into three groups: voltage-gated potassium channels ( $K_v$ ), inward rectifying potassium channels ( $K_v$ ) and calcium-activated potassium channels ( $K_v$ ). Of these,  $K_v$  channels are the largest group, and they are encoded by 40 genes and further divided into 12 subfamilies ( $K_v$ 1-12).

The K<sub>2</sub>11 family name, "ether-à-go-go" (EAG) coined by Kaplan and Trout in 1969, was a humorous reference to the similarity between the go-go dancers of the 1960s and the way the legs of a mutant fly shake under ether anesthesia.<sup>5</sup> In this family, the K\_11.1 channel, of which the pore-forming subunit is encoded by the human ether-à-go-go related gene (hERG), has received the greatest scientific scrutiny due to its association with potentially life-threatening cardiac arrhythmias. The K<sub>2</sub>11.1 channel is often referred to by several partial synonyms: KCNH2 and hERG to the gene,  $I_{\kappa_r}$  to the native current, and  $K_{\nu}11.1$  to the fully assembled channel. Missense mutations on human chromosome 7 where KCNH2 is located or pharmacological blockade of the K<sub>2</sub>11.1 channel lead to a prolongation of the action potential (AP) duration (APD) that could translate into long QT syndrome (LQTS), Torsades de Pointes (TdP) and sudden death.<sup>5,6</sup> Electrophysiological studies on these mutants and K, 11.1 blockers have revealed that gating kinetics of the channel play an important role in  $I_{K_r}$  suppression.<sup>7-9</sup> Moreover, the assessment of K<sub>2</sub>11.1 liability for drug candidates only with IC<sub>50</sub> values is considered to be unrefined. Therefore, more attention should be paid to the binding kinetics of ligands at the channel in order to achieve a better understanding of the underlying mechanisms of drug actions and subsequently to evaluate more comprehensively their proarrhythmic profiles. 10 In this context, this review will focus on gating kinetics and ligand binding kinetics investigated on the K\_11.1 channel as well as a general introduction of the channel production cycle, its structure, physiological functions, and pharmacological applications during the drug development process.

#### Production and degradation of the K\_11.1 channel

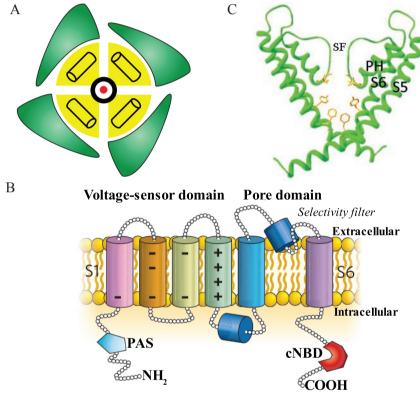
The production of the K\_11.1 channel begins in the nucleus with transcribing KCNH2 to mRNA, which then acts with the ribosome to synthesize polypeptides. The resulting polypeptide chains are assisted by cytosolic chaperones (heat shock protein 70 [Hsp70] and 90 [Hsp90]) to prevent misfolding or protein degradation, and are further used in the formation of channel subunits in the endoplasmic reticulum.11 The produced subunits assemble into a tetramer (immature channel) that is exported to the Golgi apparatus, where the channel undergoes a complex glycosylation to obtain a mature and stable conformation before being trafficked to the plasma membrane. The K\_11.1 trafficking is an essential process that contributes to the maturation of the channel ensuring its proper physiological functions.<sup>11</sup> In some cases, the defective trafficking of mutant channels results in a QT interval prolongation, which can be rescued by pharmacological chaperones that stabilize the K<sub>2</sub>11.1 channel by binding to specific sites of the channel inner pore.5 On the other side, degradation of mature channels occurs in lysosomes after being internalized in endocytic vesicles and tagged with ubiquitin. In addition, when the K\_11.1 maturation process is defective, its subunits are also tagged by ubiquitination for degradation and further transferred to the proteasomes or recycled via recycling endosomes. The K\_11.1 channel density at the cell surface is determined by the balance between the transport of mature channels to the cell membrane and their corresponding degradation.

#### Structural features and physiological functions of the K<sub>2</sub>11.1 channel

#### **Structure**

Consistent with other K<sub>y</sub> channels, the K<sub>y</sub>11.1 channel has a tetrameric structure formed by four  $\alpha$ -subunits (**Figure 1A**) and auxiliary subunits. As shown in **Fig**ure 1B, each α-subunit is composed of six transmembrane segments that constitute two major domains with helices S1-S4 contributing to the voltage-sensing domain (VSD) and helices S5-S6 together with the pore loop outlining the pore domain. The VSD undergoes conformational changes upon perturbation of the transmembrane potential through an array of positively charged residues located within S4 helices, whereby the lengthy S5-Pore linker, also known as the "turret", is a peculiar feature of the K<sub>2</sub>11.1 channel. The α-subunits co-assemble along the pore axis to form the cavity and selectivity filter (SF) of the channel to allow the central ion conduction. The stability of the SF (Figure 1C) is dominated by the 28

interaction of surrounding pore helices with filter residues.<sup>6</sup> With regard to the auxiliary subunits, a  $\beta$ -subunit, MinK-related peptide 1 (MiRP1), has an important role in accelerating  $I_{\rm Kr}$  deactivation and modulating the response of the K $_{\rm v}$ 11.1 channel to drugs.<sup>12</sup> Missense mutations in the KCNE2 gene encoding the MiRP1 have been identified to associate with LQTS and ventricular fibrillations.<sup>13</sup> Notably, the expression levels of MiRP1 in atria and ventricular muscle appear to be low, and thus, exert minor effects on biophysical and pharmacological properties of the channel.<sup>14</sup> However, it is highly expressed in Purkinje fibers of the ventricular conduction system and atrial pacemaker cells.



**Figure 1. (A)** The tetrameric structure of the  $K_v11.1$  channel with the voltage-sensing domain (VSD) colored in green and pore region in yellow, and the red dot representing the potassium ions flow.<sup>5</sup> (**B**) Schematic diagram of a single  $K_v11.1$  subunit containing 6 α-helical transmembrane domains (S1-S6).<sup>6</sup> (**C**) Homology model of the selectivity filter (SF) of the  $K_v11.1$  channel based on the crystal structure of the  $K_v11.2$  channel.<sup>15</sup>

Like all other potassium channels, the structure of the  $K_v11.1$  channel is mainly divided into transmembrane and cytoplasmic regions. The four  $\alpha$ -subunits and auxiliary subunits are located in the transmembrane segment, while the NH<sub>2</sub>- and COOH-terminals are arranged at the cytoplasmic side. The extracellular loop

between S5 and the pore region contains a glycosylation site that is important for maturation and trafficking of the channel protein.8 The NH<sub>2</sub>-terminal contains a Per-Arnt-Sim (PAS) domain and a stretch of 240 amino acids that are responsible for protein trafficking and definition of the K\_11 family, respectively.<sup>5, 16</sup> On the other hand, the COOH-terminal linked to the S6 helice consists of a cyclic nucleotide binding site (cNBD), contributing to the opening and closure of the permeation pathway for ions. A major difference between the K<sub>2</sub>11.1 channel and other 6-transmembrane domain potassium channels is derived from the vestibule site. The K\_11.1 channel has a larger central cavity due to the absence of a Pro-X-Pro (PXP) motif from the S6 domain that causes the inner helices to bend in other potassium channels.<sup>17</sup> This structural openness is also supported by drug-trapping experiments which indicate that the K<sub>2</sub>11.1 channel can accommodate a wide variety of structurally diverse drugs and large molecules. 18 These K, 11.1 blockers are able to access the binding sites inside the pore, within S6 domains and the end of pore helices, where the important residues for high-affinity blockade of compounds are located.<sup>3</sup>

#### Physiological functions

The K\_11.1 channel is expressed in multiple organs including heart, central nervous system (CNS), gastrointestinal tract (GIT) and endocrine system, and its physiological functions differ according to their distributions in these specific tissues. 6, 19 In the CNS of mammals this channel maintains the membrane potential and development of neurons of the spinal cord and carotid glomus cells, whereas it regulates the gut motility in the GIT. In the endocrine system the K\_11.1 channel has been found to be involved in insulin secretion and epinephrine release in chromaffin cells. Apart from these functions, the K<sub>x</sub>11.1 channel has a dominant presence in normal human cardiomyocytes where its physiological function is best characterized.

As shown in Figure 2, the generation of the myocardial AP reflects the sequential activation and inactivation of a series of ion channels, and the AP can be divided into five different phases. 6, 20, 21 In Phase 0, a rapid depolarization occurs when the openings of potassium channels decrease and fast sodium channels open. During Phase 1, there is an early and fast repolarization mainly due to the inactivation of sodium channels and activation of the transient outward potassium channels  $(I_{to})$ . Thereafter, the activation of the long opening calcium channels (L-type  $\operatorname{Ca}^{2+}$ ,  $I_{\operatorname{Ca}}$ ) produces a small steady influx of  $\operatorname{Ca}^{2+}$ , which counteracts the outward K<sup>+</sup> currents induced by an ultra-rapid potassium delayed rectifier. This leads to the formation of a prolonged plateau called Phase 2. The repolarization

in Phase 3 is rapid and robust, and it is initiated by deactivation of sodium and calcium channels and activation of  $K_v 11.1 (I_{K_r})$  and  $K_v 7.1 (I_{K_r})$  channels. The K\_11.1 channel plays a very important role in Phase 3 repolarization because of its unique gating kinetics: slow activation and fast inactivation coupled to fast recovery from inactivation and slow deactivation. This rapid and delayed rectifier potassium currents  $(I_{\nu})$  terminate the cardiac AP and return the membrane potential to baseline (Phase 4). Since the K<sub>2</sub>11.1 channel closes (de-activates) slowly towards the end of the repolarization, there is a constant efflux of potassium ions through the channel. Thus, the resting cell membrane potential is close to the Nernst equilibrium potential for potassium. This also contributes to the refractoriness of cardiac cells in the early diastolic period immediately after repolarization.<sup>5</sup> When a premature beat occurs during the early diastolic period, a large outward current antagonizes the depolarization of the cell and thus propagation of the beat. In this case, functional reduction of  $I_{Kr}$  by "loss-of-function" mutations or pharmacological blockade on the K\_11.1 channel translates into a clinical setting in patients prone to arrhythmias determined by premature beats. As shown in Figure 2, different waveforms in the electrocardiogram (ECG) reflect distinct phases of the APD in a cardiomyocyte.

#### Target and antitarget of the K<sub>2</sub>11.1 channel in drug development

#### Roles in cancer

Changes in the expression of the K<sub>y</sub>11.1 channel have been associated with dysregulations of cell proliferation in a plethora of tumor types from different cell lineages. The K\_11.1 channel in tumor cells has been postulated to facilitate cell migration in diverse hematopoietic neoplasm through an integrin-related signaling pathway and to enhance the proliferation, invasiveness and lymph node dissemination of cells as well as to reduce cell differentiation.<sup>23, 24</sup> By contrast, the K<sub>2</sub>11.1 channel has not been expressed at a significantly high levels in the corresponding non-cancerous cells. Therefore, the K<sub>2</sub>11.1 channel, in different tissues, may be used as a biomarker for abnormal cell behaviors and proliferation. Nevertheless, the K<sub>2</sub>11.1 channel has also been unraveled to induce cell apoptosis in various cell types by different mechanisms, independent of their capacity to inhibit cell proliferation via cell cycle arrest.<sup>24</sup> Further exploration of this apoptotic pathway together with the channel up-regulation in different tumor types may lead to the development of new anti-tumor therapies in the future. For instance, leukemia is a type of cancer that originates from the bone marrow hematopoietic

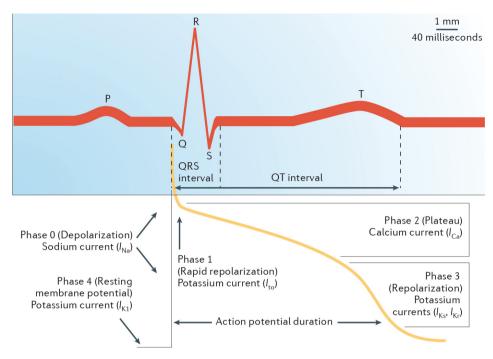


Figure 2. Standard model of the cardiomyocyte QT interval on an electrocardiogram (upper panel) and the corresponding action potential duration (lower panel). The QT interval is a measure of the duration of ventricular depolarization and subsequent repolarization.<sup>22</sup>

stem cells. K\_11.1 channels are expressed in CD34+/CD38-/CD123high cells and induce significant proliferation of leukemia cells, whereas they are not detected in normal bone marrow cells.<sup>25</sup> Thus, a specific K\_11.1 blocker could be used to potentially prevent leukemia via reducing abnormal cell proliferation without affecting healthy cells that do not express the K 11.1 channel. Taken together, the K<sub>2</sub>11.1 channel may be used as a novel diagnostic and prognostic marker in human cancers as well as a new target for anti-neoplastic treatments.

#### Roles in schizophrenia

The contribution of the native K 11.1 channel to the intrinsic electrical properties of neurons is poorly understood. K, 11.1-3.1 is the primate specific isoform of the K<sub>v</sub>11.1 channel expressed in the CNS, and it has been associated with schizophrenia.<sup>26</sup> In this isoform the first 102 amino acids of the most abundant isoform K\_11.1-1a are replaced with 6 amino acids, which results in faster channel deactivation but slower inactivation rates. Although both isomers are comparably expressed at the hippocampus and prefrontal cortex in a control group, the expression ratio of K<sub>2</sub>11.1-3.1 to K<sub>2</sub>11.1-1a demonstrates an approximate 2.5-fold increase among patients with schizophrenia. Investigations on the K\_11.1 channel in neural cells may facilitate our understanding of the pathology of schizophrenia, and would provide a new solution for treating this mental disorder in the future.

#### Roles in smooth muscle related diseases

The K\_11.1 channel is pH sensitive, implying a molecular link for regulating electrical signaling through acidity of the gastrointestinal lumen. Several antibiotics like erythromycin display such adverse reactions as cramps and diarrhea, which might be caused by unintentional blockade of the K<sub>2</sub>11.1 channel.<sup>24</sup> Akbarali et al. reported that the K<sub>2</sub>11.1 channel fine-tuned the resting membrane potential of the circular smooth muscle cells in opossum esophagi, and channel blockers initiated the contraction of muscle segments by depolarization.<sup>27</sup> In human jejunum the K\_11.1 channel was immunohistochemically detected in circular and longitudinal smooth muscle cells, where it was assumed to have a fundamental role in controlling the motility patterns by modulating the electrical behavior of muscle cell.<sup>28</sup> Administration of methanesulfonanilide K<sub>2</sub>11.1 blockers like E-4031 and MK-499 impacted the rhythmic contractions of jejunum. Altogether, these findings could help in the search for new targets for the treatment of gastro-esophageal reflux diseases, achalasia as well as intestinal motility disorders.

Human uterine contractions are triggered by an AP possessing a spike followed by prolonged depolarization.<sup>29</sup> Activation of the K<sub>2</sub>11.1 channel results in the inactivation of the AP by suppressing the amplitude and duration of contraction before labor. In obese women, enhancement of the K 11.1 activity by reducing the expression of  $\beta$ -inhibitory proteins shortened the duration of uterine APs and weakened contractions. This finding offers an explanation for the higher incidence of caesarian delivery in obesity, which should be further studied as it offers a new insight into the poor labor outcomes of obese women.

#### **Roles in endocrine dysregulations**

In human pancreatic islets, the K<sub>2</sub>11.1 channel has been proven to contribute to the regulation of insulin secretion and the firing of  $\beta$ -cells.<sup>30</sup> Blockade of the  $K_{\nu}11.1$ channel is sufficient to cause hyper-excitability during application of the insulin releasing drugs. This suggests that alterations in the channel function might be associated with some hyperinsulinemic conditions. The role of the K<sub>2</sub>11.1 channel in hyperinsulinemic conditions is awaiting further elucidation that may lead to the discovery of more effective treatment for diabetes.

#### Roles in arrhythmia

Cardiac arrhythmias, anything from mild palpitations to severe conditions that have the potential to induce cardiac arrest or sudden death, are a major mortality cause in developed countries. The K\_11.1 channel has been suggested to play a particularly pivotal role in the suppression of arrhythmias, yet it has also been found to account for 30-40% of congenital LOTS cases and is the most common cause of drug-induced arrhythmias and cardiac death all over the world. 31, 32

#### The K\_11.1 channel as a target in arrhythmology

The short QT syndrome (SQTS) is a cardiac channelopathy associated with atrial fibrillation and sudden cardiac death.<sup>4</sup> The SQTS type 1 variant (SQTS1) involves mutations at KCNH2 that encodes the α-subunit of the K<sub>..</sub>11.1 channel.<sup>33</sup>, <sup>34</sup> Brugada et al. identified two different missense mutations at position 588 in the S5-P loop region (i.e. N588K) of the K<sub>v</sub>11.1 channel, which were linked to the hereditary SQTS. These mutations resulted in a dramatic increase in  $I_{Kr}$  leading to a heterogeneous shortening of the APD and refractoriness.5 For this SQTS1, class III antiarrhythmic drugs, also known as K 11.1 blockers, may be a therapeutic approach for controlling dysregulations of the cardiac rhythm by lengthening the APD and increasing the myocardial refractoriness without significantly altering the conduction velocity.<sup>35</sup> Notably, the sensitivity of SQTS1 to distinct K<sub>v</sub>11.1 blockers differs in normalizing the shortening of QT intervals. For example, quinidine restored the shortened QT interval towards a normal range, whereby administration of sotalol and ibutilide in the same condition failed to normalize the abnormal QT interval.<sup>36</sup> This could be explained by the state-dependent inhibition of K<sub>2</sub>11.1 blockers to the channel, as the N588K-K<sub>2</sub>11.1 channel exhibits altered voltage-dependent inactivation kinetics compared to the wild type channel.

Missense mutations of the KCNH2 gene are also associated with LQTS type 2 variant (LQTS2) and an increased risk for sudden cardiac death. There are four classes of mechanisms determining the LQTS2 abnormalities, acting on K\_11.1 synthesis (class 1), intracellular transport or protein trafficking (class 2), channel gating (class 3) and potassium ion permeation (class 4). Anderson et al. performed two comprehensive analyses on 201 LQTS2-linked mutations in four structural domains (NH2- and COOH-terminus, S1-S4 [VSD], S5-S6 [pore region], extracellular and intracellular linkers), and confirmed that the most common underlying mechanism for LQTS2 was the disturbance of protein trafficking (class 2).<sup>37</sup> Therefore, correction of the K\_11.1 trafficking deficiency could become a useful therapeutic principle in rescuing LQTS2. Most mutations causing trafficking defects can be pharmacologically amended by highly selective K, 11.1 blockers like

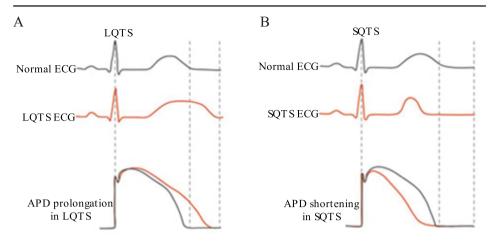
E-4031.38 This compound displayed the highest chance to successfully correct the LQTS2 in channels expressing mutations at the NH,-terminus, followed by those in the pore domain and then C-linker.<sup>37</sup> Nevertheless, when the wild type K<sub>2</sub>11.1 channel was co-expressed, the correction ability of E-4031 changed priorities to mutations in the pore domain followed by the ones in the NH<sub>3</sub>-terminus and finally these at the C-linker site. Additionally, K, 11.1 activators, which can augment the  $I_{K,k}$  offer another novel anti-arrhythmic strategy for treatment of LQTS2. The first K\_11.1 activator RPR260243 was reported in 2005, 39 and only a limited number of such compounds have been available until now. Although these activators have shown pronounced antiarrhythmic effects in various in silico, in vitro and in vivo models, they are found to induce other proarrhythmic side effects such as increasing the APD dispersion<sup>31</sup> and shortening the APD. Overall, activation of the K<sub>2</sub>11.1 channel is a new research area, which remains to be exploited in the future.

#### K\_11.1-related cardiotoxicity

As mentioned previously, mutations of the K<sub>2</sub>11.1 channel affect heart conditions characterized by "loss-of-function" or "gain-of-function", which are related to LQTS (Figure 3A) and SQTS (Figure 3B), respectively. A genetic heterogeneous disorder induced by "gain-of-function" mutations of the K,11.1 channel is the congenital SQTS, which is identified by a shortened QT interval (< 330 ms) and episodes of syncope, paroxysmal atrial fibrillation or life-threatening arrhythmias.<sup>36</sup> On the other hand, the LQTS represents a cardiac clinical condition involving the K\_11.1 channel, which is characterized by "loss-of-function" of the channel. LQTS can be either congenital or acquired. Congenital LQTS is caused by mutations disrupting the channel functions as described previously. Mutations often lead to deficiencies in protein trafficking, which reduces the amounts of functional K<sub>2</sub>11.1 channels located in the cell membrane.<sup>36</sup> Although mutations at other ion channels can also result in LQTS, the reported LQTSs are dominated by the K\_11.1 and K\_7.1 channels. Acquired LQTS refers to drug induced prolongation of the QT interval. This response may be induced by a specifically designed drug that blocks the cardiac repolarizing currents through binding to either the K\_11.1 channel (e.g., dofetilide, sotalol, ibutilide) or to other ion channels involved in cardiac repolarization (e.g., alfuzosin to the sodium channel). 40 However, almost all drugs known to cause acquired LQTS appear to preferentially block the K<sub>2</sub>11.1 channel. Furthermore, QT prolongation may arise as a side effect from compounds designed to act on non-cardiac targets. Drugs that belong to different chemical classes and possess various therapeutic indications are able to exert cardiotoxic effects by interfering with the K<sub>2</sub>11.1 channel, such as antihistamines

(e.g., terfenadine), gastrointestinal prokinetic agents (e.g., cisapride), psychoactive agents (e.g., chlorpromazine) and antibiotics (e.g., cotrimoxazole). 12 To date, the most common cause of withdrawal from or restrictions of drugs in the market is related to their potential to cause QT prolongations and TdP via blockade of the K 11.1 channel. This underlines the importance of early evaluation of possible blockade of the K, 11.1 channel for drug candidates, desirably in the early phase of drug development.

Several factors have been identified to account for the promiscuity of the K\_11.1 channel towards drug binding.<sup>6,12,41</sup> Firstly, two aromatic residues Tyr652 and Phe656 are highly conserved in the K\_11.1 channel, and they are critical for the binding of structurally diverse drugs. A number of drugs can form a cation- $\pi$ interaction between their positively charged nitrogen atoms and Tyr652 as well as a hydrophobic attraction of aromatic rings with Phe656. Secondly, the K<sub>2</sub>11.1 channel lacks a PXP motif on the S6 helix leading to a large size of the inner cavity, which makes the channel accommodate a range of K 11.1 blockers encompassing ample molecules. Lastly, multiple distinct binding sites for drugs at the K<sub>v</sub>11.1 channel have been discovered, such as an intracellular binding pocket for dofetilide and an extracellular binding region for BeKm-1.42 This provides another reasonable explanation for the promiscuity of the K<sub>2</sub>11.1 channel towards drug binding. Therefore, it is now common practice for pharmaceutical industry to screen for K\_11.1 liability in new chemical entities and conduct thorough clinical QT studies for potential drug candidates. In spite of the seriousness of K\_11.1 related-cardiotoxicty of drugs, the exact molecular and physiological interrelationships behind K<sub>2</sub>11.1 blockade, QT prolongation and TdP remain to be explored. As shown in Figure 2, depolarization and repolarization of cardiac myocytes are complex processes that result from multiple competing and complementing ionic currents. For example, the potent K<sub>2</sub>11.1 blocker verapamil does not display a repolarization liability due to its inhibition of the inward calcium current. 40, 43 In this context, a Comprehensive in vitro Proarrhythmia Assay (CiPA) has been proposed as a novel safety screening proposal intended to replace the current regulatory guideline by July 2015. 44, 45 Other strategies considering both affinity and kinetic indicators have also been proposed in order to improve the assessment of arrhythmic side effects for drug candidates. 46, 47



**Figure 3.** Schematic diagrams depicting the correlation of ventricular action potential duration (APD) to body-surface electrocardiogram (ECG) in normal, LQTS and SQTS patient groups. **(A)** Representation of an LQTS; **(B)** Representation of an SQTS.<sup>48</sup>

## Gating kinetics of the K<sub>2</sub>11.1 channel

Voltage gated potassium channels regulate the selective conduction of  $K^+$  across biological membranes by opening and closing the pore via structural rearrangement of the activation and/or inactivation gates. These gates are located at both intracellular and extracellular entrances of the SF, and the mediating process is referred to as gating kinetics.<sup>49</sup>

The two gates of the K<sub>v</sub>11.1 channel control the potassium flow through the channel. The activation gate is arranged at the intracellular side and formed by four S6 segments with slow movement to open and closed states, while the inactivation gate is located at the extracellular region of the pore and constituted by four SF segments with rapid movement to an inactivated state. The pore of the K<sub>v</sub>11.1 channel is more stable in the open conformation compared to other voltage-gated potassium channels.<sup>50</sup> The voltage-sensor of the K<sub>v</sub>11.1 channel consists of positively charged residues on the S4 domain and mediates the channel opening.<sup>5</sup> Additionally, the K<sub>v</sub>11.1 channel has three extra negative charges in the VSD, which makes it different from other voltage-gated potassium channels that possess only three negative charges on S2 and S3 segments.<sup>51</sup> All these negative charges stabilize the positive charges on the S4 segment by forming salt bridges, influencing the movement of the S4 relative to other transmembrane segments during the gating.

Compared to most voltage-gated potassium channels, the gating kinetics of the  $K_v11.1$  channel are unique: the transition rates of inactivation and recovery from inactivation are more rapid than the kinetics of activation and deactivation

(Figure 4).<sup>51</sup> The time constant of inactivation for the K\_11.1 channel aligns from 6.5 to 2.2 ms in the voltage range of -10 to +30 mV, whereas its activation time constant is more than 50 ms at a voltage of +50 mV. By comparison, the durations of inactivation and activation at the same voltage ranges for the Shaker channel were approximately 2000 ms and less than 2 ms, respectively. The quick removal of inactivation at the depolarizing pulse as well as a slow deactivation leads to the formation of a "hook" (tail) current following repolarization, which is typical for the  $K_v 11.1$  channel and can be observed in various macroscopic  $I_{K_r}$  recordings.<sup>52</sup>

#### **Activation and deactivation**

The activation and deactivation processes of the K\_11.1 channel are relatively slow compared to other voltage-gated potassium channels. The slow activation gating is caused by the slow movement of the VSD. The electrostatic interaction between Asp540 and Arg665 mediates the coupling between the VSD and S6 segment. Mutations on these residues were found to affect the activation and deactivation rates of the K\_11.1 channel, implicating that they play an important role in stabilizing the intermediate states in the activation process.<sup>53</sup> The slow deactivation kinetics is maintained by the unique cytosolic domains of the channel, in particular the proximal N-terminal domain.<sup>54</sup> Smith et al. confirmed that deleting the NH, terminus and adding a recombinant protein corresponding to the PAS domain accelerated the channel deactivation rate. Comparable to deletion of the entire NH<sub>2</sub> terminus, disruption of the α-helix at the NH<sub>2</sub> tail by substitutions of glycine or serine led to an increase of the K 11.1 deactivation rate. Similarly, K 11.1 channels, which have the N-terminal and PAS domains removed or Asp774, Glu788 and Asp803 mutated by arginine, exhibited much faster deactivation gating kinetics in contrast to the wild type channels.55 However, this amendment did not alter the activation gating kinetics of the K\_11.1 channel. Of note, voltage dependency of activation and deactivation was insensitive to the potassium concentrations at the extracellular face of the channel.<sup>52</sup>

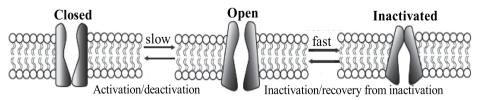


Figure 4. Different K 11.1 channel conformations in the closed, open and inactivated states.56

## Inactivation and recovery from inactivation

Inactivation kinetics of the K\_11.1 channel comprise the forward (inactivation) and backward (recovery from inactivation) transitions between the open and inactivated states. 32, 51 These procedures are extremely fast and voltage sensitive. At depolarized potentials the channel inactivation becomes faster than activation resulting in a steady-state rectification at positive voltages, whereas the channel deactivation is much slower compared to its recovery rate from inactivated state forming a hooked tail current that is specific for the K<sub>2</sub>11.1 channel.<sup>5,57</sup> The K<sub>2</sub>11.1 channel only demonstrates a C-type inactivation, which is different from the N-type inactivation existing for other voltage-gated ion channels. The N-type inactivation of potassium channels is induced by an occlusion of the ion-conducting pore by the intracellular N-terminal domain, while the C-type inactivation of the K\_11.1 channel permits the interconversion between two conformations of the stable conducting and non-conducting SFs. Kinetic analyses have underlied the early and late pore helix motions during the inactivation process of the K\_11.1 channel.<sup>58</sup> To date, the C-type inactivation of ion channels has been better understood than N-type inactivation.<sup>32</sup>

## Effect of gating kinetics on $I_{Kr}$

Ion conduction is highly dependent on the SF, which is composed of four peptides with a conformational arrangement pointing toward the pore axis of the  $K_v$ 11.1 channel. The gating kinetics influence the contributions of  $I_{K_v}$  to the AP repolarization phases.<sup>59</sup> The K<sub>2</sub>11.1 channel is activated during the AP upstroke and plateau phases, while a fast inactivation process of the channel greatly limits the amount of outward current at positive voltages (inward rectification). As the cell membrane repolarizes to negative voltages, the K\_11.1 channel recovers from inactivation and reenters the open state leading to an outward "resurgent" current that facilitates further repolarization. Thereafter, the outward  $I_{\kappa_r}$  declines at the end of the AP due to the channel deactivation as well as a decreasing driving force for outward currents. The mechanism of K<sub>2</sub>11.1 rectification is closely related to voltage-gated fast inactivation. Since the inactivation of channels takes place at a much faster rate than their activation, no significant outward currents are observable during the depolarization to a high membrane potential. The rapid inactivation of the K\_11.1 channel and the resulting rectification are partially responsible for the prolonged plateau phase which is typical of ventricular APs.<sup>7</sup>

## Factors modulating K\_11.1 gating kinetics

Apart from voltage, the gating kinetics of the K\_11.1 channel are susceptible to many other factors such as temperature, extracellular pH, K<sup>+</sup> and Ca<sup>2+</sup> concentrations. 60 An increase in temperature shifted the voltage dependence of activation to a hyperpolarizing direction, whereas that of inactivation was shifted towards a depolarizing direction. 61 Moreover, the rates of activation and deactivation increase with higher temperatures, but less markedly than inactivation and the reverse recovery rates. In addition, Vandenberg et al. emphasized that the results obtained at room temperature could not be simply extrapolated to those of 37 °C by using a single temperature scale factor for the K\_11.1 channel due to the temperature dependence of gating kinetics.<sup>61</sup> When it comes to the extracellular pH, the  $I_{\rm Kr}$  amplitude was decreased and kinetics of activation and deactivation were increased when the external pH was lowered from 7.4 to 6.0, whereas activation was accelerated by alternations of the external pH from 7.4 to 8.0.62 Regarding the extracellular potassium concentrations, increasing the K<sup>+</sup> concentration from 4 to 10 mmol·L<sup>-1</sup> produced a +10 mV shift in voltage-dependent inactivation of the K<sub>2</sub>11.1 channel and slowed the inactivation time course, while reducing the K<sup>+</sup> concentrations 4 to 1 mmol·L<sup>-1</sup> had little influence on inactivation voltage dependence, but accelerated the inactivation time course. 63 By contrast, an increase in Ca<sup>2+</sup> concentration from 1.8 to 10 mM did not significantly influence the inactivation kinetics of the K 11.1 channel, whereby it resulted in a reduction of  $I_{\rm Kr}$  amplitude and activation kinetics but increased the deactivation rate.<sup>64</sup> Thus, Ca<sup>2+</sup> may interact with externally accessible channel residues and further alter the membrane potential sensitivity of the voltage-sensor responsible for activation and deactivation, while this interaction does not change the inactivation gating mechanism.

## Influences of ligands on K,11.1 gating kinetics

 $K_v 11.1$  activators are able to increase the ventricular  $I_{K_v}$ , which can be used to treat LQTS associated ventricular arrhythmias. These activators act primarily on the C-type inactivation mechanism that is strictly related to the inward rectification. Based on different action modes, K, 11.1 activators are classified into type 1 and 2 groups. 15 The type 1 activators enhance the K<sub>v</sub>11.1 function by attenuating the C-type inactivation and slowing down the deactivation rate of the channel. The study of RPR260242 revealed that this activator exerted a concentration-dependent effect on reducing the deactivation rate of the K\_11.1 channel and also produced a positive shift in the voltage dependent inactivation, leading to an increase in  $I_{Kr}$  amplitude. 65 RPR260242 was found to interact with two clusters

of residues located on separate domains of the channel. The first cluster had its residues located in regions S5 (L553 and F557) and S6 (N658 and V659). Interaction with this cluster affects the channel's rearrangements that are associated with the deactivation and inactivation gating of the channel. On the other hand, residues for the second cluster are located in the S4-S5 linker (V549 and L550) and S6 (I662, L666 and Y667) segment. Binding of RPR260242 to this cluster is essential to mediate the slow deactivation kinetics induced by its interaction with the first cluster residues. Type 2 K\_11.1 activators augment the channel activity by primarily attenuating its C-type inactivation via slowing down the rapid onset of inactivation and shifting the half maximal inactivation voltage to more depolarized potentials. Perry et al. reported that PD-118057 shifted the half-point for inactivation of the K\_11.1 channel by +19 mV and increased the peak outward current of the K<sub>2</sub>11.1 channel by 136%.66 Scanning mutagenesis and molecular modeling suggested that PD-118057 bound to a hydrophobic pocket on two different channel subunits: L646 in the S6 domain for one subunit and L622 and F619 in the pore helix for an adjacent subunit. 15, 66

Apart from these activators, a number of K 11.1 blockers alter the gating kinetics of the K<sub>2</sub>11.1 channel. Based on the existence of distinct binding regions for ligands at the channel, K 11.1 blockers can be divided into small molecules and peptide analogs which interact with intracellular and extracellular binding pockets, respectively. The anesthetic drug midazolam was recently unraveled to inhibit  $I_{Kr}$  by binding to the same classical binding site within the channel pore as that for prototypical blockers.<sup>67</sup> This compound resulted in a small negative shift of the activation curve of the K<sub>v</sub>11.1 channel, while it did not significantly impact the channel's steady-state inactivation. On the other hand, BeKm-1 is a peptide inhibitor interacting with the extracellular amino acid residues close to the external pore region of the K<sub>...</sub>11.1 channel. <sup>42</sup> Zhang *et al.* found that BeKm-1 drastically shifted the midpoint of the activation curve towards positive voltages, decreased the activation rate but accelerated the deactivation rate.<sup>68</sup> These effects can be explained by the fact that binding of the BeKm-1 molecule to the channel contacted with the S5-P linker, impacting activation and deactivation kinetics of the K<sub>2</sub>11.1 channel.

## Binding and corresponding kinetics of ligands at the K<sub>1</sub>11.1 channel

Methods for measurements

Patch clamp technique

Voltage-gated ion channels are primarily studied using the voltage clamp/patch clamp technique, which enables the clamping of the cell membrane potential to particular voltages using square or more complex waveforms that allows the measurement of specific activities in the ion channels. <sup>69</sup> Since this method allows a direct and real-time measurement of ion channel activity, it has been regarded as the golden standard in performing electrophysiological investigations on ion channels. There are various patch clamp techniques and systems available, different from one another but relying on the same principle.

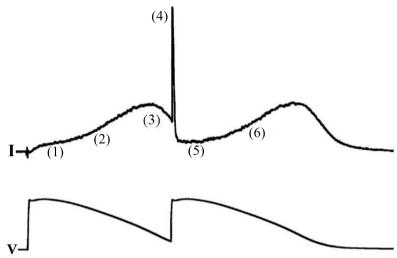


Figure 5. Example of a so-called action potential voltage clamp applied to transfected CHO cells expressing K<sub>2</sub>11.1 channels (CHOK<sub>2</sub>11.1 cells). Two pre-recorded action potentials from paced isolated ventricular myocytes (APs, lower panel) were used as the command voltage waveform for whole cell patch clamping the CHOK 11.1 cells. The first AP in this example is followed by a second one elicited by a premature stimulus after 90% repolarization of the first AP. The resulting K<sub>v</sub>11.1 currents recorded from the CHOK, 11.1 cells closely resemble the behavior of  $I_{Kr}$  in the native myocytes during the first and second AP (K<sup>+</sup>-current, upper panel).<sup>12</sup>

As shown in Figure 5, 12 the profile of current flow (upper panel) through the channel can be split into six phases: (1) the initial phase is characterized by a slow opening and fast inactivation of the K<sub>2</sub>11.1 channel, limiting the current flow; (2) decreasing the voltage enables the channels to recover from inactivation, increasing the current flow; (3) a diminishment in current flow is determined by the slow deactivation and a decrease in driving force for potassium ions. At this phase, many of the channels are still in open states but pass little current; (4) applying a premature stimulus induces an increase in outward current; (5) the outward current has a rapid decay due to the rapid channel inactivation at depolarized potentials; (6) similar to the first phase, the large outward current in phase 6 opposes the cellular depolarization and suppresses the propagation of a premature beat. The advantage of this AP voltage clamp technique is that one can study the behavior of a single type of ion channel currents during APs without being hampered by other (often larger) ion currents that are also activated during an AP in native cells. Herein, the AP voltage clamp example implicates a mechanism foundation for other standard patch clamp recordings on the K\_11.1 channel.

The major difference between the K\_11.1 channel and other voltage-gated potassium channels is the presence of a very prominent tail current during the repolarization phase. Figure 6 shows an example for the application of a standard voltage patch clamp assay in determining the concentration-dependent effects of cisapride on the K<sub>2</sub>11.1 channel in transfected CHO cells.<sup>70</sup> From a holding potential of -80 mV, the cell is shortly clamped at -50 mV to test a leak current at this potential when all K<sub>1</sub>11.1 channels are in closed states. Since this potential is not enough to activate the K<sub>2</sub>11.1 channel, this recording is performed to correct for small background currents that do not originate from the K\_11.1 channel. The recorded current at -50 mV is referred to as a baseline step or leak current. This step is followed by a depolarization to +40 mV, where the K, 11.1 channels are alternating between the open and non-conductive, inactivated states. When repolarizing back to -50 mV, the  $I_{Kr}$  reaches its highest peak also known as the "peak tail current" due to the fast recovery of inactivated ion channels. Afterwards, the channels slowly close (de-activate). A cell responding to this voltage clamp protocol (Figure 6A) with a typical current signature shown in Figure 6B is regarded to express the K<sub>2</sub>11.1 channel.

The drug concentration required to inhibit 50% of the tail current (IC<sub>50</sub>) is usually determined by fitting the Hill equation to the compiled dose-response data. After the channels are exposed to compounds, a subsequent washout experiment is performed to characterize the reversibility of the inhibition. The patch clamp technique is able to identify the activity of moderately potent  $K_v11.1$  blockers which could otherwise have been ignored in other assays leading to false negative results. Although the patch clamp technique is considered to be the "golden standard" in investigating the affinity and kinetic parameters of drugs at the K 11.1 channel, there are several limitations to this technique. For instance, Hill et al. identified a limitation of any kinetic studies on drug blockade at the K<sub>v</sub>11.1 channel, which is the lag time associated with drug application and washout. 10 Most systems used for kinetic investigations support a lag time between 10 to 40 milliseconds that still allows recording slow time constants but causes difficulties with capturing fast time constants. Thus, drugs having low affinity and faster unblock time constants at the K<sub>y</sub>11.1 channel, demonstrate more inaccurate estimates of

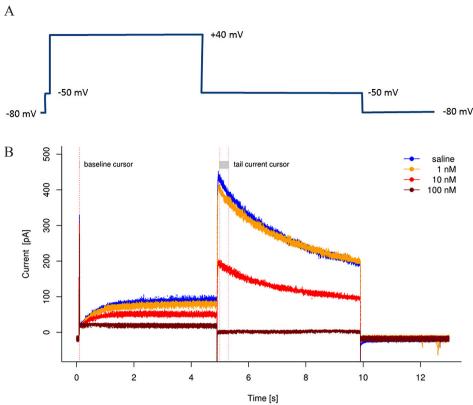


Figure 6. Example of K\_11.1 recordings showing the concentration-dependent effect (0-100 nM) of cisapride: (A) the voltage clamp protocol for  $I_{\kappa_r}$  measurements; (B)  $I_{\kappa_r}$  recordings after administration of different dosages of cisapride and saline (0 nM cisapride).<sup>70</sup>

their fast components. Additionally, this technique is low throughput and requires a skilled operator, which hamper profiling of large compound series. 71 Therefore, higher throughput assays are warranted to characterize the binding parameters of drugs candidates at the K<sub>2</sub>11.1 channel in the early phase of drug development. Recently, automated patch techniques have revolutionized the research field of experimental electrophysiology, and enabled a substantial increase in throughput without compromising the data quality. 70, 72 Moreover, the throughput of patch clamp screening can be further improved by introducing the patch clamp electrophysiology robots.<sup>70</sup>

## Radioligand binding assay

Since radioligand binding assays can provide reasonably accurate and efficient estimates of K, 11.1 affinity for compounds, they have become a different in vitro screening tools used by the pharmaceutical industries.<sup>5</sup> These assays rely on

competitive binding of test ligands with a high-affinity, radiolabelled blocker to study the affinity and kinetics of unlabeled ligands at the receptor. Up to now, radioligand binding assays at the K\_11.1 channel have been developed for several specific and potent K\_11.1 blockers including [3H]dofetilide, [3H]astemizole, [3H] MK-499 and [3H]BeKm-1. In the [3H]dofetilide binding assay, the radioligand is used in conjunction with cell lines or membrane preparations expressing the K<sub>2</sub>11.1 channel followed by incubation with the test compound which then competes for the binding site that is occupied by [3H]dofetilide. 73 This can yield the experimenter a wealth of information whether the screened compound competitively displaces [3H]dofetilide, its relative potency and, with further experimentation outlined in this thesis, the ligand's kinetics of interaction with that binding site. Diaz et al. demonstrated that the binding K, values for 22 K, 11.1 blockers obtained in a [3H]dofetilide binding assay correlated well with their functional IC<sub>50</sub> values derived from whole cell patch clamp studies.<sup>74</sup> Chiu et al. found that all their selected K 11.1 blockers inhibited the binding of [3H]astemizole in a competitive manner, and consequently, developed a [3H]astemizole binding assay that they claimed to be robust with adequate sensitivity. 75 This assay is comparable to [3H]dofetilide binding experiments in terms of measured binding affinity (K<sub>1</sub> and K<sub>2</sub> values), with only 2- to 5-fold variations. Competitive radioligand binding assays apparently provide a rapid and highly predictive tool, which can be utilized together with electrophysiological studies to identify compounds that inhibit the K\_11.1 channel. Nevertheless, the binding assays have the disadvantage of being prone to produce false negative outcomes, for instance when the examined ligand binds to the channel at an allosteric site. Additionally, these assays may also provide false positives as channel modulation may need more than compound-channel binding. Accordingly, the role of radioligand binding assays in screening the K<sub>2</sub>11.1 liability of drug candidates is still under discussion, and more researches should be conducted to improve the predictability of these assays.

#### Ion flux assay

#### Radioactive ion flux assay

Radiotracers are used to quantitate the flux of ions moving in or out of a cell through ion channels present on the cell membrane.<sup>71</sup> The radiotracers of ions are specific for each class of ion channels, and 86Rb+ is commonly used for the voltage-gated potassium channels. However, 86Rb+ flux assays are heterogeneous, and also very slow due to the inconvenient radiotracer loading and wash steps. Furthermore, the radioactive tracers can only be applied to measure the steadystate functions of ion channels. Additionally, an incomplete removal of extracellular tracers after loading or a continuous leak of the tracers out of cells results in a low signal-to-noise ratio for this kind of assays. Hence, although radioactive ion flux assays have been used for more than 30 years in ion channel investigations, their application for screening drug activity at ion channels like the K<sub>y</sub>11.1 channel has become limited.

#### Non-radioactive ion flux assay

Ion flux assays with non-radioactive ion tracers using atomic adsorbance spectrometry (AAS) have been explored for a variety of voltage-gated ion channels.<sup>71</sup> For this analysis, non-radioactive tracer ions are loaded into cells expressing the targeted ion channels (i.e. the K<sub>2</sub>11.1 channel). Ion flux is initiated after channel activation performed by depolarizing the cell membrane with a high K<sup>+</sup> concentration buffer. Then, the concentration of the tracer ions in the supernatants and/ or cells is measured, and the percentage of ion efflux (or influx) is calculated. This screening method has a moderate throughput, and it has been predominantly focused on potassium channels (e.g., Rb+ efflux assay for the K\_11.1 channel).76 The non-radioactive Rb+ efflux assay is a high-capacity method for functional analysis of the K<sub>2</sub>11.1 channel, which is particularly useful for the identification and characterization of K<sub>2</sub>11.1 blockers and activators. Since this efflux assay is a direct measure of the channel activity, it is likely not prone to disturbances that usually occur in other indirect procedures, such as false positives in radioligand binding assays. Tang et al. compared the IC<sub>50</sub> values of five known K<sub>v</sub>11.1 inhibitors (dofetilide, terfenadine, cisapride, sertindole and astemizole) obtained from both the Rb<sup>+</sup> efflux assay and patch clamp technique.<sup>77</sup> The former displayed 5- to 20-fold higher IC<sub>50</sub> values than the latter, yet the rank order of measured IC<sub>50</sub> values for these compounds was identical in these two different experiments. This illustrates the capacity of the Rb<sup>+</sup> efflux assay to rank unknown compounds that block the K\_11.1 channel by their potency, meeting the requirements for compound profiling. In addition, Chaudhary et al. used the Rb+ efflux assay to evaluate the K<sub>2</sub>11.1 liability of astemizole, E-4031, cisapride, terfenadine, risperidone, quinidine and sotalol, and compared the results to those acquired in the patch clamp technique and [3H]dofetilide binding assay.78 IC<sub>50</sub> values of test ligands gathered with the efflux assay were increased compared with those from the electrophysiological technique, while the Rb+ efflux assay was less sensitive in determining the potency of compounds but had a higher throughput to identify potent K<sub>v</sub>11.1 blockers than the radioligand binding assay. Hence, the Rb<sup>+</sup> efflux assay allows for the rapid screening of compounds in a non-invasive manner. However, this nonradioactive ion flux assay has a low temporal resolution indicative of less

information compared to the voltage clamp assay. Meanwhile, a high expression level of K\_11.1 channels is necessary to achieve an acceptable signal-to-noise ratio for this technique.<sup>79</sup> Additionally, Rb<sup>+</sup> appeared to inhibit the inactivation of the K\_11.1 channel resulting in an underestimation of potency for most drugs, which impairs its application in drug screening for K<sub>2</sub>11.1 liability.<sup>76</sup>

#### Other techniques

The electrophysiological properties of cardiomyocytes expressing various ion channels can also be investigated by micro-electrode arrays (MEAs). This novel technology is based on the integration of cardiac cell cultures or vibratome slices from cardiac tissues on microchips in order to monitor the changes in extracellular electrograms.<sup>80</sup> The detected field potential durations (FPDs) correlate well with the APDs of cardiomyocytes from patch clamp assays as well as with QT intervals in clinical ECGs, while the measured beat frequencies are in agreement with the heart rates. 81,82 Moreover, MEAs utilizing networks of cells demonstrate a higher complexity, which circumvents the danger of retrieving false positives encountered by single-cell based assays.<sup>82</sup> In addition, the non-invasive nature of MEAs allows the study of drug effects on beating rate and rhythm, and enables the detection of arrhythmia.81 However, the MEA devices are not suitable for stimulating and recording the electrophysiological parameters of single cells compared to patch clamp techniques. Furthermore, implantation of a micro-electrode array might have negative effects on the measurements due to various biological responses of cells upon the external insertion.<sup>83</sup> Altogether, screening compounds in cardiomyocytes, in particular human induced pluripotent stem cell-derived cardiac myocytes (hiPSC-CMs), with the MEAs technology is still awaiting optimization and will make a substantial contribution to safety pharmacology, which provides a promising means for preliminary drug profiling in the drug discovery pipeline. 80, 84

xCELLigence Real-Time Cell Analyzer (RTCA) Cardio System is another innovative label-free real-time platform, which can be utilized to measure the beating functions of cardiomyocytes like hiPSC-CMs based on impedance measurement.85 This assay is performed using specially designed microtiter plates with interdigitated golden electrodes to dynamically monitor the impedance changes of cardiomyocyte layers during rhythmic contractions.<sup>85, 86</sup> This new technology displays a higher throughput than patch clamping, allowing for screening the cardiotoxic liability of compounds earlier during the drug research process. Furthermore, the xCELLigence system enables the monitoring of ligand influences on cardiac functions in real time over prolonged intervals, which supports its application in revealing compounds that cause arrhythmia via inhibiting the protein trafficking of ion channels. 85, 87 In addition to such advantages, there are several challenges for the application of xCELLigence RTCA Cardio System in preclinical cardiac safety assessment.87 For example, the predictive values of proarrhythmic drugs derived from the extracted parameters in this technology are not explicit, and are required to be further elucidated in the future.

## Factors affecting the measurements

All the methods used for investigating the affinity and kinetic parameters of ligands at the K<sub>2</sub>11.1 channel are influenced by a range of external factors. Herein, we mainly focus on the impacts of assay conditions, species diversity and temperature on the data interpretation for the K\_11.1 channel.

The patch clamp technique is a highly flexible method that can be performed in a variety of configurations and modes, providing a direct, precise and detailed measurement of the activity for an ion channel.<sup>79</sup> Specific bath and pipette solutions, voltage and current protocols based on research aims are essential to obtain reliable results. For instance, Kirsch et al. found that there was a 2-fold underestimation of the potency for pimozide at the K\_11.1 channel when the step-pulse protocol was changed from 2 s to 0.5 s at 35 °C.88 In radioligand binding assays, BSA concentration in the assay buffers might considerably reduce the speed of filtration and further result in the loss of a low affinity binding component for the K<sub>v</sub>11.1 blockers. Consequently, variability amongst the results acquired in different laboratories may stem, in part, from the varied assay conditions.

Species diversity is another common factor that influences the ligands' activity at the K<sub>2</sub>11.1 channel in all techniques. Different binding and kinetic parameters of drugs at the K<sub>2</sub>11.1 channel could be observed with cells from different species or tissues. There have been reported differences in IC<sub>50</sub> values between the patch clamp measurements in Xenopus oocytes versus transfected mammalian cell cultures (i.e. HEK293 and native K\_11.1 channel in cardiomyocytes).89 Next to protocol variations, these are also attributed to the restricted drug diffusion at the oocytes surface membrane and yolk sac absorption, which does not occur in mammalian cells. Moreover, even when the K<sub>v</sub>11.1 channel is assessed in the same species but in different cell cultures, variations might be noticeable in the activity and functionality of the channel. This could be related to the varied contents or isoforms of the K, 11.1 proteins in different cell cultures. In this circumstance, tissue linearity experiments can be conducted to standardize the distinct results from different cell cultures.<sup>75</sup>

Temperature is the most important factor that impacts the interpretation of data from different assays. Physiological temperature is supposed to be more

suitable to determine drug safety at the K\_11.1 channel, since channel gating, receptor binding and drug accumulation might be temperature-dependent. In the patch clamp study the measured IC<sub>50</sub> values of sotalol and erythromycin at 35 °C were decreased by 2.9- and 7.1-fold respectively compared to these determined at room temperature. 88 Furthermore, the binding kinetics of ligands at the ion channel can be remarkably affected by the temperature. For instance, erythromycin had an extremely slow onset of block and the time required to reach half-maximal effect was 156 s at 22 °C, whereas the same compound displayed a much faster onset of block and the time needed to approach half-maximal effect was 18 s at 35 °C. 88 The incubation temperature affects the time for ion exchange in ion flux assays, 90 and also plays a paramount role in measuring the affinity and kinetic parameters of ligands in radioligand binding assays.<sup>91</sup>

## Binding of ligands and their kinetics at the K\_11.1 channel

#### Small molecules that act at the intracellular side

## Astemizole, desmethylastemizole and norastemizole

Astemizole is a second generation histamine H<sub>1</sub>-receptor antagonist. 89 In addition to antihistamine effects, it causes a QT interval prolongation and cardiac arrhythmia due to the unintended blockade of the K\_11.1 channel. Oxidative O- and N-demethylations of astemizole generate two metabolites desmethylastemizole and norastemizole, respectively. Desmethylastemizole and astemizole initiated equipotent blockade of the K\_11.1 channel, while norastemizole had weaker potency at the channel. 89 Taglialatela et al. investigated the concentration-dependent effect of astemizole on Xenopus oocytes transfected with the K<sub>2</sub>11.1 channel by an extracellular perfusion, and calculated its IC<sub>50</sub> as 480 nM.<sup>92</sup> This value was 10 times higher than that previously reported by Suessbrich et al., 93 which was explained by a longer incubation time in the latter study leading to higher intracellular drug concentrations and thus stronger potency. Binding of the parent drug and its metabolites occurs primarily in the open and inactivated states of the channel with a minimal block in the closed state.89 Blockade of astemizole and desmethylastemizole occurred faster at higher action frequencies. Recovery from astemizole blockade was minimal, while a partial recovery for desmethylastemizole was observed after a washout period of 30 minutes. However, the inhibition of norastemizole was incomplete and only occurred at high concentrations after channel activation. Astemizole and norastemizole were found to bind in the open state of the channel in the region of the hydrophobic inner vestibule after the assembly of the channel tetramers. 94 A mutation to cysteine at the F656 pore

residue lining in the inner vestibule dramatically reduced the blockade of these two compounds.

## Clozapine

Hill et al. performed the first study on directly measuring the binding and unbinding kinetics of the typical antipsychotic clozapine at the K\_11.1 channel expressed in CHO cells using the patch clamp technique with a Dynaflow automated compound delivery system. 10 They revealed two kinetically distinct blocking effects consistent with different binding kinetics to the open and inactivated states of the channel, while the binding affinity of clozapine for the two different states were almost identical. The block and unblock time courses of clozapine were not monophasic, which encompassed a fast and a slow phase. The fast component of block was related to the fast component of unblock, and vice versa. The current recovery rate from clozapine block during the washout experiment was independent of the drug dosages applied in the system, whereas it was exposure time dependent with a slower recovery rate after a longer application of the drug. The molecular mechanism of clozapine blockade at the K\_11.1 channel has not been elucidated. However, it is believed that clozapine binds at the pore site towards the lumen on residues Tyr652 and Phe656 of the S6 domain, which is comparable to other prototypical K<sub>2</sub>11.1 blockers.<sup>95</sup>

## Clofilium and ibutilide

Clofilium and ibutilide, which are two structurally related class III antiarrhythmic drugs, were found to have different time-dependent kinetics of block at the K\_11.1 channel in whole cell voltage clamped *Xenopus* oocytes. 18 The binding potency of ibutilide at the channel showed an IC<sub>50</sub> value of 28 nM, while a precise measurement of the IC<sub>50</sub> value for clofilium was difficult due to its extraordinarily slow onset of block. The steady-state inhibition of clofilium to the K<sub>2</sub>11.1 channel was not approached even after a prolonged exposure to more than 1 hour. Nevertheless, the onset of block was reported to be faster when clofilium was applied from the intracellular side of excised membranes in excised patch clamp experiments. Regarding the unblock rate, the inhibition of ibutilide was almost completely recovered after a 10-minute washout, whereby only 19%  $I_{Kr}$  was restored from clofilium block within the same washout period. This utra-slow recovery rate of clofilium from K<sub>2</sub>11.1 blockade is similar to the findings for other potent channel blockers which demonstrate slow unbinding profiles because of drug trapping within the central cavity of the channel after closure of the activated gate upon membrane repolarization.<sup>96, 97</sup> To investigate the trapping hypothesis,

both clofilium and ibutilide were tested at the K\_11.1 D540K mutant that permits untrapping at a hyperpolarized state by allowing the gate to remain open. <sup>18</sup> Again, recovery from K\_11.1 blockade in the case of ibutilide was almost complete, yet little current recovery from clofilium inhibition to the channel was observed. This ruled out the possibility of drug tapping within the channel cavity, also known as a "foot in the door" mechanism, for clofilium. The binding sites of clofilium and ibutilide at the K<sub>2</sub>11.1 channel were the same by alanine-scanning mutagenesis of Tyr652, Phe656, Thr623, Ser624 and Val625 residues. The S624A mutant located at the base of the pore helix of the K\_11.1 channel was the only one that was found to enable the rapid  $I_{Kr}$  recovery from clofilium blockade.

#### **Terfenadine**

Terfenadine is a potent blocker of the K<sub>v</sub>11.1 channel in the open state. The IC<sub>50</sub> of terfenadine in Xenopus oocytes expressing the K, 11.1 channel was 350 nM at 22 °C.93 Extracellular potassium concentrations significantly altered the K<sub>2</sub>11.1 potency of terfenadine with an IC<sub>50</sub> of 3.5 mM at 96 mM K<sup>+</sup> reduced to 350 nM in 2 mM K<sup>+</sup>.98 The steady-state block of the K\_11.1 channel by terfenadine was approached within 7 minutes superfusion of Ringer's solution containing terfenadine, and the blockade was frequency-, voltage- and use-independent. The recovery of terfenadine from block at the closed state took place after a washout period of more than 30 minutes, which can be explained by drug trapping inside of the central cavity of the K<sub>2</sub>11.1 channel.<sup>99</sup> Unlike clofilium, the closure-deficient K<sub>2</sub>11.1 D540K mutant was experimentally confirmed to facilitate the  $I_{v_r}$ recovery from terfenadine block at the K<sub>2</sub>11.1 channel. Focusing on the molecular binding perspective, terfenadine interacts with S624, Y652, F656 and T632 residues, while amino acids Y652 and F656 are the most important determinants for the K<sub>w</sub>11.1 sensitivity to terfenadine.<sup>99</sup>

#### Cisapride

Cisapride, a prokinetic agent, is a potent K<sub>v</sub>11.1 blocker.<sup>99</sup> The IC<sub>50</sub> value of cisapride at the K<sub>2</sub>11.1 channel on the *Xenopus* oocytes was 630 nM. The onset of block of cisapride was rapid and the recovery from block was complete within approximate 2 minutes.<sup>100</sup> Mutation studies for the binding site of cisapride at the K<sub>2</sub>11.1 channel implicated that  $\pi$ - $\pi$  interactions with Y652 and F656 residues and additional hydrophobic interactions with Y652 were essential for the channel sensitivity to cisapride. 101

#### Propafenone

Witchel et al. used a concentration of 50 mM propafenone to induce a profound K\_11.1 channel blockade. 102 The tail current was blocked by 96% at -70 mV while the blockade dropped to 5% in the F656A mutant in Xenopus oocytes. This indicates that the F656 residue is a critical determinant for the blockade of propafenone at the K, 11.1 channel, similar to other methanesulfonanilides like dofetilide and MK-499. Recovery of the K\_11.1 channel from propafenone blockade was very rapid in the open state, whereas it was virtually undetectable in the time scale tested in the closed state. A further study performed by Windisch et al. revealed that propagenone derivatives with short side chains were trapped in the closed channel and displayed a very slow unblock rate at the K<sub>2</sub>11.1 channel, which was in contrast to the long side chain analogues displaying insignificant trapping. 96 Altogether, the study provides direct evidence for a "foot in the door" type blockade of ligands at the K\_11.1 channel.

#### Peptides that act at the extracellular side

#### APETx1 and Saxitoxin

APETx1 is a peptide purified from sea anemone, which acts as a gate modifier of the  $K_v 11.1$  channel. APETx1 inhibited the  $I_{Kr}$  in a heterologous system with an IC<sub>50</sub> value of 34 nM at room temperature, while the inhibition was maximal at 300 nM leaving 20% current unblocked even at concentrations up to 1 µM in COS-7 cells transfected with K<sub>2</sub>11.1 channels. Its kinetics of block profile were characterized by a rapid onset of block, and the steady-state inhibition was reached within 1 minute. The block of APETx1 to the K\_11.1 channel was totally reversible upon a washout period of 100 s. Similar to the binding sites for gating modifier toxins of other voltage-gated potassium channels, mutations in the S3 region of the K<sub>2</sub>11.1 channel like G514C and E518C had dramatic impact on the responsiveness of the channel to APETx1.<sup>104</sup>

Saxitoxin inhibited the  $I_{Kr}$  by more than 90% at 3  $\mu$ M in HEK293 cells expressing the  $K_v 11.1$  channel at  $36 \pm 1$  °C. <sup>105</sup> Current reduction reached steady state quickly, and the block was reversible upon washout with dissociation characteristics opposing the simple first-order process. The concentration-dependency of saxitoxin alongside its kinetics of block onset and recovery suggested a complex binding of multiple molecules of saxitoxin to the extracellular side of the K\_11.1 channel. Co-administration of methanesulfonanilide MK-499 showed that the K 11.1 inhibition effects of the two drugs were additive, further indicating that saxitoxin did not bind to the intracellular side of the channel as MK-499.

#### BeKm-1

BeKm-1 is a scorpion toxin, which selectively blocked the  $I_{\rm Kr}$  current with an IC<sub>50</sub> value of 3.3 nM in a patch clamp study in transfected HEK93 cells at room temperature. 106 The current block and unblock rates of wile type BeKm-1 at the K 11.1 channel were 2.5 ×10<sup>6</sup> M<sup>-1</sup>·s<sup>-1</sup> and 0.017 s<sup>-1</sup>, respectively. <sup>107</sup> In radioligand binding assays the association and dissociation kinetics of [125I]-BeKm-1 were extremely fast at room temperature, while its on- and off-rate constants at 0 °C were 3.8×107 M<sup>-1</sup>·s<sup>-1</sup> and 0.00051 s<sup>-1</sup>, separately.<sup>107</sup> Additionally, wild-type BeKm-1 displaced the binding of [125I]-BeKm-1 with a half-maximal inhibitory concentration of 44 pM at 0 °C. 107 Collectively, discrepancies exist between quantitative results acquired in patch clamp and radioligand binding assays, albeit to the comparable rank orders of measured parameters from these two different techniques. BeKm-1 preferentially inhibited the  $I_{\rm Kr}$  current through a closed state channel blockade mechanism, 108 and interacted with the K<sub>v</sub>11.1 channel at the extracellular site.42

## **Conclusions**

The K<sub>2</sub>11.1 channel is a voltage-gated potassium channel that shares structural characteristics with the voltage-gated potassium channel family but displays unique gating kinetics. It is involved in a number of physiological conditions, and has the potential to become a new biomarker and therapeutic target for cancer, schizophrenia and other important diseases due to its high expression in diseased cells instead of normal cells. More importantly, the K\_11.1 channel plays an important role in maintaining cardiac stability under normal conditions being the target of class III antiarrhythmic drugs, but at the same time, when dysregulated, causes notorious side effects of severe arrhythmias. A surprisingly diverse group of drugs including cardiac and non-cardiac medications have been withdrawn from or restricted for use in the market due to their blockade of the K\_11.1 channel. Therefore, it is routine practice for pharmaceutical companies to screen the K<sub>2</sub>11.1 liability of drug candidates before they are approved to enter the market by regulatory agencies.

In this review, we summarized the generation, degradation, structure and function of the K\_11.1 channel, and its corresponding intentional and unintentional roles in drug development. This provides valuable information and theoretical foundations for investigations on the K<sub>2</sub>11.1 channel as well as a comprehensive overview of the channel.

The unique gating kinetics of the K<sub>v</sub>11.1 channel contribute to its important

functions in physiological conditions. Almost all K 11.1 ligands affect the gating kinetics of the channel during their interaction with the channel. We reviewed the closed, open and inactivated states of the channel and also the transitions among these different states. Moreover, the impact of several assay conditions and K<sub>v</sub>11.1 ligands on the channel kinetics was discussed. This emphasizes the importance of considering the gating kinetics of the K<sub>2</sub>11.1 channel in seeking for K\_11.1 activators and compounds with less K\_11.1-related cardiotoxicity. In addition, the differences between laboratories in evaluating the K<sub>v</sub>11.1 ligands should be standardized by taking distinct assay conditions into consideration.

Currently, the estimation of the risk of a drug to cause LQTS is only based on its IC<sub>50</sub> value at the K<sub>y</sub>11.1 channel. When a drug or drug candidate exhibits an IC<sub>50</sub> value 30-fold higher than its effective therapeutic plasma concentration, this drug is considered to have a safe cardiac profile. Although this rule ensures the exclusion of potentially harmful drugs, it could also lead to the preliminary withdrawal of certain drug candidates that may be unproblematic. Furthermore, drug inhibition of the K 11.1 channel is a dynamic process in which binding and unbinding events occur over time. Therefore, we also examined in this review the binding and corresponding kinetics of different series of ligands to the K 11.1 channel. Different techniques used to determine the binding and kinetic parameters of these ligands at the channel were summarized as well. All in all, we strongly plea for incorporation of kinetic assessment into ligand affinity at the channel in order to better predict the proarrhythmic propensities of K\_11.1 blockers. To date, the roles of binding (association) and unbinding (dissociation) of ligands with respect to their K<sub>y</sub>11.1 cardiac side effects have not been well elucidated, and thus, more attention should be paid to the studies on these kinetics in the future.

Overall, a thorough analysis of the K<sub>2</sub>11.1 channel, in particular the gating kinetics and ligand binding kinetics, enables the improvement of drug targeted design and reduction of unwanted arrhythmic side effects on this channel. The outcome of these investigations would also substantially contribute to the development of homology and mathematical models for better predicting the interactions of K 11.1 ligands with the channel, which could effectively facilitate the investigations on the K<sub>1</sub>11.1 channel in the near future.

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## Chapter 3

# Allosteric modulators of the K<sub>v</sub>11.1 (hERG) channel

Radioligand binding assays reveal allosteric characteristics of dofetilide analogues

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#### **Abstract**

Drugs that block the cardiac K, 11.1 channel encoded by the human ether-à-gogo gene (hERG) have been associated with QT interval prolongation leading to proarrhythmia, and in some cases, sudden cardiac death. Because of special structural features of the K<sub>1</sub>11.1 channel, it has become a promiscuous target that interacts with pharmaceuticals of widely varying chemical structures and a reason for concern in the pharmaceutical industry. The structural diversity suggests that multiple binding sites are available on the channel with possible allosteric interactions between them. In the present study, three reference compounds and nine compounds of a previously disclosed series were evaluated for their allosteric effects on the binding of [3H]astemizole and [3H]dofetilide to the K<sub>2</sub>11.1 channel. LUF6200 was identified as an allosteric inhibitor in dissociation assays with both radioligands, yielding similar EC<sub>50</sub> values in the low micromolar range. However, potassium ions increased the binding of the two radioligands in a concentration-dependent manner, and their EC<sub>50</sub> values were not significantly different, indicating that potassium ions behaved as allosteric enhancers. Furthermore, addition of potassium ions resulted in a concentration-dependent leftward shift of the LUF6200 response curve, suggesting positive cooperativity and distinct allosteric sites for them. In conclusion, our investigations provide evidence for allosteric modulation of the K<sub>2</sub>11.1 channel, which is discussed in the light of findings on other ion channels.

## Introduction

The K<sub>2</sub>11.1 channel, encoded by the hERG gene<sup>1</sup>, is responsible for the rapid delayed rectifier  $K^+$  current  $(I_{\kappa})$  that plays a critical role in the repolarization of cardiomyocytes during the cardiac action potential<sup>2</sup>. It is made up of large intracellular N- and C-terminal domains<sup>3, 4</sup> and four identical α-subunits, each of which is formed by six  $\alpha$ -helical transmembrane domains and a looping "pore region"5,6. In humans, blockade of the K\_11.1 channel by drugs can cause excessive lengthening of the action potential, which is reflected by a OT interval prolongation in the electrocardiogram (ECG)<sup>7,8</sup>. The excessive action potential prolongation may combine to produce and sustain Torsade de Pointes (TdP), which can be self-limiting or degenerate into ventricular fibrillation rapidly leading to death<sup>6,8</sup>. Therefore, it has become a routine practice in the pharmaceutical industry to test compounds for their K\_11.1 liability during early preclinical safety assessments according to FDA guidelines9. In recent years, most attention has been paid to assess the affinity for the K<sub>2</sub>11.1 channel of potential drug candidates in order to avoid and discard modest-to-high affinity compounds during the lead finding and optimization process<sup>10</sup>. However, allosteric modulation of the K<sub>2</sub>11.1 channel as an alternative way of interaction has not been studied in any details until now.

With regard to G protein-coupled receptors (GPCRs), successful drugs that mediate their effects through the allosteric modulation of target activity have already reached the market, also in view of their potential greater selectivity, potency and/or safety profile when compared to orthosteric ligands<sup>11, 12</sup>. Radioligand binding assays, in particular kinetic radioligand dissociation assays, have been widely utilized to quantify the allosteric effects of GPCR ligands<sup>13</sup>. As for ion channels, allosteric modulators have been reported for ligand-gated ion channels in particular. For example, GW791343 might be applied to treat inflammatory disorders and pain due to its allosteric inhibition of the P2X<sub>7</sub> receptor<sup>14, 15</sup>. A negative allosteric modulator of the nicotinic acetylcholine receptor, UCI-30002, had been reported to have significant benefits as a strategy for treating nicotine addiction because of its high subtype-selectivity<sup>16</sup>. It has also become increasingly apparent that new potent and selective allosteric modulators of the GABA-A receptor and 5-HT3 receptor may replace conventional antagonists and agonists for these targets<sup>17, 18</sup>. Allosteric modulation of voltage-gated ion channels such as several types of calcium, sodium and potassium channels and their possible clinical applications has been investigated as well, but to a lesser extent. As an example, a novel quinazolinone ligand (TTA-Q4) showed a positive allosteric interaction with the T-Type calcium channel since it enhanced radioligand binding, increased affinity in a saturable manner and slowed dissociation<sup>19</sup>.

Figure 1. Compounds evaluated in this study. Terfenadine is a high-affinity reference K\_11.1 blocker, fluvoxamine is a low-affinity reference K\_11.1 blocker, PD118057 is a reference K\_11.1 activator, and all the other LUF compounds are members of a previously disclosed compound series interacting with the K\_11.1 channel.

Radioligand binding assays as a means of studying the K<sub>v</sub>11.1 channel have been developed over the years. Two radioligands, [3H]astemizole and [3H]dofetilide, have mostly been used<sup>20-22</sup>. The purpose of the present study was to identify and investigate allosteric modulators of the K<sub>2</sub>11.1 channel using these two radioligands. The anti-depressant fluvoxamine<sup>23</sup> and the channel opener PD118057<sup>24</sup>, <sup>25</sup> (**Figure 1**), reported to have binding sites different from conventional  $K_v$ 11.1 blockers, were selected as representative reference compounds. We identified a number of dofetilide analogues (Figure 1) displaying allosteric modulation in a pilot experiment and these were further investigated as were the allosteric effects of potassium ions in the same binding assays. Our findings suggest potential applications for such allosteric modulators, which might provide novel solutions for drug cardiotoxicity due to blockade of the K\_11.1 channel.

## Materials and methods

## Chemicals and reagents

Astemizole, terfenadine, fluvoxamine and PD118057 were purchased from Sigma Aldrich (Zwijndrecht, The Netherlands). Dofetilide and all the LUF compounds were synthesized in our own laboratory, as published previously<sup>26</sup>. [<sup>3</sup>H]

Astemizole (specific activity 78.9 Ci·mmol<sup>-1</sup>) and [<sup>3</sup>H]dofetilide (specific activity 70.0 Ci·mmol<sup>-1</sup>) were purchased from Perkin Elmer (Groningen, The Netherlands). Bovine serum albumin (BSA, fraction V) was purchased from Sigma (St. Louis, MO). G418 was obtained from Stratagene (Cedar Creek, USA). All the other chemicals were of analytical grade and obtained from standard commercial sources. HEK293 cells stably expressing the K\_11.1 channel (HEK293K\_11.1) were kindly provided by Dr. Eckhard Ficker (University of Cleveland, USA).

## Cell culture

HEK293K 11.1 cells were cultured in a humidified atmosphere at 37 °C and 7% CO<sub>2</sub> in DMEM, containing 10% fetal calf serum, 50 IU mL<sup>-1</sup> penicillin, 50 μg·mL<sup>-1</sup> streptomycin and 1.25 μg·mL<sup>-1</sup> G418 for selection. Cells were subcultured twice a week (1:8). Then, the cells were subcultured 1:10 and transferred to large 15-cm diameter plates for membrane preparation.

## Membrane preparation

HEK293K\_11.1 cells were grown to 80-90% confluence and detached from the plates by scraping them into 5 mL of PBS. Then, the detached cells were collected and centrifuged at 250 g for 10 min. The cell pellets were pooled and resuspended in 50 mM ice-cold Tris-HCl buffer containing 2 mM MgCl<sub>2</sub>, pH 7.4. An UltraTurrax (Heidolph Instruments, Schwabach, Germany) was used to homogenize the cell suspension. Membranes and the cytosolic fraction were separated by centrifugation at 100, 000 g in an Optima LE-80K ultracentrifuge (Beckman Coulter, Fullerton, CA) at 4 °C for 20 min. The pellets were resuspended in the Tris-HCl buffer, and the homogenization and centrifugation steps were repeated. The resulting pellets were resuspended in ice-cold incubation buffer (10 mM HEPES, 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl,, 1 mM EGTA, 10 mM glucose, 0.1% BSA, pH 7.4) using the UltraTurrax. Aliquots (125 or 250 µL) were stored at -80 °C. The protein concentration of the membranes was measured using the BCA method<sup>27</sup>.

## Kinetic radioligand association and dissociation assays

The incubation temperature and protein concentration were optimized for the kinetic studies. Three different temperatures (4, 15 and 25 °C) and five protein amounts (10, 15, 20, 30 and 50 µg) were investigated at the beginning of our research program. The optimal conditions for [3H]astemizole kinetic studies proved to be using 30 µg of membrane protein at an incubation temperature of 15 °C. The

association experiments of [3H]astemizole were performed by incubating membrane aliquots containing 30 ug protein in a total volume of 100 uL incubation buffer at 15 °C for 90 min with 2 nM [3H]astemizole. The amounts of radioligand bound to the receptor were measured at various time intervals during the incubation. The dissociation experiments were conducted by preincubating membrane aliquots containing 30 µg protein in a total volume of 100 µL incubation buffer at 15 °C for 90 min with 2 nM [3H]astemizole. After preincubation, dissociation was initiated by addition of 10 µM astemizole for a total period of 60 min. The amounts of [3H]astemizole still bound to the receptor were also measured at various time intervals. Separation of bound from free radioligand was performed by rapid filtration through Whatman GF/B filters using a Brandel harvester. Filters were subsequently washed six times with 2 mL ice-cold wash buffer (25 mM Tris-HCl, 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl,, 0.05 mM CaCl,, 0.05% BSA, pH 7.4). Filter-bound radioactivity was determined by scintillation spectrometry using a liquid Scintillation Analyzer (Tri-Carb 2900TR) after addition of 3.5 mL Packard Emulsifier Safe and 2 h extraction.

For the kinetic studies of [3H]dofetilide binding we used 20 ug of membrane protein at an incubation temperature of 25 °C with 5 nM [3H]dofetilide. The association experiments were performed for 120 min, whereas the dissociation assays were induced by 10 µM dofetilide after a total period of 120 min preincubation and followed for another 60 min. Incubations were terminated, and samples were obtained and analyzed as described for [3H]astemizole.

## Kinetic radioligand dissociation assays-allosteric modulation

## Single point dissociation assays

Dissociation experiments were mainly performed as described in the above paragraph. After preincubation, dissociation was initiated by addition of 10 µM astemizole or dofetilide in the absence (control) or presence of 10 µM of other ligands. After 10 min of dissociation, samples were separated by rapid filtration through a 96-well GF/B filter plate using a Perkin Elmer Filtermate-harvester (Perkin Elmer, Groningen, The Netherlands). Filters were subsequently washed twelve times with ice-cold wash buffer. The filter-bound radioactivity was determined by scintillation spectrometry using the 1450 Microbeta Wallac Trilux scintillation counter (Perkin Elmer) after addition of 37.5 µL Microscint and 2 h extraction. The binding of [3H]astemizole or [3H]dofetilide in the control was set as 100%.

#### **Dissociation assays**

After preincubation, dissociation was initiated by addition of 10 µM astemizole or dofetilide in the absence (control) or presence of 10 µM of possible allosteric modulators for a total period of 60 min. The amounts of radioligand still bound to the receptor were measured at various time intervals. Incubations were terminated, and samples were obtained as described in "Single point dissociation assays".

#### Concentration dependency of allosteric modulators

In this setup the dissociation assay of [3H]dofetilide was performed at 15 °C and the preincubation time to achieve radioligand association was 180 min; the dissociation assay for [3H]astemizole was the same as described in "Kinetic radioligand association and dissociation assays". After preincubation, dissociation was initiated by addition of 10 µM astemizole or dofetilide in the absence (control) or presence of different concentrations of allosteric modulators. Incubations were terminated after 10 min of dissociation, and samples were obtained as described in "Single point dissociation assays".

## Equilibrium radioligand binding assays

The [3H]astemizole and [3H]dofetilide equilibrium binding assays for the K\_11.1 channel were performed as described before<sup>20-22, 28</sup> with minor modifications. In short, membrane aliquots containing 30 µg protein for [3H]astemizole or 20 µg protein for [3H]dofetilide were incubated in a total volume of 100 µl incubation buffer at 25 °C for 60 min. Binding experiments were performed using a series of concentrations of potassium ions in the presence of 2 nM [3H]astemizole or 5 nM [3H]dofetilide. Nonspecific binding was determined in the presence of 10 μM astemizole and represented approximately 15% of the total binding. [3H]astemizole and [3H]dofetilide did not show specific binding to membranes prepared from HEK293 cells lacking the expressed K\_11.1 channel. The binding of [3H] astemizole and [3H]dofetilide in the presence of 100 mM KCl was set at 100% in all experiments, whereas nonspecific binding was set at 0%. Incubations were terminated by dilution with ice-cold wash buffer, and samples were obtained as described in "Kinetic radioligand association and dissociation assays".

## Cooperativity between different allosteric modulators

The dissociation assay of [3H]dofetilide in the presence of different concentrations of KCl (2, 5, 10 and 60 mM) was performed as described in "Concentration dependency of allosteric modulators". After preincubation, dissociation was initiated by addition of 10 µM dofetilide in the absence (control) or presence of different concentrations of LUF6200. Incubations were terminated after 6 min of dissociation, and samples were obtained as described in "Single point dissociation assavs".

### Data analysis

All data of radioligand binding assays were analyzed with Prism v. 5.0 (Graph-Pad, San Diego, CA, USA). EC<sub>50</sub> values for potassium ions were directly obtained from the non-linear regression concentration-effect curves. Dissociation rate constants, k<sub>off</sub>, were obtained by computer analysis of the exponential decay of [3H] astemizole or [3H]dofetilide bound to the receptor. Association rate constants, k, were calculated according to the equation  $k_{on} = (k_{obs} - k_{off})/[L^*]$ , where the  $k_{obs}$  was obtained by computer analysis of the exponential association of [3H]astemizole or [3H]dofetilide bound to the receptor and [L\*] was the amount of radioligand used for the association experiments. EC<sub>50</sub> values from the dissociation experiments were calculated by non-linear regression analysis of concentration-effect curves of dissociation in the presence of different concentrations of unlabeled ligands. The differences of binding parameters between compounds and control were analyzed by an unpaired student t test. Significance was determined as P < 0.05. All values obtained are means of at least three independent experiments performed in duplicate.

## Results

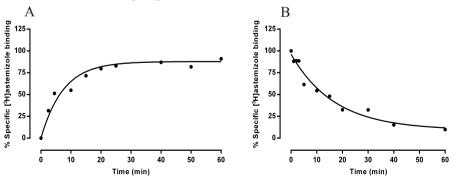
## Optimization of assay conditions for [3H]astemizole and [3H] dofetilide kinetic study

Assay conditions were optimized according to a general radioligand binding protocol in our laboratory<sup>29</sup>. Firstly, displacement assays for [<sup>3</sup>H]astemizole and [3H]dofetilide were performed at five different amounts of HEK293K\_11.1 membranes (10, 15, 20, 30 and 50 µg). A suitable window of specific [3H]astemizole binding was obtained using 30 µg of membrane protein, whereas 20 µg of protein was required for [3H]dofetilide binding to achieve a similar window. These membrane protein concentrations were also selected for the further kinetic study of two radioligands. Then, since it has been reported that compounds like AICAR or phenformine can also inhibit the K<sub>v</sub>11.1 currents via activating AMP-dependent protein kinase<sup>30</sup>, the binding of [<sup>3</sup>H]astemizole and [<sup>3</sup>H]dofetilide at membranes

prepared from HEK293 cells lacking the expressed K<sub>v</sub>11.1 channel were determined. Neither of the two radioligands showed any specific binding here, indicating the direct binding of astemizole and dofetilide to the K<sub>v</sub>11.1 channel. Subsequently, the association and dissociation of [³H]astemizole and [³H]dofetilide were determined at three different temperatures (4, 15 and 25 °C). The optimal incubation temperature was found to be 15 °C for [³H]astemizole and 25 °C for [³H]dofetilide concerning the appropriate on- and off-rates and the practical convenience of the experiments. Nevertheless, these findings were not applied too stringently as concentration-dependent effects of allosteric modulators of [³H] astemizole and [³H]dofetilide binding to the K<sub>v</sub>11.1 channel were later conducted at 15 °C and concentration-effect curves for K<sup>+</sup> ions were recorded at 25 °C in order to obtain comparable results between the two radioligand binding assays.

## Association and dissociation of [3H]astemizole from the HEK-293K,11.1 membranes

Because association and dissociation of [ $^3$ H]astemizole from the HEK293K $_v$ 11.1 membranes were both very fast at 25 °C, the kinetic study of [ $^3$ H]astemizole binding was performed at 15 °C using 30 µg HEK293K $_v$ 11.1 membrane protein. The data were best described by non-linear regression using a mono-exponential model. As is shown in **Figure 2A** and **Table 1**, the binding equilibrium was reached within 90 min with a k $_{on}$  value of  $0.068 \pm 0.010$  nM $^{-1}$  min $^{-1}$  The dissociation rate and half-life time of [ $^3$ H]astemizole were  $0.055 \pm 0.010$  min $^{-1}$  and 11  $\pm$  1.4 min, respectively (**Figure 2B** and **Table 1**). The dissociation of [ $^3$ H]astemizole at 10 min was almost half of the total binding, so the 10-min dissociation time was selected for assessing the allosteric effects of compounds on the K $_v$ 11.1 channel in the follow-up experiments.



**Figure 2.** Association and dissociation curves of [ $^{3}$ H]astemizole binding to the K $_{v}$ 11.1 channel. (**A:** Association curve; **B:** Dissociation curve) Dissociation was started by addition of 10  $\mu$ M astemizole after 90 min preincubation. Experiments were performed at 15

°C using 30 µg of HEK293K, 11.1 membrane protein.

**Table 1.** The association rate  $(k_{on})$ , dissociation rate  $(k_{off})$  and the kinetically derived  $K_d$  of [3H]astemizole and [3H]dofetilide binding to the K\_11.1 channel.

	k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> )	k <sub>off</sub> (min <sup>-1</sup> )	K <sub>d</sub> (nM)
Astemizolea	$0.068 \pm 0.010$	$0.055 \pm 0.010$	$0.85 \pm 0.045$
<b>Dofetilide</b> <sup>b</sup>	$0.032 \pm 0.0026$	$0.20\pm0.026$	$6.4 \pm 1.3$

Values are means (± S.E.M.) of at least three independent assays performed in duplicate. <sup>a</sup>The association and dissociation assays of [<sup>3</sup>H]astemizole were performed at 15 °C.

## Association and dissociation of [3H]dofetilide from the HEK-293K,11.1 membranes

As is depicted in **Figure 3**, the association of [<sup>3</sup>H]dofetilide reached equilibrium at around 30 min, and it completely dissociated from the K 11.1 channel within 60 min after addition of 10 μM dofetilide. By analyzing association and dissociation curves using a mono-exponential model, the k<sub>op</sub>, k<sub>off</sub> and t<sub>1/2</sub> values were  $0.032 \pm 0.0026 \text{ nM}^{-1} \cdot \text{min}^{-1}$  and  $0.20 \pm 0.026 \text{ min}^{-1}$ , respectively (**Table 1**). Similar to [3H]astemizole binding assay, a 10-min dissociation time point was chosen for the further study of allosteric modulators of the K<sub>y</sub>11.1 channel.

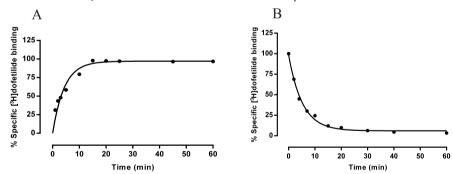


Figure 3. Association and dissociation curves of [3H]dofetilide binding to the K\_11.1 channel. (A: Association curve; B: Dissociation curve) Dissociation was started by addition of 10 µM dofetilide after 120 min preincubation. Experiments were performed at 25 °C using 20 µg of HEK293K, 11.1 membrane protein.

<sup>&</sup>lt;sup>b</sup>The association and dissociation assays of [<sup>3</sup>H]dofetilide were conducted at 25 °C.

## Screening for allosteric modulators of [3H]astemizole and [3H] dofetilide binding

**Table 2.** The specific binding of [3H]astemizole and [3H]dofetilide to the K\_11.1 channel after 10 min of dissociation in the absence ( $B_{control}$ ) or presence of 10  $\mu M$  unlabelled ligands (B).

Compound	%B/B <sub>control</sub> of [ <sup>3</sup> H]astemizole	%B/B <sub>control</sub> of [ <sup>3</sup> H]dofetilide
Control	$100 \pm 0$	$100 \pm 0$
Terfenadine	$98 \pm 1$	$93 \pm 7$
Fluvoxamine	129 ± 13 *	$95 \pm 12$
PD118057	133 ± 14 *	109 ± 2 *
LUF6145	$109 \pm 5$	$93 \pm 2$
LUF6171	$102 \pm 6$	$104 \pm 3$
LUF6173	$97 \pm 8$	71 ± 5 *
LUF6200	39 ± 4 ***	50 ± 1 ***
LUF6202	70 ± 2 ***	62 ± 4 *
LUF6206	86 ± 2 ***	70 ± 7 *
LUF6208	36 ± 6 ***	47 ± 5 **
LUF6209	39 ± 2 ***	43 ± 8 *
LUF6217	53 ± 4 ***	54 ± 5 *

Values are means (± S.E.M.) of three independent assays performed in duplicate. (\* P < 0.05; \*\* P < 0.01; \*\*\* P < 0.001 versus control)

We screened three reference ligands and an in-house library of K 11.1 blockers (LUF compounds) to identify their allosteric effects on [3H]astemizole binding to the K\_11.1 channel using a single point dissociation assay. This protocol resulted in several compounds that potentially increased (allosteric inhibitors) and decreased (allosteric enhancers) the dissociation of the radioligand. All the data are listed in Table 2 and shown in Figure 4. Fluvoxamine and PD118057 appeared to increase the specific binding of [3H]astemizole by 29 and 33% above control after 10 min, suggesting that they might behave as allosteric enhancers of [3H] astemizole binding. On the contrary, some of the compounds synthesized in our laboratory including LUF6200, LUF6208, LUF6209 and LUF6217 substantially reduced the binding of [3H]astemizole, supporting their role as allosteric inhibitors by accelerating the dissociation of the radioligand. Among them, LUF6200, 6208 and LUF6209 were the most potent allosteric inhibitors with only  $39 \pm 4, 36$  $\pm$  6 and 39  $\pm$  2% specific binding left compared with 100% for the control. In addition, LUF6202 and LUF6206 also displayed significant allosteric inhibition effects, but to a lesser extent. Other compounds (terfenadine, LUF6145, LUF6171 and LUF6173) had no significant allosteric modulatory effects in this experiment.

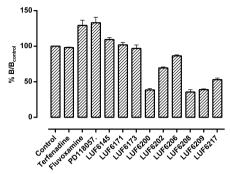


Figure 4. Bar graphs of the percentage specific binding of [3H]astemizole to the K\_11.1 channel after 10 min of dissociation induced by 10 µM astemizole in the absence (control) or presence of 10 µM of different compounds. The specific binding of [3H]astemizole in the absence of compounds was set as 100%. Experiments were performed at 15 °C using 30 µg HEK293K 11.1 membrane protein.

The effects of the same compounds in the single point dissociation experiments of [3H]dofetilide binding to the K 11.1 channel are shown in Figure 5 and **Table 2**. The results obtained from this assay were almost the same as those derived from the [3H]astemizole binding assay. The only two exceptions were that fluvoxamine exerted no allosteric enhancement here and that LUF6173 now showed significant allosteric inhibition of [3H]dofetilide binding.

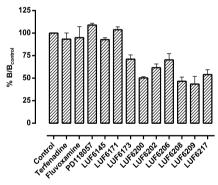
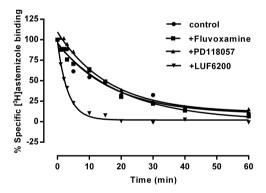


Figure 5. Bar graphs of the percentage specific binding of [3H]dofetilide to the K\_11.1 channel after 10 min of dissociation induced by 10 µM dofetilide in the absence (control) or presence of 10 µM of different compounds. The specific binding of [3H]dofetilide in the absence of compounds was set as 100%. Experiments were performed at 25 °C using 20 μg HEK293K 11.1 membrane protein.

Subsequently, fluvoxamine, PD118057 and one potent representative allosteric inhibitor (LUF6200) were selected for the further experiments to validate their allosteric modulation effects.

## Dissociation of [3H] astemizole from the K\_11.1 channel: effects of fluvoxamine, PD118057 and LUF6200

The dissociation of [3H]astemizole from the the K\_11.1 channel was induced by 10 μM astemizole in the absence (control) or presence of 10 μM of fluvoxamine, PD118057 or LUF6200 during a 60-min time course at 15 °C. The results are depicted in Figure 6. The dissociation rates of [3H]astemizole in the presence of fluvoxamine and PD118057 were  $0.040 \pm 0.0091 \text{ min}^{-1}$  and  $0.055 \pm 0.0053$ min-1, respectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which were not significantly different from the control (koeffectively, which wer =  $0.055 \pm 0.010$  min<sup>-1</sup>). Moreover, the three curves in the absence or presence of fluvoxamine and PD118057 practically overlapped with each other. Thus, in this more detailed experiment, we verified that fluvoxamine and PD118057 did not allosterically enhance [3H]astemizole binding at a concentration of 10 μM. Interestingly, LUF6200 increased the dissociation rate of radioligand binding notably to a  $k_{off}$  value of  $0.21 \pm 0.011 \text{ min}^{-1}$  (P = 0.0005), which corresponded to a 3.7-fold increase in the dissociation rate found for [3H]astemizole alone.



**Figure 6.** Dissociation curves of [<sup>3</sup>H]astemizole induced by 10 μM astemizole in the absence (control) or presence of 10 µM fluvoxamine, PD118057 and LUF6200. Experiments were performed at 15 °C using 30 µg HEK293K 11.1 membrane protein.

## Dissociation of [3H] dofetilide from the K,11.1 channel: effects of PD118057 and LUF6200

Since fluvoxamine was found to have no allosteric capacity in both the [3H]astemizole and [3H]dofetildie binding assays described above, only PD118057 and

LUF6200 were tested for their allosteric effects on [3H]dofetilide binding to the K\_11.1 channel (Figure 7). The results demonstrated that LUF6200 accelerated the dissociation rate of [ $^{3}$ H]dofetilide from  $0.20 \pm 0.026$  min $^{-1}$  to  $1.5 \pm 0.17$  min $^{-1}$ (P = 0.0016), while the  $k_{off}$  of [3H]dofetilide in the presence of 10  $\mu$ M PD118057 was  $0.20 \pm 0.015$  min<sup>-1</sup>, displaying no enhancement effects on the binding of [<sup>3</sup>H] dofetilide. This further proved that LUF6200 was an allosteric inhibitor but that PD118057 was not an allosteric enhancer of the binding of K, 11.1 radioligands.

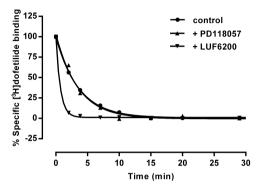


Figure 7. Dissociation curves of [3H]dofetilide by 10 μM dofetilide in the absence (control) or presence of 10 µM PD118057 and LUF6200. Experiments were was performed at 25 °C using 20 μg HEK293K, 11.1 membrane protein.

## Concentration dependence of allosteric effects of LUF6200 on [3H]astemizole and [3H]dofetilide binding

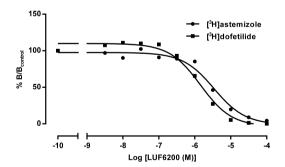


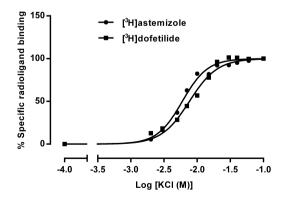
Figure 8. Concentration dependence curves of LUF6200 for accelerating the [3H]astemizole or [3H]dofetilide dissociation from the K\_11.1 channel. Membranes were first pre-equilibrated with [3H]astemizole or [3H]dofetilide, then the dissociation was induced by 10 µM astemizole or dofetilide in the absence (control) or presence of different concentrations of LUF6200 and the reaction was terminated after 10 min. The results are expressed as the ratio of the specific binding of [3H]astemizole or [3H]dofetilide in the

presence of 10  $\mu$ M astemizole or dofetilide plus various concentrations of LUF6200 (B) over that in the presence of 10  $\mu$ M astemizole or dofetilide alone (B<sub>control</sub>). Experiments were were performed at 15 °C using 30  $\mu$ g HEK293K<sub>v</sub>11.1 membrane protein for [<sup>3</sup>H] astemizole and 20  $\mu$ g membrane protein for [<sup>3</sup>H]dofetilide binding.

The influence of increasing concentrations of LUF6200 on the dissociation of [ $^3$ H]astemizole and [ $^3$ H]dofetilide induced by 10  $\mu$ M astemizole or dofetilide is shown in **Figure 8**. After 10 min of dissociation, the process was terminated by washing and rapid filtration. LUF6200 accelerated the dissociation of [ $^3$ H]astemizole and [ $^3$ H]dofetilide in a concentration-dependent manner with similar, not statistically significantly different, EC<sub>50</sub> values of 4.9  $\pm$  0.52 and 3.4  $\pm$  0.97  $\mu$ M, respectively (P = 0.45).

## Concentration dependent effects of potassium ions on [3H]astemizole and [3H]dofetilide binding

To assess the influence of increasing concentrations of potassium ions on the binding of [ ${}^{3}$ H]astemizole and [ ${}^{3}$ H]dofetilide, we determined the specific binding of [ ${}^{3}$ H]astemizole and [ ${}^{3}$ H]dofetilide at different concentrations of KCl after 60 min preincubation at 25 °C. The results are shown in **Figure 9**. The binding of both [ ${}^{3}$ H]astemizole and [ ${}^{3}$ H]dofetilide were elevated with increasing concentrations of potassium ions and reached a plateau as of a concentration of 30 mM. The mean EC<sub>50</sub> values in the two different radioligand binding assays were 7.2  $\pm$  0.60 mM and 8.1  $\pm$  0.47 mM for [ ${}^{3}$ H]astemizole and [ ${}^{3}$ H]dofetilide, respectively, which were not significantly different (P = 0.30). It is also worth noting that when no potassium ions were added to incubation mixture, the specific binding of [ ${}^{3}$ H] astemizole and [ ${}^{3}$ H]dofetilide was negligible (data not shown).



**Figure 9.** Dose-response curves of the specific binding of [<sup>3</sup>H]astemizole and [<sup>3</sup>H]dofetilide at different concentrations of potassium ions. Experiments were performed at 25 °C

using 30 µg HEK293K\_11.1 membrane protein for [3H]astemizole binding and 20 µg membrane protein for [3H]dofetilide binding.

## Effects of potassium ions on the acceleration of the [3H]dofetilide dissociation rate induced by LUF6200

In order to investigate whether the two modulators described above acted at different sites to produce these effects, the concentration-dependent effect of LUF6200 on [3H]dofetilide dissociation was studied in the presence of four different concentrations of KCl (Figure 10). The data obtained are represented in two formats. Figure 10A shows that addition of potassium ions decreased the dissociation, whereas LUF6200 dose-dependently enhanced that dissociation. Figure 10B demonstrates that addition of potassium ions resulted in a concentration-dependent leftward shift of the LUF6200 concentration-effect curve. The EC<sub>50</sub> values of LUF6200 in the presence of 2, 5, 10 and 60 mM potassium ions were 11  $\pm$ 1.7,  $6.4 \pm 0.74$ ,  $4.7 \pm 0.53$  and  $3.5 \pm 0.12$  µM, respectively. This indicates that LUF6200 and potassium ions act at distinct allosteric sites on the K<sub>2</sub>11.1 channel, and that potassium ions enhanced the allosteric effect of LUF6200.

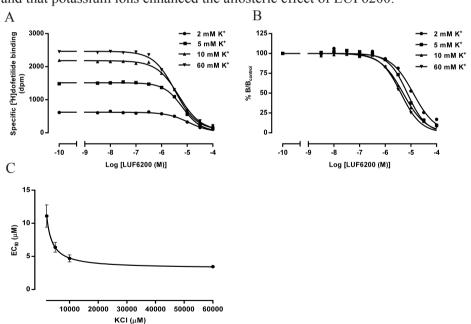


Figure 10. Concentration-effect curves of LUF6200 for accelerating the [3H]dofetilide dissociation from the K<sub>2</sub>11.1 at 2, 5, 10 and 60 mM potassium ions before (A) and after normalization (B), and the positive cooperativity effect of potassium ions on LUF6200 in enhancing [3H]dofetilide dissociation (C). Membranes were first pre-equilibrated with [3H]dofetilide at different concentrations of potassium ions, then the dissociation was in-

duced by 10 µM dofetilide in the absence (control) or presence of different concentrations of LUF6200 and the reaction was terminated after 6 min. The results in **B** are expressed as the ratio of the specific binding of [3H]dofetilide in the presence of 10 µM dofetilide plus various concentrations of LUF6200 (B) over that in the presence of 10 µM dofetilide alone (B<sub>control</sub>). Experiments were performed at 15 °C using 20 µg HEK293K<sub>y</sub>11.1 membrane protein. The experimental data of different concentrations of potassium ions affecting the modulating potency of LUF6200 are exhibited with standard error of the mean from three independent experiments (C).

Last, the cooperativity between LUF6200 and potassium ions was further analyzed according to a cooperativity model studied by us previously<sup>31</sup>, and the curve connecting all the data is displayed in Figure 10C. We discriminated three possible situations<sup>31</sup>, in which two compounds showed positive ( $\delta > 1$ ), neutral  $(\delta = 1)$  or negative cooperativity ( $\delta < 1$ ). In the present study, our data points complied with  $\delta > 1$ , thus further supporting a positive cooperativity between the binding of LUF6200 and potassium ions.

#### **Discussion**

In the present study, we evaluated the allosteric effects of a series of compounds with diverse chemical structures on the K<sub>v</sub>11.1 channel using both [<sup>3</sup>H]astemizole and [3H]dofetilide binding assays, and also addressed the influence of potassium ions on the binding of two radioligands. To the best of our knowledge, this study is the first time to utilize radioligand binding assays for the characterization of allosteric modulators for the K<sub>v</sub>11.1 channel. From these initial results, one potent allosteric inhibitor (LUF6200) was identified and several other compounds including LUF6208, LUF6209, LUF6217, LUF6202 and LUF6206 were also shown to possess allosteric inhibitory effects in two radioligand binding assays. Moreover, potassium ions increased the binding of [3H]astemizole and [3H]dofetilide to the K<sub>2</sub>11.1 channel. Additionally, fluvoxamine and PD118057, known to have binding sites distinct from most other K, 11.1 blockers, exhibited no allosteric effects in this investigation.

A common binding site for a wide spectrum of K\_11.1 blockers, including astemizole and dofetilide, has been implied through different methods<sup>32, 33</sup>. However, the amino acid residues involved in the interaction of these structurally diverse compounds may be different<sup>32</sup>. Therefore, in this study two different radioligands ([3H]astemizole and [3H]dofetilide) were adopted in order to study, validate and compare the behavior of putative allosteric modulators. Since an altered dissociation rate is unequivocally indicative of an allosteric interaction<sup>34</sup>, [<sup>3</sup>H]astemizole and [3H]dofetilide dissociation assays were firstly developed and then performed to reveal the allosteric effects of different ligands. The results in this paper show that both [3H]astemizole and [3H]dofetilide exhibit fast association rates and dissociate from the K 11.1 channel after addition of an excess of unlabelled astemizole or dofetilide, indicative of the reversible binding of the two ligands. Additionally, both the association and dissociation curves of the two radioligands are better fitted with a single-site than a two-site model, which is different from the two binding sites model for [3H]dofetilide reported by Duff et al.35 However, the latter experiments were performed on intact cells instead of membrane preparations used in our study. The diffusion into and possible trapping of [3H] dofetilide inside the intact cells might affect the association and dissociation of the radioligand significantly.

Subsequently, single time point dissociation assays were conducted to screen for allosteric modulators of the K<sub>2</sub>11.1 channel. LUF6200 was selected as a representative allosteric inhibitor from the screening experiments due to its potent effects on both [3H]astemizole and [3H]dofetilide binding. Besides, terfenadine, which occupies the same binding site as astemizole and dofetilide<sup>36</sup>, did not display allosteric effects in either of the two radioligand screening assays, suggesting the validity and accuracy of our methods. To further characterize the allosteric effects of LUF6200, a full radioligand dissociation assay was performed in the presence of this compound. LUF6200 caused a 3.7- and 7.5-fold increase in the dissociation rates of [3H]astemizole and [3H]dofetilide, respectively. This further supports the notion that LUF6200 is an allosteric inhibitor of conventional K<sub>2</sub>11.1 blockers binding to the K<sub>2</sub>11.1 channel. The EC<sub>50</sub> values from the two different radioligand dissociation assays were almost the same and in the micromolar range, which suggests that astemizole and dofetilide occupy the same binding site on the K\_11.1 channel. In a previous study from our laboratory, LUF6200 displaced [3H]astemizole binding from the K<sub>2</sub>11.1 channel under equilibrium conditions with a very low K, value of  $1.2 \pm 0.6$  nM<sup>26</sup>. As is evident from **Figure 8**, there is little or negligible allosteric inhibition observed for LUF6200 at a concentration lower than 0.1 µM. Taken together, we conclude that LUF6200 interacts with the K 11.1 channel in a complicated fashion: it behaves as an orthosteric blocker at lower concentrations and as an allosteric inhibitor at higher concentrations.

Two major binding regions have been hypothesized for the K 11.1 channel, one in the transmembrane region, the other extracellularly<sup>37, 38</sup>. Most K<sub>2</sub>11.1 blockers bind to the central transmembrane cavity of the channel, whereas agents described as facilitators, activators or enhancers of the K\_11.1 channel usually interact with a region facing away from the inner cavity<sup>37</sup>. For instance, A-935142 has recently been stated to possess a binding site responsible for K<sub>2</sub>11.1 current enhancement which is different from the pore binding site for traditional K<sub>11.1</sub> blockers<sup>39</sup>. The binding sites of several peptide blockers like saxitoxin and BeKm-1 are located in the extracellular parts of the K 11.1 channel<sup>40, 41</sup> and some metal ions bind to different regions of the pore of the channel in comparison to most other K<sub>2</sub>11.1 blockers<sup>42, 43</sup>. Additionally, it has been reported that a few nontoxin K\_11.1 blockers are unlikely to fit within the inner cavity and must bind elsewhere<sup>23, 44</sup>. Thus, multiple binding sites of these diverse compounds or ions at the K<sub>2</sub>11.1 channel imply the plausible allosteric modulation between them. They might allosterically increase (allosteric inhibitors/negative allosteric modulators) or decrease (allosteric enhancers/positive allosteric modulators) the dissociation rates of typical K\_11.1 blockers from the channel. The current study does not allow determining the location of the putative binding site of LUF6200. Its nanomolar potency as a classical blocker suggests that it interacts with the central cavity; for its allosteric property a different binding site or different conformation of the channel may play a role, and further studies such as cross linker and western blotting experiments are required to elucidate the exact binding sites of this compound.

The results in this paper also show that fluvoxamine does not behave as an allosteric modulator when tested in a rigorous kinetic assay. However, it has been reported that the binding of fluvoxamine to the K<sub>v</sub>11.1 channel did not involve amino acids such as Y652 and F656 that are crucial to the binding of other general K<sub>v</sub>11.1 blockers<sup>45, 46</sup>; hence, it was suggested that fluvoxamine probably binds to an extracellular site at the outer mouth of the channel<sup>23</sup>. In addition, the activator PD118057 did not exert allosteric enhancement in both [³H]astemizole and [³H]dofetilide dissociation assays. PD118057 has been identified to bind to a hydrophobic binding pocket instead of the shared binding site for K<sub>v</sub>11.1 blockers at the channel<sup>47</sup>. The results obtained for these two compounds suggest that even though a compound binds to a distinct region of the K<sub>v</sub>11.1 channel, its allosteric effects on the binding of other K<sub>v</sub>11.1 blockers are still uncertain and need to be validated in more detailed experiments.

Since it has been reported that the extracellular potassium concentration affected the binding of [ $^3$ H]dofetilide to neonatal mouse myocytes $^{48}$ , we also studied the allosteric effects of potassium ions and found that these monovalent cations increase both [ $^3$ H]astemizole and [ $^3$ H]dofetilide binding in a concentration-dependent manner with very similar EC $_{50}$  values. In earlier studies on a variety of ion channels, it was reported that metal ion binding to the pore of an ion channel may affect the occupancy of the binding site for other ligands and thus decrease their binding $^{42,49}$ . This is in contrast to the observation in our study. Thus, we assume that K $^+$  ions possibly function as an allosteric enhancer of K $_v$ 11.1 blockers by changing the conformation of their binding site rather than competitively oc-

cupying their binding site. Since two distinct allosteric sites on the human GnRH receptor 31 and three different allosteric sites on the dopamine D<sub>4</sub> receptor 50 have been reported, we wondered whether LUF6200 and potassium ions exert their effect through two distinct allosteric sites on the K\_11.1 channel as well. Potassium ions both enhance the allosteric effect of LUF6200 and shift the LUF6200 doseresponse curve to the left, supporting that the two modulators are indeed acting at different binding sites at the K, 11.1 channel. Furthermore, a curve described in Figure 10C indicated a positive cooperativity between LUF6200 and potassium ions. Taken together, we anticipate that there are at least two different sites of allosteric modulation for K\_11.1 blockers at the channel and that they interact with each other as well.

It is hypothesized that the ceiling effect of allosteric modulators of ligand-gated ion channels has important advantages over classic orthosterically acting compounds or open channel blockers, because it offers a much larger safety margin when it comes to drug administration and patient compliance<sup>51</sup>. A similar scenario may also be feasible for the K\_11.1 channel. For allosteric inhibitors one might argue that when non-cardiovascular drugs (e.g., astemizole) are combined with allosteric inhibitors of the K 11.1 channel, their cardiac side effects might be diminished. On the other hand, when allosteric enhancers are administered together with antiarrhythmic drugs such as dofetilide whose desired target is the K 11.1 channel, the therapeutic dose might be lowered, thereby possibly preventing adverse drug reactions such as TdP. Taken together, the discovery of allosteric modulators for the K\_11.1 channel might open new avenues for treating clinical conditions related to the K 11.1 channel as well as potentially reducing K 11.1 cardiotoxicity of other non-cardiovascular drugs by combination therapy.

In summary, the present study provides evidence for the allosteric effects of dofetilide analogues on the binding of K\_11.1 blockers to the K\_11.1 channel using both [3H]astemizole and [3H]dofetilide binding assays. LUF6200 emerges as an allosteric inhibitor with micromolar potency. Potassium ions allosterically enhance the binding of typical K\_11.1 blockers in concentrations that are physiologically relevant. Moreover, LUF6200 and potassium ions act at distinct allosteric sites on the K<sub>2</sub>11.1 channel, and these binding sites show positive cooperativity. In addition, neither fluvoxamine nor PD118057 exhibit significant allosteric effects on the binding of two radioligands. This new information may be relevant for a further analysis of K\_11.1-related cardiotoxicity and provide novel solutions for proarrhythmic side effects induced by K<sub>2</sub>11.1 blockers.

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## **Chapter 4**

Allosteric modulation of  $K_v$ 11.1 (hERG) channels protects against drug-induced ventricular arrhythmias

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#### **Abstract**

Drug-induced ventricular arrhythmias (DiVAs) due to unintentional blockade of the K\_11.1 (hERG) channel are a major safety concern in drug development. In the past years, several highly prescribed drugs have been withdrawn for their ability to cause life-threatening arrhythmias. Here, we investigated whether the proarrhythmic risk of existing drugs could be reduced by K, 11.1 allosteric modulators. Using [3H]dofetilide binding assays with membranes of human K\_11.1-expressing HEK293 cells, two existing compounds (VU0405601 and ML-T531) and a newly synthesized compound (LUF7244) were found to be negative allosteric modulators of dofetilide binding to the K 11.1 channel with LUF7244 showing the strongest effect at 10 µM. The K\_11.1 affinity of typical blockers (i.e. dofetilide, astemizole, sertindole and cisapride) were significantly decreased by LUF7244. Treatment of neonatal rat ventricular myocyte (NRVM) monolayers with astemizole or sertindole caused heterogeneous prolongation of action potential (AP) duration and a high incidence of early afterdepolarizations upon 1-Hz electrical point stimulation, occasionally leading to unstable, self-terminating reentrant tachyarrhythmias. Pretreatment of NRVMs with LUF7244 prevented these proarrhythmic effects. NRVM monolayers treated with LUF7244 alone displayed electrophysiological properties indistinguishable from those of untreated NRVM cultures. Prolonged exposure of NRVMs to LUF7244 or LUF7244 plus astemizole did not affect their viability, excitability and contractility as assessed by molecular, immunological and electrophysiological assays. Conclusively, allosteric modulation of the K<sub>2</sub>11.1 channel efficiently suppresses DiVAs in vitro by preventing potentially arrhythmogenic changes in AP characteristics, raising the possibility to resume the clinical use of unintended K<sub>2</sub>11.1 blockers via pharmacological combination therapy.

## Introduction

Drug-induced ventricular arrhythmias (DiVAs) are a frequently encountered clinical problem, which has resulted in the restricted use or market withdrawal of existing cardiac and non-cardiac drugs and still represents a major obstacle for the development of new drugs. Inhibition of the rapid component of the delayed rectifier  $K^+$  current ( $I_{Kr}$ ) has been identified as the major culprit in the development of DiVAs. The consequential slowing of cardiac repolarization, manifested in the surface electrocardiogram as a prolongation of the QT interval, increases the likelihood of early afterdepolarizations (EADs), which may give rise to ectopic beats. Drug-induced  $I_{Kr}$  blockade also increases spatial dispersion of repolarization and refractoriness due to differences in relative  $I_{Kr}$  and  $I_{Ks}$  (the slow component of the delayed rectifier  $K^+$  current) densities between cardiomyocytes in different regions of the ventricular wall. Together these electrophysiological alterations promote the development of a special type of polymorphic ventricular arrhythmias known as "Torsades de Pointes" (TdPs), which mostly resolve spontaneously but occasionally degenerate into fatal ventricular fibrillation.  $^3$ 

The molecular correlate of  $I_{\rm Kr}$  is the  $\rm K_v 11.1$  protein encoded by the  $\it KCNH2$  gene (also known as ether-à-go-go-related gene 1 [ERG or ERG1]), possibly in combination with the MiRP1 or MinK protein, which are encoded by the  $\it KCNE1$  and  $\it KCNE2$  gene, respectively.  $^4$  K $_{\rm v} 11.1$  represents the pore-forming  $\alpha$  subunit of the  $\it I_{\rm Kr}$  channel, while MiRP1 and MinK are thought to be accessory  $\beta$  subunits involved in the modulation of its activity. Several structural features of K $_{\rm v} 11.1$  render the  $\it I_{\rm Kr}$  channel particularly susceptible to blockade by a heterogeneous collection of chemical compounds including various non-cardiac drugs (e.g., astemizole, sertindole and cisapride).

An obvious strategy to reduce the proarrhythmic risk of drugs with unintended  $I_{\rm Kr}$  blockade is by lowering their  $\rm K_v 11.1$  affinity via chemical modifications. Alternatively, supplementary drugs that lessen the proarrhythmic risk of inadvertent  $\rm K_v 11.1$  blockers can be developed potentially allowing the (i) reintroduction of medicines previously recalled from the market because of their  $\rm K_v 11.1$ -related cardiotoxicity and (ii) admission of new drugs with fortuitous  $I_{\rm Kr}$ -blocking effects. Paradoxically, screening of drugs for possible  $I_{\rm Kr}$ -blocking side effects has resulted in the serendipitous discovery of various  $\rm K_v 11.1$  activators. Besides their potential usefulness in treating inherited long QT syndrome (LQTS), these  $\rm K_v 11.1$  activators may also be used to counteract  $\rm K_v 11.1$  blockade-associated DiVAs. Indeed, Kang *et al.* showed that the action potential (AP) duration (APD)-prolonging effect of the  $I_{\rm Kr}$ -blocking, class III antiarrhythmic drug dofetilide could be counterbalanced by the first identified synthetic  $\rm K_v 11.1$  activator designated

RPR260243.6 However, due to the APD-shortening effect, K 11.1 activators carry the risk of possibly inducing short QT syndrome. 5 Recently, Potet et al. described a compound designated VU0405601 that upon pretreatment significantly reduced the APD-prolonging effect of dofetilide in Langendorff-perfused rabbit hearts and dose-dependently mitigated the K\_11.1-blocking effects of seven different drugs in a fluorescence-based thallium flux assay performed with HEK293 cells stably overexpressing the human KCNH2 gene (HEK293K, 11.1 cells).7 VU0405601 exerted its effects on the K\_11.1 channel in whole-cell voltage clamp experiments using HEK293K\_11.1 cells only when applied extracellularly. This suggests that VU0405601 binds to the extracellular domain of the K\_11.1 channel rather than to its central cavity leading us to hypothesize that VU0405601 counteracts the APD-prolonging effect of K\_11.1 blockers by an allosteric mechanism.

Allosteric modulators bind to their targets at a site topologically different from that of the endogenous ligand. Of note, since the K\_11.1 channel does not have an endogenous ligand, the authors refer to the site where typical blockers (i.e. dofetilide and astemizole) bind as the orthosteric site. From this so-called allosteric or regulatory site they generally display higher selectivity across receptor subtypes and thus provide a safer pharmacological profile than ligands binding to the orthosteric or active site. 8 Whereas allosteric modulators targeting ligand-gated ion channels and G protein-coupled receptors have been well established as research tools and therapeutic agents, 8-10 very little progress has been made in the discovery and clinical development of such compounds for voltage-gated ion channels, in particular the K<sub>2</sub>11.1 channel.

In this study, different in vitro radioligand binding assays were used to investigate whether two previously reported compounds (i.e. VU04056017 and ML-T53111) and a new compound designated LUF7244 could allosterically modulate the binding of the potent K<sub>2</sub>11.1 blocker dofetilide to the channel's central cavity. 12, 13 Radioligand binding assays were also employed to study LUF7244's influence on the interaction between (i) the K\_11.1 channel and three different blockbuster drugs (i.e. astemizole, sertindole and cisapride) that have been withdrawn from the market due to their K<sub>2</sub>11.1-related cardiotoxicity<sup>14</sup> and (ii) astemizole and its intended target, the human histamine H<sub>1</sub> receptor (hH<sub>1</sub>R). The radioligand binding assays were complemented with optical voltage mapping experiments in confluent monolayers of neonatal rat ventricular myocytes (NRVMs). These experiments were performed in the absence and presence of LUF7244 and/or any of the three blockbuster drugs. Electrophysiological parameters analyzed included conduction velocity (CV), APD at 40 and 90% repolarization (APD<sub>40</sub> and APD<sub>qn</sub>, respectively), APD dispersion and EAD incidence. In addition, the effects of LUF7244 alone and together with astemizole on the viability, excitability and

contractility of the NRVMs were investigated. The results of this study indicate that negative allosteric modulation of the  $K_v$ 11.1 channel may provide a safe and effective means to prevent the proarrhythmic effects of  $I_{Kr}$  blockers that bind to the channel's central cavity.

#### Materials and methods

#### **Materials**

Dofetilide was synthesized in-house as previously described. Bovine serum albumin (BSA, fraction V) and the fortuitous K<sub>v</sub>11.1 blockers astemizole, sertindole and cisapride were purchased from Sigma-Aldrich (St. Louis, MO). Tritium-labeled dofetilide (specific activity: 82.3 Ci·mmol<sup>-1</sup>) and astemizole (specific activity: 78.4 Ci·mmol<sup>-1</sup>) were obtained from PerkinElmer (Groningen, The Netherlands). The synthesis of VU0405601, ML-T531, as well as the design and synthesis of LUF7244 and its derivatives will be detailed elsewhere. G418 was purchased from Stratagene (Cedar Creek, TX). All other chemicals were of analytical grade and obtained from standard commercial sources. HEK293K<sub>v</sub>11.1 cells were kindly provided by Dr. Eckhard Ficker (Case Western Reserve University, Cleveland, OH). Expression plasmid pcDNA3.1-hH<sub>1</sub>Rwt was a gift of Prof. Thue W. Schwartz (University of Copenhagen, Copenhagen, Denmark). All animal experiments were approved by the Animal Experiments Committee of Leiden University Medical Center and conformed to the Guide for the Care and Use of Laboratory Animals as stated by the US National Institutes of Health.

## Radioligand binding assays

#### Cell culture and membrane preparations

 $\rm HEK293K_{_{v}}11.1$  cells were cultured and membranes were prepared and stored as described previously.  $^{\rm 17}$ 

HEK293 cells were cultured in a humidified atmosphere at 37 °C and 7% CO<sub>2</sub> in Dulbecco's modified Eagle's medium (DMEM; Sigma-Aldrich, D6546) supplemented with 10% fetal bovine serum (Sarstedt, Nümbrecht, Germany), 50 IU·mL·¹ penicillin (Sarstedt) and 50 μg·mL·¹ streptomycin (Sarstedt). When the cells had reached 50-60% confluence, the culture medium was refreshed and 500 μL of 150 mM NaCl containing 15 μg of pcDNA3.1-hH<sub>1</sub>Rwt DNA and 45 μg of linear 25-kDa polyethylenimine (Polysciences Europe, Eppelheim, Germany) was added per 100-mm culture dish (Sarstedt). Cells were harvested 48 h after transfection. Membranes of HEK293 cells expressing hH<sub>1</sub>R (HEK293hH<sub>1</sub>R cells)

were prepared and stored as described for HEK293K\_11.1 cells except that incubation buffer I (50 mM Tris-HCl [pH 7.4]) was used instead of incubation buffer II (10 mM HEPES-NaOH [pH 7.4], 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl, 1 mM EGTA, 10 mM glucose, 0.1% BSA).

#### Radioligand displacement assays

[3H]Dofetilide binding assays for the K\_11.1 channel were performed in incubation buffer II as described previously.<sup>17</sup> Briefly, membrane aliquots containing 20 μg protein were incubated with 5 nM [3H]dofetilide in a total volume of 100 μl at 25 °C for 1 hour. Radioligand displacement experiments were carried out with various concentrations of the test compounds. Total binding was determined in the presence of unsupplemented incubation buffer II, whereas non-specific binding was evaluated in incubation buffer II containing 10 µM astemizole. Displacement experiments with different concentrations of dofetilide, astemizole, sertindole and cisapride were conducted in the absence (control) or presence of 10 µM LUF7244. Incubations were terminated by dilution with ice-cold wash buffer (25 mM Tris-HCl [pH 7.4], 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl<sub>2</sub>, 0.05 mM CaCl<sub>2</sub>, 0.05% BSA). Free radioligand was separated from bound [3H]dofetilide by rapid filtration through a UniFilter-96 GF/B microplate using a PerkinElmer MicroBeta Filtermate-96 harvester. The radioactivity was determined using a Wallac 1450 MicroBeta TriLux liquid scintillation counter (PerkinElmer) after extraction with 25 µl MicroScint 20 (PerkinElmer).

[3H]Astemizole binding assays for the hH,R were performed in incubation buffer I. Membrane aliquots containing 15 µg protein were incubated with 3.5 nM [3H]astemizole in a total volume of 100 µl at 25 °C for 3 h. Total binding was determined in the presence of unsupplemented incubation buffer I, whereas non-specific binding was evaluated in incubation buffer I containing 100 µM astemizole. LUF7244 displacement assays were carried out with various concentrations of this compound, while displacement assays with different concentrations of astemizole were conducted in the absence (control) or presence of 10 μM LUF7244. Incubations were terminated by dilution with ice-cold wash buffer II (50 mM Tris-HCl [pH 7.4]), and samples were obtained as mentioned above.

### Kinetic dissociation assays

Kinetic dissociation assays of [3H]dofetilide were performed as described previously<sup>17</sup> with the following modifications. Single-point dissociation experiments were conducted by addition of 10 µM dofetilide in the absence (control) or presence of 10 or 50 µM of the selected compounds after preincubating membranes

with [3H]dofetilide at 25 °C for 2 h. After 6 min of dissociation, incubations were terminated and samples were obtained as described under "Radioligand displacement assays". Traditional dissociation experiments were carried out with 10 μM dofetilide in the absence (control) or presence of 50 µM of the test compounds at 25 °C for a total period of 2 h after preincubation with the radioligand. The amounts of [3H]dofetilide still bound to the receptor were measured at various time intervals. The concentration-dependent effect of LUF7244 was determined by addition of 10 µM dofetilide in the absence (control) or presence of different concentrations of LUF7244. After 6 min of dissociation, incubations were terminated and samples were obtained as described under "Radioligand displacement assavs".

#### **Functional assays in NRVMs**

#### Isolation and culture of NRVMs

NRVMs were isolated and cultured as described previously. 18 Briefly, neonatal rats were anaesthetized with 4-5% isoflurane inhalation anesthesia. After adequate anesthesia had been confirmed by the absence of reflexes, the heart was quickly excised. Ventricles were separated from the remainder of the heart, cut into small pieces with a fine scissor and a scalpel and dissociated by collagenase type I (Worthington, Lakewood, NJ) digestion. The resulting cell suspension was applied to Primaria culture dishes (Corning Life Sciences, Amsterdam, The Netherlands) and incubated for 75 min at 37 °C in a humidified atmosphere of 5% CO, to allow preferential attachment of cardiac fibroblasts. The unattached cells (mainly cardiomyocytes) were collected, passed through a cell strainer (70-µm mesh pore size; BD Biosciences, Breda, The Netherlands) and applied at a density of 8×10<sup>5</sup> cells per well of a 24-well cell culture plate (Corning Life Sciences) to fibronectin (Sigma-Aldrich)-coated, round glass coverslips. Proliferation of residual cardiac fibroblasts (10-15%) was inhibited by incubating the cells for 2 h in culture medium containing 10 µg·mL<sup>-1</sup> mitomycin C (Sigma-Aldrich) at 24 h after seeding (i.e. at day 1 of culture). Cells were subsequently maintained in a 1:1 mixture of DMEM (Life Technologies Europe, Bleiswijk, The Netherlands; catalog number: 22320) and Ham's F10 medium (Life Technologies Europe) supplemented with 5% horse serum (Life Technologies Europe), 2% BSA and sodium ascorbate to a final concentration of 0.4 mM. This so-called NRVM medium was replaced daily.

#### Immunocytochemical analyses

NRVMs were plated on fibronectin-coated, 15-mm diameter round glass coverslips at a density of  $4\times10^4$  cells. At day 9 of culture, the cells were washed thrice with ice-cold phosphate-buffered saline (PBS), fixed for 30 min in buffered 4% formaldehyde (Added Pharma, Oss, The Netherlands) of 4 °C and permeabilized by a 10-minute incubation at room temperature (RT) with 0.1% Triton X-100 in PBS. Next, the cells were double-immunostained for K\_11.1 (rabbit polyclonal antibodies raised against the C terminus of human K 11.1, Merck Millipore, Billerica, MA, catalog number: AB5930) and α-actinin (mouse monoclonal antibody, Sigma-Aldrich, clone: EA-53). Incubation with primary antibodies (diluted 1:200 in PBS-0.1% normal donkey serum) and corresponding donkey Alexa Fluor 488/568-conjugated secondary antibodies (1:400 dilution, Life Technologies Europe) lasted for 2 h. To visualize their nuclei, the cells were incubated for 10 min at RT with 10 µg·mL<sup>-1</sup> Hoechst 33342 (Life Technologies Europe) in PBS. After each processing step, the cells were washed three times with PBS of RT. To minimize photobleaching, coverslips were mounted in VECTASHIELD mounting medium (Vector Laboratories, Burlingame, CA). Photomicrographs were obtained with the aid of a digital color camera-equipped fluorescence microscope (Nikon Eclipse 80i; Nikon Instruments Europe, Amstelveen, The Netherlands).

### Reverse transcription-quantitative polymerase chain reaction (RT-qPCR) analyses

Cultures of NRVMs and of neonatal rat cardiac fibroblasts (NRCFs; maintained in NRVM medium) were washed once with ice-cold PBS after which the cells were lysed in TRIzol Reagent (Life Technologies Europe) and total RNA was isolated using the RNeasy Mini kit (QIAGEN Benelux, Venlo, The Netherlands). The RNA was reverse transcribed with the iScript cDNA synthesis kit (Bio-Rad, Hercules, CA) and the resulting cDNA was amplified by PCR using the Bioline SensiFAST SYBR No-ROX kit (GC biotech, Alphen aan den Rijn, The Netherlands). The forward and reverse primers for the amplification of rat *Kcnh2*-specific cDNA were localized in different exons and had the following sequences, respectively: 5' TAGCCTCCTCAACATCCC 3' and 5' CCATGTCTGCACTTAGCC 3'. For normalization purposes, rat 18S rRNA (Rn18s)-specific cDNA was amplified in parallel using the following forward and reverse primer, respectively: 5' GTAACCCGTTGAACCCCATT 3' and 5' CCATCCAATCGGTAGTAGCG 3'. PCR amplifications were carried out in a CFX96 Touch Real-Time PCR detection system (Bio-Rad) using a 2-step cycling protocol (20-40 cycles of 95 °C 10 sec, 60 °C 30 sec) after a 5-minute incubation at 95 °C. Quantitative analyses were based on the  $2^{-\Delta\Delta CT}$  method using CFX Manager software (Bio-Rad). For these analyses, PCRs were performed *in triplo* on three independent samples.

#### **Optical mapping**

APs were investigated on a whole-culture scale by optical mapping using the potentiometric dye di-4-ANEPPS (Life Technologies Europe) as described previously. 19 The measurements were carried out at 37 °C on 9-day-old confluent NRVM monolayer cultures (see above). Optical signals were captured using a MiCAM ULTIMA-L imaging system (SciMedia, Costa Mesa, CA). To validate the experimental model, cells were incubated for 20 min in culture medium containing 0, 10, 30, 100 or 300 nM of the hH<sub>1</sub>R antagonist and unintended K<sub>2</sub>11.1 blocker astemizole and dimethylsulfoxide (DMSO) at a final concentration of 0.03%. In a subsequent experiment, NRVM cultures were first exposed for 30 min to 10 µM LUF7244 or its solvent (i.e. culture medium containing 0.1% DMSO). Next, astemizole (final concentration of 100 nM) or vehicle was added to the culture medium raising the DMSO concentration to 0.13%. Following an incubation period of 30 min at 37 °C, optical recordings were started in the continued presence of the appropriate vehicle/drug combinations (Figure 4A). Optical traces were analyzed using Brain Vision Analyzer 1208 software (Brainvision, Tokyo, Japan). To minimize noise artifacts, calculations were based on the average of the signals at a selected pixel and its eight nearest neighbors. CV, APD<sub>40</sub>, APD<sub>50</sub> and APD<sub>40</sub> dispersion were determined using NRVM cultures showing uniform AP propagation and 1:1 capture after 1-Hz local stimulation. Each of these electrophysiological parameters was the average of values obtained from 6 different positions equally distributed across the cell cultures. The effects of LUF7244 on the K\_11.1-inhibiting antipsychotic drug sertindole and gastroprokinetic agent cisapride were assessed in the same way as for astemizole expect that the final concentration of sertindole and cisapride was 1  $\mu$ M.

To study the effects of long-term  $K_v11.1$  allosteric modulation in the absence and presence of an unintended  $K_v11.1$  blocker on cardiac excitability, confluent NRVM cultures were incubated for 3 days in culture medium containing 100 nM astemizole and/or 10  $\mu$ M LUF7244 prior to potentiometric dye loading, short-term drug treatment (see above) and optical mapping in the continued presence of drug(s) at culture day 9. Confluent NRVM monolayers exposed to culture medium containing vehicle (i.e. DMSO at a final concentration of 0.13%) only, served as negative controls for this experiment. Just prior to optical mapping, movies of the cells were made using a Carl Zeiss Axiovert 40C microscope equipped with a LD A-Plan 20x/0.3 Ph1 objective (Zeiss Nederland, Sliedrecht, The Netherlands) to record their contractions upon electrical field stimulation at 1 Hz and 8 V using

a 15-ms rectangular pulse.

#### Apoptosis assay

To investigate the effects of long-term K\_11.1 allosteric modulation in the absence or presence of simultaneous K\_11.1 blockade on cell viability, externalized phosphatidylserine (as early marker of apoptosis) was detected using Alexa-568-conjugated annexin V (Life Technologies Europe). NRVM cultures in 24-well plates were treated cells for 72 h with 100 nM astemizole and/or 10 μM LUF7244 as described under "Optical mapping". Mock-treated cells and cells incubated for 24 h in culture medium containing 1 µM doxorubicin (Sigma-Aldrich) served as negative and positive controls, respectively. Next, the cells were washed once with ice-cold PBS and incubated for 15 min at RT with annexin V conjugate diluted 40-fold in binding buffer (10 mM HEPES-NaOH [pH 7.4], 140 mM NaCl, 2.5 mM CaCl<sub>2</sub>) of 4 °C containing 10 µg·mL<sup>-1</sup> Hoechst 33342. After a single wash with binding buffer of 4 °C, photomicrographs were taken using a Leica DMI6000 B inverted microscope equipped with a Leica DFC300 FX digital color camera (both from Leica Microsystems, Rijswijk, The Netherlands).

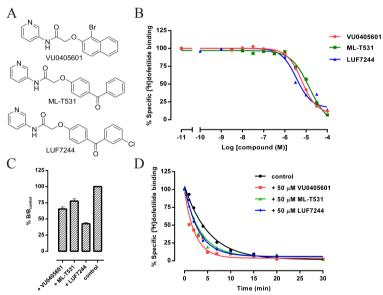
### Data analysis

Data of the radioligand binding assays were analyzed with Prism 5.0 (Graph-Pad, San Diego, CA). Half maximal inhibitory concentrations (i.e. IC<sub>50</sub> values) in displacement assays were directly obtained from non-linear regression analysis of dose-response curves. Apparent inhibitory binding constants (K, values) were derived from the IC<sub>50</sub> values according to the Cheng-Prusoff relationship<sup>20</sup>: K<sub>1</sub> =  $IC_{50}/(1 + [L^*]/K_D)$ , where [L\*] is the concentration of radioligand and  $K_D$  is its dissociation constant determined by saturation assay<sup>21</sup>. Dissociation rate constants (k<sub>off</sub>) were obtained by computer analysis of the exponential decay of [3H] dofetilide bound to the K<sub>2</sub>11.1 channel. Half maximal effective concentrations (i.e. EC<sub>50</sub> values) from kinetic dissociation assays were calculated by non-linear regression analysis of concentration-effect curves of dissociation in the presence of different concentrations of unlabeled ligands. All values obtained from radioligand binding assays in this study are means of at least three independent experiments performed in duplicate. The number of samples per experimental group in the optical mapping experiments varied from 5 to 24 as indicated. Whenever appropriate, data are expressed as mean  $\pm$  standard error of the mean (SEM) or as mean  $\pm$  standard deviation (SD). Statistical analysis was performed with a twotailed unpaired Student's t-test.

## Results

# Characterization of allosteric modulators of [ $^{3}H$ ]dofetilide binding to the $K_{\nu}11.1$ channel

The interaction of two previously reported ligands VU405601 and ML-T531 as well as the newly designed and synthesized compound LUF7244 (**Figure 1A**) with the human  $K_{\nu}11.1$  channel was studied in different [ $^3H$ ]dofetilide binding assays. As shown in **Figure 1B** and **Table 1**, all three compounds reduced [ $^3H$ ] dofetilide binding at the  $K_{\nu}11.1$  channel with relatively low affinity, i.e. with IC $_{50}$  values of 7.8  $\pm$  0.4  $\mu$ M, 12  $\pm$  1  $\mu$ M and 3.9  $\pm$  0.7  $\mu$ M for VU0405601 and ML-T531 and LUF7244, respectively. Moreover, all displacement curves demonstrated Hill coefficients significantly different from unity (data not shown), implying that VU0405601, ML-T531 and LUF7244 might not competitively displace [ $^3H$ ] dofetilide from the  $K_{\nu}11.1$  channel but may occupy distinct binding pockets in the channel protein to allosterically modulate binding of the radioligand.



**Figure 1.** Characterization of allosteric modulators of dofetilide binding to the  $K_v11.1$  channel in a [ $^3$ H]dofetilide binding assay performed with membranes of HEK293 $K_v11.1$  cells. **(A)** Chemical structures of VU0405601, ML-T531 and LUF7244. **(B)** Displacement curves of VU0405601, ML-T531 and LUF7244. **(C)** Percentage specific binding of [ $^3$ H] dofetilide to the  $K_v11.1$  channel after 6 min of dissociation induced by 10  $\mu$ M dofetilide in the absence (control) or presence of 10  $\mu$ M of VU0405601, ML-T531 or LUF7244. The specific binding of [ $^3$ H]dofetilide in the absence of the test compounds was set as  $B_{control}$  while the specific binding in their presence was set as B. **(D)** Time-dependent dissociation

of [3H]dofetilide induced by 10 uM dofetilide in the absence (control) or presence of 50 uM VU0405601, ML-T531 or LUF7244. Experiments were performed at 25 °C using 20 µg of HEK293K\_11.1 membrane protein.

Subsequently, single-point dissociation assays were performed to screen for allosteric effects of these compounds on the binding of [3H]dofetilide to the K\_11.1 channel. The results are summarized in Figure 1C and Table 1. At a concentration of 10 µM, VU0405601, ML-T531 and LUF7244 significantly increased the dissociation of [3H]dofetilide from the K\_11.1 channel, indicating that these compounds are negative allosteric modulators of dofetilide binding to the channel, LUF7244 appeared to be the most potent negative allosteric modulator with  $43 \pm 2\%$  dofetilide binding left compared to control conditions, while 10  $\mu$ M VU0405601 and ML-T531 reduced dofetilide binding to  $66 \pm 3\%$  and  $78 \pm 3\%$ , respectively.

The allosteric effects of VU0405601, ML-T531 and LUF7244 on the K<sub>v</sub>11.1 channel were further investigated in traditional radioligand dissociation experiments to determine whether co-administration of these compounds with an excess unlabeled dofetilde would change the dissociation rate of [3H]dofetilide from the K\_11.1 channel. In order to obtain larger effects, the three compounds were tested at a concentration of 50 μM instead of 10 μM as used in the single-point dissociation assay. As shown in **Figure 1D** and **Table 1**, all compounds significantly accelerated the dissociation of dofetilide, in line with the results from the single-point dissociation experiments. The off-rate of [3H]dofetilide was allosterically increased 2.2-fold (from  $0.22 \pm 0.02$  to  $0.49 \pm 0.09$  min<sup>-1</sup>) with 50  $\mu M$  VU0405601. The  $k_{off}$  value of [3H]dofetilide rose to  $0.33 \pm 0.03$  min<sup>-1</sup> in the presence of 50 µM LUF7244, which was comparable to the effect of ML-T531  $(k_{off dofetilide} = 0.30 \pm 0.02 \text{ min}^{-1}).$ 

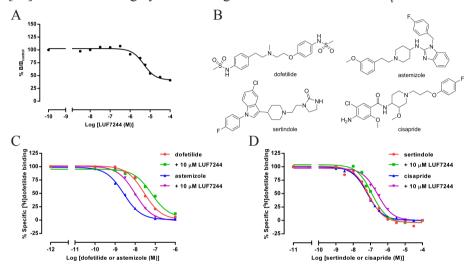
**Table 1.** The half maximal inhibitory concentrations (IC<sub>50</sub>), percentage specific binding of [3H]dofetilide to the K, 11.1 channel after 6 minutes of dissociation in the absence or presence of 10  $\mu$ M of the indicated compounds (%B/B<sub>control</sub>), and dissociation rates (k<sub>off</sub>) of [<sup>3</sup>H]dofetilide in the absence or presence of 50 μM of the indicated compounds.

	IC <sub>50</sub> (nM)	%B/B <sub>control</sub>	k <sub>off</sub> (min <sup>-1</sup> )
Dofetilide	-	$100 \pm 0$	$0.22 \pm 0.02$
+ VU0405601	$7757 \pm 350$	66 ± 3 (***)	$0.49 \pm 0.09 \ (**)$
+ ML-T531	$12000 \pm 1213$	78 ± 3 (***)	$0.30 \pm 0.02 \ (**)$
+ LUF7244	$3855 \pm 724$	43 ± 2 (***)	$0.33 \pm 0.03 \ (**)$

Values are means (± SEM) of at least three independent assays performed in duplicate. (\*\* P < 0.01 versus control; \*\*\* P < 0.001 versus control)

## Effects of LUF7244 on the binding of typical K<sub>v</sub>11.1 blockers to the channel

Since LUF7244 was the most potent amongst the three allosteric modulators at the lower test concentration of 10  $\mu$ M and may therefore have the best safety profile, its potency to increase the dissociation of [ $^3$ H]dofetilide from the K $_v$ 11.1 channel was investigated. The corresponding concentration-effect curve is shown in **Figure 2A**. From this graph, the modulatory potency (i.e. the EC $_{50}$ ) of LUF7244 was calculated to be  $4.6 \pm 0.8 \,\mu$ M. Notably, LUF7244 could not completely abrogate [ $^3$ H]dofetilide binding by accelerating its dissociation from the K $_v$ 11.1 channel.



**Figure 2.** Assessment of the effects of LUF7244 on the binding of K<sub>v</sub>11.1 blockers to the channel in a [³H]dofetilide binding assay performed with membranes of HEK293K<sub>v</sub>11.1 cells. **(A)** Effect of different concentrations of LUF7244 on the dofetilide-induced dissociation of [³H]dofetilide from the K<sub>v</sub>11.1 channel. After preincubating of HEK293K<sub>v</sub>11.1 cell membranes with [³H]dofetilide, radioligand dissociation was induced by 10 μM dofetilide in the absence or presence of different concentrations of LUF7244 and the incubation was terminated after 6 min. The results are expressed as the ratio of the specific binding of [³H]dofetilide in the presence of 10 μM dofetilide plus various concentrations of negative allosteric modulators (*B*) over that in the presence of 10 μM dofetilide alone ( $B_{control}$ ). **(B)** Chemical structures of dofetilide, astemizole, sertindole and cisapride. **(C)** Displacement curves of dofetilide and astemizole in the absence or presence of 10 μM LUF7244. **(D)** Displacement curves of sertindole and cisapride in the absence or presence of 10 μM LUF7244. Experiments were performed at 25 °C using 20 μg of HEK-293K<sub>v</sub>11.1 membrane protein.

To investigate the effects of LUF7244 on the binding affinity of other com-

pounds besides dofetilide to the K\_11.1 channel, three additional K\_11.1 blockers (i.e. astemizole, sertindole and cisapride) from distinct therapeutic classes were selected (Figure 2B). As shown in Figure 2C and 2D, the [3H]dofetilide displacement curves of all four drugs were shifted rightwards in the presence of 10 μM LUF7244, implicating that their K 11.1 affinity were diminished by this negative allosteric modulator. The K<sub>s</sub> values of dofetilide, astemizole, sertindole and cisapride in the absence or presence of 10 µM LUF7244 are listed in **Table** 2. LUF7244 most strongly modulated cisapride binding to the K\_11.1 channel, increasing its K<sub>1</sub> value by 4.0-fold from  $21 \pm 1$  to  $85 \pm 7$  nM. Similarly, the K\_11.1 affinity of astemizole, dofetilide and sertindole were reduced by 3.2-, 2.6and 2.1-fold in the presence of LUF7244. Thus, the negative allosteric effect of LUF7244 on the K<sub>2</sub>11.1 channel significantly lowered the channel's affinity for several chemically and therapeutically distinct K<sub>y</sub>11.1 blockers.

**Table 2.** The K\_11.1 affinity (K<sub>1</sub>) of dofetilide, astemizole, sertindole and cisapride in the absence (control) or presence of 10 µM LUF7244.

	K <sub>i</sub> (nM)	K <sub>i,+10 µM LUF7244</sub> (nM)	Fold Change
Dofetilide	$6.0 \pm 0.5$	16 ± 4 (**)	2.6
Astemizole	$1.2 \pm 0.2$	$3.7 \pm 0.1 \ (***)$	3.2
Sertindole	$20 \pm 2$	$40 \pm 4 \ (**)$	2.1
Cisapride	$21 \pm 1$	85 ± 7 (***)	4.0

Values are means (± SEM) of at least three independent assays performed in duplicate. (\*\* P < 0.01 versus control; \*\*\* P < 0.001 versus control)

## Effects of LUF7244 on the binding of [3H] astemizole to the hH<sub>1</sub>R

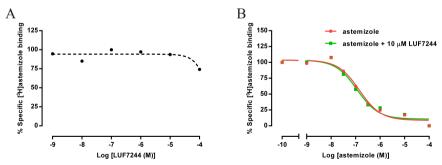
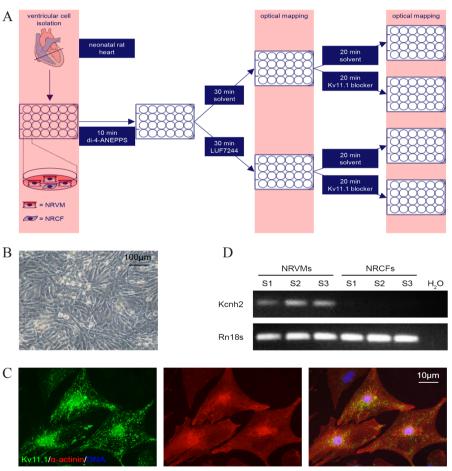


Figure 3. Assessment of the effects of LUF7244 on the binding of astemizole to the hH<sub>1</sub>R in a [3H]astemizole binding assay performed with membranes of HEK293hH, R cells. (A) Displacement curve of LUF7244. (B) Displacement curves of astemizole in the absence or presence of 10 µM LUF7244. Experiments were performed at 25 °C using 15 µg of HEK293hH<sub>1</sub>R membrane protein.

To be of practical use in preventing DiVAs, LUF7244 should not interfere with the desired activities of unintended  $K_v11.1$  blockers. We therefore tested whether LUF7244 affected binding of the antihistamine drug astemizole to the hH<sub>1</sub>R. As displayed in **Figure 3A**, LUF7244 did not decrease the binding of [ $^3$ H]astemizole to the hH<sub>1</sub>R. Furthermore, the displacement curve of astemizole binding to the hH<sub>1</sub>R was not significantly shifted rightwards by LUF7244 (**Figure 3B**). Thus, LUF7244 reduces the affinity of the  $K_v11.1$  channel for astemizole without affecting the binding of astemizole to its intended target receptor (i.e. the hH<sub>1</sub>R).

## Analysis of K, 11.1 protein expression in NRVMs



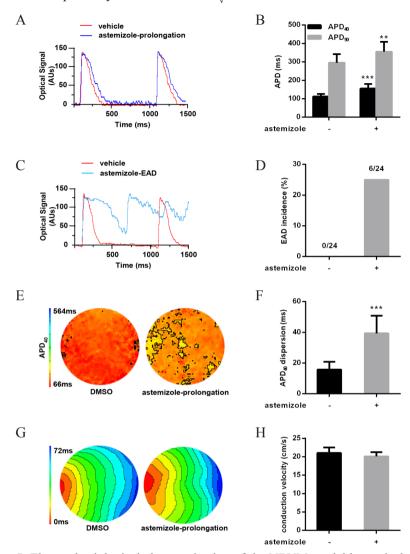
**Figure 4.** Biochemical characterization of the NRVM model. **(A)** Basic experimental setup of the optical voltage mapping experiments. **(B)** Phase-contrast image of a typical confluent NRVM monolayer used for optical voltage mapping. **(C)** Immunocytological analysis of  $K_v$ 11.1 protein expression in NRVM cultures. The  $K_v$ 11.1 protein (green) is mainly located around the nucleus (blue) and at the sarcolemma of the  $\alpha$ -actinin (red)-positive

NRVMs and hardly detectable in the  $\alpha$ -actinin-negative non-cardiomyocytes. (D) Analysis of Kcnh2 mRNA levels in NRVMs and NRCFs by RT-PCR.

Next, the electrophysiological consequences of allosteric modulation of the binding of typical K<sub>v</sub>11.1 blockers at the channel by LUF7244 (see Figure 4A) was examined in confluent monolayer cultures of NRVMs (Figure 4B) as relevant in vitro model for studying cardiac arrhythmias.<sup>22</sup> Double immunostaining for K<sub>2</sub>11.1 and sarcomeric α-actinin showed that all cardiomyocytes in the NRVM cultures expressed the K\_11.1 protein (Figure 4C). The K\_11.1 signal had a punctate or linear appearance and was concentrated around nuclei and along the sarcolemma (Figure 4C). No significant K\_11.1 protein expression was observed in the low percentage of α-actinin cells (mainly NRCFs) present in the NRVM cultures. Consistently, comparison of Kcnh2 transcript levels between NRVMs and NRCFs by RT-qPCR demonstrated 39-fold higher (n=3, P < 0.01) Kcnh2 mRNA expression in NRVMs than in NRCFs (Figure 4D).

## Electrophysiological consequences of K\_11.1 blockade by astemizole in NRVMs

Due to its relatively high specificity for the K\_11.1 channel, 23 astemizole was selected to study the electrophysiological consequences of K 11.1 blockade by optical voltage mapping. NRVMs treated with astemizole displayed a concentration-dependent lengthening of APD and EAD incidence (Figure 5 and Figure 6). As displayed in Figure 5A and 5B, the APD<sub>40</sub> and APD<sub>90</sub> values of NRVMs were increased from  $112 \pm 16$  and  $295 \pm 62$  ms in vehicle-treated cultures (n = 24) to  $156 \pm 39$  and  $355 \pm 66$  ms in cultures containing 100 nM astemizole (n = 24, P < 0.001 and P < 0.01), respectively. Furthermore, exposure to 100 nM astemizole resulted in the occurrence of EADs in 25% of the NRVM cultures, whereas no EADs were observed under control conditions (Figure 5C and 5D). Intriguingly, results presented in Figure 5E and 5F demonstrated that the APD<sub>40</sub> dispersion between NRVMs in the presence of 100 nM astemizole was significantly higher than that in its absence (39  $\pm$  11 versus 16  $\pm$  5 ms; n = 24, P < 0.001), indicative of aggravated repolarization heterogeneity due to the inhibition of  $I_{\rm Kr}$  by astemizole. Occasionally, the astemizole-induced EADs resulted in short episodes of reentrant excitation (Figure 7) reminiscent of spontaneously terminating TdP episodes. Nonetheless, as indicated by the activation maps shown in Figure 5G and corresponding quantitative analysis depicted in Figure 5H, the CV in NRVM cultures was not significantly influenced by 100 nM astemizole ( $20 \pm 4 \text{ cm} \cdot \text{s}^{-1} \text{ vs}$  $21 \pm 3$  cm·s<sup>-1</sup> in control cultures; n = 10, P = 0.48). Overall, these data validate the utility of NRVM cultures as an *in vitro* model for investigating the AP-prolonging and associated proarrhythmic effects of K\_11.1 blockers like astemizole.



**Figure 5.** Electrophysiological characterization of the NRVM model by optical voltage mapping following 1-Hz electrical stimulation. Representative filtered optical signal traces (**A** and **C**), corresponding APD dispersion and activation maps (**E** and **G**, respectively) and quantitative analyses (bar graphs in **B**, **D**, **F** and **H**) of NRVM cultures exposed for 30 min to 100 nM astemizole or vehicle (control) and subjected to optical voltage mapping immediately afterwards. Astemizole treatment resulted in APD prolongation (APD<sub>40</sub> and APD<sub>90</sub>, **A** and **B**), the occurrence of EADs (**C** and **D**) and an increase in APD<sub>40</sub> dispersion (**E** and **F**) but did not significantly alter CV (**G** and **H**). \*\* P < 0.01 and \*\*\* P < 0.001.

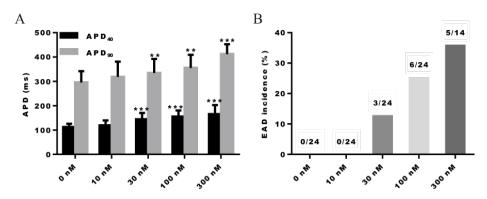


Figure 6. Assessment by optical voltage mapping of the effect of different concentrations of astemizole on the APD (APD<sub>40</sub> and APD<sub>90</sub>, **A**) and EAD incidence (**B**) in NRVM monolayers. \*\* P < 0.01 and \*\*\* P < 0.001.

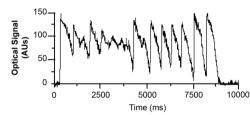


Figure 7. Optical voltage trace showing an unstable, self-terminating tachyarrhythmia in an astemizole-treated NRVM culture following 1-Hz electrical point stimulation.

# Effects of LUF7244 on K\_11.1 blockade-associated proarrhythmic changes in NRVM cultures

To investigate the functional consequences of negative allosteric modulation of K\_11.1's interaction with typical K\_11.1 blockers, di-4-ANNEPS-loaded NRVM cultures were incubated for 20 min in culture medium containing 10 µM LUF7244 before addition of astemizole, sertindole or cisapride. Since the K<sub>1</sub> values of sertindole and cisapride for K<sub>2</sub>11.1 are more than 10 times higher than that of astemizole (Table 2), these drugs were tested at a final concentration of 1 μM instead of 100 nM as was used for astemizole. After incubation for 30 min, the NRVMs were optically mapped. As shown in **Figure 8A**, the APD-prolonging and EAD-promoting effects of astemizole were effectively suppressed by LUF7244. The APD<sub>40</sub> and APD<sub>90</sub> were significantly shortened from  $156 \pm 39$  ms to  $118 \pm 18$ ms (n = 24, P < 0.001) and from  $355 \pm 66$  ms to  $282 \pm 63$  ms (n = 24, P < 0.01), respectively, i.e. LUF7244 was able to reduce APD<sub>40</sub> and APD<sub>60</sub> to control values (Figure 8B). Moreover, in the presence of 10 μM LUF7244 EADs were no longer observed in NRVM cultures exposed to 100 nM astemizole (Figure 8C).



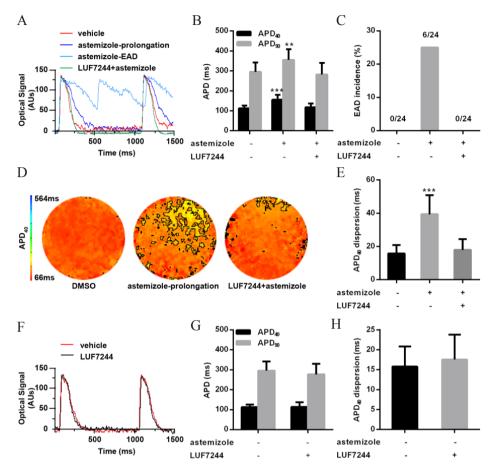


Figure 8. Assessment by optical voltage mapping of the ability of LUF7244 to counteract the proarrhythmic effects of astemizole on NRVMs. Representative filtered optical signal traces (A and F), APD dispersion maps (D) and quantitative analysis (bar graphs in B, C, E, G and H) of control NRVM cultures and of NRVM cultures that had been treated with 100 nM astemizole or with 10 µM LUF7244 followed by 100 nM astemizole immediately before optical mapping. Pretreatment of NRVM cultures with LUF7244 completely prevented the astemizole-induced APD prolongation (APD $_{40}$  and APD $_{90}$ , **B**), occurrence of EADs (C) and increase in APD<sub>40</sub> dispersion (D and E). Treatment of NRVM cultures with 10 μM LUF7244 only did not change AP morphology (F), APD (G) or APD<sub>40</sub> dispersion (H). \*\* P < 0.01 and \*\*\* P < 0.001.

LUF7244 also prevented the increase in APD<sub>40</sub> dispersion caused by astemizole (Figure 8D and 8E). Reassuringly, the CVs in the NRVM cultures treated with vehicle, LUF7244, astemizole or LUF7244 plus astemizole did not significantly differ (data not shown). Also, at a final concentration of 10 µM LUF7244 per se did not reduce APD or significantly affect APD<sub>40</sub> dispersion (Figure 8F-8H), suggesting that this negative allosteric modulator poses very little, if any, risk for the development of arrhythmias associated with abnormal APD shortening. Very similar results of LUF7244 were obtained in NRVM cultures exposed to sertindole or cisapride (Figure 9). LUF7244 can thus suppress the proarrhythmic side effects of drugs from different therapeutic classes by allosteric modulation of the K 11.1 channel without exerting, by itself, any obvious adverse electrophysiological effects on cardiomyocytes.

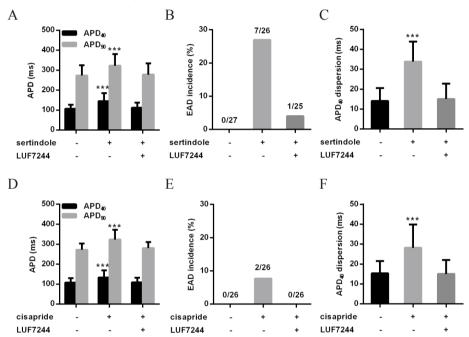
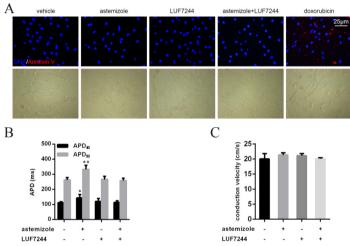


Figure 9. Assessment by optical voltage mapping of the ability of LUF7244 to counteract the proarrhythmic effects of sertindole and cisapride on NRVMs. Pretreatment of NRVM cultures with LUF7244 completely prevented the increase in APD (APD<sub>40</sub> and APD<sub>90</sub>, A and D), EAD incidence (B and E) and APD40 dispersion (C and F) induced by exposure of the cells to 1  $\mu$ M sertindole or cisapride. \*\* P < 0.01 and \*\*\* P < 0.001.

# Effects of prolonged exposure of NRVMs to LUF7244 and/or astemizole on cell viability, excitability and contractility

To study the effects of prolonged allosteric modulation and/or blockade of the K\_11.1 channel on the viability of cardiomyocytes, 6-day-old NRVM cultures were incubated for 72 h in culture medium containing 10 μM LUF7244 and/or 100 nM astemizole. Subsequent staining with fluorescently labeled annexin V to detect cell surface-exposed phosphatidylserine yielded only very weak signals for cells exposed to vehicle, LUF7244, astemizole or LUF7244 plus astemizole, while NRVMs incubated for 24 h with the proapoptotic drug doxorubicin displayed strong annexin V fluorescence (**Figure 10A**). These results indicate that prolonged exposure to 10  $\mu$ M LUF7244 alone or in combination with 100 nM astemizole does not trigger programmed cell death.

Optical voltage mapping of 9-day-old NRVM cultures that had been mock-treated or treated for 72 h with 10  $\mu$ M LUF7244 and/or 100 nM astemizole produced very similar results as were obtained for cells that had only briefly been exposed to these compounds (compare **Figure 5** and **8** with **Figure 10B** and **10C**) and provided no indications for a drug-related reduction in excitability. Moreover, after electrical field stimulation cells in all four treatment groups displayed rhythmic contractions following the pacing frequency of 1 Hz (data not shown).



**Figure 10.** Effects of prolonged exposure of NRVMs to LUF7244 and/or astemizole on cell viability and excitability. **A:** Analysis by Alexa-568-conjugated annexin V staining of cell surface expression of phosphatidylserine as early marker of apoptosis. NRVM cultures exposed for 72 hours to vehicle, 10 μM LUF7244, 100 nM astemizole or 10 μM LUF7244 plus 100 nM astemizole showed very weak signals, while NRVMs incubated for 24 hours with 1 μM of the proapoptotic drug doxorubicin displayed strong annexin V fluorescence. **B:** Quantitative comparison of APDs (APD<sub>40</sub> and APD<sub>90</sub>) in NRVM cultures exposed for 72 hours to vehicle, 10 μM LUF7244, 100 nM astemizole or 10 μM LUF7244 plus 100 nM astemizole. **C:** Quantitative comparison of CV in NRVM cultures exposed for 72 hours to vehicle, 10 μM LUF7244, 100 nM astemizole or 10 μM LUF7244 plus 100 nM astemizole.

## Discussion

## Major findings

In radioligand binding assays the structurally related compounds VU0405601, ML-T531 and LUF7244 were found to weaken the interaction between the human K, 11.1 channel and the class III antiarrhythmic agent dofetilide as well as the unintended K<sub>2</sub>11.1 blockers astemizole, sertindole and cisapride. VU405601, ML-T531 and LUF7244 exerted their negative effects on the binding of typical K\_11.1 blockers to the channel's central cavity by an allosteric mechanism. Importantly, LUF7244 decreased the K\_11.1 affinity of astemizole without influencing its affinity at the hH<sub>1</sub>R, being astemizole's intended target. Optical voltage mapping showed that incubation of NRVM monolayers with astemizole, sertindole or cisapride led to a significant increase in APD, APD dispersion and, except for cisapride, EAD incidence demonstrating the usefulness of this cellular model system for studying K\_11.1 blockade-related proarrhythmic risk. Pretreatment of the NRVMs with LUF7244 effectively prevented the proarrhythmic changes induced by astemizole, sertindole and cisapride without significantly shortening APD by itself and without adversely affecting NRVM viability, excitability and contractility. These findings provide a rationale for further exploring allosteric modulation as a strategy to prevent DiVAs.

## NRVM monolayers as model for studying drug-induced LQTS

Many different methods have been exploited to assess drugs for their K\_11.1 blockade-associated arrhythmogenicity.<sup>3, 24</sup> The interaction between a drug and the K\_11.1 channel is usually first investigated by computational and biochemical assays. Next, the K<sub>v</sub>11.1 liability of suspicious drugs are typically evaluated by electrophysiological measurements on CHO or HEK293 cells expressing the human K<sub>2</sub>11.1 channel. Despite their practical advantages, these non-excitable cellular models do not recapitulate the complex regulatory circuits governing K<sub>v</sub>11.1 channel activity in cardiomyocytes and are unsuitable for studying AP generation and propagation. We hence employed NRVM monolayers to study the effects of LUF7244 on the proarrhythmic potential of three inadvertent K\_11.1 blockers. The reasons to specifically use NRVMs for this purpose are that these cells (i) can be relatively easily isolated and cultured, (ii) are well-characterized and (iii) do express functional K<sub>v</sub>11.1 channels.<sup>25</sup> However, to the best of our knowledge, it is the first time that this model is being used to test compounds for their ability to prevent DiVAs resulting from unintended K, 11.1 blockade.

Although mechanistic insight into DiVAs is limited, there is broad consen-

sus about spatial dispersion of repolarization and EAD-induced triggered activity providing the substrate and trigger for the genesis of drug-induced TdP, respectively. The EADs typically arise during phase 2 of the cardiac AP due to drug-dependent decreases in  $I_{\rm Kr}$  causing increases of APD and QT interval. In support of the validity of our model, astemizole dose-dependently increased the APD and EAD incidence in monolayers of NRVMs (**Figure 6**). Also sertindole and cisapride had APD-prolonging effects (**Figure 9A** and **9D**) and each of the three fortuitous  $K_v$ 11.1 blockers increased APD dispersion (**Figure 5F**, **Figure 9C** and **9F**). While 100 nM astemizole, 1  $\mu$ M sertindole and 1  $\mu$ M cisapride increased APD and APD dispersion to a similar extent, cisapride did not significantly increase EAD incidence in contrast to the other two drugs. This finding is in line with a previous study<sup>26</sup> and may be explained by cisapride's inhibitory effect on the  $I_{\rm CaL}$  in NRVMs.

Initially, EAD-dependent ectopic activity at multiple competing foci was thought to generate the undulating electrocardiographic patterns of TdP. Recently, meandering  $I_{Cal}$ -mediated reentrant circuits initiated by EADs at single foci have been proposed as an alternative explanation for the highly characteristic electrocardiographic signature of TdP.<sup>27, 28</sup> In support of the latter idea, exposure of NRVMs to K, 11.1 blockers occasionally gave rise to unstable, self-terminating reentrant tachyarrhythmias reminiscent of spontaneously terminating TdP episodes. Pretreatment with LUF7244 rendered NRVM monolayers unsusceptible to the K\_11.1-blocking effects of astemizole, sertindole and cisapride. In the presence of LUF7244, these K\_11.1 blockers no longer caused heterogeneous APD prolongation and, due to the reduced opportunity for L-type Ca<sup>2+</sup> channel reactivation, no longer gave rise to EADs. This raises the perspective to use LUF7244 as an antiarrhythmic additive to drugs with unintended  $I_{\kappa r}$ -suppressing effects. Thus, radioligand binding assays in combination with optical voltage mapping experiments of NRVM cultures offer convenient preclinical test systems for evaluating chemical entities that can potentially reduce K<sub>v</sub>11.1-related cardiotoxicity.

#### LUF7244's mode of action

The chemical structure of LUF7244 resembles those of ML-T531 and VU0405601. In a recent study, 10  $\mu$ M of ML-T531 was shown to reduce the APD of human induced pluripotent stem cells (hiPSC)-derived cardiomyocytes from an LQT1 patient to that of control cells by augmenting  $I_{\rm Kr}^{-11}$  However, the effects of ML-T531 on the APD of hiPSC-derived cardiomyocytes from a healthy individual and the ability of ML-T531 to inhibit the APD-prolonging effects of unintended  $K_{\nu}11.1$  blockers were not investigated. Voltage clamp recordings of  $K_{\nu}11.1$ -ex-

pressing CHO cells showed that ML-T531 reduces the deactivation rate of the K\_11.1 channel and causes a shift of its inactivation curve towards more positive voltages. Shortly after the discovery of ML-T531, VU0405601 was identified as a compound that, at a final concentration of 5 µM, protected Langendorff-perfused rabbit hearts from the proarrhythmic effects of exposure to 100 nM dofetilide. Although VU0405601 only partially reversed the dofetilide-dependent increase in APD, its administration prior to dofetilide strongly reduced the pacing-induced arrhythmia incidence from 42% to 4%, which was very close to the 2% of pacing-induced premature ventricular contractions observed in untreated hearts. Exposure of isolated rabbit ventricular myocytes to 5 µM of VU0405601 only marginally reduced APD, which is consonant with our finding that 10 µM LUF7244 did not noticeably affect the APD of NRVMs. However, at a final concentration of 50  $\mu$ M, VU0405601 decreased the APD<sub>50</sub> and APD<sub>90</sub> of rabbit ventricular myocytes by  $35 \pm 6\%$  and  $32 \pm 4\%$ , respectively. Patch clamp analysis of HEK293K, 11.1 cells linked the APD-shortening effect of 50 µM VU0405601 to shifts in the V<sub>1</sub> of activation and inactivation and to changes in the kinetics of (de)activation and (de)inactivation causing an increase in  $I_{\rm Kr}$ .

In this study, we found that VU0405601, ML-T531 and LUF7244 displayed comparatively low K 11.1 affinity with Hill coefficients significantly different from unity for their [3H]dofetilide displacement curves. This suggests that these ligands bind to the K 11.1 channel at sites distinct from that of dofetilide, indicative of an allosteric mode of action. 10, 29, 30 The binding of a drug to a receptor at an allosteric site (i.e. a site topologically distinct from that of the test ligand) triggers a conformational change within the receptor ultimately causing an alteration of the ligand's dissociation rate from its cognate (i.e. orthosteric) binding site. 10, <sup>31</sup> Altered ligand dissociation rates have been found representative of allosteric interactions in various drug targets such as muscarinic and adenosine receptors. 10, <sup>32-34</sup> Consistently, VU0405601, ML-T531 and LUF7244 significantly accelerated the dissociation of dofetilide from the K\_11.1 channel, strengthening the conclusion that they are negative allosteric modulators of dofetilide binding to the K<sub>v</sub>11.1 channel. Our finding for VU0405601 is in agreement with the results of Potet et al., who presented indirect evidence that VU0405601 binds from outside to the ectodomain of the K<sub>2</sub>11.1 channel, while astemizole, cisapride, dofetilide and sertindole all bind to the channel's central cavity from inside<sup>35-37</sup>. In addition, the [3H]dofetilide displacement curves of dofetilide, astemizole, sertindole and cisapride were shifted rightwards by LUF7244 (Figure 2C and 2D) providing further proof for its negative allosteric effect. 10, 34 Collectively, the results of the different radioligand binding assays provide strong evidence that VU0405601, ML-T531 and LUF7244 are negative allosteric modulators of dofetilide binding to the  $K_v$ 11.1 channel. Binding of VU0405601, ML-T531 and LUF7244 likely alters the 3D structure of the  $K_v$ 11.1 channel, which decreases its affinity for typical  $K_v$ 11.1 blockers by increasing the dissociation rates of these blockers from the channel. Of note, the vast majority of fortuitous  $K_v$ 11.1 blockers exert their effects by occupying the central cavity of the  $K_v$ 11.1 channel and thereby obstructing the transport of  $K^+$  ions through the channel's pore. There are, however, also examples of drugs that reduce  $I_{Kr}$  by inhibiting, directly or indirectly, the trafficking of  $K_v$ 11.1 to the plasma membrane. Given their specific mode of action, it is unlikely that the  $I_{Kr}$ -inhibiting effects of these drugs can be abolished by LUF7244 or a related compound.

Since both VU0405601 and ML-T531 dose-dependently increased the K\_11.1 current density and given the structural similarity of these compounds with LUF7244, it is very likely that also LUF7244 can directly act as a (allosteric) K\_11.1 activator. Accordingly, the ability of LUF7244 to counteract astemizole-, sertindole- and cisapride-related arrythmogenesis may be the combined result of its inhibitory effect on the binding of these unintended K\_11.1 blockers to the channel and of its direct enhancing effect on the K\_11.1 channel's activity. However, the fact that exposure of NRVMs to 10 µM LUF7244 alone did not significantly change AP characteristics suggests that LUF7244's K\_11.1-activating activity is not of critical importance for its ability to suppress the proarrhythmic effects of inadvertent K\_11.1 blockers. Moreover, possible effects of LUF7244 on other cardiac ion channels besides the K<sub>2</sub>11.1 channel must be considered. The absence of a noticeable change in AP shape and duration following exposure of NRVMs to LUF7244 also argues against a possible effect of this allosteric modulator on cardiac ion channels different from the  $K_v11.1$  channel. In keeping with this notion,  $Na_v 1.5$ ,  $I_{Ks}$  and  $K_v 1.5$  activities were not affected or only slightly reduced by 50 μM VU0405601.7 Likewise, ML-T531 at a final concentration of 10  $\mu$ M had a minor suppressive effect on  $I_{Ks}$  and did not influence Na<sub>v</sub>1.5, Ca<sub>v</sub>1.2, K<sub>v</sub>4.3 or Kir2.1 activities.<sup>11</sup> Thus, the antiarrhythmic propensity of LUF7244 is dominated by its negative allosteric impact on the binding of typical K<sub>v</sub>11.1 blockers to the channel.

The fact that LUF7244 significantly decreased the K<sub>v</sub>11.1 affinity of drugs with very different chemical structures (**Figure 2B**) in membrane radioligand binding assays and prevented these drugs from causing APD prolongation in NRVMs makes it likely that LUF7244 will be effective in reducing the cardiotoxicity of a broad range of K<sub>v</sub>11.1 blockers. Further support for this notion comes from the fact that VU0405601, which was here shown to inhibit dofetilide interaction with the K<sub>v</sub>11.1 channel by a similar mechanism to LUF7244, could abolish the blockade of K<sub>v</sub>11.1 by seven different drugs.<sup>7</sup> The different degree

to which LUF7244 increased the K, values of dofetilide, astemizole, sertindole and cisapride suggests that the sensitivity of different K 11.1 blockers to conformational changes in the K\_11.1 proteins differs. Accordingly, different LUF7244 concentrations may be required to abrogate the proarrhythmic effects of distinct K 11.1 blockers.

Although at a final concentration of 50 µM LUF7244's ability to weaken the interaction between [3H]dofetilide and the K\_11.1 channel was similar to those of VU0405601 and ML-T531 (Figure 1D), the new modulator was much more effective than the other two compounds at a final concentration of 10 μM (Figure 1C). Moreover, even 50 μM VU0405601 only modestly inhibited the APD-prolonging effect of 1 µM dofetilide (K, for human K,11.1: 6.0 nM) on rabbit ventricular myocytes, 7 while 10 μM LUF7244 totally blocked the APD prolongation caused by exposure of NRVMs to 100 nM astemizole (K, for human K, 11.1: 1.2 nM). These findings together with the substantial decrease in the APD of normal rabbit ventricular myocytes caused by 50 µM VU0405601 (see above), suggests that LUF7244 may possess a more favorable safety profile than VU0405601 or ML-T531.

## Clinical applicability of allosteric K\_11.1 modulators

A major concern and obstacle for medicinal chemists involved in the development of drugs to remedy congenital and acquired LQTS is excessive APD shortening resulting in the short QT syndrome (SQTS).<sup>5, 40</sup> For instance, at a concentration of 10 µM, the potent K 11.1 activator ICA-105574 shortened the APD in Langendorff-perfused guinea pig hearts by 34% and induced non-TdP like ventricular arrhythmias in 2 out of 8 animals. 41 Likewise, in a transgenic rabbit model of LQT1, shortening of the QTc interval by another K<sub>v</sub>11.1 activator designated NS1643 was accompanied by an increased incidence of pacing-induced ventricular fibrillation.<sup>42</sup> However, the fact that LUF7244 at a concentration that completely abolished the proarrhythmic effects of astemizole, sertindole and cisapride does not shorten the APD of NRVMs and therefore may not give rise to drug-induced short QT syndrome (SQTS) raises hope for its clinical applicability.

For the experiments in NRVM monolayers, the final astemizole concentration was 100 nM in most cases, while sertindole and cisapride were used at a final concentration of 1 µM. These concentrations are much higher than the therapeutic blood concentrations of these fortuitous K<sub>2</sub>11.1 blockers, which are 2-50 μg·L<sup>-1</sup> (i.e. 4.4-109 nM) for astimizole, 40-80 μg·L<sup>-1</sup> (i.e. 86-172 nM) for cisapride and 50-100 µg·L<sup>-1</sup> (i.e. 113-227 nM) for sertindole.<sup>43</sup> The proarrhythmic effects of these drugs on NRVMs were completely prevented by 10 μM (i.e. 3.7

mg·L<sup>-1</sup>) LUF7244, a concentration that did not negatively influence the binding of astemizole to the hH $_{\rm l}$ R and did not noticeably affect the viability, contractility and excitability of the cardiomyocytes. Accordingly, blood LUF7244 concentrations well below 10  $\mu$ M may suffice to counteract the proarrhythmic effects of unintended K $_{\rm v}$ 11.1 blockers in humans. However, before LUF7244 or other allosteric modulators can be clinically applied as "co-drugs" or otherwise, they should be subjected to rigorous preclinical and clinical testing with appropriate pharmacokinetic and pharmacodynamics assessments to determine their benefit-risk profiles.

#### Limitations

Due to the difficulty to obtain and culture adult human ventricular myocytes, in this study NRVMs were used as 2D model system to investigate the effects of LUF7244 on K\_11.1 blockade-associated proarrhythmic changes in cardiac electrophysiology. However, adult human and neonatal rat ventricular myocytes have different AP morphologies because of qualitative and quantitative differences in the molecular components shaping the APs. Also, changes in cardiomyocyte electrophysiological properties may work out differently in a 2D cell layer than in the 3D heart. Nonetheless, studies on NRVM monolayers have greatly contributed to our current understanding of cardiac electrophysiology. Moreover, in spite of the differences in ventricular ion channel composition between humans and rats, their K<sub>2</sub>11.1 proteins are very similar showing 95% amino acid identity for the largest isoforms. Consistently, the results of the radioligand binding assays, which were carried out with the human K\_11.1 protein, correlated very well with those of the optical mapping studies using NRVMs. Yet, ultimately, the ability of LUF7244 to counteract the proarrhythmic effects of unintended K<sub>2</sub>11.1 blockers should be investigated in human subjects.

As mentioned above, LUF7244 did not inhibit the binding of astemizole to its intended target in radioligand binding assays and did not compromise the viability, excitability or contractility of NRVMs at a concentration sufficient to fully abrogate the proarrhythmic consequences of drug-induced  $K_v11.1$  blockade in these cells. Despite these encouraging results, certainty about the lack of specific adverse/interfering effects of this negative allosteric modulator of  $K_v11.1$  channel in humans can only be obtained through clinical studies. Moreover, now that the ability of LUF7244 to reduce the channel binding affinity of  $K_v11.1$  blockers has been established, allosteric modulators with higher safety and/or efficacy than LUF7244 are likely to arise in the near future. The design of such compounds may benefit from the identification of the precise binding site of LUF7244 at the

K 11.1 channel.

## **Conclusions**

Allosteric modulators at the K\_11.1 channel could provide a new pharmacological treatment for drug-induced LQTS by preventing the potentially arrhythmogenic changes in AP characteristics caused by unintended K\_11.1 blockers. Through combined administration with a negative allosteric modulator, use of "old" drugs that have been banned because of their "K\_11.1 liability" may be resumed and new drugs with K\_11.1-blocking effects may not have to be excluded from clinical application.

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# Chapter 5

Synthesis and biological evaluation of negative allosteric modulators of the  $K_{\rm v}11.1$  (hERG) channel

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#### **Abstract**

We synthesized and evaluated a series of compounds for their allosteric modulation at the  $K_v11.1$  (hERG) channel. Most compounds were negative allosteric modulators of [³H]dofetilide binding to the channel, in particular **7f**, **7h-j** and **7p**. Compounds **7f** and **7p** were the most potent negative allosteric modulators amongst all ligands, dramatically increasing the dissociation rate of dofetilide in the radioligand kinetic binding assay, while remarkably reducing the affinity of dofetilide and astemizole in a competitive displacement assay. Additionally, both **7f** and **7p** displayed peculiar displacement characteristics with Hill coefficients significantly distinct from unity as shown by e.g., dofetilide, further indicative of their allosteric effects on dofetilide binding. Our findings in this investigation yielded several promising negative allosteric modulators for future functional and clinical research with respect to their antiarrhythmic propensities, either alone or in combination with known  $K_v11.1$  blockers.

#### Introduction

The past decades have witnessed an emerging interest in the identification of allosteric modulators of G protein-coupled receptors (GPCRs), ligand-gated ion channels (LGICs) and enzymes. 1-3 Allosteric modulators are ligands that bind to a binding site on a target protein that is not overlapping with and spatially distinct from the so-called orthosteric binding site, i.e. the site where e.g., hormones and neurotransmitters bind.<sup>4</sup> Negative allosteric modulators reduce receptor binding and/or function associated with orthosteric ligands, whereas positive allosteric modulators enhance these.<sup>2, 4, 5</sup> The therapeutic potential of allosteric modulators has been demonstrated in clinical trials, with a number of such drugs now on the market. Allosterically acting ligands provide novel opportunities for drug discovery due to a possible higher subtype-selectivity (allosteric sites are usually less conserved than orthosteric binding sites) and fewer side effects compared with traditional orthosteric compounds due to their 'ceiling' effect.<sup>2, 3</sup> However, research on the allosteric modulation of voltage-gated ion channels (VGICs) is largely lacking, in particular for the K<sub>2</sub>11.1 (hERG) channel.

The K\_11.1 channel, encoded by the human ether-à-go-go-related gene (hERG), plays a vital role in regulating cardiac repolarization of the action potential of human ventricular myocytes.<sup>6,7</sup> Genetic dysfunction or pharmacological inhibition of the K\_11.1 channel leads to prolongation of the action potential duration (APD), QT interval lengthening, and the development of Torsades de Pointes (TdP). A staggering array of drugs including antiarrhythmic agents, antihistamines, antibiotics and antipsychotics are known to block the K<sub>v</sub>11.1 channel via a common, promiscuous binding region within the aqueous inner cavity of the pore.<sup>8, 9</sup> More recently, a small number of compounds, referred to as K<sub>2</sub>11.1 activators, have been proposed to remediate repolarization disorders in the heart, including acquired and congenital long QT syndromes, due to their up-regulation of K<sub>2</sub>11.1 currents.<sup>7, 10, 11</sup> These activators were found to bind at sites that are distinct from each other and also different from the binding residues of prototypical K<sub>2</sub>11.1 blockers.9 For instance, RPR260243 ((3R,4R)-4-[3-(6-methoxyquinolin-4-yl)-3-oxopropyl]-1-[3-(2,3,5-trifluorophenyl)prop-2ynyl] piperidine-3-carboxylic acid) that was designated as a type 1 activator interacted with a putative binding site at the cytoplasmic end facing away from the inner cavity of the channel, while type 2 activators like NS1643 (1,3-bis[2-hydroxy-5-(trifluoromethyl)phenyl]urea) appeared to interact with residues located on the outer mouth of the pore.<sup>8, 12, 13</sup> Furthermore, some of these activators were found to reduce the affinity and subsequently reverse the action potential prolongation produced by K\_11.1 blockers. Pretreatment of cells expressing the K\_11.1

channel with RPR260243 slightly reduced the affinity of the reference hERG blocker dofetilide (N-[4-(2-{[2-(4-methane sulfonamidophenoxy)ethyl](methyl) amino ethyl) phenyl] methanesulfonamide), leading to a reduction of dofetilide's prolongation of the action potential. 12, 14 Likewise, NS3623 (1-[4-bromo-2-(2Htetrazol-5-vl)phenvl]-3-[3-(trifluoro- methyl)phenvl]urea) produced a reversal (N-[4-[1-[2-(6-Methyl-2-pyridinyl)ethyl]-4-piperidinyl]carbonyl]phenyl|methanesulfonamide dihydrochloride)-induced QT prolongation in both anaesthetized and conscious guinea pigs.<sup>15</sup> In this sense, the so-called activators might be negative allosteric modulators of the action of K\_11.1 blockers, and so normalize the pharmacologically induced action potential prolongation. Indeed, the allosteric terminology at the K 11.1 channel has been introduced in our previous study and a recent review.<sup>4,16</sup> Classic methods for screening allosteric modulators have been based on techniques used to identify K<sub>v</sub>11.1 blockers, such as voltage-dependent fluorescence experiments and laborious patch clamp assays. However, these methods do not allow high throughput screening in the search for new hit and lead molecules,.<sup>17</sup> Moreover, although negative allosteric modulators of the K 11.1 channel may be therapeutically promising, the current generation of modulators is not very potent and this raises concerns about their selectivity and liability for cardiac and non-cardiac side effects. 9, 11 Therefore, it would be of particular importance to design and synthesize more novel allosteric modulators with higher potencies at the K 11.1 channel.

Recently, Zhang et al. characterized a potent modulator (ML-T531, compound 7a in this study) for the K\_11.1 channel after screening a large compound library using an automated electrophysiological technique, and found that it normalized APD prolongation induced by dysfunction of the K<sub>2</sub>7.1 channel.<sup>18</sup> This was the first experimental evidence that K\_11.1 modulators can prevent long QT type 1 (LQT1) in addition to type 2 (LQT2) syndrome. The concept that K\_7.1 and K\_11.1 activators can mutually rescue K\_7.1-related LQT1 and K\_11.1-induced LQT2 syndromes has been put forward previously, and was also validated by using a highly selective K<sub>2</sub>7.1 modulator, R-L3 ((3R)-5-(2-fluorophenyl)-3-(1*H*-indol-3-ylmethyl)-1-methyl-3*H*-1,4-benzodiazepin-2-one)), pressed early afterdepolarizations initiated by the K\_11.1 blocker dofetilide in rabbit myocytes. 17, 19, 20 Apart from ML-T531, a chemically similar compound (VU0405601, compound 7r) was discovered in a fluorescence-based thallium influx assay, and this ligand allosterically diminished the affinity of dofetilide at the channel and relieved the arrhythmia induced by dofetilide. 21 Altogether, negative allosteric modulators of the K<sub>x</sub>11.1 channel could be beneficial in LQT syndromes induced by genetic loss-of-function or pharmacological inhibition of both K\_11.1 and K\_7.1 channels. In the present study, we selected ML-T531 and

VU0405601 as our starting point, and synthesized a series of new ligands with modifications at the three aromatic rings present in the basic scaffold of the lead compounds. Based on the results from a previously validated single point dissociation assay, 16 their structure-activity relationships were analyzed. Subsequently, a selection of potent negative allosteric modulators were comprehensively characterized in [3H]dofetilide dissociation and displacement assays, and their influences on affinity of two specific K\_11.1 blockers (dofetildie and astemizole (1-[(4-fluorophenyl)methyl]-N-[1-[2-(4-methoxyphenyl)ethyl]- 4-piperidyl]benzoimidazol-2-amine)) were evaluated as well.

#### Results and discussion

## **Chemistry**

The synthetic route of all final compounds is depicted in **Scheme 1** except that compound 7r was synthesized as previously described. 21 Small systematic chemical variations were made by introducing different functional groups at the three aromatic rings of the reference compound (ML-T531, 7a). Compounds based on modifications at the two phenyl rings (7a-p) were synthesized as follows. Friedel-Crafts acylation of substituted anisols (1a-c) and benzoyl chlorides (2d-l and **2n-p**) resulted in the formation of different substituted-benzophenone derivatives (3b-l and 3n-p).<sup>22</sup> Apart from acid chlorides 20 and 2p that were synthesized from the corresponding carboxylic acids 80 and 8p (Scheme 2), all other intermediates were commercially available. Demethylation of the methoxy group for compounds 3b-l and 3n-p with AlCl, or BBr, in the case of 3c led to 4b-l and **4n-p** with high yields (65-97%).<sup>22, 23</sup> Notably, demethylation of **3c** using AlCl, conditions resulted in the undesired debrominated product 4a. Subsequently, reaction of 4a-p with 2-methylbromoacetate and hydrolysis of 5a-p with LiOH produced compounds 6a-p. Eventually, final compounds 7a-p were obtained via a peptide coupling, using EDCI\*HCl (7e and 7l) or the superior HATU, between the acids (6a-p) and 3-aminopyridine in yields of 23-98%.<sup>21</sup> Fluorenone analogue 7q was synthesized as follows: i) an oxidative dehydrogenative cyclization of benzophenone 4a resulted in fluorenone intermediate 4q, and ii) subsequent HATU peptide coupling conditions led to the formation of final compound 7q.<sup>24</sup> To further investigate the influences of the 3-pyridine ring on the allosteric modulation capacities of ligands, a variety of substituted 3-aminopyridines (7f1-f8) and other nitrogen-containing heterocyclic rings (7f9-f11) were synthesized in a similar approach as described above. All intermediates were commercially available except intermediate 10f5, which was synthesized from 5-bromo-3-methoxypyridine (9f5) through an amination using ammonia and copper sulphate pentahydrate (Scheme 2).<sup>25</sup>

Scheme 1. Synthesis of compounds 7a-q and 7f1-f11.

a) AlCl<sub>3</sub>, DCM, 0 °C to r.t., 2 h; b) AlCl<sub>3</sub>, toluene, 130 °C, 2 h; c) 2-methylbromoacetate, K,CO,, DMF, 65 °C, overnight; d) LiOH, THF, MeOH, 105 °C, 30 min; e) 3-aminopyridine, Et,N, HATU or EDCI\*HCl (7e and 7l), DMF, overnight.

Scheme 2. Synthesis of intermediates (20, 2p and 10f5)

a) SOCl<sub>2</sub>, 100 °C, 2h; b) NH<sub>3</sub> (aq), CuSO<sub>4</sub>·5H<sub>2</sub>O, 140 °C, 20 h.

## Structure-activity relationships

Initially, all synthesized compounds were tested for their abilities to increase or decrease the dissociation of [3H]dofetilide from the HEK293K\_11.1 cell membrane preparations at 10 µM in a single point dissociation assay, 16 and the results are summarized in **Table 1-3**. Compound **7a**, also referred to as ML-T531 in literature, has been reported to be the most potent K\_11.1 activator, which normalizes the prolonged APDs of patient-derived cardiomyocytes. 18 Thus, a series of compounds with different substituents on the phenylcarbonyl ring were synthesized and evaluated for their allosteric modulation at the K\_11.1 channel (Table 1). Similar to compound 7a, all compounds behaved as negative allosteric modulators of the K 11.1 channel as they significantly increased the dissociation of dofetilide from the channel except 71 and 7n. The latter observation is in agreement with the previous finding that the effect of 7n on K, 11.1 currents was insignificant and negligible, 18 indicative of the reliability of our high-throughput single point dissociation assay. Introduction of halogens at the ortho position (7d, 7e and 7g) did not significantly impact the negative allosteric effect of the lead compound 7a, whereas halogen substituents at the meta position (7h-k) displayed a more prominent enhancement of dofetilide dissociation compared to 7a. Notably, introduction of a bromo substituent at the ortho position (7f) dramatically decreased the relative binding of dofetilide to  $44 \pm 1\%$  from  $78 \pm 3\%$  for 7a, which implicated that ortho-bromo derivatives could be the starting point for designing more potent allosteric modulators of the K\_11.1 channel in the following step. With respect to the para position, methyl and chloro substituents (71 and 7n) totally abolished the negative allosteric modulation properties of this series of compounds, while the 4-fluoro derivative slightly increased the allosteric effect (7m versus 7a). The bigger size of the methyl and chloro substituents at the para-position may be the reason for this decrease in effect. Additionally, electron donating (71) and withdrawing (7n) groups seemed to reduce the allosteric profiles of 7a to the same extent, implying a negligible role of electronic effects in affecting allostery of compounds at the K\_11.1 channel. When bromo and fluoro substituents were simultaneously introduced to the ortho and meta positions, respectively, additional allosteric inhibition was observed for 7p but 7o exerted a much lower allosteric effect compared to compounds 7f and 7h. This confirms that steric hindrance at the phenyl ring might not be beneficial for the potency of these negative allosteric modulators.

**Table 1.** Allosteric modulation of [<sup>3</sup>H]dofetilide binding at the K\_11.1 channel by 10 μM of compounds 7a and 7d-p.

Compd	R	%B/B <sub>control</sub>	Compd	R	%B/B <sub>control</sub>
7a	Н	78 ± 3	7j	3-Br	$57 \pm 5$
7 <b>d</b>	2-F	$85 \pm 4$	7k	3-I	$62 \pm 3$
7e	2-C1	$71 \pm 2$	71	$4-CH_3$	$97 \pm 5$
<b>7</b> f	2-Br	$44 \pm 1$	7 <b>m</b>	4-F	$65 \pm 2$
7g	2-I	$72 \pm 2$	7n	4-C1	$104 \pm 2$
7h	3-F	$47 \pm 2$	<b>7o</b>	2-Br, 3-F	$85 \pm 1$
7i	3-C1	$43 \pm 2$	<b>7</b> p	2-Br, 5-F	$38 \pm 3$

Since compounds with a bromo substituent (7f and 7p) at the ortho position of the phenylcarbonyl ring were among the most potent modulators, a variety of compounds with the same ortho-bromo substitution but different nitrogen-containing heterocyclic aromatic rings were designed to investigate the influences of the pyridine moiety on allosteric modulation of the K\_11.1 channel. As displayed in Table 2, none of these compounds showed more prominent negative allosteric activities than the starting compound 7f. By contrast, introduction of methyl, methoxyl, chlorine and phenyl groups amongst 3-pyridyl analogues (7f1f8) abrogated the negative allosteric effects, and compounds 7f2 and 7f4 even appeared to decrease the dissociation of dofetilide albeit not significantly. It is worth noting that substituents with opposite electronic effects such as 7f1 and 7f3 could not be distinguished with respect to their allosteric activities, which is in agreement with our earlier finding for derivatives 71 and 7n shown in Table 1. This further proves that electronic effects were not the determinants of allosteric characteristics for this series of ligands. In addition, replacement of the pyridine ring with chlorine-substituted diazine (7f9), pyrimidine (7f10) and pyrazine (7f11) moieties led to significant decreases of the negative allosteric effects on the K<sub>v</sub>11.1 channel compared to 7f. Altogether, a non-substituted pyridine moiety is preferred for binding to the allosteric sites at the K<sub>2</sub>11.1 channel. As compound 7r (VU0405601) was found to significantly increase the IC<sub>70</sub> value of dofetilide and thus prevent its K<sub>y</sub>11.1 inhibition,<sup>21</sup> substituents at the phenolic ring were also explored in Table 3. Introducing bromine at the phenolic ring wholly eliminated their allosteric effects (7b-c versus 7a), whereby linkage of the two phenyl rings

(7q) resulted in a comparably negative allosteric potency to the lead compound 7a. Consistent with the previous publication, 21 substitution of the general benzophenone moiety by a bromine-substituted naphthyl ring moderately increased the negative allosteric effect (7r versus 7a).

**Table 2.** Allosteric modulation of [<sup>3</sup>H]dofetilide binding at the K<sub>2</sub>11.1 channel by 10 μM of compounds 7f1-f11.

Compd	R	%B/B <sub>control</sub>	Compd	R	%B/B <sub>control</sub>
7f1		$105 \pm 12$	7 <b>f</b> 7		$105 \pm 12$
7f2		$124 \pm 13$	7 <b>f</b> 8		87 ± 12
7f3	CI	95 ± 12	7 <b>f</b> 9	CI N. N. z.c.	$92 \pm 2$
7 <b>f</b> 4	N Joseph Market	121 ± 15	7f10	N P	87 ± 9
7f5	O contract	$107 \pm 15$	7f11	N	92 ± 2
7f6	N Jes	$108 \pm 16$			

Taken together, all three aromatic rings played critical roles in modifying the allosteric effects of this series of compounds at the K<sub>y</sub>11.1 channel. In general, introducing halogen substituents with comparatively less steric hindrance at the phenylcarbonyl ring, and avoiding substituents at the pyridine and phenolic rings were favorable for enhancing the negative allosteric potencies of these modulators. Among all compounds shown in Table 1-3, 7f, 7i and 7p were the most potent negative allosteric modulators by reducing [3H]dofetilide binding during dissociation to  $44 \pm 1$ ,  $43 \pm 2$  and  $38 \pm 3\%$  respectively, which demonstrated their higher potencies than reference compounds 7a ( $78 \pm 3\%$ ) and 7r ( $66 \pm 3\%$ ).

Table 3. Allosteric modulation of [3H]dofetilide binding at the K\_11.1 channel by 10 μM of compounds 7b, 7c, 7q and 7r.

Compd R %B/B<sub>control</sub> Compd R %B/B<sub>control</sub>

7b 
$$96 \pm 1$$
 7q  $79 \pm 6$ 

7c  $96 \pm 7$  7r  $8r$   $66 \pm 3$ 

Subsequently, several of the more potent allosteric modulators were selected in [3H]dofetilide dissociation and displacement assays to exploit their pharmacological characteristics at the K<sub>2</sub>11.1 channel in more detail. Since the disintegration characteristics of a radioligand-receptor complex can only be altered by the binding of a compound to a site distinct from the radioligand binding site, the effects of these synthesized compounds on the dissociation rate of [3H]dofetilide can be unequivocally indicative of their allosteric actions.<sup>26</sup> In this respect, the dissociation behavior of [3H]dofetilide was investigated in the absence (control) or presence of potent modulators (Figure 1 and Table 4). As concentration-dependent allosteric modulation has been observed in a plethora of receptors, 5, 27-29 we evaluated a higher concentration (50 mM) of modulators in this assay, also to assess whether we could surpass the effects measured at 10 µM. The dissociation rate of the radioligand induced by an excess concentration of unlabeled dofetilide alone was  $0.19 \pm 0.01$  min<sup>-1</sup>, which was significantly increased by the selected compounds (7f, 7h-j and 7p), once more illustrating their negative allosteric modulation of the K<sub>2</sub>11.1 channel. Compound 7p was the most potent modulator that accelerated the  $k_{off}$  of dofetilide to  $0.78 \pm 0.20$  min<sup>-1</sup>, while 7j was least efficacious with a dissociation rate of  $0.27 \pm 0.03$  min<sup>-1</sup> for dofetilide. The ranking order of all five compounds in enhancing the dissociation rate of dofetilde was identical to the one obtained in the single point dissociation assay except for compound 7f, which had a more prominent negative allosteric action in this assay. Next, the concentration-dependent effects of two representative compounds at increasing the dissociation of [3H]dofetilide from the K<sub>v</sub>11.1 channel were assessed as in Figure 2, and the determined EC<sub>50</sub> values for 7f and 7p were  $12 \pm 2$ 

and  $4.6 \pm 0.4 \mu M$ , respectively. This is in agreement with their activities in the other kinetic assays. It should be mentioned that a full concentration-effect curve of 7f could not be recorded due to its limited solubility at higher concentrations. Furthermore, there was still 25% [3H]dofetilide binding at the K\_11.1 channel left in the presence of 100  $\mu$ M 7p, which indicates that the binding of  $K_{\nu}11.1$  blockers cannot be completely displaced via conformational changes of the channel caused by these negative allosteric modulators. More recently, curve-shifts that deviate from a simple competitive interaction at an equilibrium situation were revealed to be indicative of allostery.<sup>4</sup> Consequently, we determined the effect of the potent modulators (7f, 7h-i and 7p) on the equilibrium affinity of two prototypical K\_11.1 blockers, dofetilide and astemizole (Figure 3 and Table 5).

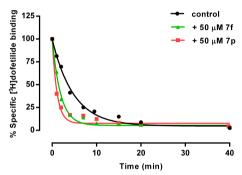


Figure 1. Dissociation curves of [3H]dofetilide by 10 μM dofetilide in the absence (control) or presence of 50 µM 7f and 7p. Experiments were performed at 25 °C with 20 µg HEK293K 11.1 membranes.

Table 4. Dissociation rates of [3H]dofetilide in the absence (control) or presence of 50 μM 7f, 7h-j and 7p, and EC<sub>50</sub> values of two representative compounds (7f and 7p) at accelerating dissociation of [3H]dofetilide from the K\_11.1 channel.

Compd	$\mathbf{k}_{\mathrm{off,dofetilide}}(\mathbf{min}^{\text{-1}})$	fold	$EC_{50}(\mu M)$
control	$0.19 \pm 0.01$	-	-
+ 7 <b>f</b>	$0.57 \pm 0.05***$	3.0	$12 \pm 2$
+ 7h	$0.30 \pm 0.04$ *	1.6	-
+ 7i	$0.33 \pm 0.03**$	1.7	-
+ 7j	$0.27 \pm 0.03*$	1.4	-
+ 7 <b>p</b>	$0.78 \pm 0.20$ *	4.1	$4.6\pm0.4$

Values are means (± SEM) of at least three independent experiments performed in duplicate (\* P < 0.05, \*\* P < 0.01, \*\*\* P < 0.001 versus control).

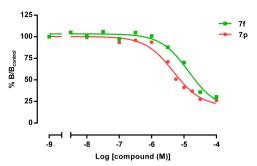


Figure 2. Concentration-dependent effects of 7f and 7p in accelerating dissociation of [3H]dofetilide from the K 11.1 channel. Membrane proteins were first pre-equilibrated with [3H]dofetilide, then the dissociation was induced by 10 µM dofetilide in the absence or presence of different concentrations of compounds and the incubation was terminated after 6 min. The results are expressed as the ratio of the specific binding of [3H]dofetilide in the presence of 10 µM dofetilide plus various concentrations of negative allosteric modulators (B) over that in the presence of 10  $\mu$ M dofetilide alone ( $B_{control}$ ). Experiments were performed at 25 °C with 20 μg HEK293K, 11.1 membranes.

As shown in Figure 3A and 3B, the displacement curves of both dofetilide and astemizole were right-ward shifted by compound 7f and 7p at 10 µM, which is another indication of the negative allosteric properties of 7f and 7p. The affinity of dofetilide ( $K_1 = 4.8 \pm 0.5 \text{ nM}$ ) and astemizole ( $K_1 = 1.3 \pm 0.1 \text{ nM}$ ) were comparable to our previous findings,30 and their apparent K, values were increased in the presence of all tested compounds. The  $K_1$  values of dofetilide shifted to 6.8  $\pm$ 0.5 nM (7j) and  $22 \pm 5$  nM (7p), while the values for astemizole were  $1.8 \pm 0.5$ nM (7j) and  $8.2 \pm 1.5$  nM (7p). Apart from compound 7f, the fold increase in K values for dofetilide and astemizole by these modulators was very comparable, and fully in line with our findings in the other dissociation assays. This diminishing of the affinity of K\_11.1 blockers through negative allosteric modulation is very appealing and may even help in mitigating acquired LQT syndromes via relieving drug-induced K\_11.1 blockade. The [3H]dofetilide equilibrium displacement curves of 7f and 7p shown in Figure 4 were rather steep with pseudo Hill coefficients (-1.8 for both compounds) much larger than unity, further indicating their allosteric actions on dofetilide binding at the K\_11.1 channel.<sup>29</sup> For the sake of comparison, the displacement curve of dofetilide was also included in Figure 4, and the derived Hill slope was equal to -1.0 illustrating the competitive and 'orthosteric' binding of dofetilide and the reliability of this assay.

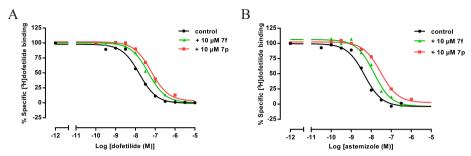


Figure 3. Displacement curves of dofetilide (A) and astemizole (B) in the absence (control) or presence of 10 µM 7f and 7p in a [3H]dofetilide binding assay. Experiments were performed at 25 °C with 20 µg HEK293K, 11.1 membranes.

Table 5. The K<sub>v</sub>11.1 ffinities of dofetilide and astemizole in the absence (control) or presence of 10 µM 7f, 7h-i and 7p.

Compd	$K_{i, dofetilide}(nM)$	fold	$K_{i, astemizole}(nM)$	fold
control	$4.8 \pm 0.5$	-	$1.3 \pm 0.1$	-
+ 7f	$12 \pm 3*$	2.5	$5.3 \pm 1.3*$	4.1
+ 7h	$8.9 \pm 0.8*$	1.9	$2.5 \pm 0.9$	1.9
+ 7i	$16 \pm 4*$	3.3	$3.7 \pm 0.1***$	2.8
+ 7j	$6.8 \pm 0.5$ *	1.4	$1.8 \pm 0.5$	1.4
+ 7p	$22 \pm 5*$	4.6	$8.2 \pm 1.5*$	6.3

Values are means (± SEM) of at least three independent experiments performed in duplicate (\* P < 0.05, \*\*\* P < 0.001 versus control).

K\_11.1 activators have been introduced as a new potential anti-arrhythmic strategy based on augmentation of the repolarization reserve of cardiomyocytes.<sup>31</sup> In this context, reference compounds 7a and 7r have also been defined as activators. 18, 21 Compound 7r had been found to decrease the affinity of K, 11.1 blockers dofetilide and droperidol, and to reduce the action potential prolongation by dofetilide in isolated rabbit ventricular cardiomyocytes. In a similar fashion, RPR26024, the first known K\_11.1 activator, dose-dependently reversed the action potential-prolonging effects of dofetilide in guinea pig myocytes.<sup>12</sup> We hypothesized that the K\_11.1 activators might exert their antiarrhythmic effects through negative allosteric modulation of the binding of K, 11.1 blockers to the channel, and verified that this was the case for compound 7r. Additionally, compound 7a had been found to relieve APD prolonged by a genetic dysfunction of the K<sub>.</sub>7.1 channel.<sup>18</sup> Several negative allosteric modulators synthesized in this study displayed higher potencies than 7a and 7r, in particular 7f and 7p, which

implicates their roles as lead compounds in eventually treating patients with LQT syndromes induced by pharmacological blockade of the K\_11.1 channel or due to genetic defects of the potassium channels.

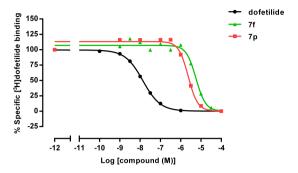


Figure 4. Displacement curves of [3H]dofetilide by dofetilide (control), 7f and 7p. Experiments were performed at 25 °C with 20 µg HEK293K\_11.1 membranes.

#### Conclusion

In summary, modifications of the three aromatic rings of the basic scaffold from reference compounds 7a and 7r lead to a series of compounds comprising of novel negative allosteric modulators of dofetilide binding to the K\_11.1 channel. Structure-activity relationships demonstrate that all the three aromatic rings play pivotal roles in determining the allosteric effects of these ligands at the K\_11.1 channel. Introducing halogen substituents at the meta-position of phenylcarbonyl ring together with non-substituted pyridine and phenolic rings enhances the negative allosteric effects of these modulators. In the kinetic dissociation assays, these compounds significantly accelerate the dissociation of [3H]dofetilide from the K<sub>1</sub>1.1 channel. Moreover, several potent modulators shift the displacement curves of prototypical K, 11.1 blockers (dofetilide and astemizole) to the right, and thus, diminish their K 11.1 affinity at the channel. This is another indication of their negative allosteric properties, and also implicates their potential antiarrhythmic propensities in reducing acquired LQT syndromes induced by pharmacological blockade. Furthermore, these negative allosteric modulators may also become a new class of medicines for alleviating congenital LQT syndromes linked to both K\_11.1 and K\_7.1 channels like compound 7a. Since compounds 7f and 7p are more potent than reference compounds 7a and 7r, they may serve as lead compounds for further optimization to relieve action potential prolongation through K<sub>y</sub>11.1 channels or other potassium channels.

# **Experimental section**

#### **Chemistry**

All solvents and reagents were purchased from commercial sources and were of analytical grade. Demineralised water is simply referred to as H<sub>2</sub>O, as was used in all cases unless stated otherwise (i.e. brine). <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker AV 400 liquid spectrometer (1H NMR, 400 MHz; <sup>13</sup>C NMR, 100 MHz) at ambient temperature. Chemical shifts are reported in parts per million (ppm), are designated by  $\delta$  and are downfield to the internal standard tetramethylsilane (TMS) in CDCl<sub>2</sub>. Coupling-constants are reported in Hz and are designated as J. Analytical purity of the final compounds was determined by high pressure liquid chromatography (HPLC) with a Phenomenex Gemini 3m C18 110A column (50 x 4.6 mm, 3 µm), measuring UV absorbance at 254 nm. Sample preparation and HPLC method was - unless stated otherwise - as follows: 0.3-0.8 mg of compound was dissolved in 1 mL of a 1:1:1 mixture of CH<sub>2</sub>CN/H<sub>2</sub>O/tBuOH and eluted from the column within 15 minutes, with a three component system of H<sub>2</sub>O/CH<sub>3</sub>CN/1% TFA in H<sub>2</sub>O, decreasing polarity of the solvent mixture in time from 80/10/10 to 0/90/10. All compounds showed a single peak at the designated retention time and are at least 95% pure. Liquid chromatography-mass spectrometry (LC-MS) analyses were performed using Thermo Finnigan Surveyor - LCQ Advantage Max LC-MS system and a Gemini C18 Phenomenex column ( $50 \times 4.6$  mm, 3 µm). The sample preparation was the same as for HPLC analysis. The elution method was set up as follows: 1-4 min isocratic system of H<sub>2</sub>O/CH<sub>2</sub>CN/1% TFA in H<sub>2</sub>O, 80:10:10, from the 4th min, a gradient was applied from 80:10:10 to 0:90:10 within 9 min, followed by 1 min of equilibration at 0:90:10 and 1 min at 80:10:10. Thin-layer chromatography (TLC) was routinely performed to monitor the progress of reactions, using aluminum coated Merck silica gel F254 plates. Purification by column chromatography was achieved by use of Grace Davison Davisil silica column material (LC60A 30-200 micron). Solutions were concentrated using a Heidolph laborota W8 2000 efficient rotary evaporation apparatus and by a high vacuum on a Binder APT line Vacuum Drying Oven. Microwave reactions were carried out in a Biotage Initiator using sealed tubes and at a set reaction temperature. The procedure for a series of similar compounds is given as a general procedure for all within that series, annotated by the numbers of the compounds.

Chlorination of bromo-fluoro-substituted-benzoic acids. To 2-bromo-5-fluorobenzoic acid 8p (500 mg, 2.28 mmol) was added thionyl chloride (207  $\mu$ L, 2.85 mmol). The mixture was heated for 1.5 h at 100 °C (an additional 1 mL of SOCl, was added after 20 min). The thionyl chloride was evaporated and the product, 2-bromo-5-fluorobenzoyl chloride (2p), was used without further purification. 454 mg yield = 84%.

2-Bromo-3-fluorobenzovl chloride (20). Methode chlorination of bromo--fluoro-substituted-benzoic acids. Continued without further purification, 436 mg yield = 81%.

3-Amino-5-methoxy pyridine (10f5). 3-Bromo-5-methoxypyridine (9f5) (500 mg, 2.66 mmol), ammonium hydroxide (15 mL) and coppersulfate pentahydrate (33 mg, 0.13 mmol) were heated at 140 °C in the microwave in a closed tube for 20 h. The reaction mixture was cooled to room temperature and the crude material separated between ethyl acetate and water. The organic layer was washed with water (3 times) and brine, dried over MgSO<sub>4</sub>, filtered and concentrated in vacuo to give 84 mg, yield 25% of the desired product as solids. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.72 (d, J = 5.4 Hz, 2H), 6.51 (t, J = 2.4 Hz, 1H), 3.79 (s, 3H) ppm; HPLC-MS: m/z 125.0

#### General method for the Friedel Crafts acylation (3b-l and 3n-p).

To a cooled mixture of the substituted benzoyl chloride 2b-l and 2n-q (5.71 mmol, 1.0 equiv.) and the (substituted)-anisole **1a-c** (7.14 mmol, 1.25 equiv.) in CH<sub>2</sub>Cl<sub>2</sub> (0.2 M) or carbon disulfide (3b) was added AlCl<sub>3</sub> (7.14 mmol, 1.25 equiv.) in portions over 20 min under a nitrogen atmosphere. After 24 h at room temperature full conversion was observed with TLC (Pet. ether/EtOAc 19/1) and the reaction mixture was poured into a 3 M aq. HCl solution. The organics were extracted 3 times from the aqueous layer with CH<sub>2</sub>Cl<sub>2</sub>, washed with water and brine, dried over MgSO<sub>4</sub> and concentrated in vacuo. The obtained crude material was triturated with petroleum ether to obtain the pure products as white solids.

- (2-Bromo-4-methoxyphenyl)-phenylmethanone (3b). Started from 3-bromoanisole (1b) and benzoyl chloride. Yellow oil 2.20 g after column chromatography (2.5-5% EtOAc/Pet. ether), yield = 75%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>): δ 7.80 (d, J = 7.2 Hz, 2H), 7.58 (d, J = 7.2 Hz, 1H), 7.45 (t, J = 8.0 Hz, 2H), 7.32 (d, J = 8.8 Hz, 1H), 7.19 (d, J = 2.0 Hz, 1H), 6.92 (dd, J = 8.6, 2.2 Hz, 1H), 3.86(s, 3H) ppm. NMR in accordance with literature.<sup>32</sup>
- (3-Bromo-4-methoxyphenyl)-phenylmethanone (3c).<sup>33</sup> Started from 2-bromoanisole (1c) and benzoyl chloride. White solids 2.78 g, yield = 64%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.07 (d, J = 2.0 Hz, 1H), 7.80 (dd, J = 8.4, 2.0 Hz, 1H), 7.77-7.73 (m, 2H), 7.60 (tt, J = 7.2, 1.2 Hz, 1H), 7.49 (t, J = 7.2 Hz, 2H), 6.96 (d, J = 8.8 Hz, 1H), 3.99 (s, 3H), 3.86 (s, 3H) ppm. NMR in accordance with literature data.34
  - (2-Fluorophenyl)(4-methoxyphenyl)methanone (3d). Started from anisole

- (1a) and 2-fluorobenzoyl chloride (2d). White solids 365 mg, yield = 34%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.83 (d, J = 8.4 Hz, 2H), 7.52-7.46 (m, 2H), 7.25 (t, J = 7.6 Hz, 1H), 7.14 (t, J = 8.4 Hz, 1H), 6.94 (d, J = 8.4 Hz, 2H), 3.86 (s, 3H) ppm. NMR in accordance with literature data.<sup>35</sup>
- (2-Chlorophenyl)(4-methoxyphenyl)methanone (3e). Started from anisole (1a) and 2-fluorobenzovl chloride (2e). Pinkish solids 2.43 g. vield = 66%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.79 (dt, J = 8.8, 2.0 Hz, 2H), 7.47-7.38 (m, 2H),7.36-7.33 (m, 2H), 6.94 (dt, J = 8.8, 2.0 Hz, 2H), 3.88 (s, 3H) ppm. NMR in accordance with literature data.<sup>36</sup>
- (2-Bromophenyl)(4-methoxyphenyl)methanone (3f). Started from anisole (1a) and 2-bromobenzovl chloride (2f). White solids 586 mg, yield = 88%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.79 (d, J = 8.4 Hz, 2H), 7.64 (d, J = 7.6 Hz, 1H), 7.41 (t, J = 7.6 Hz, 1H), 7.35-7.31 (m, 2H), 6.94 (d, J = 8.4 Hz, 2H), 3.88 (s, 3H) ppm. NMR in accordance with literature data.<sup>36</sup>
- (2-Iodophenyl)(4-methoxyphenyl)methanone (3g). Started from anisole (1a) and 2-iodobenzoyl chloride (2g). White solids 422 mg, yield = 66%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.91 (d, J = 8.0 Hz, 1H), 7.78 (d, J = 8.8 Hz, 2H), 7.44 (t, J = 7.2 Hz, 1H), 7.28 (d, J = 6.8 Hz, 1H), 7.17 (t, J = 7.6 Hz, 1H), 6.94 (d, J = 8.8 Hz, 2H), 3.88 (s, 3H) ppm. NMR in accordance with literature data.<sup>37</sup>
- (3-Fluorophenyl)(4-methoxyphenyl)methanone (3h). Started from anisole (1a) and 3-fluorobenzoyl chloride (2h). White solids 576 mg, yield = 53%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.82 (d, J = 8.8 Hz, 2H), 7.53 (d, J = 7.6 Hz, 1H), 7.47-7.42 (m, 2H), 7.28-7.24 (m, 1H), 6.97 (d, J = 8.8 Hz, 2H), 3.89 (s, 3H) ppm. NMR in accordance with literature data.<sup>38</sup>
- (3-Chlorophenyl)(4-methoxyphenyl)methanone  $(3i).^{39}$ Started anisole (1a) and 3-chlorobenzoyl chloride (2i). Pink solids 2.64 g, yield = 71%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.80 (dt, J = 8.8, 2.0 Hz, 2H), 7.73 (t, J = 1.6 Hz, 1H), 7.62 (dt, J = 8.0, 1.2 Hz, 1H), 7.53 (dd, J = 8.0, 1.2 Hz, 1H), 7.41 (t, J = 8.0Hz, 2H), 6.97 (dt, J = 9.2, 1.2 Hz, 2H), 3.89 (s, 3H) ppm.
- (3-Bromophenyl)(4-methoxyphenyl)methanone (3j). Started from anisole (1a) and 3-bromobenzoyl chloride (2j). White solids 649 mg, yield = 49%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.88 (t, J = 1.6 Hz, 1H), 7.80 (d, J = 8.8 Hz, 2H), 7.70-7.65 (m, 2H), 7.37 (t, J = 7.6, 1H), 6.97 (d, J = 8.8 Hz, 2H), 3.89 (s, 3H) ppm. NMR in accordance with literature data.<sup>35</sup>
- (3-Iodophenyl)(4-methoxyphenyl)methanone (3k). Started from anisole (1a) and 3-iodobenzoyl chloride (2k). White solids 300 mg, yield = 47%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.08 (s, 1H), 7.90 (d, J = 8.0 Hz, 1H), 7.80 (d, J = 8.8 Hz, 2H), 7.70 (d, J = 7.6 Hz, 1H), 7.22 (t, J = 8.0 Hz, 1H), 6.98 (d, J = 8.8 Hz, 2H), 3.90 (s, 3H) ppm. NMR in accordance with literature data.<sup>40</sup>

- (4-Methoxyphenyl)(4-methylphenyl)methanone (31). Started from anisole (1a) and 4-methylbenzovl chloride (21). White solids 2.31 g, yield = 68%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>1</sub>):  $\delta$  8.81 (dt, J = 8.8, 2.0 Hz 2H), 7.67 (d, J = 8.0 Hz, 2H), 7.27 (d, J = 8.4 Hz, 2H), 6.96 (dt, J = 8.8, 2.0 Hz, 2H), 3.88 (s, 3H) ppm. NMR in accordance with literature data.35
- (4-Chlorophenyl)(4-methoxyphenyl)methanone (3n). Started from anisole (1a) and 4-chlorobenzovl chloride (2n). White solids 1.02 g, yield = 73%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.80 (dt, J = 8.8, 2.0 Hz, 2H), 7.71 (d, J = 8.4 Hz, 2H), 7.45 (d, J = 8.2 Hz, 2H), 6.96 (d, J = 8.4 Hz, 2H), 3.89 (s, 3H) ppm. NMR in accordance with literature data.35
- (2-Bromo-3-fluorophenyl)(4-methoxyphenyl)methanone (30). from anisole (1a) and 2-bromo-3-fluorobenzoyl chloride (2o). White solids 316 mg, yield = 56%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.78 (d, J = 8.8 Hz, 2H), 7.42-7.36 (m, 1H), 7.23 (td, J = 8.2, 1.2 Hz, 1H), 7.12 (dd, J = 7.6, 0.4 Hz, 1H), 6.95 (d, J = 8.8 Hz, 2H), 3.89 (s, 3H) ppm.
- (2-Bromo-5-fluorophenyl)(4-methoxyphenyl)methanone (3p). from anisole (1a) and 2-bromo-5-fluorobenzovl chloride (2p). White solids 438 mg, yield = 74%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.79 (d, J = 8.8 Hz, 2H), 7.61-7.58 (m, 1H), 7.06 (d, J = 7.6 Hz, 2H), 6.95 (d, J = 9.2 Hz, 2H), 3.89 (s, 3H) ppm

#### General method for demethylation of 3b, 3d-l and 3n-p with aluminum trichloride (4b, 4d-l and 4n-p).<sup>21</sup>

- AlCl<sub>3</sub> (2.5 equiv.) was slowly added to a solution of **3b**, **3d-l** and **3n-p** (1.0 equiv.) in toluene (0.15 M) under a N<sub>2</sub> atmosphere. The mixture was refluxed at 130 °C for 2 h and by TLC (Pet. ether/EtOAc 4/1) full conversion was shown. The cooled mixture was poured into an 3 M HCl (aq.) solution, the organics were extracted with ethyl acetate (3 times), washed with brine, dried over MgSO, and concentrated to give the desired 4-(benzoyl)phenols (4b, 4d-l and 4n-p) as solids.
- (2-Bromo-4-hydroxyphenyl)-phenylmethanone (4b). Brownish solids 1.36 g, yield = 65%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.81 (dd, J = 8.4, 1.2 Hz, 2H), 7.60 (t, J = 7.4 Hz, 1H), 7.46 (t, J = 7.6 Hz, 2H), 7.27 (d, J = 6.8 Hz, 1H), 7.16 (d, J = 6.8 Hz, 1H), 7.16 (d, J = 6.8 Hz, 1H), 7.16 (d, J = 6.8 Hz, 1Hz), 7.16 (d, J = 6.8 Hz), 7.2.4 Hz, 1H), 6.86 (dd, J = 8.4, 2.4 Hz, 1H), 6.36 (s br, 1H) ppm.
- (2-Fluorophenyl)(4-hydroxyphenyl)methanone (4d).<sup>41</sup> White solids 343 mg, yield = 100%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.78 (d, J = 8.4 Hz, 2H), 7.51-7.48 (m, 2H), 7.26-7.24 (m, 1H), 7.18-7.12 (m, 1H), 6.91 (d, J = 8.4 Hz, 2H), 6.58 (s br, 1H) ppm.
- (2-Chlorophenyl)(4-hydroxyphenyl)methanone (4e).<sup>42</sup> White solids 2.17 g, yield = 95%.  ${}^{1}$ H NMR (400 MHz, DMSO):  $\delta$  10.62 (s, 1H), 7.60-7.56 (m, 3H), 7.53 (td, J = 8.4, 2.0 Hz, 1H), 7.47 (td, J = 8.4, 1.6 Hz, 1H), 7.42 (dd, J = 7.6, 1.6

- Hz, 1H), 6.88 (d, J = 8.8 Hz, 2H) ppm.
- 4-(2-Bromophenyl)(4-hydroxyphenyl)methanone (4f).<sup>43</sup> Brown oil, yield = quantitative. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.73 (d, J = 8.4 Hz, 2H), 7.63 (d, J = 7.6 Hz, 1H), 7.40 (t, J = 7.6 Hz, 1H), 7.33 (t, J = 8.8 Hz, 1H), 6.89 (d, J = 8.4 HzHz, 2H), 6.69 (s br, 1H) ppm.
- 4-(4-Hydroxyphenyl)(2-iodophenyl)methanone (4g). White solids 393 mg, yield = 97%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.90 (d, J = 8.0 Hz, 1H), 7.71 (d, J= 8.8 Hz, 2H), 7.43 (td, J = 7.6, 0.8 Hz, 1H), 7.27 (dd, J = 7.6, 1.6 Hz, 1H), 6.90 (d, J = 8.8 Hz, 2H) ppm. NMR data in accordance with literature.<sup>44</sup>
- 4-(3-Fluorophenyl)(4-hydroxyphenyl)methanone (4h). White solids 172 mg, yield = 73%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.77 (d, J = 8.4 Hz, 2H), 7.53 (d, J = 8.0 Hz, 1H), 7.48-7.43 (m, 2H), 7.30-7.18 (m, 1H), 6.95 (d, J = 8.0, 0.8)Hz, 2H) ppm. NMR data in accordance with literature.<sup>45</sup>
- 4-(3-Chlorophenyl)(4-hydroxyphenyl)methanone (4i). Off white solids 2.35 g, yield = 95%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  10.54 (s, 1H), 7.71-7.63 (m, 4H), 7.60-7.53 (m, 2H), 6.90 (dt, J = 8.8, 2.0 Hz, 2H) ppm. NMR data in accordance with literature.46
- 4-(3-Bromophenyl)(4-hydroxyphenyl)methanone (4j). White solids 168 mg, yield = 88%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.88 (t, J = 1.6 Hz, 1H), 7.77 (d, J = 8.8 Hz, 2H), 7.70 (dt, J = 8.0, 1.2 Hz, 1H), 7.66 (d, J = 7.6 Hz, 1H), 7.36(t, J = 8.0 Hz, 1H), 6.93 (d, J = 8.4 Hz, 2H), 6.92 (s br, 1H) ppm. NMR data inaccordance with literature.47
- 4-(4-Hydroxyphenyl)(3-iodophenyl)methanone (4k). Brown solids 209 mg, yield = 73%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.08 (s, 1H), 7.90 (d, J = 8.0 Hz, 1H), 7.77 (d, J = 8.4 Hz, 2H), 7.70 (d, J = 7.6 Hz, 1H), 7.22 (t, J = 8.0 Hz, 1H), 6.93 (d, J = 8.8 Hz, 2H), 5.74 (s br, 1H) ppm. NMR data in accordance with literature.46
- 4-(4-Hvdroxyphenyl)(4-methylphenyl)methanone (41).<sup>48</sup> Brown solids 2.03 g, yield = 94%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  10.39 (s,1H), 7.64 (d, J = 8.4 Hs, 2H), 7.57 (d, J = 8.0 Hz, 2H), 7.33 (d, J = 8.0 Hz, 2H), 7.88 (d, J = 8.8Hz, 2H), 2.39 (s, 3H) ppm.
- 4-(4-Chlorophenyl)(4-hydroxyphenyl)methanone (4n). Off white solids 821 mg, yield = 85%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.76 (d, J = 8.8 Hz, 2H), 7.71 (d, J = 8.8 Hz, 2H), 7.46 (d, J = 8.4 Hz, 2H), 6.93 (d, J = 8.8 Hz, 2H), 6.15 (s br, 1H). NMR data in accordance with literature.<sup>49</sup>
- (2-Bromo-3-fluorophenyl)(4-hydroxyphenyl)methanone (40). Off white solids 301 mg, yield = 100 %.  ${}^{1}H$  NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.32 (s br, 1H), 7.72 (dd, J = 6.8, 2.0 Hz, 2H), 7.41-7.35 (m, 1H), 7.22 (td, J = 8.4, 2.0 Hz, 1H), 7.11 (dd, J = 7.6, 0.8 Hz, 1H), 6.92 (dd, J = 7.2, 1.6 Hz, 2H) ppm.

(2-Bromo-5-fluorophenyl)(4-hydroxyphenyl)methanone (4p). White solids 236 mg, yield = 56%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.49 (s br, 1H), 7.72 (dt, J = 8.8, 2.4 Hz, 2H, 7.60-7.57 (m, 1H), 7.09-7.04 (m, 2H), 6.93 (dd, J = 8.8, 2.4 mHz, 2H) ppm.

(3-Bromo-4-hydroxyphenyl)-phenylmethanone (4c). At -78 °C 3c (873 mg, 3.00 mmol, 1.0 equiv.) was dissolved in 20 mL of CH<sub>2</sub>Cl and a solution of BBr, in CH<sub>2</sub>Cl (1 M) (15 mL, 15 mmol, 5 equiv.) was added under a N<sub>2</sub> atmosphere. The mixture was stirred at room temperature and after 24 h cooled again to -78 °C and quenched with water. The organics were extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 times) from the aqueous layer, dried over MgSO<sub>4</sub> and concentrated. The pure 4c (584 mg of a yellowish solid yield = 70%) was obtained by column chromatography eluting with a gradient of CH<sub>2</sub>Cl<sub>2</sub> to 2% MeOH in CH<sub>2</sub>Cl<sub>2</sub>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.03 (d, J = 2.0 Hz, 1H), 7.77-7.18 (m, 3H), 7.59 (tt, J = 7.6, 1.6 Hz, 1H), 7.50 (t, J = 7.6 Hz, 2H), 7.10 (d, J = 8.4 Hz, 1H), 6.00 (s, 1H) ppm.

#### General method for *O*-alkylation yielding compounds 5a-q.

To a solution of **5a-q** (1.0 eq.) in DMF (0.2 M) was added K<sub>2</sub>CO<sub>2</sub> (2.0 equiv.) and 2-methyl bromoacetate (2.0 equiv.). The mixture was stirred at 65 °C for 24 h, after which it was cooled to room temperature and the mixture was separated between ethyl acetate and water. The organic layer was washed with water (3 times), brine, dried over MgSO<sub>4</sub>, and concentrated. This gave the desired O-alkylated products (5a-q) as one spot on TLC (1 EtOAc/4 Pet. ether) and used as crude material in the next reaction or purified as specified in the individual examples.

Methyl 2-(4-benzoylphenoxy)acetate (5a). 50 Recrystallized from EtOAc/ Pet. ether. White solids 1.52 g, yield = 56%.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.83 (d, J = 8.8 Hz, 2H), 7.76 (d, J = 8.0 Hz, 2H), 7.57 (t, J = 8.0 Hz, 1H), 7.48 (t, J = 8.0 Hz, 2H)8.0 Hz, 2H), 6.97 (d, J = 8.8 Hz, 2H), 4.73 (s, 2H), 3.83 (s, 3H) ppm.

Methyl 2-(4-benzoyl-3-bromophenoxy)acetate (5b). Used as crude in the next reaction, yield = quantitative. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.79 (d, J = 7.6 Hz, 2H), 7.59 (t, J = 7.2 Hz, 1H), 7.46 (t, J = 7.6 Hz, 2H), 7.32 (d, J = 8.4 Hz, 1H), 7.20 (d, J = 1.6 Hz, 1H), 6.93 (dd, J = 8.4, 1.6 Hz, 1H), 4.69 (s, 2H), 3.83 (s, 3H) ppm.

Methyl 2-(4-benzoyl-2-bromophenoxy)acetate (5c). Started from 4c (2.11 mmol) and this gave 606 mg of the product as a colorless oil after column chromatography using the gradient 1/1 Pet. ether/CH<sub>2</sub>Cl<sub>2</sub> to CH<sub>2</sub>Cl<sub>2</sub>, yield = 82%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.08 (d, J= 2.0 Hz, 1H), 7.78-7.74 (m, 3H), 7.60 (t, J = 7.6 Hz, 1H), 7.49 (t, J = 7.6 Hz, 2H), 6.83 (d, J = 7.6 Hz, 1H), 4.82 (s, 2H), 3.83 (s, 3H) ppm.

Methyl 2-(4-(2-fluorobenzoyl)phenoxy)acetate (5d). Used as crude in the next reaction, yield = quantitative. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.84 (d, J = 8.4 Hz, 2H), 7.54-7.48 (m, 2H), 7.28-7.24 (m, 1H), 7.16 (t, J = 9.2 Hz, 1H), 6.96 (d, J = 8.8 Hz, 2H, 4.72 (s, 2H), 3.82 (s, 3H) ppm.

Methyl 2-(4-(2-chlorobenzovl)phenoxy)acetate (5e). Colorless oil 2.36 g. yield = 83%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.76 (dt, J = 8.8, 1.6 Hz, 2H), 7.61-7.54 (m, 2H), 7.51-7.45 (m, 2H), 7.09 (dt, J = 8.8, 2.0 Hz, 2H), 4.94 (s, 2H), 3.70 (s, 3H) ppm.

Methyl 2-(4-(2-bromobenzoyl)phenoxy)acetate (5f). Colorless oil after column chromatography using the gradient 4/1 Pet. ether/EtOAc 341 mg, yield = 75%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.79 (d, J = 8.8 Hz, 2H), 7.64 (d, J = 7.6 Hz, 1H), 7.41 (t, J = 7.2 Hz, 1H), 7.36-7.30 (m, 2H), 6.94 (d, J = 8.8 Hz, 2H), 4.71 (s, 2H), 3.82 (s, 3H) ppm.

Methyl 2-(4-(2-iodobenzoyl)phenoxy)acetate (5g). Used as crude material in the next reaction, brown oil 444 mg, yield = 93%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.91 (d, J= 8.0 Hz, 1H), 7.78 (d, J= 8.8 Hz, 2H), 7.44 (t, J= 7.6 Hz, 1H), 7.27 (dd, J = 7.6, 1.2 Hz, 1H), 7.17 (td, J = 7.6, 1.6 Hz, 1H), 6.95 (d, J = 9.2 Hz, 2H),4.73 (s, 2H), 3.81 (s, 3H) ppm.

Methyl 2-(4-(3-fluorobenzoyl)phenoxy)acetate (5h). Used as crude material in the next reaction, brown oil 244 mg, yield = quantitative. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.82 (dd, J = 8.8, 2.8 Hz, 2H), 7.52 (dd, J = 7.6, 0.8 Hz, 1H), 7.48-7.44 (m, 2H), 7.28 (d, J = 7.2 Hz, 1H), 6.98 (dd, J = 8.8, 2.4 Hz, 2H), 4.72(s, 2H), 3.82 (s, 3H) ppm.

Methyl 2-(4-(3-chlorobenzoyl)phenoxy)acetate (5i). Washed crude mixture with Pet. ether, off white solids 2.86 g, yield = 93%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.78-7.70 (m, 3H), 7.68 (t, J = 1.6 Hz, 1H), 7.62 (dt, J = 8.0, 1.6 Hz, 1H), 7.58 (t, J = 7.6 Hz, 1H), 7.11 (d, J = 8.8 Hz, 2H), 4.95 (s, 2H), 3.72 (s, 3H) ppm.

Methyl 2-(4-(3-bromobenzovl)phenoxy)acetate (5i). Washed crude mixture with Pet. ether, off white solids 253 mg, yield = 99%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.89 (t, J = 2.0 Hz, 1H), 7.81 (d, J = 8.8 Hz, 2H), 7.72-7.66 (m, 2H), 7.36 (t, J = 7.6 Hz, 1H), 6.98 (d, J = 8.8 Hz, 2H), 4.74 (s, 2H), 3.84 (s, 3H) ppm.

Methyl 2-(4-(3-iodobenzoyl)phenoxy)acetate (5k). Washed crude mixture with Pet. ether, off white solids 243 mg, yield = 99%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.08 (t, J = 1.6 Hz, 1H), 7.90 (d, J = 7.6 Hz, 1H), 7.81 (dt, J = 8.2, 2.0 Hz, 2H), 7.70 (dt, J = 8.0, 1.2 Hz, 1H), 7.22 (t, J = 8.0 Hz, 1H), 6.98 (d, J = 8.8Hz, 2H), 4.73 (s, 2H), 3.83 (s, 3H) ppm.

Methyl 2-(4-(4-methylbenzoyl)phenoxy)acetate (51). Washed crude mixture with Pet. ether, off white solids 2.71 g, yield = 99%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.80 (d, J = 8.4 Hz, 2H), 7.67 (d, J = 8.0 Hz, 2H), 7.27 (d, J = 8.0 Hz, 2H), 6.96 (d, J = 8.4 Hz, 2H), 4.72 (s, 2H), 3.83 (s, 3H), 2.44 (s, 3H) ppm.

Methyl 2-(4-(4-fluorobenzovl)phenoxy)acetate (5m). Started from the commercially available 4m. Washed crude mixture with Pet. ether, white solids 2.53 g, yield = 95%. <sup>1</sup>H NMR (400 MHz, DMSO): δ 7.80-7.76 (m, 2H), 7.73 (dt, J = 8.8, 2.0 Hz, 2H), 7.38 (tt, J = 9.2, 2.4 Hz, 2H), 7.09 (dt, J = 8.8, 2.0 Hz, 2H), 4.95 (s. 2H), 3.72 (s. 3H) ppm.

Methyl 2-(4-(4-chlorobenzoyl)phenoxy)acetate (5n). Used as crude material in the next reaction, yield was quantitative. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>): δ 7.80 (d, J = 8.8 Hz, 2H), 7.71 (d, J = 8.4 Hz, 2H), 7.46 (d, J = 8.4 Hz, 2H), 6.98 (d,= 8.8 Hz, 2H, 4.73 (s, 2H), 3.83 (s, 3H) ppm.

Methyl 2-(4-(2-bromo-3-fluorobenzoyl)phenoxy)acetate (50). Off white solids 275 mg, yield = 73%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.77 (d, J = 8.8 Hz, 2H), 7.43-7.38 (m, 1H), 7.24 (td, J = 8.4, 1.2 Hz, 1H), 7.11 (d, J = 7.6 Hz, 1H), 6.95 (d, J = 8.8 Hz, 2H), 4.73 (s, 2H), 3.79 (s, 3H) ppm.

Methyl 2-(4-(2-bromo-5-fluorobenzoyl)phenoxy)acetate (5p). Used as crude material in the next reaction, yield = 98%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>): δ 7.78 (d, J = 8.4 Hz, 2H), 7.62-7.57 (m, 1H), 7.10-7.04 (m, 2H), 6.96 (d, J = 8.4 Hz)Hz, 2H), 4.73 (s, 2H), 3.80 (s, 3H) ppm.

Methyl 2-((9-oxo-9H-fluoren-3-yl)oxy)acetate (5q). Started from 4b (2.11 mmol) and this gave 309 mg as yellow solids after column chromatography using CH<sub>2</sub>Cl<sub>2</sub> as an eluent, yield = 92%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  7.64-7.60 (m, 2H), 7.48-7.45 (m, 2H), 7.33-7.27 (m, 1H), 7.07 (d, J = 2.4 Hz, 1H), 6.71 (dd, J = 2.4 Hz, 1H), 6. = 8.0, 2.0 Hz, 1H, 4.74 (s, 2H), 3.84 (s, 3H) ppm.

#### General saponification procedure to obtain acids 6a-q.

To a solution of **5a-i** (1.0 eq.) in a mixture of equal amounts of THF and methanol (0.5 M solution) was added an agueous 1 M solution of LiOH (5.0 eq.). After 1 hour at 100 °C the saponification of the esters 6a-q was completed shown by TLC (EtOAc/Pet. ether 1/3). While the mixture was cooled on ice, the pH was adjusted to pH = 1 using a 2 M HCl solution (aq.). The resulting precipitate was collected by filtration, washed with water, Pet. ether and co-evaporated with acetone to dryness.

**2-(4-Benzoylphenoxy)acetic acid (6a)**. <sup>51</sup> White solids 1.37 g, yield = 95%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub> + drop of DMSO):  $\delta$  9.50 (s br, 1H), 7.82 (d, J = 8.4Hz, 2H), 7.75 (d, J = 7.6 Hz, 2H), 7.58 (t, J = 7.6 Hz, 1H), 7.47 (t, J = 7.6 Hz, 2H), 7.00 (d, J = 8.8 Hz, 2H), 4.69 (s, 2H) ppm.

2-(4-Benzovl-2-bromophenoxy)acetic acid (6b). White solids 548 mg. yield = 95%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.06 (d, J = 2.0 Hz, 1H), 7.77-7.34 (m, 3H), 6.60 (t, J = 7.6 Hz, 1H), 7.50 (t, J = 8.0 Hz, 2H), 6.89 (d, J = 8.4 Hz,

- 1H), 4.78 (s, 2H) ppm.
- 2-(4-Benzoyl-3-bromophenoxy)acetic acid (6c). Brown solids 1.28 g, yield = 78%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.88 (s, br, 1H), 7.80 (d, J = 8.0 Hz, 2H), 7.63-7.58 (m, 1H), 7.50-7.44 (m, 2H), 7.34 (d, J = 8.4 Hz, 1H), 7.23 (d, J = 2.4Hz, 1H), 6.96 (dd, J = 8.8, 2.4 Hz, 1H), 4.76 (s, 2H) ppm.
- 2-(4-(2-Fluorobenzoyl)phenoxy)acetic acid (6d). Off white solids 312 mg, yield = 67%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  13.12 (s br, 1H), 7.73 (d, J = 8.4 Hz, 2H), 7.67-7.61 (m, 1H), 7.53 (td, J = 7.8, 1.2 Hz, 1H), 7.40-7.34 (m, 2H), 7.07 (d, J = 8.8 Hz, 2H, 4.82 (s, 2H) ppm.
- 2-(4-(2-Chlorobenzoyl)phenoxy)acetic acid (6e). White solids 1.77 g, yield = 79%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.67 (dt, J = 8.8, 2.0 Hz, 2H), 7.62-7.53 (m, 2H), 7.51-7.43 (m, 2H), 7.05 (d, J = 9.2 Hz, 2H), 4.81 (s, 2H) ppm.
- 2-(4-(2-Bromobenzoyl)phenoxy)acetic acid (6f). Off white solids 8.60 g, yield = 86%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  9.17 (s, br, 1H), 7.80 (d, J = 8.4 Hz, 2H), 7.63 (d, J = 8.0 Hz, 1H), 7.43-7.30 (m, 3H), 6.95 (d, J = 8.8 Hz, 2H), 4.75 (s, 2H) ppm.
- 2-(4-(2-Iodobenzovl)phenoxy)acetic acid (6g). White solids 302 mg, yield = 71%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.96 (d, J = 7.6 Hz, 1H), 7.64 (d, J = 8.8 Hz, 2H), 7.54 (t, J = 7.6 Hz, 1H), 7.35 (d, J = 7.6 Hz, 1H), 7.28 (t, J = 7.6 Hz, 1H), 7.06 (d, J = 8.8 Hz, 2H), 4.81 (s, 2H) ppm.
- 2-(4-(3-Fluorobenzoyl)phenoxy)acetic acid (6h). Brown solids 94 mg, yield = 45%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.76 (d, J = 8.8 Hz, 2H), 7.63-7.56 (m, 1H), 7.52-7.46 (m, 3H), 7.08 (d, J = 9.2 Hz, 2H), 4.82 (s, 2H) ppm.
- 2-(4-(3-Chlorobenzoyl)phenoxy)acetic acid (6i). White solids 2.65 g, yield = 97%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.75-7.69 (m, 3H), 7.67 (s, 1H), 7.61 (t, J = 7.6 Hz, 1H, 7.57 (t, J = 7.6 Hz, 1H), 7.04 (d, J = 8.8 Hz, 2H), 4.71 (s, 2H) ppm.
- 2-(4-(3-Bromobenzoyl)phenoxy)acetic acid (6j). White solids 146 mg, yield = 61%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.86 (d, J = 8.0 Hz, 1H), 7.81 (s, 1H), 7.74 (d, J = 8.0 Hz, 2H), 7.67 (d, J = 7.2 Hz, 1H), 7.51 (t, J = 8.0 Hz, 1H), 7.08 (J = 8.0 Hz, 2H), 4.81 (s, 2H) ppm.
- 2-(4-(3-Iodobenzoyl)phenoxy)acetic acid (6k). White solids 118 mg, yield = 51%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  8.01 (d, J = 7.6 Hz, 1H), 7.97 (s, 1H), 7.73 (d, J = 8.4 Hz, 2H), 7.67 (d, J = 7.6 Hz, 1H), 7.35 (t, J = 7.6 Hz, 1H), 7.08 (d, J = 8.8 Hz, 2H), 4.82 (s, 2H) ppm.
- 2-(4-(4-Methylbenzoyl)phenoxy)acetic acid (61). Off white solids 2.16 g. yield = 81%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.70 (d, J = 8.8 Hz, 2H), 7.60 (d, J = 8.0 Hz, 2H, 7.34 (d, J = 8.0 Hz, 2H), 7.04 (d, J = 9.2 Hz, 2H), 4.76 (s, 2H),2.40 (s, 3H) ppm. NMR data in accordance with literature.<sup>52</sup>
  - 2-(4-(4-Fluorobenzoyl)phenoxy)acetic acid (6m). White solids after recrys-

tallization from MeOH 1.55 g, yield = 64%. <sup>1</sup>H NMR (400 MHz, DMSO): δ 13.20 (s br, 1H), 7.80-7.75 (m, 2H), 7.73 (d, J = 8.8 Hz, 2H), 7.37 (t, J = 8.8 Hz, 2H), 7.07 (d, J = 8.8 Hz, 2H), 4.81 (s, 2H) ppm.

- 2-(4-(4-Chlorobenzovl)phenoxy)acetic acid (6n).<sup>53</sup> White solids 730 mg. vield 71%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.73-7.70 (m, 4H), 7.61 (d, J = 7.6Hz, 2H), 7.05 (d, J = 8.0 Hz, 2H), 4.74 (s, 2H) ppm.
- 2-(4-(2-Bromo-3-fluorobenzoyl)phenoxy)acetic acid (60). White solids 225 mg, yield = 85%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.89 (s br, 1H), 7.80 (d, J = 8.8 Hz, 2H), 7.43-7.37 (m, 1H), 7.25 (td, J = 8.4, 1.6 Hz, 1H), 7.12 (d, J = 7.2 Hz, 1H), 6.97 (d, J = 8.8 Hz, 2H), 4.77 (s, 2H) ppm.
- 2-(4-(2-Bromo-5-fluorobenzoyl)phenoxy)acetic acid (6p). White solid 231 mg, yield = 83%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  10.07 (s, 1H), 7.78 (d, J = 8.8 Hz, 2H), 7.61-7.56 (m, 1H), 7.10-7.03 (m, 2H), 6.96 (d, J = 8.8 Hz, 2H), 4.75 (s, 2H) ppm.
- 2-((9-Oxo-9H-fluoren-3-yl)oxy)acetic acid (6q). Yellow solids 255 mg, yield = 87%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  13.20 (br s, 1H), 7.81 (d, J = 7.6 Hz, 1H), 7.62-7.54 (m, 3H), 7.46 (s, 1H), 7.38 (t, J = 7.6 Hz, 1H), 6.84 (d, J = 6.8Hz, 1H), 4.87 (s, 2H) ppm.

#### General peptide coupling method (7a-r and 7f1-f11).

To a solution of 6a-r (1.0 equiv.) and Et<sub>2</sub>N (1.5 equiv.) in DMF (0.125 M) was added 3-aminopyridine (1.1 equiv.) and HATU (1.1 equiv.) or EDCI\*HCl (7e and 71). The mixture was stirred at room temperature for 20 h. The mixture was separated between ethyl acetate and water. The organic layer was washed with water twice, brine, dried over MgSO<sub>4</sub> and concentrated. Column chromatography using mixtures of 5% methanol/dichloromethane or EtOAc:Pet.ether 2:1 gave the pure desired products.

- 2-(4-Benzoylphenoxy)-N-(pyridin-3-yl)acetamide (7a). White solid, 211 mg, yield = 69%.  ${}^{1}H$  NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.70 (s br, 1H), 8.44 (s br, 1H), 8.40 (s br, 1H), 8.22 (d, J = 8.0 Hz, 1H), 7.89 (d, J = 8.4 Hz, 2H), 7.77 (d, J = 7.6Hz, 2H), 7.60 (t, J = 7.6 Hz, 1H), 7.50 (t, J = 7.6 Hz, 2H), 7.34 (d, J = 7.2 Hz, 1H), 7.09 (d, J = 8.8 Hz, 2H), 4.74 (s, 2H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl3):  $\delta$ 195.3, 166.4, 160.2, 137.4, 132.4, 132.2, 131.2, 129.5, 128.1, 127.8, 114.1, 67.2 ppm.; HPLC  $t_p = 7.28 \text{ min. purity } 100\%; ESI-MS: 333.13 [M+H]^+.$
- 2-(4-Benzoyl-3-bromophenoxy)-N-(pyridin-3-yl)acetamide (7b). White solid, 158 mg, yield = 51%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.71 (s, 1H), 8.43 (d, J = 4.8 Hz, 2H), 8.24 (d, J = 8.0 Hz, 1H), 7.80 (d, J = 7.6 Hz, 2H), 7.61 (t, J = 7.6 Hz) Hz, 1H), 7.47 (t, J = 7.8 Hz, 2H), 7.39-7.31 (m, 3H), 7.03 (d, J = 8.4 Hz, 1H), 4.71 (s, 2H) ppm; HPLC  $t_p = 7.15$  min. purity 98%; ESI-MS: 411.13 [M+H]<sup>+</sup>.

- 2-(4-Benzoyl-2-bromophenoxy)-N-(pyridin-3-yl)acetamide (7c). White solid, 293 mg, yield = 71%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.80 (s, 1H), 8.74 (d, J = 2.4 Hz, 1H), 8.44 (d, J = 4.8 Hz, 1H), 8.27-8.23 (m, 1H), 8.15 (d, J = 1.6 Hz, 1H), 7.83 (dd, J = 8.4, 2.0 Hz, 1H), 7.80-7.74 (m, 2H), 7.63(t, J = 7.6 Hz, 1H), 7.52 (t, J = 7.6 Hz, 2H), 7.35 (dd, J = 8.4, 4.8 Hz, 1H), 7.00 (d, J = 8.8 Hz, 1H), 4.78 (s, 2H) ppm; HPLC  $t_p = 7.29$  min. purity 100%; ESI-MS: 411.07 [M+H]<sup>+</sup>.
- 2-[4-(2-Fluorobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7d). White solid, 150 mg, yield = 78%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.71 (s, 2H), 8.39 (d, J = 3.6 Hz, 1H), 8.20 (d, J = 8.0 Hz, 1H), 7.84 (d, J = 8.4 Hz, 2H), 7.55-7.49 (m, 2H), 7.31-7.24 (m, 2H), 7.15 (t, J = 8.4 Hz, 1H), 7.02 (d, J = 9.2 Hz, 2H), 4.70 (s, 2H) ppm; HPLC  $t_R = 6.67$  min. purity 98%; ESI-MS: 351.13 [M+H]<sup>+</sup>.
- 2-[4-(2-Chlorobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7e). Used EDCI\*HCl instead of HATU. White solid, 72 mg, yield = 20%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.70-8.60 (m, 2H), 8.39 (d, J = 4.4 Hz, 1H), 8.19 (d, J = 8.0 Hz, 1H), 7.81 (d, J = 8.4 Hz, 2H), 7.45-7.28 (m, 5H), 7.02 (d, J = 8.4 Hz, 2H), 4.71 (s, 2H) ppm; HPLC  $t_p = 7.04$  min. purity 100%; ESI-MS: 367.13 [M+H]<sup>+</sup>.
- 2-[4-(2-Bromobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7f). White solid, 9 mg, yield = 23%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.68 (d, J = 2.0 Hz, 1H), 8.42 (d, J = 4.4 Hz, 1H), 8.36 (s, 1H), 8.22 (dt, J = 8.4, 0.8 Hz, 1H), 7.85 (d, J = 8.4, 0.8 Hz, 1H)8.8 Hz, 2H), 7.65 (dd, J = 7.6, 0.8 Hz, 1H), 7.43 (td, J = 7.4, 1.2 Hz, 1H), 7.38-7.31 (m, 3H), 7.06 (d, J = 8.8 Hz, 2H), 4.72 (s, 2H) ppm; HPLC  $t_p = 7.09$  min. purity 98%; ESI-MS: 411.07 [M+H]+.
- 2-[4-(2-Iodobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7g). White solid, 113 mg, yield = 63%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.68 (s, 1H), 8.43 (d, J = 4.4 Hz, 1H), 8.34 (s, 1H), 8.23 (d, J = 8.4 Hz, 1H), 7.93 (d, J = 8.0 Hz, 1H), 7.85 (d, J = 8.4 Hz, 2H), 7.46 (t, J = 7.6 Hz, 1H), 7.33 (dd, J = 8.4 Hz, 4.8 Hz,1H), 7.29 (d, J = 7.6 Hz, 1H), 7.20 (td, J = 7.6 Hz, 1.2 Hz, 1H), 7.07 (d, J = 8.8Hz, 2H), 4.73 (s, 2H) ppm; HPLC  $t_R = 7.19$  min. purity 100%; ESI-MS: 459.00  $[M+H]^+$ .
- 2-[4-(3-Fluorobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7h). White solid, 80 mg, yield = 67%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.83 (s, 1H), 8.73 (d, J = 2.4 Hz, 1H, 8.39 (d, J = 4.0 Hz, 1H), 8.21 (d, J = 8.8 Hz, 1H), 7.82 (d, J = 8.8 Hz, 1H)Hz, 2H), 7.52-7.42 (m, 3H), 7.32-7.27 (m, 2H), 7.04 (d, J = 8.8 Hz, 2H), 4.73 (s, 2H) ppm; HPLC  $t_R = 6.88$  min. purity 100%; ESI-MS: 351.13 [M+H]<sup>+</sup>.
- 2-[4-(3-Chlorobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7i). White solid, 319 mg, yield = 87%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.72 (d, J = 2.0 Hz, 1H), 8.44-8.40 (m, 2H), 8.27 (dd, J = 8.4, 2.0 Hz, 1H), 7.88 (d, J = 8.8 Hz, 2H), 7.74 (t, J = 1.6 Hz, 1H), 7.63 (dt, J = 7.6, 1.6 Hz, 1H), 7.58-7.55 (m, 1H), 7.44 (t, J = 7.6 Hz, 1H), 7.36 (dd, J = 8.4, 4.8 Hz, 1H), 7.11 (d, J = 8.8 Hz, 2H), 4.75 (s,

- 2H) ppm; HPLC  $t_p = 7.33$  min. purity 100%; ESI-MS: 367.07 [M+H]<sup>+</sup>.
- 2-[4-(3-Bromobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7j). White solid, 63 mg, yield = 25%.  $^{1}$ H NMR (400 MHz, DMSO):  $\delta$  10.42 (d, 1H), 8.79 (d, J = 2.0 Hz, 1H), 8.30 (d, J = 4.0 Hz, 1H), 8.07 (d, J = 8.4 Hz, 1H), 7.85 (d, J =8.0 Hz, 1H), 7.81 (s, 1H), 7.78 (d, J = 8.8 Hz, 2H), 7.67 (d, J = 8.0 Hz, 1H), 7.51 (t, J = 7.6 Hz, 1H), 7.39-7.36 (m, 1H), 7.18 (d, J = 8.8 Hz, 2H), 4.90 (s, 2H) ppm;HPLC  $t_p = 7.42 \text{ min. purity } 98\%; \text{ ESI-MS: } 411.13 \text{ } [\text{M+H}]^+.$
- 2-[4-(3-Iodobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7k). White solid, 91 mg, yield = 71%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  8.79 (d, J = 2.4 Hz, 1H), 8.30 (dd, J = 4.8, 1.2 Hz, 1H), 8.07 (dt, J = 8.0, 1.6 Hz, 1H), 8.01 (d, J = 8.0 Hz, 1H), 7.98 (t, J = 1.2 Hz, 1H), 7.77 (d, J = 8.8 Hz, 2H), 7.68 (d, J = 7.6 Hz, 1H), 7.39-7.33 (m, 2H), 7.17 (d, J = 8.8 Hz, 2H), 4.89 (s, 2H) ppm; HPLC  $t_p = 7.50$ min. purity 98%; ESI-MS: 459.07 [M+H]+.
- 2-[4-(4-Methylbenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (71). Used EDCI\*HCl instead of HATU. White solid, 81 mg, yield = 26%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.69 (d, J = 1.6 Hz, 1H), 8.45-8.38 (m, 2H), 8.23 (dd, J = 7.2, 1.2 Hz, 1H), 7.85 (d, J = 8.8 Hz, 2H), 7.68 (d, J = 8.0 Hz, 2H), 7.33 (dd, J = 8.4, 4.8 Hz, 1H), 7.28 (d, J = 7.2 Hz, 2H), 7.07 (d, J = 8.8 Hz, 2H), 4.73 (s, 2H), 2.45 (s, 3H) ppm; HPLC  $t_p = 7.09$  min. purity 98%; ESI-MS: 347.13 [M+H]<sup>+</sup>.
- 2-(4-(4-Fluorobenzoyl)phenoxy)-N-(pyridin-3-yl)acetamide (7m). White solid, 187 mg, yield = 53%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.72 (d, J = 2.4 Hz, 1H), 8.44-8.40 (m, 2H), 8.30-8.23 (m, 1H), 7.85 (d, J = 8.8 Hz, 2H), 7.83-7.77 (m, 2H), 7.35 (dd, J = 8.4, 4.8 Hz, 1H), 7.17 (t, J = 8.8 Hz, 2H), 7.09 (d, J = 8.8Hz, 2H), 4.74 (s, 2H) ppm; HPLC  $t_p = 6.89$  min. purity 99%; ESI-MS: 351.13  $[M+H]^+$ .
- 2-[4-(4-Chlorobenzoyl)phenoxy]-N-(pyridin-3-yl)acetamide (7n). White solid, 903 mg, yield = 98%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  10.41 (s, 1H), 8.79 (d, J = 2.4 Hz, 1H), 8.30 (dd, J = 4.4, 1.2 Hz, 1H), 8.07 (dt, J = 8.0, 1.6 Hz, 1H),7.77 (d, J = 8.8 Hz, 2H), 7.72 (dd, J = 8.8, 2.0 Hz, 2H), 7.62 (d, J = 8.4 Hz, 2H), 7.38 (dd, J = 8.4, 4.8 Hz, 1H), 7.17 (d, J = 8.8 Hz, 2H), 4.89 (s, 2H) ppm; HPLC  $t_p = 7.27 \text{ min. purity } 100\%; \text{ ESI-MS: } 367.13 \text{ [M+H]}^+.$
- 2-[4-(2-Bromo-3-fluorobenzoyl)phenoxy]-N-(5-methoxypyridin-3-yl) acetamide (70). White solid, 69 mg, yield = 25%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.68 (d, J = 2.0 Hz, 1H), 8.42 (d, J = 4.4 Hz, 1H), 8.37 (s, 1H), 8.22 (d, J = 8.4Hz, 1H), 7.85 (d, J = 8.4 Hz, 2H), 7.44-7.39 (m, 1H), 7.33 (dd, J = 8.2, 4.8 Hz, 1H), 7.26 (t, J = 8.0 Hz, 1H), 7.13 (d, J = 7.6 Hz, 1H), 7.07 (d, J = 8.8 Hz, 2H), 4.72 (s, 2H) ppm; HPLC  $t_R = 7.25$  min. purity 98%; ESI-MS: 429.07 [M+H]<sup>+</sup>.
- 2-[4-(2-Bromo-5-fluorobenzoyl)phenoxy]-N-(5-methoxypyridin-3-yl) acetamide (7p). White solid, 108 mg, yield = 39%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):

 $\delta$  8.67 (s, 1H), 8.43 (d, J = 4.4 Hz, 1H), 8.28 (s, 1H), 8.21 (d, J = 8.4 Hz, 1H), 7.86 (d, J = 8.8 Hz, 2H), 7.61 (dd, J = 8.4, 4.8 Hz, 1H), 7.33 (dd, J = 8.0, 4.8 Hz, 1H), 7.14-7.03 (m, 4H), 4.73 (s, 2H) ppm; HPLC  $t_p = 7.22$  min. purity 100%; ESI-MS: 429.07 [M+H]+.

2-[(9-Oxo-9H-fluoren-3-yl)oxy]-N-(pyridin-3-yl)acetamide (7q). Started from 4q and 3-aminopyridine. Yellow solid, 170 mg, yield = 51%. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  10.52 (s, 1H), 8.89 (d, J = 2.4 Hz, 1H), 8.37 (dd, J = 4.8, 1.2 Hz, 1H), 8.17 (d, J = 8.4 Hz, 1H), 7.81 (d, J = 7.2 Hz, 1H), 7.65-7.55 (m, 3H), 7.53-7.47 (m, 2H), 7.39(t, J = 7.2 Hz, 1H), 6.95 (dd, J = 8.0, 2.0 Hz, 1H), 4.95 (s, 2H) ppm; HPLC  $t_p = 6.73$  min. purity 99%; ESI-MS: 331.07 [M+H]<sup>+</sup>.

2-[(1-Bromonaphthalen-2-yl)oxy]-N-(pyridin-3-yl)acetamide (7r).21 White solid, 235 mg, yield = 74%.  ${}^{1}H$  NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  9.07 (s br, 1H), 1.2 Hz, 1H), 8.24 (d, J = 8.8 Hz, 1H), 7.89 (d, J = 8.8 Hz, 1H), 7.84 (d, J = 8.0Hz, 1H), 7.65 (dd, J = 6.8, 0.8 Hz, 1H), 7.49 (dd, J = 7.2, 0.8 Hz, 1H), 7.34 (dd, J = 8.4, 3.6 Hz, 1H), 7.25 (d, J = 8.4 Hz, 1H), 4.83 (s, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>): 8 166.2, 151.2, 145.9, 141.4, 134.0, 132.8, 130.6, 129.8, 128.4, 128.3, 127.0, 126.3, 125.5, 123.8, 114.7, 109.9, 69.0 ppm. HPLC  $t_R = 7.95$  min. purity 99%; ESI-MS: 357.00 [M+H]+.

2-[4-(2-Bromobenzoyl)phenoxy]-N-(6-methylpyridin-3-yl)acetamide (7f1). Started from 6f and 5-amino-2-methylpyridine. White solid, 47 mg, yield = 73%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.54 (d, J = 2.4 Hz, 1H), 8.35 (s, 1H), 8.07 (dd, J = 8.4, 2.4 Hz, 1H), 7.84 (d, J = 8.8 Hz, 2H), 7.65 (d, J = 7.6 Hz, 1H), 7.43(td, J = 7.2, 0.8 Hz, 1H), 7.37-7.33 (m, 2H), 7.17 (d, J = 8.4 Hz, 1H), 7.05 (d, J = 8.4 Hz, 1H)= 8.8 Hz, 2H), 4.70 (s, 2H), 2.54 (s, 3H) ppm; HPLC  $t_R$  = 7.09 min. purity 99%; ESI-MS: 425.07 [M+H]+.

2-[4-(2-Bromobenzoyl)phenoxy]-N-(6-methoxypyridin-3-yl)acetamide (7f2). Started from 6f and 5-amino-2-methoxypyridine. White solid, 50 mg, yield = 94%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.31-8.20 (m, 2H), 7.91 (dd, J = 7.8, 2.0 Hz, 1H), 7.83 (d, J = 8.4 Hz, 2H), 7.64 (d, J = 7.6 Hz, 1H), 7.42 (t, J = 7.2 Hz, 1H), 7.37-7.32 (m, 2H), 7.04 (d, J = 8.8 Hz, 2H), 6.75 (d, J = 8.8 Hz, 1H), 4.69 (s, 2H), 3.92 (s, 3H) ppm; HPLC  $t_p = 9.06$  min. purity 98%; ESI-MS: 441.07  $[M+H]^+$ .

2-[4-(2-Bromobenzovl)phenoxy]-N-(6-chloropyridin-3-yl)acetamide (7f3). Started from 6f and 5-amino-2-chloropyridine. Yellow solid, 51 mg, yield = 82%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.47 (d, J = 2.8 Hz, 1H), 8.30 (s, 1H), 8.21 (dd, J = 8.8, 2.8 Hz, 1H), 7.85 (d, J = 8.0 Hz, 2H), 7.65 (d, J = 8.0 Hz, 1H),7.43-7.32 (m, 4H), 7.05 (d, J = 8.8 Hz, 2H), 4.71 (s, 2H) ppm; HPLC  $t_R = 9.65$ min. purity 100%; ESI-MS: 445.07 [M+H]+.

- 2-[4-(2-Bromobenzovl)phenoxy]-N-(5-methylpyridin-3-yl)acetamide (7f4). Started from 6f and 3-amino-5-methylpyridine. White solid, 55 mg, yield = 86%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  8.47 (d, J = 2.0 Hz, 1H), 8.40 (s, 1H), 8.24 (s, 1H), 8.04 (s, 1H), 7.83 (dd, J = 7.8, 2.0 Hz, 2H), 7.65 (dd, J = 7.2, 0.8 Hz, 1H), 7.43 (td, J = 7.0, 1.2 Hz, 1H), 7.37-7.33 (m, 2H), 7.05 (d, J = 8.8 Hz, 2H), 4.71 (s, 2H), 2.36 (s, 3H) ppm. HPLC  $t_p = 7.18$  min. purity 99%; ESI-MS: 425.07 [M+H]+.
- 2-[4-(2-Bromobenzoyl)phenoxy]-N-(5-methoxypyridin-3-yl)acetamide (7f5). Started from 6f and 3-amino-5-methoxypyridine. White solid, 274 mg, yield = 100%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.78 (s, 1H), 8.23 (d, J = 1.6 Hz, 1H), 8.09 (d, J = 2.4 Hz, 1H), 7.90 (t, J = 2.4 Hz, 1H), 7.80 (dt, J = 9.2, 2.0 Hz, 2H), 7.63 (dd, J = 7.8, 1.2 Hz, 1H), 7.41 (td, J = 7.8, 1.2 Hz, 1H), 7.34 (td, J = 7.8, 1.2 Hz, 1H), 7.35 (td, J = 7.8, 1.2 Hz, 1H), 7.36 (td, J = 7.8, 1.2 Hz, 1H), 7.37 (td, J = 7.8, 1.2 Hz, 1H), 7.38 (td, J = 7.8, 1.2 Hz, 1H), 7.39 (td, J = 7.8, 1.2 Hz, 1H), 7.39 (td, J = 7.8, 1.2 Hz, 1H), 7.31 (td, J = 7.8, 1.2 Hz, 1H), 7.31 (td, J = 7.8, 1.2 Hz, 1H), 7.32 (td, J = 7.8, 1.2 Hz, 1H), 7.34 (td, J = 7.8, 1.2 Hz, 1H), 7.35 (td, J = 7.8, 1.2 Hz, 1H), 7.36 (td, J = 7.8, 1.2 Hz, 1H), 7.37 (td, J = 7.8, 1.2 Hz, 1H), 7.38 (td, J = 7.8, 1.2 Hz, 1H), 7.39 (td, J = 7.8, 1.2 Hz, 1H), 7.31 (td, J = 7.8, 1H), 7.32 (td, J = 7.8, 1H), 7.33 (td, J = 7.8, 1H), 7.34 (td, J = 7.8, 1H), 7.35 (td, J = 7.87.8, 2.0 Hz, 1H), 7.30 (dd, J = 7.6, 2.0 Hz, 1H), 7.02 (dt, J = 8.8, 2.4 Hz, 2H), 4.70 (s, 2H), 3.85 (s, 3H) ppm. HPLC  $t_p = 7.44$  min. purity 99%; ESI-MS: 441.07  $[M+H]^{+}$ .
- 2-[4-(2-Bromobenzovl)phenoxy]-N-(4-methylpyridin-3-yl)acetamide (7f6). Started from 6f and 3-amino-4-methylpyridine. White solid, 90 mg, yield = 70%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.92 (s, 1H), 8.34 (d, J = 4.4 Hz, 1H), 8.28 (s, 1H), 7.84 (d, J = 8.8 Hz, 2H), 7.64 (d, J = 7.6 Hz, 1H), 7.43 (td, J = 7.2, 0.8 Hz, 1H), 7.35 (td, J = 7.8, 2.0 Hz, 1H), 7.33 (dd, J = 7.4, 2.0 Hz, 1H), 7.16 (d, J= 4.8 Hz, 1H, 7.05 (d, J = 8.8 Hz, 2H), 4.76 (s, 2H), 2.25 (s, 3H) ppm. HPLC t= 7.05 min. purity 100%; ESI-MS: 425.13 [M+H]+.
- 2-[4-(2-Bromobenzoyl)phenoxy]-N-(4-methoxypyridin-3-yl)acetamide (7f7). Started from 6f and 3-amino-4-methoxypyridine. Yellow solid, 35 mg, yield = 72%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>1</sub>):  $\delta$  9.48 (s, 1H), 8.66 (s, 1H), 8.33 (d, J = 5.2 Hz, 1H), 7.85 (d, J = 8.8 Hz, 2H), 7.65 (d, J = 8.0 Hz, 1H), 7.44-7.32 (m, 3H), 7.05 (d, J = 8.8 Hz, 2H), 6.85 (d, J = 5.6 Hz, 1H), 4.72 (s, 2H), 3.95 (s, 3H) ppm. HPLC  $t_p = 7.13$  min. purity 99%; ESI-MS: 441.13 [M+H]<sup>+</sup>.
- 2-(4-(2-Bromobenzoyl)phenoxy)-N-(quinolin-3-yl)acetamide (7f8). Started from 6f and 3-aminoquinoline. White solid, 44 mg, yield = 83%. <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{CDCl}_2)$ :  $\delta 8.85 \text{ (d, } J = 2.4 \text{ Hz}, 1\text{H)}, 8.79 \text{ (s, 1H)}, 8.60 \text{ (s, 1H)}, 8.06 \text{ (d, } 1.80 \text{$ J = 8.4 Hz, 1H), 7.86-7.81 (m, 3H), 7.68-7.63 (m, 2H), 7.55 (t, J = 8.0 Hz, 1H), 7.40 (t, J = 7.6 Hz, 1H), 7.37-7,31 (m, 2H), 7.07 (d, J = 7.6 Hz, 2H), 4.76 (s, 2H) ppm. HPLC  $t_R = 8.10$  min. purity 99%; ESI-MS: 461.07 [M+H]<sup>+</sup>.
- 2-[4-(2-Bromobenzoyl)phenoxy]-N-(6-chloropyridazin-3-yl)acetamide (7f9). Started from 6f and 3-amino-6-chloro-pyridazine. White solid, 32 mg, yield = 6%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>)  $\delta$  9.49 (s, 1H), 8.57 (d, J = 8.8 Hz, 1H), 7.86 (d, J = 7.6 Hz, 2H), 7.65 (d, J = 8.0 Hz, 1H), 7.57 (d, J = 9.2 Hz, 1H), 7.43-7.30 (m, 3H), 7.07 (d, J = 7.1 Hz, 2H), 4.77 (s, 2H) ppm. HPLC  $t_p = 9.37$  min.

purity 97%; ESI-MS: 445.93 [M+H]+.

2-[4-(2-Bromobenzoyl)phenoxy]-N-(pyrimidin-5-yl)acetamide (7f10). Started from 6f and 5-aminopyrimidine. White solid, 24 mg, yield = 26%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>):  $\delta$  9.08 (s, 2H), 9.03 (s, 1H), 8.45 (s, 1H), 7.81 (d, J =8.8 Hz, 2H), 7.65 (d, J = 8.0 Hz, 1H), 7.43 (td, J = 7.4, 1.2 Hz, 1H), 7.39-7.31 (m, 2H), 7.05 (d, J = 8.8 Hz, 2H), 4.75 (s, 2H) ppm. HPLC  $t_R = 8.27$  min. purity 99%; ESI-MS: 412.07 [M+H]+.

2-(4-(2-Bromobenzoyl)phenoxy)-N-(pyrazin-2-yl)acetamide (7f11). Started from 6f and 2-aminopyrazine. White solid, 14 mg, yield = 53%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>)  $\delta$  9.62 (s, 1H), 8.87 (s, 1H), 8.43 (s, 1H), 8.31 (s, 1H), 7.85 (dt, J = 8.8, 2.0 Hz, 2H), 7.65 (dd, J = 7.6, 0.8 Hz, 1H), 7.43 (td, J = 7.6, 1.2 Hz, 1H),7.39-7.32 (m, 2H), 7.06 (dt, J = 9.2, 2.8 Hz, 2H), 4.75 (s, 2H) ppm. HPLC  $t_R =$ 8.68 min. purity 98%; ESI-MS: 412.00 [M+H]+.

#### **Biology**

#### Materials and methods

Dofetilide was synthesized in our own laboratory,<sup>54</sup> and astemizole was purchased from Sigma Aldrich (Zwijndrecht, The Netherlands). Tritium-labeled dofetilide (specific activity 82.3 Ci·mmol-1) was purchased from PerkinElmer (Groningen, The Netherlands). Bovine serum albumin (BSA, fraction V) was purchased from Sigma (St. Louis, MO, USA). G418 was obtained from Stratagene (Cedar Creek, USA). All the other chemicals were of analytical grade and obtained from standard commercial sources. HEK293 cells stably expressing the K\_11.1 channel (HEK293K\_11.1) were kindly provided by Dr. Eckhard Ficker (University of Cleveland, USA).

#### Cell culture and membrane preparation

HEK293K, 11.1 cells were cultured, and membranes were prepared and stored as described previously.16

#### Radioligand kinetic dissociation assays

Kinetic dissociation assays of [3H]dofetilide were performed in incubation buffer (10 mM HEPES, 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl<sub>2</sub>, 1 mM EGTA, 10 mM glucose, 0.1% BSA, pH 7.4) as described previously with the following modifications. 16 Single point dissociation experiments were conducted by addition of 10 µM dofetilide in the absence (control) or presence of 10 µM synthesized

compounds after preincubation at 25 °C for 2 h. After 6 minutes of dissociation, incubations were terminated by dilution with ice-cold wash buffer (25 mM Tris-HCl, 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl,, 0.05 mM CaCl,, 0.05 % BSA, pH 7.4). Separation of bound from free radioligand was performed by rapid filtration through a 96-well GF/B filter plate using a PerkinElmer Filtermate-harvester (PerkinElmer, Groningen, The Netherlands). The filter-bound radioactivity was determined by scintillation spectrometry using the P-E 1450 Microbeta Wallac Trilux scintillation counter (PerkinElmer) after addition of 25 µL Microscint and extraction. Full dissociation assays were carried out with 10 µM dofetilide in the absence (control) or presence of 50 µM selected compounds for a total period of 2 h after preincubation. The amounts of radioligand still bound to the receptor were measured at various time intervals. Concentration-dependent effects of compounds 7f and 7p were determined by addition of 10 µM dofetilide in the absence (control) or presence of different concentrations of 7f and 7p. After 6 min of dissociation, the incubations were terminated and samples were obtained as described above.

#### Radioligand displacement assay

[3H]Dofetilide binding assays for the K\_11.1 channel were performed in incubation buffer as described previously.16 Briefly, membrane aliquots containing 20  $\mu$ g protein were incubated with 5 nM [3H]dofetilide in a total volume of 100  $\mu$ l incubation buffer at 25 °C for 1 h. Radioligand displacement experiments were carried out with various concentrations of tested compounds. Total binding was determined in the presence of incubation buffer, whereas nonspecific binding was evaluated with 10 µM astemizole. Incubations were terminated by dilution with ice-cold wash buffer, and samples were obtained as described in the "radioligand kinetic dissociation assays". The displacement assays of dofetilide and astemizole were conducted in the absence (control) or presence of 10 μM 7f, 7h-j and 7p.

#### Data analysis

All data of radioligand binding assays were analyzed with Prism v. 5.0 (Graph-Pad, San Diego, CA, USA). Dissociation rate constants, k<sub>off</sub>, were obtained by computer analysis of the exponential decay of [3H] dofetilide bound to the K<sub>v</sub>11.1 channel. EC<sub>50</sub> values from kinetic dissociation assays were calculated by non-linear regression analysis of concentration-effect curves of dissociation in the presence of different concentrations of unlabeled ligands.  $IC_{50}$  values in displacement assays were directly obtained from non-linear regression analysis of dose-re-

sponse curves. Apparent inhibitory binding constants (K. values) were derived from the IC<sub>50</sub> values according to the Cheng-Prusoff relationship: $^{55}$  K<sub>i</sub> = IC<sub>50</sub>/(1 +  $[L^*]/K_D$ , where  $[L^*]$  is the concentration of radioligand and  $K_D$  its dissociation constant from the saturation assay. 30 All values obtained from radioligand binding assays in this study are means of at least three independent experiments performed in duplicate, and data are presented as mean  $\pm$  SEM. Statistical analysis was performed with a two-tailed unpaired Student's t-test.

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# Chapter 6

 $K_v$ 11.1 (hERG)-induced cardiotoxicity: a molecular insight from a binding kinetics study of prototypic  $K_v$ 11.1 inhibitors

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Adapted from Br. J. Pharmacol. 2015, 172, 940-945.

#### **Abstract**

Drug-induced arrhythmia due to blockade of the K\_11.1 channel (also known as the hERG K<sup>+</sup> channel) is a frequent side effect and arouses major safety concerns during the drug development process. Previous studies have primarily focused on equilibrium parameters, i.e. affinity or potency, of drug candidates at the channel. The aim of this study was to determine the kinetics of the interaction with the channel for a number of known K<sub>2</sub>11.1 blockers and to explore a possible correlation with the affinity or physicochemical properties of these compounds. In this context, the affinity and kinetic parameters of fifteen prototypic K\_11.1 inhibitors were evaluated in a number of [3H]dofetilide binding assays. The lipophilicity (logK<sub>w-C8</sub>) and membrane partitioning (logK<sub>w-IAM</sub>) of these compounds were determined by means of HPLC analysis. Accordingly, a novel [3H]dofetilide competition association assay was set up and validated, which allowed us to determine the binding kinetics of the K<sub>2</sub>11.1 blockers used in this study. Interestingly, the compounds' affinity (K, values) were correlated to their association rates rather than dissociation rates. Overall lipophicility or membrane partitioning of the compounds were not correlated to their affinity and rate constants for the channel. In conclusion, a compound's affinity for the K<sub>2</sub>11.1 channel is determined by the rate of association with the channel, while overall lipophilicity and membrane affinity are not. In more general terms, our findings provide novel insights into the mechanism of action for a compound's activity at the K<sub>v</sub>11.1 channel. This may yield further understanding in how K<sub>v</sub>11.1-induced cardiotoxicty is governed and how it can be circumvented in the future.

#### Introduction

The K\_11.1 channel, a voltage-gated potassium channel previously known as hERG (human ether-à-go-go related gene), encodes the pore-forming subunit of the rapid component of the delayed rectifier  $K^+$  channel,  $I_{K^+}$  which contributes to phase 3 repolarization in cardiac action potentials<sup>1-4</sup>. Genetic defects or drug blockade of the K\_11.1 channel normally cause delayed repolarization of cardiac action potentials and Torsades de Pointes (TdP)5. In some cases, TdP can degenerate into ventricular fibrillation and lead to sudden death<sup>6, 7</sup>. Currently, a number of drugs have been withdrawn from the market due to their blockade of the K\_11.1 channel, including astemizole and cisapride<sup>8</sup>. In addition, either 'black box' warnings or significant restrictions have been rendered for several other drugs, such as droperidol and ibutilide8. Withdrawing or restricting drugs due to their K<sub>2</sub>11.1 liability in less than 1% of the patient population causes unpredictable and huge costs to industry, and brings troubles for patients as well owing that some effective drugs would be beneficial for the vast majority of patients 8,9.

In the past decades, a series of techniques including whole cell patch-clamp technologies, radioligand binding assays and ion flux assays have been developed and used to screen for K\_11.1 liability of compounds<sup>5, 10</sup>. From these, equilibrium radioligand displacement assays have been suggested as a low-cost and high throughput technology alternative to other available methods and are widely employed to obtain the affinity of compounds at the K<sub>v</sub>11.1 channel<sup>11-13</sup>. These studies provide equilibrium parameters, such as IC<sub>50</sub> or K<sub>1</sub> values. For instance, a 30-fold margin between K<sub>v</sub>11.1 IC<sub>50</sub> and maximum free plasma concentration (C<sub>max</sub>) has been recommended to assess drug safety with respect to arrhythmogenesis <sup>14</sup>. However, the kinetics of the interaction between a drug and the K 11.1 channel may be equally, or even more, important. Knowledge of association and dissociation rate constants has already led to better insights in the action of drugs at other targets, such as enzymes and G protein-coupled receptors<sup>15, 16</sup>.

In the present study, we hypothesized that a slow association rate (k<sub>op</sub>) to and/or a fast dissociation rate constant (koff) from the channel may be beneficial to reduce or avoid K<sub>v</sub>11.1-related cardiovascular side effects. Hence, a detailed understanding of a ligand's binding kinetics at the channel may provide clues to enable the optimization of its kinetic profiles, and potentially 'rescue' it from removal of the drug discovery pipeline. In this context, we describe the development and validation of a [3H]dofetilide competition association assay in HEK293K, 11.1 cell membranes to determine the kinetic binding parameters of 15 unlabelled reference compounds from different "Redfern" categories<sup>14</sup> (Table 1). We systematically investigated both equilibrium affinity and kinetic binding

data of those compounds, and compared these parameters with the compounds' physicochemical properties. Since the interaction of drugs with membrane phospholipids is not only driven by lipophilicity, but also electrostatic interactions with both acidic and basic moieties on phosphate head groups<sup>17, 18</sup>, a regular column and an immobilized artificial membrane (IAM) column, were used to derive two hydrophobic parameters for these compounds, i.e. logK<sub>w,C8</sub> and logK<sub>w,IAM</sub>, respectively. Finally, both hydrophobic parameters were compared with the compounds' affinity and kinetic data. Taken together, the present study provides a novel approach to study the kinetics of the interaction between a drug and the K\_11.1 channel, which may help to more precisely assess K\_11.1 liability of drug candidates in the future.

#### **Methods**

#### **Materials**

Astemizole, sertindole, terfenadine, moxifloxacin, amiodarone, chlorpromazine, ibutilide, clofilium, pimozide, cisapride, ranolazine, sotalol, thioridazine and all the solutes for HPLC determinations were purchased from Sigma Aldrich (Zwiindrecht, The Netherlands). Dofetilide and E-4031 were synthesized in our own laboratory<sup>19, 20</sup>. Tritium-labeled dofetilide (specific activity 65~87 Ci·mmol<sup>-1</sup>) was purchased from Perkin Elmer (Groningen, The Netherlands). Bovine serum albumin (BSA, fraction V) was purchased from Sigma (St. Louis, MO, USA). G418 was obtained from Stratagene (Cedar Creek, USA). The chemicals for HPLC were of HPLC grade; all the other chemicals were of analytical grade and achieved from standard commercial sources. HEK293 cells stably expressing the K\_11.1 channel (HEK293K\_11.1) were kindly provided by Dr Eckhard Ficker (University of Cleveland, USA). The molecular target nomenclature (K\_11.1) conforms to 'The Concise Guide to PHARMACOLOGY 2013/14: Ion Channels'21.

## Cell culture and membrane preparation

HEK293K\_11.1 cells were cultured and membranes were prepared and stored as described previously<sup>22</sup>.

**Table 1.** Chemical structures and information on the use of 15  $\rm K_v 11.1$  blockers examined in this study.

Name	Chemical structure	Therapeutic class	TdPa	Redfern category <sup>b</sup>	Status in the US <sup>a</sup>
Dofetilide		antiarrhythmic	++	1	restricted
Astemizole		antihistamine	++	2	withdrawn
E-4031	O H	antiarrhythmic	++	1*	not approved
Sertindole	CI ONH	antipsychotic	+	2	not approved
Terfenadine	OH N OH	antihistamine	++	2	withdrawn
Moxifloxacin	HN OH	antibacterial	++	3*	restricted
Amiodarone		antiarrhythmic	++	1	restricted
Chlorpromazine	S N_	antipsychotic	++	3*	restricted
Ibutilide	S H	antiarrhythmic	++	1	restricted
Clofilium	CI	antiarrhythmic	++	1*	not approved
Pimozide	F	antipsychotic	++	3	restricted

Table		continued
Table	1	Commuca

Name	Chemical structure	Therapeutic class	TdPa	Redfern category <sup>b</sup>	Status in the US <sup>a</sup>
Cisapride		gastroprokinetic	++	2	withdrawn
Ranolazine	H OH OH	antianginal	+	5*	restricted
Sotalol	S H OH H	antiarrhythmic	++	1	restricted
Thioridazine		antipsychotic	++	3	restricted

a+ drugs with possible TdP risk, and ++ drugs with known TdP risk. Information on these compounds was retrieved from CredibleMeds® available at: http://crediblemeds.org/ (accessed 10 April 2014).

<sup>b</sup>Redfern categories for most compounds were derived from the literature (Redfern et al., 2003); they are 1: Class Ia and III antiarrythmics; 2: Withdrawn from market due to TdP; 3: Measurable incidence / numerous reports of TdP in humans; 4: Isolated reports of TdP in humans; 5: No reports of TdP in humans. \* Redfern categories for these compounds were deduced according to the definition for different categories.

## Radioligand saturation assay

Membrane aliquots containing 20 µg of protein were incubated in a total volume of 100 µL of incubation buffer (10 mM HEPES, 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl<sub>2</sub>, 1 mM EGTA, 10 mM Glucose, 0.1% BSA, pH 7.4) at 25 °C for 120 min to ensure that the equilibrium was reached at all concentrations of radioligand. Total binding was determined at a range of concentrations (0.2  $\sim$ 22 nM) of [3H]dofetilide, whereas nonspecific binding was determined at three different concentrations of radioligand in the presence of 10 µM astemizole and analyzed by linear regression. Incubations were terminated by dilution with icecold wash buffer (25 mM Tris-HCl, 130 mM NaCl, 60 mM KCl, 0.8 mM MgCl<sub>2</sub>, 0.05 mM CaCl<sub>2</sub>, 0.05% BSA, pH 7.4). Separation of bound from free radioligand was performed by rapid filtration through Whatman GF/B filters using a Brandel harvester. Filters were subsequently washed 6 times with 2 mL ice-cold wash buffer. Filter-bound radioactivity was determined by scintillation spectrometry using a liquid Scintillation Analyzer (Tri-Carb 2900TR) after addition of 3.5 mL of Packard Emulsifier Safe and 2 h extraction.

#### Radioligand association and dissociation assay

Kinetic association experiments were performed by incubating membrane aliquots containing 20 μg of protein in a total volume of 100 μL incubation buffer at 25 °C for 120 min with 16 different concentrations (0.7~16 nM) of [3H]dofetilide. The amount of radioligand bound to the receptor was measured at various time intervals during the incubation. Incubations were terminated, and samples were obtained and analyzed as described in "radioligand saturation assay". Further traditional association and dissociation assays were performed as described previously<sup>22</sup>.

#### Radioligand displacement assay

The [3H]dofetilide binding assay for the K\_11.1 channel was performed as described previously<sup>22</sup>. In short, membrane aliquots containing 20  $\mu$ g protein were incubated in a total volume of 100 µl incubation buffer at 25 °C for 60 min. Radioligand displacement experiments were conducted using 11 concentrations of the competing ligand in the presence of 5 nM [3H]dofetilide. At this concentration, total radioligand binding did not exceed 10% of the radioligand added to prevent ligand depletion. Nonspecific binding was determined in the presence of 10 µM astemizole and represented approximately 15% of the total binding. [3H]Dofetilide did not bind to membranes prepared from empty HEK293 cells lacking the K\_11.1 channel (data not shown). Total binding was determined in the presence of incubation buffer and was set at 100% in all experiments, whereas nonspecific binding was set at 0%. Incubations were terminated by dilution with ice-cold wash buffer. Separation of bound from free radioligand was performed by rapid filtration through a 96-well GF/B filter plate using a Perkin Elmer Filtermate-harvester (Perkin Elmer, Groningen, The Netherlands). Filters were subsequently washed 12 times with ice-cold wash buffer. The filter-bound radioactivity was determined by scintillation spectrometry using the P-E 1450 Microbeta Wallac Trilux scintillation counter (Perkin Elmer) after addition of 25 µL Microscint and 2 h extraction.

## Radioligand competition association assay

The binding kinetics of unlabelled reference compounds were determined at 25 °C using the competition association assay according to a previously published method<sup>23</sup>. In a standard assay, three different concentrations (0.3, 1- and 3-fold of their K, values) of unlabelled dofetilide, astemizole and E-4031 were tested. We assessed the binding kinetics of all other unlabelled reference compounds in a simplified one-concentration competition association assay based on Guo et al. <sup>24</sup>. The experiments were initiated by incubating membrane aliquots containing 20 μg of protein in a total volume of 100 μL of incubation buffer in the absence (control) or presence of a certain concentration of unlabelled ligands at 25 °C for 120 min with 5 nM [3H]dofetilide. The amounts of radioligand bound to the receptor were measured at various time intervals during the incubation. Incubations were terminated, and samples were obtained and analyzed as described in "radioligand displacement assay".

# Determination of $log K_{w-cs}$ and $log K_{w-lam}$ parameters by HPLC

 $LogK_{w.cs}$  values were measured on a Supelcosil LC-ABZ, 5 cm  $\times$  4.6 mm, 5  $\mu m$ column according to a methodology described previously<sup>25, 26</sup>. In short, retention times of the compounds were determined at three different methanol percentages. These retention times were converted to k values by using the formula  $k = (t_p$  $t_0/t_R$  in which  $t_R$  is the retention time and  $t_0$  the retention time of a "nondelayed" compound (pure methanol). The calculated log k values were plotted against the methanol concentrations and extrapolated to a 0% methanol situation yielding the logK<sub>w.cs</sub> values for 15 reference compounds (intercept of y axis).

An isocratic method was applied to measure the  $logK_{W-lam}$  values of all tested compounds on a 10 cm x 4.6 mm, 10 µm Regis IAM PC DD2 column<sup>27</sup>. Retention times of the compounds were determined at three different concentrations of acetonitrile. The  $k_{IAM}$  values were calculated by the equation  $k_{IAM} = (t_R - t_0)/t_R$  in which t<sub>R</sub> represents retention times of tested compounds, whereas t<sub>0</sub> is determined by injecting a sodium nitrate solution in the HPLC system The log k<sub>IAM</sub> values for a compound were plotted against the applied acetonitrile concentrations. The intercept with the Y-axis of the straight line through these data points yielded the extrapolated logK<sub>W-IAM</sub> values for the 15 reference compounds.

#### Data analysis

All data of radioligand binding assays were analyzed using the non-linear regression curve fitting program Prism v. 5.1 (GraphPad, San Diego, CA, USA). K and B<sub>max</sub> values of [3H]dofetilide at HEK293K<sub>v</sub>11.1 membranes were obtained by computational analysis of saturation curves. Apparent inhibitory binding constants (K<sub>i</sub> values) were derived from the IC<sub>50</sub> values according to the Cheng and Prusoff equation  $K_i = IC_{50}/(1 + [L^*]/K_D)$ , where [L\*] was the concentration of radioligand and K<sub>D</sub> was its dissociation constant from the saturation assay <sup>28</sup>. In

the kinetic association experiments, the on- and off-rates were derived from the linear regression analysis using the equation  $k_{obs} = k_{on}[L^*] + k_{off}$  where the  $k_{obs}$ value was obtained by computer analysis of the exponential association of [3H] dofetilide bound to the receptor with [L\*] being the concentration of radioligand. The association and dissociation rates were used to calculate the kinetic K<sub>D</sub> value using the following equation  $K_{\rm D} = k_{\rm off}/k_{\rm on}$ . The association and dissociation rates for unlabelled compounds were calculated by fitting the data into the competition association model using "kinetics of competitive binding"23:

$$\begin{split} &K_{A} = k_{1}[L] + k_{2} \\ &K_{B} = k_{3}[I] + k_{4} \\ &S = \sqrt{(K_{A} - K_{B})^{2} + 4k_{1}k_{3}LI10^{-18}} \\ &K_{F} = 0.5 (K_{A} + K_{B} + S) \\ &K_{S} = 0.5 (K_{A} + K_{B} - S) \\ &Q = \frac{B_{max}k_{1}L10^{-9}}{K_{F} - K_{S}} \\ &Y = Q \left( \frac{k_{4}(K_{F} - K_{S})}{K_{F}K_{S}} + \frac{k_{4} - K_{F}}{K_{F}} e^{(-K_{F}X)} - \frac{k_{4} - K_{S}}{K_{S}} e^{(-K_{S}X)} \right) \end{split}$$

Where X is the time (min), Y the specific binding of [3H]dofetilide, k, and  $k_2$  are the  $k_{on}$  (M<sup>-1</sup>·min<sup>-1</sup>) and  $k_{off}$  (min<sup>-1</sup>) of [<sup>3</sup>H]dofetilide obtained from the traditional association and dissociation assay, L the concentration of [3H]dofetilide (nM), B<sub>max</sub> the maximum specific binding (dpm) and I the concentration of the unlabelled compound (nM). Fixing these parameters allowed the following parameters to be calculated:  $k_3$ , which is the  $k_{on}$  value (M-1·min-1) of the unlabelled compound and k<sub>4</sub>, which is the k<sub>off</sub> value (min<sup>-1</sup>) of the unlabelled compound. LogK<sub>w-C8</sub> and logK<sub>w-IAM</sub> values were derived from linear regression analysis as mentioned above. The molecular weight (M.W.), logP and pK values were calculated using a structure-based calculation plug-in provided by ChemAxon (Budapest, Hungary). All values obtained from radioligand binding assays in this study are means of at least three independent experiments performed in duplicate. Statistical analysis was performed with a two-tailed unpaired Student's t-test.

#### **Results**

#### [3H]Dofetilide saturation assay

The binding of [3H]dofetilide to HEK293K\_11.1 cell membranes was saturable and best described by a one-site binding model. A representative saturation curve and the averaged data of three independent experiments performed in duplicate are shown in Figure 1 and Table 2, respectively. The K<sub>D</sub> and B<sub>max</sub> values obtained from this assay were  $2.4 \pm 0.1$  nM and  $1.6 \pm 0.1$  pmol·mg<sup>-1</sup> protein, respectively. The K<sub>D</sub> value for [3H]dofetilide from this assay was used to calculate K<sub>3</sub> rather than IC<sub>50</sub> values from the displacement assay for 15 reference compounds.

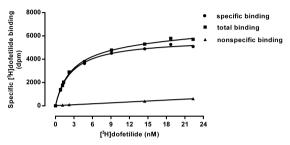


Figure 1. Saturation of [3H]dofetilide binding to HEK293K\_11.1 membranes. Total binding was determined at increasing concentrations of [3H]dofetilide. Nonspecific binding was determined at three concentrations of [3H]dofetilide, and nonspecific binding at other concentrations of radioligand was extrapolated by linear regression. Specific binding was calculated as the difference between the total and nonspecific binding. The K<sub>D</sub> value was  $2.4 \pm 0.1$  nM, and the  $B_{max}$  value was  $1.6 \pm 0.1$  pmol·mg<sup>-1</sup> protein. Experiments were performed at 25 °C using 20 µg of HEK293K, 11.1 membrane protein.

#### [3H]Dofetilide association and dissociation assay

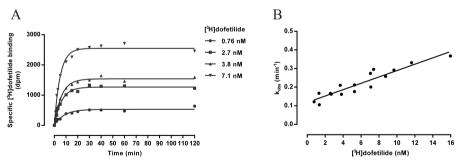


Figure 2: Characterization of the association and dissociation rates of [3H]dofetilide to HEK293K 11.1 membranes in the kinetic association assay. (A) Representative association curves of [3H]dofetilide at four different concentrations; (B) A plot of k<sub>obs</sub> values versus the concentration of [3H]dofetilide. Experiments were performed at 25 °C using 20 μg of HEK293K 11.1 membrane protein.

Initial experiments were performed to fully characterize the association and dissociation rates of [3H]dofetilide to and from HEK293K, 11.1 membranes, respec-

tively. As the association rate of a ligand is dependent upon the concentration used, kinetic association experiments with a range of [3H]dofetilide concentrations were conducted. In Figure 2A, curves are shown for four of such concentrations (0.76, 2.7, 3.8 and 7.1 nM). A plot of the k<sub>obs</sub> values against more, including higher, concentrations of [3H]dofetilide (Figure 2B) was consistent with a linear correlation ( $r^2 = 0.86$ , P < 0.0001), indicating that the binding of [ $^3$ H]dofetilide to the K<sub>2</sub>11.1 channel followed the law of mass action for a simple bimolecular interaction and that the equation  $k_{on} = (k_{obs} - k_{off})/[L^*]$  was applicable in this study. The  $k_{\mbox{\tiny on}}$  and  $k_{\mbox{\tiny off}}$  values obtained from this plot were 0.017  $nM^{\mbox{\tiny -1}}{\cdot}min^{\mbox{\tiny -1}}$  and 0.12  $min^{-1}$ , respectively (Table 2). When  $k_{off}$  was divided by  $k_{on}$ , a kinetically derived K<sub>D</sub> value of 7.1 nM was obtained. These values were in agreement with values for apparent on- and off-rates and kinetic K<sub>D</sub> of [3H]dofetilide assessed at one (5 nM) concentration ( $k_{on} = 0.032 \pm 0.003 \text{ nM}^{-1} \cdot \text{min}^{-1}, k_{off} = 0.20 \pm 0.03 \text{ min}^{-1} \text{ and } K_{D}$ =  $6.4 \pm 1.3$  nM) derived from the traditional association and dissociation assays published previously<sup>22</sup>.

Table 2. Binding parameters of dofetilide from different equilibrium and kinetic binding assays.

Binding parameters	Saturation assay	Kinetic association assay <sup>a</sup>	Traditional association and dissociation assay <sup>b</sup>	Displacement assay	Competition association assay
k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> )	-	0.017	$0.032 \pm 0.003$	-	$0.048 \pm 0.011$
$\mathbf{k}_{\mathrm{off}}$ (min <sup>-1</sup> )	-	0.12	$0.20\pm0.03$	-	$0.13 \pm 0.02$
$K_{D}(nM)$	$2.4\pm0.1$	7.1°	$6.4 \pm 1.3^{c}$	-	$2.7 \pm 0.3^{\rm c}$
$K_{i}(nM)$	-	-	-	$5.4 \pm 0.8$	

Values are means (± S.E.M.) of three independent assays performed in duplicate.

## [3H]Dofetilide displacement assay

Competition binding assays were performed to generate K, values for 15 reference compounds. Compounds were selected based on structural diversity and having a wide range of K 11.1 binding affinity, and included both anti-arrhythmic drugs and drugs for other therapeutic areas (e.g., astemizole and terfenadine for the treatment of allergic conditions). All compounds produced a concentration-dependent inhibition of specific [3H]dofetilide binding, and their displace-

<sup>&</sup>lt;sup>a</sup>Data were derived from linear regression of one independent association assay of [<sup>3</sup>H] dofetilide at different concentrations.

<sup>&</sup>lt;sup>b</sup>Data from our previous study<sup>22</sup>.

<sup>&</sup>lt;sup>c</sup>Kinetic  $K_D = k_{off}/k_{on}$ .

ment curves were best described by a one-site competition model (Figure 3). All K values are listed in **Table 3**. Among the 15 compounds, clofilium had the highest affinity to the K<sub>2</sub>11.1 channel, displacing [<sup>3</sup>H]dofetilde with a K<sub>3</sub> value of 0.55  $\pm$  0.09 nM, whereas moxifloxacin exhibited the lowest affinity of 252  $\pm$  121  $\mu$ M. Ranolazine and sotalol showed similar and relatively weak inhibition of the channel with K, values of  $21 \pm 6$  and  $25 \pm 1$   $\mu$ M, respectively. Additionally, amiodarone, thioridazine and chlorpromazine displayed modest K\_11.1 blockade with K\_values from 0.3 to 3 µM. All other compounds demonstrated relatively high affinity to the K\_11.1 channel, between  $2.5 \pm 0.2$  nM (astemizole) and  $63 \pm 4$  nM (terfenadine).

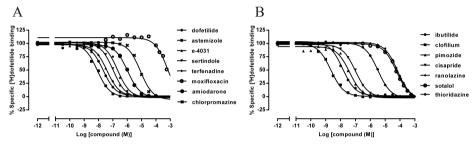


Figure 3. Displacement curves of [3H]dofetilide from HEK293K, 11.1 membranes by different known K 11.1 channel blockers. (A) Dofetilide, astemizole, E-4031, sertindole, terfenadine, moxifloxacin, amiodarone and chlorpromazine; (B) Ibutilide, clofilium, pimozide, cisapride, ranolazine, sotalol and thioridazine. Experiments were performed at 25 °C using 20 μg of HEK293K, 11.1 membrane protein.

## [3H]Dofetilide competition association assay

With the  $k_{on}(k_1)$  and  $k_{off}(k_2)$  values of [3H]dofetilide obtained from the traditional association and dissociation assays, it was possible to determine the k<sub>on</sub> (k<sub>3</sub>) and  $k_{_{off}}(k_{_{4}})$  values of unlabelled compounds by performing so-called competition association experiments. Firstly, we validated the competition association assay at the K\_11.1 channel using three concentrations of 'cold' dofetilide equivalent to 0.3-, 1- and 3-fold of its K, value. A representative experiment is shown in Figure 4. The  $k_{on}$  ( $k_3$ ) and  $k_{off}$  ( $k_4$ ) values for dofetilide determined in this assay were  $0.048 \pm 0.011 \text{ nM}^{-1} \text{ min}^{-1}$  and  $0.13 \pm 0.02 \text{ min}^{-1}$ , respectively, and in good agreement with the kinetic parameters determined in the traditional association and dissociation assays, as shown in **Table 2**. Furthermore, the kinetic K<sub>D</sub> value  $(2.7 \pm 0.3 \text{ nM})$  derived from this assay was similar to the K<sub>D</sub> value  $(2.4 \pm 0.1 \text{ nM})$ obtained from the [3H]dofetilide saturation assay. In addition, these K<sub>D</sub> values were in the same range as dofetilide's affinity constant that stemmed from the displacement assay ( $K_i = 5.4 \pm 0.8 \text{ nM}$ ) and the kinetically derived  $K_D$  values from kinetic association assay and traditional kinetic experiments (7.1 nM or 6.4

± 1.3 nM, respectively). Overall, the results presented here demonstrated that the competition association assay could be applied to determine the association and dissociation rates of other unlabeled ligands at the K\_11.1 channel. It is noteworthy that a good experimental window was achieved using a concentration of dofetilide at 1-fold its K, value in the competition association assay, as displayed in Figure 4. Reassuringly, when astemizole and E-4031 were tested in this standard three-concentration assay, similar findings were observed as well (data not shown). Thus, the other unlabelled ligands were only tested at 1-fold their K. values rather than three different concentrations in the further competition association experiments in order to improve the throughput of this method.

Table 3. The affinity constants and kinetic parameters of 15 compounds at the K 11.1 channel obtained from the [3H]dofetilide displacement and competition association assay.

Compound	$K_{i}(nM)^{a}$	k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> ) <sup>b</sup>	k <sub>off</sub> (min <sup>-1</sup> ) <sup>b</sup>	K <sub>D</sub> (nM) <sup>c</sup>
Dofetilide	$5.4 \pm 0.8$	$0.048 \pm 0.011$	$0.13 \pm 0.02$	$2.7 \pm 0.3$
Astemizole	$2.5\pm0.2$	$0.17 \pm 0.03$	$0.083 \pm 0.003$	$0.53 \pm 0.07$
E-4031	$13 \pm 0.7$	$0.026 \pm 0.003$	$0.27\pm0.02$	$10 \pm 1$
Sertindole	$34 \pm 5$	$0.048 \pm 0.007$	$0.86 \pm 0.17$	$18 \pm 1$
Terfenadine	$63 \pm 4$	$0.0071 \pm 0.0025$	$0.25\pm0.03$	$39 \pm 7$
Moxifloxacin	$252347 \pm 120995$	$(4.7 \pm 1.0) \times 10^{-6}$	$0.28 \pm 0.06$	$64531 \pm 10276$
Amiodarone	$308\pm33$	$(6.0 \pm 0.7) \times 10^{-4}$	$0.23 \pm 0.02$	$387\pm37$
Chlorpromazine	$2518 \pm 301$	$(1.3 \pm 0.2) \times 10^{-4}$	$0.36 \pm 0.05$	$2714 \pm 154$
Ibutilide	$5.1 \pm 0.4$	$0.046 \pm 0.006$	$0.20\pm0.02$	$4.6 \pm 1.1$
Clofilium	$0.55 \pm 0.09$	$0.23 \pm 0.07$	$0.10\pm0.01$	$0.54 \pm 0.17$
Pimozide	$28 \pm 7$	$0.071 \pm 0.020$	$0.22 \pm 0.07$	$3.2\pm0.4$
Cisapride	$54 \pm 10$	$0.031 \pm 0.009$	$0.59 \pm 0.14$	$20 \pm 2$
Ranolazine	$21379 \pm 5776$	$(1.5 \pm 0.4) \times 10^{-5}$	$0.23 \pm 0.04$	$16672 \pm 1820$
Sotalol	$24663 \pm 1379$	$(1.7 \pm 0.3) \times 10^{-5}$	$0.32\pm0.03$	$19740 \pm 1591$
Thioridazine	$1065 \pm 41$	$(2.3 \pm 0.4) \times 10^{-4}$	$0.24 \pm 0.05$	$1050\pm48$

Values are means (± S.E.M.) of three independent assays performed in duplicate.

<sup>&</sup>lt;sup>a</sup>K<sub>i</sub> values were derived from the [<sup>3</sup>H]dofetilide displacement assay.

 $<sup>{}^{</sup>b}k_{on}^{}$  (k<sub>3</sub>) and k<sub>off</sub> (k<sub>4</sub>) values of unlabelled compounds were determined in the [ ${}^{3}H$ ]dofetilide competition association assay.

<sup>&</sup>lt;sup>c</sup>Kinetic  $K_D = k_{off}/k_{on}$ .

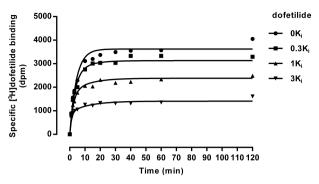


Figure 4. Competition association assay of [3H]dofetilide in the absence (control) or presence of 0.3-, 1- and 3-fold of unlabeled dofetilide's K, value. Experiments were performed at 25 °C using 20 μg of HEK293K, 11.1 membrane protein.

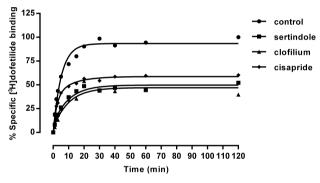


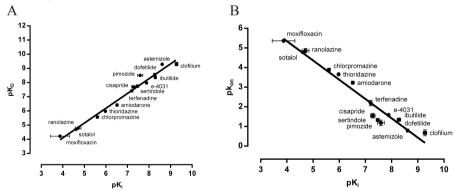
Figure 5. Representative competition association curves of [3H]dofetilide in the absence (control) or presence of unlabelled sertindole, clofilium and cisapride at a concentration of 1-fold their K, values. Experiments were performed at 25 °C using 20 µg of HEK-293K\_11.1 membrane protein.

Subsequently, kinetic parameters of 12 other known K\_11.1 blockers were evaluated and representative normalized curves for several compounds are depicted in Figure 5. The on- and off-rates of all compounds determined by these experiments are shown in Table 3. The association rates for all the compounds were quite distinct with  $k_{on}$  values ranging from  $(4.7 \pm 1.0) \times 10^{-6} \text{ nM}^{-1} \cdot \text{min}^{-1}$ (moxifloxacin) to  $0.23 \pm 0.07$  nM<sup>-1</sup>·min<sup>-1</sup> (clofilium), i.e. an almost 50,000-fold difference between the fastest and slowest associating compounds. On the other hand, the dissociation rates of these 15 compounds were more similar, with the highest value of  $0.86 \pm 0.17 \text{ min}^{-1}$  for sertindole and lowest  $k_{\text{off}}$  of  $0.083 \pm 0.003$ min<sup>-1</sup> for astemizole, i.e. only a 10-fold difference. Considering the kinetically-derived K<sub>D</sub> values shown in **Table 3**, clofilium was the most potent inhibitor to

the  $K_p$ 11.1 channel with a kinetic  $K_p$  value of 0.54  $\pm$  0.17 nM, while moxifloxacin had the lowest affinity ( $K_p = 65 \pm 10 \mu M$ ) to the channel. These results were in the same rank order as the K<sub>1</sub> values derived from the equilibrium displacement assay.

## Correlations of equilibrium $K_i$ with kinetic $K_D$ , $k_{on}$ , and $k_{off}$ values

A plot of the logarithms of kinetic  $K_D$  values (i.e.  $k_{off}/k_{on}$ ) derived from the [ $^3$ H] dofetilide competition association assays and the logarithms of equilibrium K, values obtained from the displacement experiments was made, and a significant correlation (Figure 6A) was observed. This showed an excellent consistence of the results from two different methods and indicated a high reliability of the [3H] dofetilide competition association assay. More interestingly, a significant inverse relationship was also found between pk, and pK, values for unlabelled compounds in Figure 6B. On the contrary, there was no significant linear relationship between pk<sub>off</sub> and pK<sub>i</sub> values ( $r^2 = 0.15$ , P = 0.15, data not shown). Together, this suggested that the [3H]dofetilide competition association assay was successfully validated for assessing the kinetics of other unlabelled competitive compounds, and that the affinity of these compounds at the K\_11.1 channel was mainly controlled by their on-rates rather than off-rates.



**Figure 6.** Correlations between the affinity constant (K<sub>i</sub>) and (A) the kinetically derived equilibrium dissociation constants,  $K_p$  ( $r^2 = 0.98$ , P < 0.0001) and (B) the association rates,  $k_{on}$  for unlabelled ligands at the  $K_v11.1$  channel ( $r^2 = 0.95$ , P < 0.0001).  $K_i$ ,  $K_D$  and k<sub>on</sub> values are listed in **Table 3**.

# Lipophilicity (logK<sub>w.cs</sub>) and membrane partition coefficient (logK<sub>W-IAM</sub>) of K<sub>v</sub>11.1 blockers

The isocratical logK<sub>w.cs</sub> values ('lipophilicity') were evaluated at pH 7.4 and are detailed in **Table 4**. The lipophilicity of 15 reference compounds covered a wide numerical range, varying from 0.56 (sotalol) to 5.52 (amiodarone). We also calculated logP values as a measure for lipophilicity, and plotted these against the  $logK_{w,cs}$  data (**Table 4**). A significant correlation was found between them ( $r^2 =$ 0.81, P < 0.0001), which implied that for this series of compounds calculated logP values can be used interchangeably with the experimentally determined values.

Table 4. The lipophicility, membrane partition coefficients and other physicochemical properties of 15 K, 11.1 blockers.

Compound	logK <sub>W-C8</sub> <sup>a</sup>	logK <sub>W-IAM</sub>	M.W.c	logPc	pK <sub>a</sub> <sup>c</sup>
Dofetilide	0.84	2.08	441.57	0.59	8.99
Astemizole	3.52	3.40	458.57	5.39	8.75
E-4031	1.29	1.98	401.52	1.73	8.01
Sertindole	3.97	3.38	440.94	3.77	8.59
Terfenadine	4.05	4.01	471.67	6.48	9.02
Moxifloxacin	1.12	1.57	401.43	1.97	9.42
Amiodarone	5.52	3.30	645.31	7.64	8.47
Chlorpromazine	3.39	3.36	318.86	4.54	9.20
Ibutilide	0.90	2.53	384.58	3.25	10.40
Clofilium	2.00	-0.35	338.98	2.91	n.a.d
Pimozide	4.69	3.80	461.55	5.83	8.38
Cisapride	3.12	2.66	465.95	2.49	8.24
Ranolazine	2.17	2.39	427.54	2.83	7.17
Sotalol	0.56	0.66	272.36	0.05	9.43
Thioridazine	3.53	3.80	370.58	5.47	8.93

 $<sup>^{</sup>a}$ LogK $_{w-cs}$  values were derived from HPLC experiments on a Supelcosil LC-ABZ, 5 cm × 4.6 mm, 5 μm column.

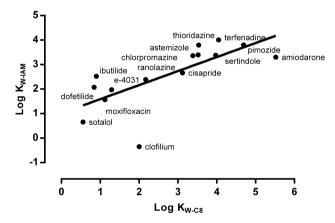
 $<sup>^</sup>bLogK_{w\text{-}IAM}$  values were derived from HPLC experiments on a 10 cm x 4.6 mm, 10  $\mu m$ Regis IAM PC DD2 column.

<sup>&</sup>lt;sup>c</sup>Values were derived from the structure-based calculation plug-in by ChemAxon.

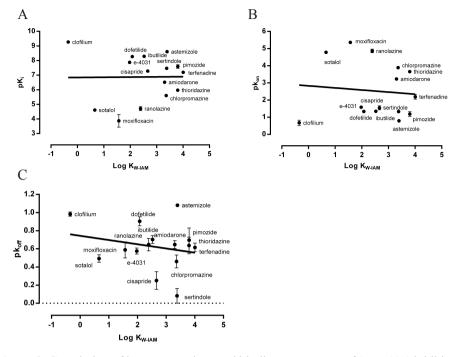
<sup>&</sup>lt;sup>d</sup>n.a. not applicable; this compound is permanently charged.

To mimic the interactions of our ligands with membrane phospholipids, an IAM HPLC column that is a reflection of the lipid environment of a fluid cell membrane on a solid matrix was used to determine membrane partition coefficients (logK $_{\text{W-IAM}}$ ) for all reference compounds. Their logK $_{\text{W-IAM}}$  values were measured at pH 7.4 and are summarized in **Table 4**. Terfenadine had the highest logK $_{\text{W-IAM}}$  value of 4.01, indicating that this compound possessed the highest affinity for membrane phospholipids. On the contrary, the logK $_{\text{W-IAM}}$  for clofilium was only - 0.35, most likely due to its quaternary ammonium moiety, which demonstrated that this compound hardly interacted with phospholipid membranes.

Subsequently, the possible correlation between  $\log K_{W-C8}$  and  $\log K_{W-IAM}$  was studied, and the result is shown in **Figure 7**. A significant linear relationship ( $r^2 = 0.52$ , P < 0.0024) was observed for these two parameters even when including the outlier clofilium. Obviously, a similar significant correlation was also found between calculated logP and  $\log K_{W-IAM}$  values ( $r^2 = 0.52$ , P = 0.0022, data not shown). Next, the relationship between affinity constants or kinetic rate constants of the 15  $K_v$ 11.1 inhibitors and their membrane interactions were investigated, as shown in **Figure 8A – 8C**. Apparently, no relationships were found for any of them (P > 0.05), demonstrating that membrane interactions did not affect affinity and binding kinetics of these ligands at the  $K_v$ 11.1 channel. Similarly, there were no correlations between  $\log K_{W-C8}$  and  $pK_i$ ,  $pk_{on}$  or  $pk_{off}$  values (data not shown).



**Figure 7.** Correlation between  $logK_{w-C8}$  and  $logK_{w-IAM}$  values at pH 7.4 for 15  $K_v$ 11.1 inhibitors ( $r^2 = 0.52$ , P < 0.0024).  $logK_{w-C8}$  and  $logK_{w-IAM}$  values are listed in **Table 4**.



**Figure 8.** Correlation of  $logK_{W-lam}$  values and binding parameters of 15  $K_v$ 11.1 inhibitors. No significant relationship between  $logK_{w,lam}$  and values of (A)  $pK_i$  ( $r^2 = 0.00011$ , P =0.97), **(B)**  $pk_{op}(r^2 = 0.0087, P = 0.74)$  or **(C)**  $pk_{off}$  was observed  $(r^2 = 0.051, P = 0.42)$ . All data used in these plots are listed in Table 3 and Table 4.

# Role of other calculated physicochemical properties in ligand-receptor binding kinetics

Lastly, two other physicochemical properties of the unlabelled compounds, molecular weight (M.W.) and acid/base constant (pK<sub>2</sub>) (Table 4), were compared with their on-/off- rates and dissociation constants. As depicted in Figure 9, no significant correlations were observed between the logarithms of on-rates of the compounds in this study and their molecular properties. Moreover, there were no obvious relationships between the logarithms of off-rates or equilibrium K, values and these physicochemical properties either (P > 0.05, data not shown). These results implied that the binding kinetics of compounds at the K\_11.1 channel was not governed by their overall, macroscopic, physicochemical properties.

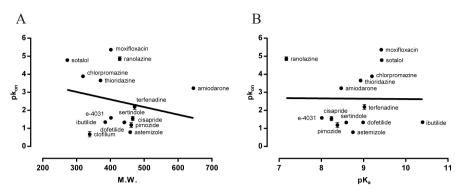


Figure 9. Lack of correlation between the association rates (pkon values) of selected compounds and their physicochemical properties. (A) No significant correlation was observed with M.W. ( $r^2 = 0.049$ , P = 0.43); (B) No significant correlation was observed with pK<sub>a</sub> ( $r^2$ = 0.000076, P = 0.98). M.W. and pK<sub>2</sub> values are shown in **Table 4**.

#### **Discussion and conclusions**

Since the introduction of the competition association assay 23, more and more researchers have utilized this method to study the binding kinetics of unlabelled ligands at their targets, such as muscarinic receptors and the adenosine A<sub>2A</sub> receptor<sup>24, 29, 30</sup>. In addition, a few examples in literature have been reported to apply this technique for the determination of ligand binding kinetics at ligand-gated ion channels<sup>31</sup>. To the best of our knowledge, the current study is the first to assess binding kinetics of ligands interacting with voltage-gated ion channels, in particular the K\_11.1 channel.

It might be argued that our study has some limitations. The stably-transfected HEK293K\_11.1 cells lack the (co-)expression of two ancillary β-subunits, minK and MiRP13. Discrepancies could also exist between membrane binding assays and whole cell experiments, which are more relevant to the channel's natural orientation. For instance, the kinetics of the drug-K, 11.1 interaction have been reported to be use- and frequency-dependent in whole cell patch clamp assays due to special gating kinetics of the channel<sup>32,33</sup>. In our membrane binding assays, we speculate that the channels maintain the same configurations, and thus the kinetic parameters in this investigation should resemble the actual binding process between K\_11.1 blockers and the channel. To perform straightforward and accurate kinetic determinations, experiments were carried out at 25 °C with membrane preparations of stably-transfected HEK293K\_11.1 cells. We reason that all parameters obtained at 25 °C in this investigation lead to a similar compound rank order as those at physiological temperature. In fact, this has been shown for another target based on van't Hoff and Eyring equations<sup>34</sup>.

The linear plot of k<sub>obs</sub> values obtained at increasing concentrations of [3H] dofetilide (Figure 2B) strongly supported a pseudo-first-order kinetic behavior of its binding to the K 11.1 channel, which provided the theoretical foundation for all kinetic analyses in our present study. Other mechanisms of action such as induced-fit or conformational selection would have resulted in deviations from linearity; in such cases, the Motulsky and Mahan model and equations would not have been applicable<sup>35, 36</sup>. All 15 reference compounds had a pseudo-Hill coefficient close to unity in the [3H]dofetilide displacement assay (data not shown), which indicated a competitive mode of inhibition with regard to the radioligand. This is another indication of the simple bimolecular interaction model and corroborated a pseudo-first-order kinetic behavior of the binding to the K\_11.1 channel. This finding was in accordance with previous studies describing the presence of a single affinity state for K<sub>2</sub>11.1 inhibitors at the channel<sup>37, 38</sup>. Subsequently, a newly developed [3H]dofetilide competition association assay was applied to study the binding kinetics of 15 unlabelled reference compounds with diverse chemical structures at the  $K_{\nu}11.1$  channel (**Table 1**). The  $k_{on}$ ,  $k_{off}$  and  $K_{D}$  values of dofetilide from this assay were comparable to the values derived from saturation and kinetic association experiments and traditional association and dissociation assays<sup>22</sup>, indicative of the accuracy and reliability of this new assay. Moreover, the excellent linear correlation between these kinetically derived K<sub>D</sub> values of 15 reference compounds and their K, values from the equilibrium displacement assay further supported the latter (Figure 6A). Taken together, this led us to conclude that a novel [3H]dofetilide competition association assay was successfully developed and validated.

Correlations between the affinity data (K<sub>i</sub>) and the kinetic parameters (onand off-rates) were also investigated in the present study. Surprisingly, there was a significant inverse correlation between pK<sub>1</sub> and pk<sub>0n</sub> values for reference compounds (Figure 6B), whereas no relationship was found between pK, and pk values (data not shown). Although this correlation between affinity and association rates has been shown to be the case for  $\beta_2$ -adrenoceptor agonists<sup>39</sup> and OX2-receptor antagonists40, this phenomenon is supposed to be unusual and counter-intuitive. Dogma has it that the collision theory sets the maximum value for  $k_{\mbox{\tiny on}}$  as the diffusion limit for a ligand and a target, which is about  $10^8 \sim 10^9$ M<sup>-1</sup>·s<sup>-1</sup><sup>41</sup>. In this view, there can only be small variations in the on-rate constants and thus the equilibrium affinity changes are mainly dictated by the off-rates of ligands<sup>15, 41, 42</sup>. In contrast to these classical assumptions, association rates of our selected structurally diverse compounds varied around 50,000-fold, whereas their dissociation rates differed only 10-fold. This clearly indicated that apart from

off-rates, on-rates also play a pivotal role in regulating affinity of ligands to their targeted receptors, at least for the K\_11.1 channel.

Recently, it has been reported that drug-receptor association rates and corresponding affinity were enhanced due to concentrating effects and lateral diffusion of membrane associated  $\beta_2$ -adrenoceptor ligands and  $k_{on}$  values were positively correlated to their lipophilicity and membrane interactions<sup>18, 43</sup>. Furthermore, Sykes et al. 18 observed that dissociation rates were much more correlated to the corrected 'true' affinity, which considers drug concentration gradients in the local environment due to membrane affinity, than the apparent affinity. Additionally, it was recommended that any prediction of pharmacodynamic properties should take membrane interactions as well as lipophilicity into account<sup>17</sup>. Hence, the potential influence of membrane affinity on ligand binding to the K\_11.1 channel was investigated by the determination of both  $logK_{W-C8}$  and  $logK_{W-IAM}$  values. However, when pK<sub>1</sub> and pk<sub>0n</sub> were compared to logK<sub>W-C8</sub> and logK<sub>W-IAM</sub>, no significant relationships were observed in our case (Figure 8). This indicated that membrane interactions of our ligands did not influence their association rates and binding affinity to the K<sub>v</sub>11.1 channel, and that the apparent affinity of these reference compounds is their 'true' affinity. Interestingly, when logK<sub>w,C8</sub> and logK<sub>w</sub>. NAME were compared to each other, clofilium was observed to deviate from the significant linear correlation (Figure 7). Herein, we hypothesized that the lipophilic alkyl chains of clofilium compensated for its hydrophilic quaternary ammonium group and thus dominated its lipophilicity in the octanol-water system, while the positive charge at the nitrogen played a more critical role in the IAM system and weakened its membrane interactions. With regard to dissociation rates, no correlations were found to either  $logK_{W\text{-}C8}$  or  $logK_{W\text{-}IAM}$  values as well (data not shown). This was in accordance with previous findings<sup>18, 44</sup> and demonstrated that the dissociation rate of a drug was not affected by its local concentrations and thus independent of membrane affinity. Of note, typical K, 11.1 inhibitors are known to bind to the inner cavity in the intracellular part of the channel<sup>3</sup>. However, all experiments in the present study were performed with HEK293K\_11.1 membranes instead of intact cells and were thus independent of transmembrane transport of ligands. From Figure 6B, it follows that a significant relationship between affinity and association rates for 15 K, 11.1 inhibitors was found, which showed that ligand-K\_11.1 rather than ligand-membrane association controlled their affinity. Apparently, an aqueous entry pathway predominates the binding of K<sub>2</sub>11.1 blockers to the channel, and an additional compartment induced by the lipid membranes does not play a significant role.

We questioned whether it would be possible to correlate other general physicochemical properties of these molecules to their biological profile (Figure 9). This was not the case, since typical features such as molecular weight and the basicity of the K\_11.1 blockers were not correlated to any of the equilibrium or kinetic binding parameters tested. However, it has been stated previously that physicochemical properties of drugs are relevant for their kinetic parameters<sup>41,45</sup>. For instance, Miller et al. 46 reported that molecular weight was one of the most important factors to affect the dissociation kinetics of ligands from their biological targets including enzymes (kinases) and G protein-coupled receptors. From the present study, it follows that the effects of general molecular properties on the binding kinetics of a ligand to K 11.1 channel are negligible compared to other targets.

Previous studies have also suggested that k<sub>off</sub> values, or drug-target residence times (RTs, the reciprocal of k<sub>off</sub>), play an important role in the duration of pharmacological effects<sup>15, 24, 42</sup>. In other words, an increased drug-target RT could elicit an improved or prolonged drug effect 47. Nevertheless, when adverse effects result directly from drug occupancy at the pharmacological target, such as chlorpromazine binding to D, receptor or clopidogrel to P2Y, receptor, a long RT would result in so-called 'on-target' drug toxicity, and thus a short RT is favored<sup>48</sup>. In our hands, however, the dissociation rates and hence residence times of selected 15 K, 11.1 inhibitors were very similar (Table 3). On the contrary, association rates of our reference compounds varied widely, and are therefore potentially more important to predict the safety of K 11.1 inhibitors.

Previously, Redfern et al.14 have assigned drugs into five categories of torsadogenic propensity and proposed a "30-fold safety margin" between C<sub>max</sub> and K<sub>v</sub>11.1 IC<sub>so</sub> values as a marker to predict TdP. However, it is known that TdP are not only induced by inhibition of the  $K_{\nu}11.1$  channel but also regulated by other potassium channels, sodium and calcium channels<sup>14, 49</sup>. Therefore, a Comprehensive in vitro Proarrhythmia Assay (CiPA) has recently been introduced to assess drug-induced proarrhythmic risk more efficiently and accurately<sup>49, 50</sup>. The kinetics of drug block and unblock were suggested to be incorporated together with drug potency at ion channels in arrhythmia evaluation during the interpretation of this paradigm<sup>49</sup>. Although the present study focuses on the K<sub>2</sub>11.1 channel only, it provides a new medium throughput method to determine the association and dissociation rates of K, 11.1 blockers, which can be used subsequently for other ion channels as well, if a radioligand is available. Interestingly, Veroli et al.51 derived from mathematical models that fast binding K 11.1 blockers in the untrapped configuration and trapped blockers induced greater action potential and QT prolongation. This strengthens the application of our method in the future, and suggests that using slow association and/or fast dissociation characteristics as a "novel marker" might be beneficial for reducing K\_11.1 cardiotoxicity of drug candidates.

In conclusion, a novel [3H]dofetilide competition association assay has been successfully developed and validated to characterize the kinetic binding parameters of unlabelled compounds at the K<sub>2</sub>11.1 channel. Importantly, association rates of K, 11.1 blockers were divergent, i.e. not diffusion limited, and excellently correlated to their affinity values. In addition, membrane interactions and other molecular properties do not influence the affinity and kinetic binding parameters of ligands at the K<sub>2</sub>11.1 channel. Altogether, this is quite unlike the mechanisms of interaction proposed for other drug target classes, such as kinases and G protein-coupled receptors. Hence, we postulate that association rates can be used to assess a compound's K\_11.1 liability, apart from its affinity. However, further studies involving compounds with a wide range of k of values are required to assess the effect of residence time on drug-induced K\_11.1 cardiotoxicity. Overall, we believe that this research provides novel insights into the kinetic study of ion channels, which can hopefully help to avoid K,11.1-induced cardiotoxicity of drug candidates in the future.

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# Chapter 7

Structure-affinity relationships (SARs) and structure-kinetics relationships (SKRs) of  $K_{\rm v}11.1$  (hERG) blockers

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#### **Abstract**

We explored structure-kinetics relationships and structure-affinity relationships in four series of  $\rm K_v 11.1$  (hERG) blockers. We learned that despite dramatic differences in affinity and association rates, there were hardly any variations in the dissociation rate constants of these molecules with residence times (RTs) of a few minutes only. Hence, we synthesized sixteen novel molecules, in particular in the pyridinium class of compounds, to further address this peculiar phenomenon. We found molecules with very short RTs (< 1 min) and much longer RTs (> 100 min). This enabled us to construct a  $\rm k_{on}$ - $\rm k_{off}$ - $\rm K_{D}$  kinetic map for all compounds, providing a possible framework for a further and more precise categorization of  $\rm K_v 11.1$  blockers. Additionally, two representative compounds were tested in patch clamp assays, and their RTs were compared with patch clamp IC $_{50}$  values. Our findings strongly suggest that the simultaneous study of ligand affinity and kinetic parameters may help to explain and predict  $\rm K_v 11.1$ -mediated cardiotoxicity.

# Introduction

Cardiac safety has become a major concern facing the pharmaceutical industries and regulatory agencies over the last several decades. A substantial number of drugs, including both cardiac and non-cardiac medications, have been restricted in their applications or withdrawn from the market due to their K\_11.1-induced cardiotoxicity.<sup>1, 2</sup> The K<sub>2</sub>11.1 channel, often referred to as the human ether-à-gogo-related gene (hERG) K+ channel, mediates the most important component of phase 3 repolarization of human ventricular myocytes.<sup>3</sup> Blockade of the K\_11.1 channel can lead to the prolongation of action potential durations (APDs) and an elevated risk of lethal arrhythmias such as Torsade de Pointes (TdP).4 As a consequence, measurement of K<sub>2</sub>11.1 liability of drugs is a critical component of regulatory guidelines for preclinical QT studies.<sup>5, 6</sup> In this regard, a 30-fold safety margin between  $K_v11.1~IC_{50}$  and  $C_{max}$  values (maximum free plasma concentrations) has been proposed to define the cardiac safety of compounds.7 However, this approach of considering IC<sub>50</sub> values solely is rather arbitrary and crude, which might explain unsatisfactory quantitative predictions and sometimes excludes useful medications that are not problematic. <sup>6,8,9</sup> For instance, the IC<sub>50</sub> value of ketoconazole  $(1-[4-(4-\{[(2R,4S)-2-(2,4-Dichlorophenyl)-2-(1H-imidazol-$ 1-ylmethyl)-1,3-dioxolan-4-yl] methoxy}phenyl)piperazin-1-yl]ethan-1-one) at the K 11.1 channel is relatively close to its effective therapeutic plasma concentration (11-fold difference), yet it has been proven to be a safe drug in clinical trials due to the fact that this drug has a slow binding rate and is not trapped into the channel. 9, 10 Therefore, knowledge of association and dissociation kinetics of drugs at the K<sub>2</sub>11.1 channel may be important in defining their propensities to prolong the QT interval.

The concept of structure-affinity relationships (SARs) describing the interdependency between compound structures and binding affinity has been extensively explored over the past decades. 11, 12 However, similar studies illustrating the dependency between compound structures and binding kinetics, coined as structure-kinetics relationships (SKRs), have been less well investigated to date. Recently, drug-target kinetics, in particular receptor-ligand residence times (RTs), have been shown to also determine the therapeutic potential of drug candidates in vivo, and to be predictive of drug efficacy and safety. 13, 14 Consequently, there is an emerging awareness of the importance of measuring the kinetics of drug-target interactions. SKRs are now being reported and discussed for ligand binding to G protein-coupled receptors (GPCRs) and enzymes. 11, 15-17 With regard to ion channels, measuring the interaction kinetics of drugs with the K\_11.1 channel has already been recommended to be incorporated into Comprehensive in vitro Proarrhythmia Assay (CiPA) strategies to hopefully provide significant improvements of proarrhythmic assessments. 18, 19 Recent studies have demonstrated that K<sub>v</sub>11.1 blockers with similar potency (IC<sub>50</sub> values) but distinct binding kinetics can have markedly diverse proarrhythmic potential, suggesting the necessity to extend K<sub>v</sub>11.1 inhibition assays with studies on investigating the dynamics of drug-channel interactions. 1, 8, 9 In addition, particular interest has arisen in the so-called trapping phenomenon, characterized by capture of a drug within the K\_11.1 channel and thus a slow dissociation rate from the channel, which harbors an enhanced proarrhythmic risk caused by the drug. 20, 21 Moreover, a substantial underestimation of cardiac risks has been observed in assessing the ventricular action potential due to neglecting drug-binding kinetics at cardiac K\_11.1 channels.8 Therefore, there is an urgent need to add the dimension of binding and unbinding kinetics to the more routinely determined affinity values in order to evaluate the arrhythmogenic potential of drug candidates in vitro.

In our previous studies, we have successfully validated a [3H]dofetilide competition association assay that can determine the association/dissociation rates and RTs of compounds accurately and efficiently.<sup>2</sup> In the present study a structurally diverse set of K 11.1 blockers with four different scaffolds were newly synthesized or selected from our in-house library published previously, <sup>22-25</sup> and their kinetic parameters (k<sub>on</sub>, k<sub>off</sub>, K<sub>D</sub> and RTs) as well as affinity (K<sub>i</sub>) were determined in [3H]dofetilide binding assays. A k<sub>on</sub>-k<sub>off</sub>-K<sub>D</sub> 'kinetic map' was delineated for all compounds together with three reference compounds: astemizole (1-[(4-fluorophenyl)methyl]-N-[1-[2-(4-methoxyphenyl)ethyl]-4-piperidyl]benzoimidazol-2-amine, a notorious K<sub>v</sub>11.1 blocker), dofetilide (N-[4-(2-{[2-(4-methane sulfonamidophenoxy)ethyl] (methyl)amino}ethyl)phenyl]methanesulfonamide, a specific K 11.1 blocker with restricted usage in the market) and ranolazine ((RS)-N-(2,6-dim-ethylphenyl)-2-[4-[2-hydroxy-3-(2-methoxyphenoxy)-propyl] pipe-razin-1-yl]acetamide, a safe K,11.1 blocker).2, 26-28 Two compounds were further tested in a manual patch clamp assay. The SARs and SKRs derived for the compounds in this study reveal new features and aspects of the blockers' interactions with the K<sub>v</sub>11.1 channel. This information is expected to provide valuable information for circumventing K<sub>2</sub>11.1-induced cardiotoxicity during the preclinical stages of the drug discovery pipeline.

# Results and discussion

## Chemistry

Compounds 19, 20, 24, 25, 27-30 and 39-46 were newly synthesized in order to

obtain compounds with various RT values besides a diverse range of K, and k values. All other compounds were selected from our in-house library based on their distinct affinity and kinetic parameters, and synthesis of these compounds has been reported in our previously published studies.<sup>22-25</sup>

As shown in **Scheme 1**, clofilium (4-(4-Chlorophenyl)butyl-diethyl-heptylammonium) derivatives 19 and 20 were prepared from 4-iodotoluene in a 3-step sequence as following: Sonogashira cross-coupling of 4-iodotoluene and propargyl alcohol, formation of the corresponding tosylate and displacement of the tosylate with the corresponding secondary amine.<sup>24</sup>

Scheme 1. Synthesis of clofilium derivatives (19 and 20)

i) PdCl<sub>2</sub>(PPh<sub>2</sub>)<sub>2</sub>, CuI, Et<sub>2</sub>N, THF, r.t.; ii) tosyl chloride, KOH, Et<sub>2</sub>O, 0 °C to r.t.; iii) K<sub>2</sub>CO<sub>2</sub>, DMF, r.t.

To investigate the effect of flexibility in the side chains of aliphatic pyridinium derivatives, several symmetrical 2,6-substituted N-methylated-pyridine compounds with different alkyl chain lengths (24, 25 and 27-30) were synthesized according to Scheme 2. The first series of compounds (S23a-e) were derived from commercially available 2,6-dibromo-pyridine and 1-alkynes with various lengths via a Sonogashira coupling reaction.<sup>25</sup> The final compounds (24, 25 and 27-29) were synthesized in relatively high yields by reduction of triple bonds to single bonds and methylation of neutral pyridines. 30 was obtained directly from S23e by methylation of the central nitrogen with a good yield of 94%.

Next, influences of different substituents at the phenyl rings on affinity and kinetic parameters of K\_11.1 blockers were explored, and a series of substituted biphenyl pyridines (39-46) were synthesized. As displayed in Scheme 3, compound S32 was obtained by Sonogashira reaction of 2,6-dibromo-pyridine and ethynyltrimethylsilane with Pd(PPh<sub>3</sub>)<sub>4</sub> and CuI as catalysts instead of PdCl<sub>2</sub>(PPh<sub>2</sub>)<sub>2</sub> and CuI displayed in **Scheme 2**. <sup>29</sup> After one-pot deprotection and Castro-Stephens coupling with 1-bromo-4-iodobenzene, <sup>30</sup> S32 was converted to S33 in a low yield that was probably caused by formation of undesired oxidative

**Scheme 2.** Synthesis of symmetrical 2,6-substituted *N*-methylated pyridines with various alkyl chain lengths

i) PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, CuI, Et<sub>2</sub>N, r.t.; ii) 10% Pd/C, H<sub>2</sub>, THF, MeOH, r.t.; iii) CF<sub>3</sub>SO<sub>2</sub>OCH<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to r.t.

**Scheme 3.** Synthesis of symmetrical 2,6-substituted *N*-methylated pyridines with different biphenyl substituents

A) Synthesis of compounds 39, 40, 43, 44 and 46: i) Pd(PPh<sub>3</sub>)<sub>4</sub>, CuI, diisopropylamine, toluene, r.t.; ii) CuCl ,PPh3, potassium benzoate, 1-bromo-4-iodobenzene, DMI, 120 °C; iii) Pd(PPh<sub>3</sub>)<sub>4</sub>, K<sub>2</sub>CO<sub>3</sub>, toluene, ethanol, substituted-phenylboronic acid, 50 °C; iv) CF<sub>3</sub>SO<sub>2</sub>OCH<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to r.t.

**B)** Synthesis of compound **45**: i) BBr<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C to r.t.

Glaser homo-coupling and intra-polymerization of reacting molecules.<sup>31</sup> Suzuki reactions were further performed with \$33 and different substituted phenylboronic acids to yield the neutral pyridine compounds (S34a-e).<sup>32</sup> from which final compounds (39, 40, 43, 44 and 46) were derived via methylation of the central nitrogen in moderate to high yields. Subsequently, demethylation of 44 with boron tribromide led to the formation of compound 45. Notably, synthetic routes mentioned above were not suitable for producing neutral biphenyl pyridines with methoxy and chlorine substituents at the meta-position of the second benzene ring. Hence, 41 and 42 were synthesized by different methods shown in Scheme 4. In the first step, Suzuki reactions were applied between the substituted phenylboronic acid and 1-bromo-4-iodobenzene in the presence of different palladium catalysts to produce S43a and S43b in moderate yields of 27% and 40%, respectively. S43a and S43b further reacted with ethynyltrimethylsilane through Sonogashira reaction,<sup>33</sup> resulting in high yields of S44a and S44b. Compound 41 was eventually obtained from S44a with a yield of 36% through three steps: deprotection, Sonogashira reaction and methylation, whereas 42 was obtained from S44b via a one-pot Sonogashira reaction and methylation in a yield of 45%.

Scheme 4. Synthesis of symmetrical 2,6-substituted N-methylated pyridines (41 and 42)

A) Synthesis of compounds S44a-b: i) 2M K<sub>2</sub>CO<sub>3</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub>, toluene, ethanol, 80 °C; ii) 2M K,CO<sub>3</sub>, Pd(OAc)<sub>2</sub>, PPh<sub>3</sub>, toluene, ethanol, 80 °C; iii) PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>3</sub>, CuI, PPh<sub>3</sub>, THF, piperidine, 120 °C.

B) Synthesis of compound 41: i) 2M NaOH, diethyl ether, methanol; ii) PdCl<sub>2</sub>(PPh<sub>2</sub>)<sub>2</sub>, CuI, Et<sub>3</sub>N, 2, 6-dibromo-pyridine; iii) CF<sub>3</sub>SO<sub>2</sub>OCH<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to r.t.

C) Synthesis of compound 42: i) CuCl, PPh<sub>2</sub>, potassium benzoate, 2, 6-dibromo-pyridine, DMI, 120 °C; ii) CF<sub>3</sub>SO<sub>2</sub>OCH<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to r.t.

# **Biology**

The binding affinity and kinetic parameters of all compounds were determined in [3H]dofetilide competitive displacement and competition association assays on HEK293 cell membranes stably transfected with the K\_11.1 channel (HEK-293K, 11.1) as described previously by our group.<sup>2</sup> As shown in **Figure 1**, compounds with a wide range of IC<sub>50</sub> values demonstrated a monophasic binding behavior instead of a biphasic pattern reported for a [3H]astemizole binding assay, <sup>23, 25</sup> indicating that the binding process of compounds followed a single-step bimolecular interaction scheme, and thus, fulfilled the theoretical requirements

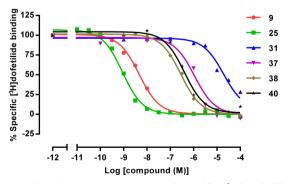


Figure 1. Representative displacement curves of specific [3H]dofetilide binding by 9, 25, 31, 37, 38 and 40. Experiments were performed at 25 °C using 20 μg of HEK293K 11.1 membrane protein.

for the Motulsky-Mahan approach to evaluate the binding kinetics of unlabeled ligands.<sup>34</sup> Figure 2 shows the competition association assay for three representative compounds (31, 37 and 38) with different binding kinetics, in particular varying RTs. Compound 37 had a shorter RT than [3H]dofetilide with an association curve slowly and monotonically approaching equilibrium versus time, whereas 38 possessed a longer RT compared to the radioligand, indicated by a typical 'overshoot' and then a decline in the association curve. 15 Meanwhile, 31 exhibited a RT similar to dofetilide, evidenced by the same curve shape compared to the association curve of radioligand alone. As the K, values of all tested compounds calculated from the IC<sub>50</sub> values mentioned above, <sup>35</sup> were in good agreement with kinetically derived  $K_D$  data  $(K_D = k_{off}/k_{on}$ , Figure 3A), only  $K_i$  values were used to summarize the SARs in this study. The consistency of K, and K<sub>D</sub> values extended our previous finding to a more general observation,<sup>2</sup> proving the reliability of the [3H]dofetilide competition association assay in assessing the kinetics of unlabeled ligands at the K<sub>v</sub>11.1 channel. In addition, the k<sub>on</sub> values for all ligands were significantly correlated to their K, values (P < 0.0001, Figure 3B), which is in agreement with our published data.<sup>2</sup> In the same publication affinity and kinetic parameters of the parental compounds dofetilide, E-4031 and clofilium were also measured and reported.<sup>2</sup> Furthermore, a provisional k<sub>on</sub>-k<sub>off</sub>-K<sub>D</sub> 'kinetic map' was constructed based on the compounds' varying affinity and kinetic parameters in order to divide these compounds into four different groups (quadrants) of potentially negligible (quadrant IV), moderate (quadrant I and III) and high arrhythmic (quadrant II) side effects (Figure 4). Finally, we selected two compounds (21) and 38) with comparable binding affinity but distinct association and dissociation rates or RTs to be studied in a whole-cell patch clamp assay using the same HEK-293K, 11.1 cells (Figure 5).

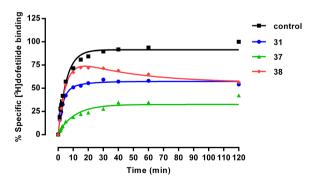
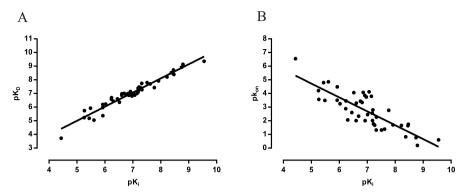


Figure 2. Representative competition association curves of [3H]dofetilide in the absence (control) or presence of unlabeled 31, 37 and 38 at their IC<sub>50</sub> value concentrations. Experiments were performed at 25 °C using 20 µg of HEK293K, 11.1 membrane protein.



**Figure 3.** Correlation between pK<sub>1</sub> and pK<sub>D</sub> (**A**,  $r^2 = 0.95$ , P < 0.0001) or pk<sub>on</sub> (**B**,  $r^2 = 0.68$ , P < 0.0001) values of all tested compounds as obtained from [3H]dofetilide competitive displacement experiments and competition association assays, respectively.

# Structure-affinity relationships (SARs) and structure-kinetics relationships (SKRs)

#### **Derivatives of dofetilide**

First, we explored the effects of pK<sub>a</sub> values of the central nitrogen on affinity, association rates and RTs of compounds 1-8 (Table 1), in which the pK values were taken from our previous publication.<sup>22</sup> Briefly, increasing the pK<sub>a</sub> values via different N-alkyl substituents resulted in an increase of affinity of these compounds (1-5, 7 and 8) at the K\_11.1 channel. Association rates of these compounds were less sensitive to pK<sub>a</sub> alterations but showed a similar trend as the affinity. For instance, compound 8 (pK  $_{a}$  = -4.95) displayed the lowest affinity and smallest  $k_{on}$  value, while 3 (pK<sub>a</sub> = 9.61) had the highest affinity and fastest association rate at the channel. These changes corresponded rather well with the principle that reducing pK<sub>2</sub> values of the central nitrogen of K<sub>2</sub>11.1 blockers decreases the proportion of molecules in the protonated form at physiological pH and further make fewer active species available for the productive  $\pi$ -cation interactions with amino acid Tyr652 lining the inner cavity of the channel. 36, 37 Additionally, zwitterionic compound 6 with a CH<sub>2</sub>COOH substituent at the central nitrogen had a lower affinity and slower association rate than 8, indicating that the negatively charged carboxylate group caused unfavorable interactions of the K<sub>v</sub>11.1 blocker with the channel.<sup>38</sup> Overall, however, the RTs (1/k<sub>off</sub>) of **1-8** were not much impacted by pK<sub>a</sub> values of the central nitrogen and zwitterionic propensity of molecules, with values between  $0.56 \pm 0.06$  (7) and  $3.8 \pm 1.0$  min (3).

**Table 1.** Binding affinity and kinetic parameters of dofetilide derivatives

$$R_1 = O \underbrace{\stackrel{R_3}{\underset{n_1}{\bigvee}} R_2}_{N_{n_1} \underbrace{\stackrel{\parallel}{\underset{n_2}{\bigvee}}} R_2}$$

Compd	$\mathbf{R}_{_{1}}$	R <sub>2</sub>	R <sub>3</sub>	n <sub>1</sub>	n <sub>2</sub>	K <sub>i</sub> (nM)	k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> )	k <sub>off</sub> (min <sup>-1</sup> )	K <sub>D</sub> (nM)	RT (min)
1	4-NO <sub>2</sub>	4-NO <sub>2</sub>	Н	2	2	47±1	0.0054±0.0015	0.27±0.06	53±6	3.7±0.8
2	4-NO <sub>2</sub>	4-NO <sub>2</sub>	$CH_3$	2	2	24±2	0.041±0.004	$0.82 \pm 0.09$	20±4	1.2±0.1
3	4-NO <sub>2</sub>	4-NO <sub>2</sub>	$CH_2CH_3$	2	2	4.2±0.5	0.15±0.06	0.26±0.07	3.0±0.2	3.8±1.0
4	4-NO <sub>2</sub>	4-NO <sub>2</sub>	$\mathrm{CH_2CH_2F}$	2	2	65±13	0.0096±0.0019	0.30±0.08	41±11	3.3±0.9
5	4-NO <sub>2</sub>	4-NO <sub>2</sub>	CH <sub>2</sub> CN	2	2	203±6	0.0046±0.0012	$0.40\pm0.05$	99±26	2.5±0.3
6	4-NO <sub>2</sub>	4-NO <sub>2</sub>	$\mathrm{CH_{2}COOH}$	2	2	588±31	0.0013±0.0001	0.35±0.10	260±52	2.9±0.8
7	4-NO <sub>2</sub>	4-NO <sub>2</sub>	COCH <sub>3</sub>	2	2	251±32	0.0098±0.0034	1.8±0.2	132±5	0.56±0.06
8	4-NO <sub>2</sub>	4-NO <sub>2</sub>	COCF <sub>3</sub>	2	2	355±14	0.0025±0.0003	0.61±0.09	247±27	1.6±0.3
9	4-NO <sub>2</sub>	4-NO <sub>2</sub>	$CH_3$	4	2	1.6±0.3	0.64±0.29	0.26±0.05	0.73±0.39	3.8±0.7
10	4-NO <sub>2</sub>	4-NO <sub>2</sub>	$CH_3$	4	1	66±6	$0.0028 \pm 0.0008$	0.22±0.05	85±7	4.5±1.0
11	3, 4-di-Cl	3, 4-di-Cl	$CH_3$	2	2	5571±507	(6.2±3.0)x10 <sup>-5</sup>	0.19±0.01	5807±228	5.3±0.2
12	Н	4-OCH <sub>3</sub>	CH <sub>3</sub>	2	2	944±58	(5.6±0.9)x10 <sup>-4</sup>	0.30±0.07	591±197	3.3±0.7

Next, the lengths of side alkyl linkers and substituents on the phenyl rings were investigated for the analysis of SARs and SKRs. Comparable to our own studies on distance-related flexibility, 24, 39 elongation of the alkyl chain between the central nitrogen and phenoxy moiety enhanced the K<sub>2</sub>11.1 affinity and association rates (9 versus 2), while shortening the other alkyl linker with the phenyl ring lowered the affinity and association rates (10 versus 2 and 9). Nonetheless, 9 (RT =  $3.8 \pm 0.7$  min) and 10 (RT =  $4.5 \pm 1.0$  min) demonstrated similar dissociation characteristics at the K\_11.1 channel compared to other compounds in Table 1, suggesting that varying chain lengths between the central nitrogen and the two peripheral aromatic rings exerted negligible effects on the dissociation process of this type of ligands from the K<sub>v</sub>11.1 channel. In addition, replacement of NO<sub>v</sub> groups on the phenoxy and phenyl rings with 3,4-di-Cl or other groups like OCH, dramatically decreased the K, 11.1 affinity and association rates of ligands (11 and 12 versus 2). Compound 11 had the lowest affinity ( $K_i = 5571 \pm 507$  nM), slowest association rate ( $k_{on} = 6.2 \pm 3.0$ )× $10^{-5}$  nM<sup>-1</sup>·min<sup>-1</sup>) but longest RT of  $5.3 \pm 0.2$  min at the K\_11.1 channel amongst all dofetilide derivatives (1-12). This implies that addition of electron-withdrawing groups on the aromatic rings reduces the affinity and association kinetics of  $K_{\nu}11.1$  blockers by hampering their  $\pi$ -stacking and hydrophobic interactions with Phe656 and Tyr652 residues lining the pore cavity of the K\_11.1 channel, 36, 40 whereby this hindered interaction did not obviously disturb the dissociation of ligands. Overall, the association rates of this series of compounds were (negatively) correlated to their K<sub>i</sub> values (see also Figure 3B), which is in full agreement with our previous finding that ligand affinity for the

K<sub>v</sub>11.1 channel are mainly regulated by the compounds' association rates.<sup>2</sup> Unlike the widely varying K, and k an values, RTs of these dofetilide analogues were largely similar.

#### **Derivatives of E-4031**

Benzoylpiperidine analogues (13, 14 and 16) demonstrated comparable K, and k<sub>on</sub> values at the K<sub>v</sub>11.1 channel. This implicates that: i) electron withdrawing or donating groups at the two peripheral rings, ii) slightly lengthening the carbon chain between the central nitrogen atom and the phenyl ring and iii) permanent protonation of the basic nitrogen, had much smaller effects on K, 11.1 affinity and association kinetics of E-4031 (N-[4-[1-[2-(6-Methylpyridin-2-yl)ethyl]piperidine-4-carbonyl]phenyl]) derivatives (**Table 2**) compared to the dofetilide series (Table 1). However, when the benzoylpiperidine was replaced by benzoylpiperazine (15), the  $K_i$  value was drastically increased to 5418  $\pm$  374 nM and its association rate was substantially decreased to  $(2.7 \pm 0.6) \times 10^{-4} \text{ nM}^{-1} \cdot \text{min}^{-1}$ . Distinct from the variations in affinity and association rates, all E-4031 derivatives displayed comparably short RTs between  $2.0 \pm 0.5$  (14) and  $2.9 \pm 0.8$  min (16) at the K\_11.1 channel.

Table 2. Binding affinity and kinetic parameters of E-4031 derivatives

$$R_1$$

Compd	R <sub>1</sub>	R <sub>2</sub>	X	R <sub>3</sub>	n	K <sub>i</sub> (nM)	k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> )	k <sub>off</sub> (min <sup>-1</sup> )	K <sub>D</sub> (nM)	RT (min)
13	4-Cl	3, 4-di-Cl	СН	-	2	54±3	0.021±0.009	0.39±0.06	36±22	2.6±0.4
14	4-CH <sub>3</sub>	Н	СН	-	3	60±2	0.017±0.003	0.49±0.12	34±14	2.0±0.5
15	Н	Н	N	-	2	5418±374	(2.7±0.6)x10 <sup>-4</sup>	$0.46\pm0.03$	1812±313	2.2±0.1
16	4-Cl	Н	СН	$\mathrm{CH}_3$	2	48±1	$0.048 \pm 0.022$	0.35±0.09	18±9	2.9±0.8

#### **Derivatives of clofilium**

Comparable to our own and other studies, 24, 25, 41 increasing the rigidity by introduction of alkene and alkyne moieties in the chain between the phenyl ring and central nitrogen atom diminished the K, 11.1 affinity and association rates of clofilium derivatives (18 and 21 versus 17, Table 3), whereby permanent protonation of the central nitrogen atom with a CH2CH3 group resulted in a dramatic enhancement of the affinity and association rate ( $K_i = 1.8 \pm 0.2$  nM and  $k_{on} =$  $0.17 \pm 0.07$  nM<sup>-1</sup>·min<sup>-1</sup> for **22**). Therefore, rigidity and protonation play pivotal roles in determining the affinity and association rates of clofilium analogues at the K\_11.1 channel. However, permanently protonating the central nitrogen obviously imposed more significant effects on K<sub>2</sub>11.1 affinity and association rates, and thus compensated for the negative influences exerted by increasing rigidity via introduction of an alkyne bond (22 versus 17). According to different in-silico  $K_v^{-1}$ 11.1 models, protonation of the central nitrogen atom facilitates the  $\pi$ -cation interaction of ligands with the crucial binding residue (Tyr652) that faces into the S6 domain of the channel.<sup>36, 42-44</sup> With regard to dissociation characteristics, compounds 17 and 18 had similar RTs at the K\_11.1 channel (2.7  $\pm$  0.3 and 2.0  $\pm$  0.5 min for 17 and 18, respectively), while 21 and 22 having an alkyne moiety displayed longer RTs. Subsequently, two novel compounds with an alkyne group but different side chain lengths (19 and 20) were synthesized in order to obtain more variations in dissociation rate constants or RTs. Both compounds exhibited comparably low affinity and slow association rates at the K<sub>v</sub>11.1 channel, indicating the flexibility adjusted by side chain lengths imposed insignificant effects on weakening K<sub>2</sub>11.1 affinity and association kinetics of this series of ligands. Likewise, the dissociation rates of 19 and 20 were identical with comparatively longer RTs of  $7.7 \pm 1.6$  and  $6.7 \pm 2.0$  min, respectively. Altogether, introduction of an alkyne group in these clofilium analogues slightly increased their RTs at the K<sub>2</sub>11.1 channel (19-22 versus 17-18). With all these clofilium derivatives (Table 3), their association rates illustrated the same overall tendency as the affinity shown in Figure 3B. Most importantly, compound 22 showed the highest affinity, fastest association rate and longest RT, implying a strong and prolonged inhibition of the K\_11.1channel.

**Table 3.** Binding affinity and kinetic parameters of clofilium derivatives

$R_1$ $R_2$ $N$ $N$											
Compd	R <sub>1</sub>	$\mathbf{R}_2$	A	n	K <sub>i</sub> (nM)	k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> )	k <sub>off</sub> (min <sup>-1</sup> )	K <sub>D</sub> (nM)	RT (min)		
17	4-Cl-C <sub>6</sub> H <sub>4</sub> CH <sub>2</sub>	-	—CH <sub>2</sub> CH <sub>2</sub> —	6	31±5	0.048±0.019	0.37±0.04	16±10	2.7±0.3		
18	$C_6H_5CH_2$	-	-HC:CH-	6	119±2	0.0098±0.0042	0.50±0.12	86±45	2.0±0.5		
19	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>5</sub>	-	_C≣C_	6	3282±613	(3.2±2.0)x10 <sup>-4</sup>	0.13±0.03	1188±122	7.7±1.6		
20	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>5</sub>	-	_C≣C_	2	1196±141	(3.1±0.7)x10 <sup>-4</sup>	0.15±0.05	637±339	6.7±2.0		
21	$(C_6H_5)_2CH$	-	_C≣C—	6	97±13	0.0035±0.0008	0.27±0.06	100±52	3.7±0.8		
22	4-Cl-C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> CH <sub>3</sub>	_C≣C—	6	1.8±0.2	0.17±0.07	0.13±0.04	1.2±0.6	7.7±2.6		

# **Derivatives of pyridinium**

As compounds with high lipophilicity can bind to residue Phe656 of the K<sub>v</sub>11.1 channel through hydrophobic van der Waals interactions as well, peripheral aro-

maticity is not essential for their K\_11.1 binding. 25, 37 In this context, eight aliphatic pyridinium analogues with various side chains (23-30) were synthesized or selected to define their SARs and SKRs (Table 4). Several general observations can be made: i) permanently protonating the central nitrogen atom by a methyl substituent (23 versus other compounds), and ii) increasing flexibility via stretching the side alkyl chain from n = 3 to n = 6 (24 versus 25-27) enhanced the K\_11.1 affinity of compounds. However, the binding affinity were decreased with iii) the continuous prolongation of side chains up to n=7 (24 versus 28-29) and iv) replacement of alkane with alkyne (29 versus 30). All these results are in excellent agreement with protonation and rigidity rules, 45 and with the fact that the large aperture at the bottom of the K\_11.1 channel cavity has boundaries that accommodate molecules with defined sizes. 46 Likewise, protonation, flexibility of side chains and rigidity by an alkyne moiety exerted the same overall effects on association rates of these aliphatic pyridines, which implicates that the affinity of aliphatic pyridinium blockers were correlated to their association rate constants at the K\_11.1 channel (see also for dofetilide, E-4031 and clofilium derivatives, Figure 3B). Intriguingly, RTs in this series of compounds tended to be longer and different from the three other compound classes mentioned above as well as from our previous data.<sup>2</sup> The newly synthesized compound 29 had the longest RT of 36  $\pm$  2 min, whereas 24 showed the shortest RT with  $4.3 \pm 0.8$  min. In addition, pyridines with *n*-octyl side chains (23, RT =  $19 \pm 1$  min) presented a comparable RT to 26 (25  $\pm$  1 min) and 28 (16  $\pm$  4 min) at the K\_11.1 channel. Apparently, permanent protonation of the central nitrogen atom (23 versus 26) and prolongation of side alkyl chains (28 versus 23 and 26) had much less influence on RTs compared to their roles in overall affinity and association rates. It is noteworthy that more rigidity induced by an alkyne moiety diminished the RT to  $8.3 \pm 1.7$  min for 30. Altogether, increasing rigidity of aliphatic pyridine blockers can decrease their affinity, association rates and RTs simultaneously (30 versus 29), and thus, it may provide an efficient solution for circumventing K<sub>2</sub>11.1 side effects of drug candidates. In a second series of pyridinium analogues (31-33), the K 11.1 affinity and association rates were also increased by permanent protonation of the central nitrogen atom (32 versus 31) but decreased with an alkyne moiety (33 versus 32). However, the RTs of **31-33** were comparable between  $7.7 \pm 0.7$  (**32**) and  $9.1 \pm 1.7$ min (33), indicating different dissociation profiles of these compounds compared to aliphatic pyridinium analogues of which rigidity affected the RTs significantly.

Next, a series of biphenyl substituted pyridinium derivatives (34, 35 and 38-46) together with their precursors (36 and 37) were synthesized or selected to achieve more variations in RTs. As shown in Table 4, permanent protonation and rigidification by alkyne imposed the same effects on K 11.1 affinity as our

findings for other derivatives (34 versus 35-46 and 35 versus 36-46, respectively). Additionally, decreasing bulkiness via removal of the second benzene rings or replacing the aromatic phenyl rings close to the central nitrogen atom with cyclohexyl diminished the affinity of K 11.1 blockers (36-37 versus 38), possibly caused by reduced  $\pi$ -stacking and hydrophobic interactions.<sup>36</sup> Furthermore, introduction of different groups at the second benzene rings lowered their K\_11.1 affinity compared to the non-substituted compound (39-45 versus 38), indicating a possible steric hindrance of these groups in ligand binding. <sup>22, 25</sup> Although the k<sub>on</sub> values showed a comparable decline to affinity because of the increased rigidity by an alkyne moiety (35 versus 38-46) and reduction of the aromatic rings (36 versus 37), the negative relationship between K, and k<sub>on</sub> values was not conspicuous for this series of compounds. For instance, compound 38 showed a smaller K, value as well as slower association rate than 39. On the contrary, the RTs of these compounds were significantly increased by replacing ethyl with alkyne (35 versus 38-46). Moreover, decreasing bulkiness of these pyridinium analogues led to shorter RTs (36-37 versus 38-46). Compound 36 and 37 showed the shortest RTs amongst all these derivatives, providing evidence that reduced bulkiness facilitates unbinding dynamics of K<sub>2</sub>11.1 blockers at the channel. Our findings are contradictory to the classic 'foot-in-door' mechanism, in which bulky residues of K<sub>2</sub>11.1 blockers prevent the closure of the channel gate and thus promote the channel recovery from blockade.<sup>20</sup> The explanation for our results is that less hydrophobic interactions and smaller sizes of molecules can accelerate the dissociation of such ligands from the channel. Compound 38 without any substituents on the benzene rings had the longest RT of  $105 \pm 4$  min at the K<sub>2</sub>11.1 channel, while substituents at the second benzene rings (39-46) often had distinct and lowering effects on RTs. While introducing CH<sub>2</sub> (40) and Cl (42) at the meta-position did not affect RTs significantly, a substituent at the ortho-position (39) of the benzene rings reduced the RT, like at the para-position (43-46) but to a lesser extent (43 versus 39). Secondly, a methoxy group at the meta-position of the second benzene rings was less favorable for maintaining long RTs (41 versus 40 and 42). A similar observation was made for the two compounds with 4-OCH<sub>3</sub> (44) and 4-OH (45) groups at the second benzene rings, of which RTs were decreased to  $24 \pm 4$  and  $19 \pm 3$  min, respectively, particularly when compared with 46 (63  $\pm$  4 min) and 43 (30  $\pm$  6 min). Collectively, we have successfully obtained compounds with divergent RTs (i.e. approx. 300-fold difference, between 37 and 38) next to varied affinity and association rates, which provides the possibility to investigate the influence of RTs on in vivo cardiotoxicity of K 11.1 blockers.

Table 4. Binding affinity and kinetic parameters of pyridinium analogues

Compd	A	R	n	K <sub>i</sub> (nM)	k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> )	k <sub>off</sub> (min <sup>-1</sup> )	K <sub>D</sub> (nM)	RT (min)
23	-CH <sub>2</sub> CH <sub>2</sub> -	-	5	36870±10861	(2.8±0.4)x10 <sup>-7</sup>	0.052±0.002	190546±27585	19±1
24	—CH <sub>2</sub> CH <sub>2</sub> —	$\mathrm{CH}_3$	3	12±1	0.020±0.005	0.23±0.04	12±0	4.3±0.8
25	—CH <sub>2</sub> CH <sub>2</sub> —	$\mathrm{CH}_{_3}$	4	0.28±0.01	0.25±0.08	$0.10\pm0.02$	0.44±0.05	10± 2
26	—CH <sub>2</sub> CH <sub>2</sub> —	$\mathrm{CH}_{_3}$	5	3.5±0.3	0.023±0.004	0.040±0.002	1.9±0.5	25±1
27	—CH <sub>2</sub> CH <sub>2</sub> —	$\mathrm{CH}_{_3}$	6	3.4±0.9	$0.018 \pm 0.003$	0.069±0.005	4.0±0.6	14±1
28	$-\!\!-\!$	$\mathrm{CH}_{_3}$	7	17±3	$0.0017 \pm 0.0002$	0.064±0.015	37±5	16±4
29	—CH <sub>2</sub> CH <sub>2</sub> —	$\mathrm{CH}_3$	9	108±10	$(1.8\pm0.2)$ x $10^{-4}$	0.028±0.001	153±10	36±2
30	_C≣C_	$\mathrm{CH}_3$	9	1179±297	(3.3±1.1)x10 <sup>-5</sup>	0.12±0.02	4258±1139	8.3±1.7

Compd	A	R	K <sub>i</sub> (nM)	k <sub>on</sub> (nM <sup>-1</sup> ·min <sup>-1</sup> )	k <sub>off</sub> (min <sup>-</sup>	K <sub>D</sub> (nM)	RT (min)
31	-CH <sub>2</sub> CH <sub>2</sub> -	-	3654±81	(1.6±0.3)x10 <sup>-5</sup>	0.11±0.01	6631±536	9.1±0.8
32	-CH <sub>2</sub> CH <sub>2</sub> -	$\mathrm{CH}_3$	6.0±0.5	0.022±0.003	0.13±0.01	6.1±1.2	7.7±0.7
33	_C≣C_	$\mathrm{CH}_3$	2460±262	(1.4±0.4)x10 <sup>-5</sup>	0.11±0.02	8877±1904	9.1±1.7

$$R_2$$
  $B$   $R_1$   $B$   $R_2$ 

Compd	R <sub>1</sub>	A	$\mathbf{R}_{2}$	В	K <sub>i</sub> (nM)	$k_{_{0n}}(nM^{\text{-}1}\!\cdot\!min^{\text{-}1})$	k <sub>off</sub> (min <sup>-1</sup> )	K <sub>D</sub> (nM)	RT (min)
34	-	_C≣C_	$C_6H_5$	$C_6H_4$	>100 µM	n.d.	n.d.	n.d.	n.d.
35	$\mathrm{CH}_{_3}$	—CH <sub>2</sub> CH <sub>2</sub> —	$C_6H_5$	$\mathrm{C_6H_4}$	63±9	0.0016±0.0003	0.098±0.021	60±4	10±2
36	$\mathrm{CH}_3$	—C≣C—	Н	$\mathrm{C_6H}_{10}$	1208±36	(3.3±1.2)×10 <sup>-4</sup>	0.30±0.05	1074±268	3.3±0.6
37	$\mathrm{CH}_3$	—C≣C—	Н	$\mathrm{C_6H_4}$	498±107	0.0086±0.0020	2.9±0.3	368±61	0.34±0.04
38	$\mathrm{CH}_3$	—C≣C—	$C_6H_5$	$\mathrm{C_6H_4}$	86±12	(7.5±1.0) ×10 <sup>-5</sup>	0.0095±0.0003	132±19	105±4
39	$\mathrm{CH}_3$	—C≣C—	2-CH <sub>3</sub> -C <sub>6</sub> H <sub>5</sub>	$\mathrm{C_6H_4}$	577±41	(3.5±0.3)×1°-4	0.071±0.004	203±19	14±1
40	$\mathrm{CH}_3$	—C≣C—	3-CH <sub>3</sub> -C <sub>6</sub> H <sub>5</sub>	$\mathrm{C_6H_4}$	122±15	(1.4±0.2)×10 <sup>-4</sup>	0.010±0.003	72±18	100±30
41	$\mathrm{CH}_3$	—C≣C—	3-OCH <sub>3</sub> -C <sub>6</sub> H <sub>5</sub>	$\mathrm{C_6H_4}$	176±10	(3.7±0.4)×10 <sup>-4</sup>	0.047±0.009	125±13	21± 4
42	$\mathrm{CH}_{_3}$	_C≣C_	3-Cl-C <sub>6</sub> H <sub>5</sub>	$\mathrm{C_6H_4}$	139±20	(8.3±0.9)×10 <sup>-5</sup>	0.011±0.002	135±14	91±17
43	$\mathrm{CH}_3$	—C≣C—	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>5</sub>	$\mathrm{C_6H_4}$	303±49	(8.6±1.0)×10 <sup>-5</sup>	0.033±0.006	440±78	30±6
44	$\mathrm{CH}_{_3}$	_C≣C_	4-OCH <sub>3</sub> -C <sub>6</sub> H <sub>5</sub>	$\mathrm{C_6H_4}$	247±44	(4.5±0.2)×10 <sup>-4</sup>	0.042±0.007	105±16	24± 4
45	$\mathrm{CH}_3$	—C≣C—	4-OH-C <sub>6</sub> H <sub>5</sub>	$\mathrm{C_6H_4}$	153±32	(5.1±0.7)×10 <sup>-4</sup>	0.053±0.008	103±3	19±3
46	CH <sub>3</sub>	_C≣C	4-Cl-C <sub>6</sub> H <sub>5</sub>	$C_6H_4$	71±9	(1.6±0.2)×10 <sup>-4</sup>	0.016±0.001	100±10	63±4

n.d.: not detectable.

## Kinetic map

Using the kinetic data in **Tables 1-4**, we plotted the dissociation rates (pk<sub>off</sub>, y axis) against association rates (pkge, x axis) of all our compounds together with three reference compounds (astemizole, dofetilide and ranolazine) into a kinetic map (Figure 4). The parallel diagonal dashed lines represent the kinetically derived  $K_D$  values. In this  $k_{on}$ - $k_{off}$ - $K_D$  map, compounds with the same affinity or association rates may have divergent dissociation rates or RTs, and vice versa. Therefore, risk assessment and possible mitigation of inhibitory efficacies of K 11.1 blockers should take association rates (k<sub>on</sub>) and RTs into account rather than affinity alone. In this context, all ligands were divided into four quadrants (I, II, III and IV) according to our hypothesis that high affinity, fast association rates and slow dissociation rates (long RTs) contribute together to serious K, 11.1-induced cardiotoxicity. The affinity and kinetic parameters of astemizole, dofetilide and ranolazine were derived from our previous investigation.<sup>2</sup> Astemizole was distributed into quadrant II, as it is a specific K<sub>2</sub>11.1 blocker and has been withdrawn from the market.<sup>26, 27</sup> On the other hand, ranolazine was assigned to quadrant **IV**, since it inhibits the K<sub>v</sub>11.1 channel but does not cause proarrhythmic events.<sup>28</sup> In addition, dofetilide is restrictedly used in the United States due to its low and finite TdP by blocking the  $K_v$ 11.1 channel, <sup>27, 47, 48</sup> and consequently, it was scattered into quadrant III. Obviously, compounds in quadrant II have high affinity (K<sub>p</sub> < 316 nM), fast association rates ( $k_{on} > 3.2 \times 10^{-4} \text{ nM}^{-1} \cdot \text{min}^{-1}$ ) and slow dissociation rates or long RTs (> 10 min) at the K\_11.1 channel, indicating that they may cause serious arrhythmias via blockade of the channel. On the other hand, quadrant IV comprises compounds with low affinity (K<sub>D</sub> > 316 nM), slow association rates  $(k_{on} \le 3.2 \times 10^{-4} \text{ nM}^{-1} \cdot \text{min}^{-1})$  and fast dissociation rates or short RTs ( $\le 10 \text{ min}$ ) at the channel, suggesting their possible cardiac safety in terms of K\_11.1 inhibition. Quadrants I and III encompass moderately active K\_11.1 blockers with either relatively low affinity, slow  $k_{_{on}}~(\le 3.2\times 10^{\text{-4}}~\text{nM}^{\text{-1}}\cdot\text{min}^{\text{-1}})$  and  $k_{_{off}}~(RT \ge 10~\text{min},$ I) or comparatively high affinity, fast  $k_{on}$  (> 3.2×10<sup>-4</sup> nM<sup>-1</sup>·min<sup>-1</sup>) and  $k_{off}$  (RT < 10 min, III). Most likely these two different classes of compounds have modest K<sub>2</sub>11.1-related cardiac side effects. Taken together, this arbitrary and provisional kinetic map that can be further extended with other suspected K<sub>v</sub>11.1 blockers sheds light on K<sub>2</sub>11.1-induced cardiotoxicity of drugs by bringing together their affinity as the traditional safety parameter with their kinetic parameters of association and dissociation rates (or RTs) at the channel. Therefore, a similar map would provide valuable information on how to accurately and comprehensively evaluate and further circumvent K<sub>v</sub>11.1 liability of drug candidates in the future.

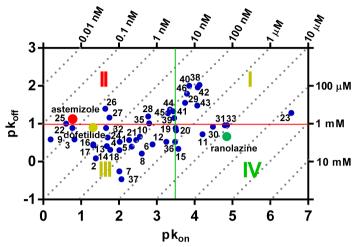


Figure 4. Kinetic map of all tested compounds in which the dissociation rate (pk<sub>off</sub> k<sub>off</sub>. min<sup>-1</sup>) is plotted on the y axis, whereas the association rate (pk<sub>on,</sub> k<sub>on:</sub> nM<sup>-1</sup>·min<sup>-1</sup>) is plotted on the x axis. Identical K<sub>D</sub> values can result from different combinations of k<sub>on</sub> and k<sub>off</sub> values  $(K_D = k_{off}/k_{on}, diagonal dashed lines)$ . Quadrants (I, II, III, IV) are arbitrarily defined by red horizontal and green vertical lines according to the varied  $k_{on}$ ,  $k_{off}$  and  $K_{D}$  values in combination with three reference compounds astemizole (red dot), dofetilide (yellow dot) and ranolazine (green dot). The intersection of red and green lines are arranged on the diagonal dashed line representing a K<sub>D</sub> value of 316 nM, while their intercepts with y and x axis are 1 (RT = 10 min) and 3.5 ( $k_{on} = 3.2 \times 10^{-4} \text{ nM}^{-1} \cdot \text{min}^{-1}$ ), respectively. Note that the kinetic parameters of compound 34 could not be determined due to its rather low K<sub>2</sub>11.1 affinity and thus is excluded in this map.

# Patch clamp study

A whole-cell voltage patch clamp assay was applied to determine the functional hERG current blockade by 21 and 38, two compounds with similar affinity but different RTs. As shown in Figure 5A, the IC<sub>50</sub> values for 21 and 38 from their concentration-effect curves in the patch clamp assay were  $182 \pm 6$  and  $107 \pm 6$ nM, respectively. However, 21 and 38 had almost identical affinity in the radioligand binding assay with  $K_i$  values of 97 ± 13 and 86 ± 12 nM, respectively. RTs of ligands have been found to be correlated to their efficacies in functional assays rather than affinity at the adenosine A<sub>2A</sub> receptor.<sup>49</sup> Thus, we surmised that the difference in IC<sub>50</sub> values obtained in functional patch clamp assays for these two compounds might be caused by their different RTs, with 38 (long RT of  $105 \pm 4$ min) displaying a higher potency in K<sub>v</sub>11.1 current inhibition than 21 (short RT

of  $3.7 \pm 0.8$  min). The time-dependent inhibition of the K 11.1 current for 21 and 38 was also assessed in the whole-cell patch clamp assay. As shown in Figure **5B**, compound **21** with a faster association rate  $(0.0035 \pm 0.0008 \text{ nM}^{-1} \cdot \text{min}^{-1})$  in the radioligand binding assay almost completely inhibited K\_11.1 currents within 10 min, whereas 15 min was required for **38** ( $k_{on} = (7.5 \pm 1.0) \times 10^{-5} \text{ nM}^{-1} \cdot \text{min}^{-1}$ ) to approach the equilibrated current inhibition. Therefore, association rates measured in the [3H]dofetilide competition association assay may be linked to the current inhibition rates of compounds in the functional patch clamp experiments. Although different wash-in times for K\_11.1 blockers have been observed on an automated patch clamp platform before, 23 this is the first time to interpolate the radioligand binding parameters, especially for the association and dissociation rates, into functional data derived from patch clamp assays.

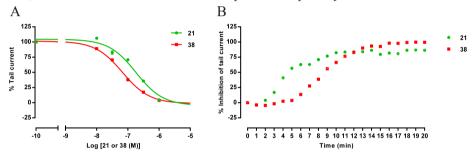


Figure 5. (A) Concentration-dependent inhibition of 21 and 38 for the K\_11.1 tail current in electrophysiological experiments. (B) Time-dependent inhibition of the K\_11.1 tail current by 21 and 38 (1 µM). Experiments were performed at room temperature using HEK293K\_11.1 cells.

# **Conclusion**

Although affinity and association rates of compounds at the K\_11.1 channel have been found to differ vastly and are correlated to each other, this study is the first to verify that K\_11.1 blockers can possess very different RTs, with up to a more than 300-fold difference (37 versus 38) at the channel. Furthermore, this comprehensive investigation on derivatives of dofetilide, E-4031, clofilium and pyridinium over a range of scaffolds is the first to illustrate the impact of several parameters outlined above on association rates and RTs of compounds at the K 11.1 channel in addition to their binding affinity. Notably, unlike the consistency of SARs for all compounds, SKRs are more chemical scaffold-dependent, in particular the structure-RT relationships. According to these findings and helped by the substantial differences in affinity, association and dissociation rate constants or RTs in the current sets of compounds, we delineated a k<sub>on</sub>-k<sub>off</sub>-K<sub>D</sub> kinetic

map, and divided all these compounds into four quadrants (toxic for quadrant II, safe for quadrant IV and suspicious for quadrant I and III in Figure 4), which might provide valuable information for pharmaceutical researchers to devise better strategies for assessing  $K_v11.1$  liability. Additionally, the association rates of two selected compounds were found to correspond well with their current block rates in the patch clamp studies, and their distinct RTs are a possible reason for their differences in functional potency when compared to their binding affinity. Although far from complete this analysis further indicates that kinetic parameters ( $k_{on}$  and RTs) should be taken into consideration as well as affinity ( $K_i$ ) during the investigation of  $K_v11.1$ -mediated adverse reactions. To conclude, the new finding of large variations in RTs promotes the extensive analysis of structure-RT relationships, and the combination of SARs and SKRs opens a new avenue for medicinal chemists to more efficiently design new chemical entities with low or negligible  $K_v11.1$ -induced cardiotoxicity.

# **Experimental section**

# Chemistry

All commercially available chemicals and solvents were used without purification, except for ethyl acetate, which was distilled before usage. Demineralized water will be referred to as H<sub>2</sub>O, unless otherwise stated. Reactions were monitored by thin-layer chromatography (TLC) using TLC silica gel 60 F254 aluminum sheets. TLC visualization was performed by a LAMAG UV-light at a wavelength of 254 or 366 nm. Grace Davison Davisil silica column material (LC60A 30-200 micron) was used for purification by column chromatography. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker AV400 or AV600 spectrometer. Deuterated solutions CDCl<sub>2</sub>, DMSO or MeOD were used for sample preparation. Chemical shifts (δ) are given in ppm and coupling constants (J) in Hertz. Trimethylsilane was used as internal standard for calibrating chemical shift for <sup>1</sup>H, <sup>13</sup>C NMR spectroscopy. HPLC-analysis was carried out on a Phenomenex Gemini reversed-phase C18 column (50 mm  $\times$  4.6 mm  $\times$  3  $\mu$ m) coupled to a UV-detector at 254 nm. Final compounds were dissolved in either a mixture of acetonitrile:tert-butanol:water (1:1:1) or in pure acetonitrile and were eluted from the column at a flow rate of 1.3 mL·min<sup>-1</sup> with 10/90 acetonitrile/water + 10% TFA, decreasing polarity of this mixture in time. LC-MS analysis was performed on an LCQ Advantage Max (Thermo Finnigan) ion-trap spectrometer (ESI<sup>+</sup>) coupled to a Surveyor HPLC system (Thermo Finnigan) equipped with a C18 column (Phenomenex Gemini, 50 mm x 4.6 mm, 5 μm). A linear gradient of 10/90 acetonitrile/water + a constant

10% of an 1% TFA in water solution was used as the mobile system. The purity of all final compounds was > 95%. High-resolution mass spectral analysis (HRMS) was performed by Leiden Institute of Chemistry by direct injection (2 µL of a 2 uM solution in water/MeCN; 50/50; v/v and 0.1% formicacid) on a LTO-Orbitrap FTMS operated in a positive ionization mode with an electrospray ionization (ESI) source (source voltage 3.5 kV, sheath gas flow 10%, capillary temperature 275 °C) with resolution R = 60000 at m/z 400 (mass range m/z = 150-2000) and calibrated for dioctylphthalate (m/z = 391.28428). All microwave reactions were performed in a Biotage initiator 2.5 in a sealed microwave vial.

**3-(P-tolyl)prop-2-vn-1-ol (S13).** To a solution of 4-iodotoluene (5.45 g, 25 mmol) in THF (33 mL) and Et<sub>2</sub>N (7.3 mL) were added PdCl<sub>2</sub>(PPh<sub>2</sub>)<sub>2</sub> (35 mg, 0.5 mmol) and CuI (327 mg, 1.7 mmol). The mixture was stirred for 30 minutes and a solution of propargyl alcohol (1.48 mL, 25 mmol) in THF (10 mL) was added dropwise. The mixture was stirred at r.t. for 12 hours and filtered on Celite. The solvents were evaporated and the crude residue was purified by flash chromatography on silica gel (33% EtOAc-Petroleum ether) to afford pure S13 (2.54 g. 70%) as an oil. Spectral data were in agreement with literature.<sup>50</sup>

3-(P-tolyl)prop-2-yn-1-yl 4-methylbenzenesulfonate (S14). To a stirred solution of S13 (1 g, 6.8 mmol) in diethylether (45 mL) at 0 °C was added tosyl chloride (1.6 g, 8.2 mmol) followed, portionwise, by freshly powdered KOH (3.8 g, 68 mmol). After the addition was complete, the reaction mixture was stirred at room temperature for 3.5 h and then poured into water (90 mL). The layers were separated and the aqueous phase was extracted with EtOAc (3 × 40 mL). The combined organic extract was dried, filtered, concentrated in vacuo and chromatographed (25% EtOAc - Petroleum ether) to obtain the title compound S14 (1.9 g, 91%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>)  $\delta$  7.87 (d, J = 8.2 Hz, 2H), 7.34 (d, J = 8.2 Hz, 2H), 7.17 (d, J = 8.1 Hz, 2H), 7.11 (d, J = 8.1 Hz, 2H), 4.97 (s, 2H), 2.42 (s, 3H), 2.36 (s, 3H) ppm.

#### General Sonogashira coupling procedure (S23a-e).

PdCl<sub>2</sub>(PPh<sub>2</sub>)<sub>2</sub> (148 mg, 0.21 mmol, 0.05 eq.) and CuI (121 mg, 0.63 mmol, 0.15 eq.) was added to a solution of 2,6-dibromopyridine (1000 mg, 4.22 mmol, 1.0 eq.) in Et,N (10 mL) and stirred for 30 min under a nitrogen atmosphere. The appropriate alkyne (10.1 mmol, 2.4 eq.) was added drop wise, and the mixture was stirred overnight at room temperature after which full conversion was shown by TLC. The reaction mixture was adsorbed on silica and purification by FCC vielded the desired compound.

2,6-di(hex-1-ynyl)pyridine (S23a). Prepared from hex-1-yne. FCC (petroleum ether: ethyl acetate 60:1 to 20:1) gave the desired product in a yield of (635

- mg) 63%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.52 (t, J = 8.0 Hz, 1H), 7.25 (d, J = 8.0 Hz, 2H), 2.42 (t, J = 7.0 Hz, 4H), 1.64-1.56 (m, 4H), 1.52-1.42 (m, 4H), 0.93 (t, J = 7.4 Hz, 6H).
- 2,6-Di(hept-1-vnvl)pyridine (S23b). Prepared from hept-1-vne. FCC (petroleum ether: dichloromethane 4:1 tot 2:1) gave the desired product in a yield of (241 mg) 21%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.53 (t, J = 7.6 Hz, 1H), 7.25 (d, J = 7.6 Hz, 2H), 2.41 (t, J = 6.8 Hz, 4H), 1.64-1.60 (m, 4H), 1.44-1.31 (m, 8H), 0.91 (t, J = 7.2 Hz, 6H).
- **2,6-Di(non-1-ynyl)pyridine (S23c).** Prepared from non-1-yne. FCC (petroleum ether:dichloromethane 30:1 tot 4:1) gave the desired product in a yield of (247 mg) 18%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.52 (t, J = 7.6 Hz, 1H), 7.25 (d, J = 8.0 Hz, 2H), 2.41 (t, J = 7.2 Hz, 4H), 1.65-1.57 (m, 4H), 1.45-1.39 (m, 4H), 1.35-1.26 (m, 12H), 0.89 (t, J = 6.4 Hz, 6H).
- 2,6-Di(dec-1-ynyl)pyridine (S23d). Prepared from dec-1-yne. FCC (petroleum ether: dichloromethane 5:1 to 3:1) gave the desired product in a yield of (210 mg) 14%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.53 (t, J = 8.0 Hz, 1H), 7.25 (d, J = 7.6 Hz, 2H), 2.41 (t, J = 7.2, 4H), 1.65-1.57 (m, 4H), 1.45-1.39 (m, 4H), 1.30-1.24 (m, 16H), 0.88 (t, J = 6.8 Hz, 6H).
- 2,6-Di(dodec-1-ynyl)pyridine (S23e). Prepared from dodec-1-yne. FCC (petroleum ether: dichloromethane 5:1 to 2:1) gave the desired product in a yield of 79% (1131 mg). <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.53 (t, J = 7.6 Hz, 1H), 7.25  $(d, J = 8.0 \text{ Hz}, 2H), 2.41 \text{ (t, } J = 7.2 \text{ Hz}, 4H), 1.64-1.57 \text{ (m, 4H)}, 1.44-1.39 \text{ (m, } J = 8.0 \text{ Hz}, 2H), 1.44-1.39 \text{ (m, } J = 8.0 \text{ ($ 4H), 1.32-1.26 (m, 24H), 0.88 (t, J = 6.4 Hz, 6H).

## General procedure for reduction of pyridine alkynes (\$24a-e).

The appropriate alkyne (S23a) (600 mg, 2.52 mmol) was dissolved in THF (3 mL) and MeOH (3 mL) under a N<sub>2</sub> atmosphere. 10% Pd/C (120 mg, 1.14 mmol) was added and H<sub>2</sub> was introduced via a balloon. The mixture was stirred overnight at r.t. after which the reaction mixture was filtered over Celite. The filtrate were concentrated under reduced pressure and FCC (petroleum ether:ethyl acetate 20:1) gave the desired compound.

- 2,6-Dihexylpyridine (S24a). Prepared from S23a and purified by FCC (petroleum ether:ethyl acetate 20:1) which gave the desired product in a yield of (469 mg) 75%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.48 (t, J = 7.6 Hz, 1H), 6.94 (d, J= 7.6 Hz, 2H, 2.75 (t, J = 7.6 Hz, 4H), 1.73-1.65 (m, 4H), 1.39-1.26 (m, 12H),0.88 (t, J = 6.8 Hz, 6H).
- 2,6-Diheptylpyridine (S24b). Prepared from S23b and purified by FCC (petroleum ether:dichloromethane 4:1 to 1:2) which gave the desired product in a yield of (106 mg) 43%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.47 (t, J = 7.6 Hz, 1H),

- 6.93 (d, J = 7.6 Hz, 2H), 2.75 (t, J = 7.6 Hz, 4H), 1.71-1.66 (m, 4H), 1.34-1.23 (m, 16H), 0.87 (t, J = 6.8 Hz, 6H).
- 2,6-Dinonylpyridine (S24c). Prepared from S23c and purified by FCC (petroleum ether:dichloromethane 4:1 to 1:2) which gave the desired product in a yield of (91 mg) 78%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.47 (t, J = 7.6 Hz, 1H), 6.93 (d, J = 7.6 Hz, 2H), 2.75 (t, J = 8.0 Hz, 4H), 1.73-1.65 (m, 4H), 1.36-1.16 (m, 24H), 0.86 (t, J = 4.4 Hz, 6H).
- 2,6-Didecylpyridine (S24d). Prepared from S23d and purified by FCC (petroleum ether:dichloromethane 3:1 to 1:1) which gave the desired product in a yield of (153 mg) 71%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  7.48 (t, J = 7.6 Hz, 1H), 6.94 (d, J = 7.6 Hz, 2H), 2.74 (t, J = 8.0 Hz, 4H), 1.73-1.65 (m, 4H), 1.32-1.25(m, 28H), 0.88 (t, J = 6.8 Hz, 6H).
- 2,6-Didodecylpyridine (S24e). Prepared from S23e and purified by FCC (petroleum ether:dichloromethane 4:1 to 1:1) which gave the desired product in a yield of (440 mg) 86%. <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>):  $\delta$  7.48 (t, J = 7.6 Hz, 1H), 6.94 (d, J = 7.6 Hz, 2H), 2.75 (t, J = 7.6 Hz, 4H), 1.73-1.65 (m, 4H), 1.35-1.25(m, 36H), 0.88 (t, J = 6.8 Hz, 6H).
- 2,6-Bis((trimethylsilyl)ethynyl)pyridine (S32). 2,6-dibromopyridine (9.00 g, 38.0 mmol), Pd(PPh<sub>2</sub>)<sub>4</sub> (2.20 g, 1.90 mmol), CuI (361 mg, 1.90 mmol), diisopropylamine (9 mL) and toluene (100 mL) were stirred at r.t. and trimethylsilylacetylene (13.0 mL, 91.2 mmol) was added. The mixture was stirred overnight followed by filtration over Celite using CH<sub>2</sub>Cl<sub>2</sub>. The filtrate was concentrated under reduced pressure, dissolved in CH2Cl2, and washed with an 1M aqueous NH4Cl solution. The organic layer was dried over MgSO<sub>4</sub>, filtered and evaporated to dryness. Yield quantitative and used in the next step without further purification. <sup>1</sup>H NMR (CDCl<sub>2</sub>):  $\delta$  7.61 (t, J = 8.0 Hz, 1H), 7.38 (d, J = 7.6 Hz, 2H), 0.25 (s, 18H).
- 2,6-Bis((4-bromophenyl)ethynyl)pyridine (S33). To a solution of CuCl (346 mg, 3.50 mmol), PPh, (918 mg, 3.50 mmol) and potassium benzoate (11.3 g, 70.7 mmol) in 38 mL DMI (1,3-dimethyl-2-imidazolidinone) were added compound S32 (9.60 g, 35.4 mmol) and 1-bromo-4-iodobenzene (20.0 g, 70.7 mmol) at r.t. The mixture was stirred at 120 °C for 4 h after which full consumption of S32 was seen. The mixture was cooled down, quenched with a 3M HCl (aq.) solution. The organics were extracted with CH<sub>2</sub>Cl<sub>2</sub>, and washed with subsequently NaHCO<sub>3</sub> and brine. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvents were evaporated under vacuum. FCC (petroleum ether:dichloromethane 10:1 to 2:1) gave the desired product. Yield (4.44 g) 29%; <sup>1</sup>H NMR (CDCl<sub>2</sub>): δ 7.70 (t, J = 8.0 Hz, 1H), 7.52-7.45 (m, 10H).

#### General Suzuki reaction procedure yielding compounds S34a-e.

Compound S33 (0.2 mmol, 1.0 eq.), the respective substituted-phenylboronic acid (0.4 mmol, 2.0 eq.), 2M K<sub>2</sub>CO<sub>2</sub> (10 eq.) (aq.) were dissolved in a mixture of 3 mL toluene and 0.4 mL of ethanol. Pd(PPh<sub>2</sub>)<sub>4</sub> (0.02 mmol, 0.1 eq.) were added and the mixture was heated in the microwave at 100 °C under a nitrogen atmosphere for 24 h. After cooling down to room temperature, EtOAc was added and a precipitate was formed (S34b-e). The precipitate was collected by filtration, resulting in the desired products.

- 2,6-Bis((2'-methylbiphenyl-4-yl)ethynyl)pyridine (S34a). Prepared from 2-methylphenyl boronic acid according to the general procedure, but a precipitate was not formed after addition of EtOAc. The organic layer was washed with a 2 × 0.5 M NaOH solution, dried over MgSO<sub>4</sub>, filtered and the solvents were evaporated in vacuo. Used this as a crude mixture in the next reaction. Yield (37 mg) 50%. ESI-MS: 460.2 [M+H]+.
- 2,6-Bis((3'-methylbiphenyl-4-yl)ethynyl)pyridine (S34b). Prepared from 3-methylphenyl boronic acid according to the general procedure. Yield (17 mg) 16%; <sup>1</sup>H NMR (CDCl<sub>2</sub>):  $\delta$  7.72-7.67 (m, 5H), 7.60 (d, J = 8.4 Hz, 4H), 7.50 (d, J= 8.0 Hz, 2H, 7.42 (d, J = 8.8 Hz, 4H), 7.35 (t, J = 7.2 Hz, 2H), 7.19 (d, J = 7.2 Hz, 2Hz)Hz, 2H), 2.43 (s, 6H).
- 2,6-Bis((4'-methylbiphenyl-4-yl)ethynyl)pyridine (S34c) was obtained as a white solid. Yield (13 mg) 18%; <sup>1</sup>H NMR (CDCl<sub>2</sub>): δ 7.72-7.70 (m, 5H), 7.59 (d, J = 8.0 Hz, 4H), 7.52-7.49 (m, 6H), 7.30-7.24 (m, H), 2.41 (s, 6H).
- 2,6-Bis((4'-methoxybiphenyl-4-yl)ethynyl)pyridine (S34d). Prepared from 4-methoxyphenyl-boronic acid according to the general procedure. Yield (55 mg) 49%; <sup>1</sup>H NMR (CDCl<sub>1</sub>):  $\delta$  7.72-7.64 (m, 5H), 7.57-7.45 (m, 9H), 7.00 (d, J = 8.0Hz, 4H), 7.30-7.24 (m, H), 2.41 (s, 6H).
- 2,6-Bis((4'-chlorobiphenyl-4-yl)ethynyl)pyridine (S34e). Prepared from 4-chlorophenyl boronic acid according to the general procedure. Yield (56 mg) 49%; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.74-7.65 (m, 5H), 7.59-7.47 (m, 10H), 7.43 (d, J = 8.4 Hz, 4H).
- 4'-Bromo-3-methoxybiphenyl (S43a). Palladium acetate (40.0 mg, 0.18 mmol) was added to mixture of 1-bromo-4-iodobenze (500 mg, 1,77 mmol), 3-methoxy-phenylboronic acid (269 mg, 1.77 mmol), triphenylphosphine (139 mg, 0.53 mmol), a 2M K<sub>2</sub>CO<sub>2</sub> aqueous solution (5 mL) in toluene (15 mL) and EtOH (2 mL). This mixture was heated at 50 °C for 24 h., after which it was cooled to room temperature and partitioned between ethyl acetate and water. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, filtered and evaporated. FCC (petroleum ether:ethyl acetate 1:0 to 4:1) gave the desired product. Yield

(126 mg) 27%; <sup>1</sup>H NMR (CDCl<sub>2</sub>):  $\delta$  7.55 (d, J = 8.4 Hz, 2H), 7.45 (d, J = 8.8 Hz, 2H), 7.36 (t, J = 8.0 Hz, 1H), 7.15-7.13 (m, 1H), 7.08 (t, J = 2.4 Hz, 1H), 6.93-6.90 (m, 1H), 3.86 (s, 3H).

4'-Bromo-3-chlorobiphenyl (S43b). Prepared from 1-bromo-4-iodobenzene and 3-chloro-phenylboronic following the procedure of 4-bromo-3-methoxybiphenyl (S43a). The reaction was performed at 80° C for 20 h. FCC (petroleum ether) gave the compound 190 mg, 40%. <sup>1</sup>H NMR (CDCl<sub>2</sub>):  $\delta$  7.56 (d, J = 8.4 Hz, 2H), 7.53 (t, J = 1.6 Hz, 1H), 7.47-7.43 (m, 3H), 7.36 (t, J = 8.0 Hz, 1H), 7.34-7.31 (m, 1H).

General procedure for Sonogashira coupling between 4'-bromo-3-substituted-biphenyl and ethynyltrimethylsilane. Compound S43a (126 mg, 0.48 mmol), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (7.0 mg, 0.01 mmol), CuI (4.0 mg, 0.02 mmol), PPh<sub>3</sub> (5.0 mg, 0.02 mmol) and trimethylsilylacetylene (123 µL, 0.86 mmol) were dissolved in THF (1 mL) and piperidine (200 µl) was added. This was heated in the microwave at 120 °C for 12 min. The mixture was extracted with diethyl ether and washed with aq. H<sub>2</sub>SO<sub>4</sub> (10%), water and brine. The organic layer was dried over MgSO<sub>4</sub>, filtered and evaporated. FCC (petroleum ether:ethyl acetate 1:0 to 10:1) gave the desired ((3'-Methoxybiphenyl-4-yl)ethynyl)trimethylsilane (S44a). Yield (104 mg) 78%. <sup>1</sup>H NMR (CDCl<sub>2</sub>):  $\delta$  7.53 (s, 4H), 7.35 (t, J = 8.0 Hz, 1H), 7.18-7.15 (m, 1H), 7.12-7.00 (m, 1H), 6.90 (dd,  $J^1 = 8.0$  Hz,  $J^2 = 2.8$  Hz, 1H), 3.86 (s, 3H), 0.27 (s, 9H).

- ((3'-Chlorobiphenyl-4-yl)ethynyl)trimethylsilane (S44b). Following the procedure of **S44a** starting from **S43b**. Yield (49 mg) 64%. <sup>1</sup>H NMR (CDCl<sub>2</sub>): δ 7.56 (t, J = 2.0 Hz, 1H), 7.53 (d, J = 8.4 Hz, 2H), 7.50 (d, J = 8.8 Hz, 2H), 7.45  $(dt, J^1 = 7.6 \text{ Hz}, J^2 = 1.6 \text{ Hz}, 1\text{H}), 7.36 (t, J = 7.6 \text{ Hz}, 1\text{H}), 7.32 (dt, J^1 = 8.0 \text{ Hz}, 1\text{H})$  $J^2 = 2.0 \text{ Hz}, 1\text{H}, 0.27 \text{ (s, 9H)}.$
- 4'-Ethynyl-3-methoxybiphenyl (S45). To a solution of S44a (104 mg, 0.37 mmol) in diethyl ether (3 mL) was added methanol (3 mL) and 2M NaOH (1 mL). After 10 min of stirring at room temperature the mixture was neutralized with 2M HCl (aq.). The layers were separated and the organic layer was washed with brine and water, dried, filtered over MgSO<sub>4</sub> and concentrated. The product was used without further purification.
- 2,6-Bis((2'-methoxybiphenyl-4-yl)ethynyl)pyridine (S46). Reaction conditions according to the general Sonogashira coupling procedure (S23a-e), starting from \$45 and \$21. Purified by FCC (petroleum ether:ethyl acetate 4:1 to 1:1). Yield (23 mg) 25%. <sup>1</sup>H NMR (CDCl<sub>2</sub>):  $\delta$  7.72-7.65 (m, 5H), 7.60 (d, J = 8.4 Hz, 4H), 7.51 (d, J = 7.6 Hz, 2H), 7.38 (t, J = 8.4 Hz, 2H), 7.20 (d, J = 8.0 Hz, 2H), 7.14 (s, 2H), 6.93 (dd,  $J^1 = 8.2$  Hz,  $J^2 = 2.0$  Hz, 2H), 3.88 (s, 6H).
  - 2,6-Bis((2'-chlorobiphenyl-4-yl)ethynyl)pyridine (S47). Starting from

compounds **S21** and **S44b**. FCC (petroleum ether:dichloromethane 20:1 to 1:2). Yield (16 mg) 19%. <sup>1</sup>H NMR (CDCl<sub>2</sub>): δ 7.70-7.66 (m, 4H), 7.59-7.55 (m, 6H), 7.53 (d, J = 5.2 Hz, 1H), 7.50-7.44 (m, 4H), 7.39 (t, J = 8.0 Hz, 2H), 7.34 (d, J =8.0 Hz, 2H).

N-Ethyl-N-(3-(p-tolyl)prop-2-vn-1-vl)heptan-1-amine (19). To a solution of **S14** (300 mg, 1 mmol) in DMF (5 mL) was added K<sub>2</sub>CO<sub>2</sub> (415 mg, 3 mmol) followed by N-ethylheptylamine (280 µL, 1.5 mmol). The mixture was stirred at r.t. for 1 hour and diluted with water (5 mL), extracted with EtOAc (3  $\times$  10 mL) and the organic layers were dried on MgSO<sub>4</sub>, filtered and the solvent was removed in vacuo. The crude residue was purified by flash chromatography on silica gel (20% EtOAc-Petroleum ether) to afford pure 19 as an oil. Yield (228 mg) 84%; <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>)  $\delta$  7.32 (d, J = 8.0 Hz, 2H), 7.10 (d, J =8.0 Hz, 2H), 3.62 (s, 2H), 2.61 (q, J = 7.2 Hz, 2H), 2.53 (t, J = 7.6 Hz, 2H), 2.34 (s, 3H), 1.52-1.49 (m, 2H), 1.31-1.26 (m, 8H), 1.11 (t, J = 7.2 Hz, 3H), 0.88 (t, J = 7.2 Hz, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>)  $\delta$  137.9, 131.6, 129.0, 120.3, 85.0, 83.8, 53.6, 47.7, 42.1, 31.9, 29.3, 27.6, 27.5, 22.7, 21.4, 14.1, 12.7 ppm. HPLC:  $t_R = 7.39$  min; HRMS (ESI):  $m/z [M + H]^+$  calcd for  $C_{19}H_{29}N$ : 272.2373, found: 272.2370.

N-Ethyl-N-propyl-3-(p-tolyl)prop-2-yn-1-amine (20). To a solution of S14 (300 mg, 1 mmol) in DMF (5 mL) was added K<sub>2</sub>CO<sub>3</sub> (415 mg, 3 mmol) followed by N-ethylpropylamine (180 μL, 1.5 mmol). The mixture was stirred at r.t. for 1 hour and diluted with water (5 mL), extracted with EtOAc (3  $\times$  10 mL) and the organic layers were dried on MgSO<sub>4</sub>, filtered and the solvent was removed in vacuo. The crude residue was purified by flash chromatography on silica gel (20% EtOAc- Petroleum ether) to afford pure 20 as oil. Yield (170 mg) 79%; <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{CDCl}_2) \delta 7.32 \text{ (d, } J = 8.0 \text{ Hz}, \text{ 2H)}, 7.10 \text{ (d, } J = 8.0 \text{ Hz}, \text{ 2H)}, 3.62 \text{ (s, } J = 8.0 \text{ Hz}, \text{ 2H)}$ 2H), 2.61 (q, J = 7.2 Hz, 2H), 2.50 (t, J = 7.6 Hz, 2H), 2.34 (s, 3H), 1.53 (sex, J= 7.6 Hz, 2H), 1.11 (t, J = 7.2 Hz, 3H), 0.93 (t, J = 7.6 Hz, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>) δ 137.9, 131.6, 129.0, 120.3, 85.0, 83.8, 55.6, 47.6, 42.1, 21.4, 20.8, 12.7, 12.0 ppm. HPLC:  $t_R = 5.70$  min; HRMS (ESI):  $m/z [M + H]^+$  calcd for  $C_{15}H_{21}$ : 216.1747, found: 216.1745.

#### General methylation procedure to yield pyridines 24, 25, 27-30, 39-44 and **46.**

The respective compound S23e, S24a-e, S34a-e, S46 and S47 (1.0 eq.) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (40 mL per mmol) and stirred at 0 °C under a N<sub>2</sub> atmosphere. Methyl trifluoromethanesulfonate (2.8 eq.) was added and the mixture was stirred overnight. After adsorption of the crude on silica, FCC gave the desired compound eluting with a mixture of dichloromethane:methanol.

- 2,6-Dihexyl-1-methylpyridinium trifluoromethanesulfonate (24). Prepared from S24a and purified by FCC (dichloromethane:methanol 60:1 to 20:1) to give the desired product. Yield (660 mg) 85%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>): δ 8.20 (t, J = 8.0 Hz, 1H), 7.64 (d, J = 8.0 Hz, 2H), 4.21 (s, 3H), 3.12 (t, J = 7.6 Hz, 2Hz)4H), 1.81-1.74 (m, 4H), 1.52-1.45 (m, 4H), 1.39-1.29 (m, 8H), 0.91 (t, J = 6.8Hz, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>): δ 159.7, 144.2, 126.0, 39.7, 34.5, 31.4, 28.9, 27.6, 22.5, 14.1 ppm. HPLC purity: 99.3% (t<sub>b</sub> 8.24 min); ESI-MS: 262.27 [M].
- 2,6-Diheptyl-1-methylpyridinium trifluoromethanesulfonate (25). Prepared from **S24b** and purified by FCC (dichloromethane:methanol 30:1 to 20:1) to give the desired product. Yield (143 mg) 86%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>): δ 8.18 (t, J = 7.6 Hz, 1H), 7.63 (d, J = 8.0 Hz, d), 4.22 (s, 3H), 3.13 (t, J = 8.0 Hz, d)4H), 1.82-1.74 (m, 4H), 1.52-1.44 (m, 4H), 1.41-1.36 (m, 12H), 0.90 (t, J = 6.4Hz, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>): δ 159.8, 144.1, 126.0, 122.4, 119.2, 39.7, 34.6, 31.7, 29.3, 29.0, 27.7, 22.7, 14.2 ppm. HPLC purity: 97% (t<sub>p</sub> 8.96 min); ESI-MS: 290.33 [M].
- 2,6-Dinonvl-1-methylpyridinium trifluoromethanesulfonate (27). Prepared from S24c and purified by FCC (dichloromethane:methanol 30:1 to 15:1) to give the desired product. Yield (87 mg) 65%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>): δ 8.18 (t, J = 8.0 Hz, 1H), 7.63 (d, J = 8.0 Hz, 2H), 4.22 (s, 3H), 3.12 (t, J = 7.6 Hz, 2H)4H), 1.82-1.74 (m, 4H), 1.50-1.46 (m, 4H), 1.44-1.28 (m, 20H), 0.89 (t, J = 6.4Hz, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>): δ 159.9, 144.0, 126.0, 39.8, 34.6, 32.0, 29.5, 29.4, 27.7, 22.8, 14.2 ppm. HPLC purity: 100% (t<sub>R</sub> 10.21 min); ESI-MS: 346.40 [M].
- 2,6-Didecyl-1-methylpyridinium trifluoromethanesulfonate (28). Prepared from S24d and purified by FCC (dichloromethane:methanol 60:1 to 10:1) to give the desired product. Yield (166 mg) 74%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>): δ 8.18 (t, J = 8.0 Hz, 1H), 7.63 (d, J = 8.0 Hz, 2H), 4.22 (s, 3H), 3.13 (t, J = 8.0 Hz, 2H)4H), 1.82-1.74 (m, 4H), 1.50-1.44 (m, 4H), 1.39-1.28 (m, 24H), 0.89 (t, J = 6.8Hz, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>): δ 159.7, 144.0, 125.9, 39.6, 34.4, 31.9, 29.5, 29.4, 29.3, 29.2, 27.6, 22.7, 14.1 ppm. HPLC purity: 100% (t<sub>R</sub> 10.77 min); ESI-MS: 374.40 [M].
- 2,6-Didodecyl-1-methylpyridinium trifluoromethanesulfonate (29). Prepared from **S24e** and purified by FCC (dichloromethane:methanol 30:1 to 20:1) to give the desired product. Yield (536 mg) 87%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>): δ 8.19 (t, J = 8.0 Hz, 1H), 7.63 (d, J = 7.6 Hz, 2H), 4.21 (s, 3H), 3.12 (t, J = 8.0 Hz, 4H), 1.81-1.73 (m, 4H), 1.51-1.44 (m, 4H), 1.38-1.27 (m, 32H), 0.88 (t, J = 6.8Hz, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>): δ 159.8, 144.1, 126.0, 39.7, 34.6, 32.0, 29.8, 29.7, 29.7, 29.6, 29.5, 29.4, 29.3, 27.7, 22.8, 14.2 ppm. HPLC purity: 100%

(t<sub>p</sub> 11.84 min); ESI-MS: 430.53 [M].

- 2,6-Di(dodec-1-ynyl)-1-methylpyridinium trifluoromethanesulfonate (30). Prepared from S23e and purified by FCC (dichloromethane:methanol 40:1 to 20:1) to give the desired product. Yield (675 mg) 94%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  8.38 (t, J = 8.0 Hz, 1H), 7.90 (d, J = 8.0 Hz, 2H), 4.49 (s, 3H), 2.65 (t, J= 7.2 Hz, 4H), 1.74-1.65 (m, 4H), 1.48-1.41 (m, 4H), 1.30-1.27 (m, 24H), 0.88 (t, J = 6.8 Hz, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>):  $\delta$  144.3, 139.5, 130.7, 112.8, 72.9, 45.1, 32.0, 29.7, 29.5, 29.4, 29.2, 29.1, 27.6, 22.9, 20.3, 14.2 ppm. HPLC purity: 100% (t<sub>p.</sub> 11.57 min); ESI-MS: 422.40 [M].
- 2,6-Di([4-(2-methylphenyl)phenyl]ethynyl)-1-methylpyridinium trifluoromethanesulfonate (39). Prepared from S34a and purified by FCC (dichloromethane:methanol 30:1 to 20:1) and preparative HPLC to give the desired product. Yield (1 mg) 96%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  8.47 (t, J = 8.0 Hz, 1H), 8.10 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 8.0 Hz, 4H), 7.46 (d, J = 8.0 Hz, 4H),7.31-7.09 (m, 8H), 4.74 (s, 3H), 2.30 (s, 6H). HPLC purity: 96% (t<sub>p</sub> 9.97 min); ESI-MS: 474.33 [M].
- 2,6-Di([4-(3-methylphenyl)phenyl]ethynyl)-1-methylpyridinium trifluoromethanesulfonate (40). Prepared from S34b and purified by FCC (dichloromethane:methanol 30:1 to 20:1) to give the desired product. Yield (18 mg) 78%. <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  8.40 (t, J = 8.0 Hz, 1H), 8.03 (d, J = 8.0 Hz, 2H), 7.73 (d, J = 8.0 Hz, 4H), 7.65 (d, J = 8.0 Hz, 4H), 7.41-7.37 (m, 4H), 7.35 (t, J =7.6 Hz, 2H), 7.22 (d, J = 7.2 Hz, 2H), 4.65 (s, 3H), 2.41 (s, 6H);  ${}^{13}$ C NMR (101 MHz, CDCl<sub>2</sub>): δ 145.0, 143.9, 139.5, 138.9, 133.5, 129.4, 129.1, 128.1, 127.7, 124.4, 117.5, 109.0, 81.2, 45.5, 21.7 ppm. HPLC purity: 98.65% (t<sub>R</sub> 10.24 min); ESI-MS: 474.33 [M].
- 2,6-Di([4-(3-methoxyphenyl)phenyl]ethynyl)-1-methylpyridinium fluoromethanesulfonate (41). Prepared from S46 and purified by FCC (dichloromethane: methanol 20:1 to 10:1) to give the desired product. Yield (12 mg) 36%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  8.39 (t, J = 8.0 Hz, 1H), 8.02 (d, J = 8.0 Hz, 2H), 7.73 (d, J = 8.0 Hz, 4H), 7.65 (d, J = 8.4 Hz, 4H), 7.37 (t, J = 7.6 Hz, 2H), 7.18 (d, J = 8.0 Hz, 2H), 7.11 (s, 2H), 6.94 (dd, J' = 8.2 Hz,  $J^2 = 2.0$  Hz, 2H), 4.63 (s, 3H), 3.86 (s, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>): δ 160.3, 144.6, 143.9, 140.9, 139.3, 133.5, 130.8, 130.3, 127.7, 119.7, 117.7, 114.0, 113.0, 108.8, 81.3, 77.4, 55.5, 45.5, 29.9 ppm. HPLC purity: 99% (t<sub>R</sub> 9.56 min); ESI-MS: 506.33 [M].
- 2,6-Di([4-(3-chlorophenyl)phenyl]ethynyl)-1-methylpyridinium trifluoromethanesulfonate (42). Prepared from S47 and purified by FCC (dichloromethane:methanol 20:1 to 10:1) to give the desired product. Yield (9 mg) 45%; <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>):  $\delta$  <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.43 (t, J = 8.0 Hz, 1H), 8.08 (d, J = 8.0 Hz, 2H), 7.78 (d, J = 8.0 Hz, 4H), 7.65 (d, J = 8.0 Hz, 4H),

7.57 (s, 2H), 7.48 (d, J = 7.2 Hz, 2H), 7.42-7.37 (m, 4H), 4.70 (s, 3H);  $^{13}$ C NMR (101 MHz, CDCl<sub>2</sub>/MeOD): δ 143.8, 143.4, 141.0, 139.3, 134.9, 133.3, 130.7, 130.3, 128.5, 127.6, 127.1, 125.3, 117.9, 108.5, 80.8, 45.0, 29.5 ppm. HPLC purity: 99% (t<sub>p</sub> 10.20 min); ESI-MS: 514.33 [M].

2,6-Di(4-(4-methylphenyl)phenylethynyl)-1-methylpyridinium trifluoromethanesulfonate (43). Prepared from S34c and purified by FCC (dichloromethane:methanol 30:1 to 5:1) to give the desired product. Yield (27 mg) 48%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  8.43 (t, J = 8.0 Hz, 1H), 8.06 (d, J = 8.0 Hz, 2H), 7.75 (d, J = 8.4 Hz, 4H), 7.67 (d, J = 8.4 Hz, 4H), 7.52 (d, J = 8.0 Hz, 4H), 7.29-7.26(m, 4H), 4.68 (s, 3H), 2.41 (s, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>/MeOD): δ 144.9, 143.9, 139.4, 138.8, 136.3, 133.4, 133.3, 130.6, 129.8, 127.4, 127.0, 117.0, 109.1, 80.9, 77.3, 21.1 ppm. HPLC purity: 99% (t<sub>p</sub> 10.28 min); ESI-MS: 474.40 [M].

2,6-Di([4-(4-methoxyphenyl)phenyl]ethynyl)-1-methylpyridinium fluoromethanesulfonate (44). Prepared from S34d and purified by FCC (dichloromethane:methanol 30:1 to 10:1) to give the desired product. Yield (42 mg) 58%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  8.42 (t, J = 8.0 Hz, 1H), 8.04 (d, J = 8.0 Hz, 2H), 7.73 (d, J = 8.4 Hz, 4H), 7.65 (d, J = 8.4 Hz, 4H), 7.57 (d, J = 8.8 Hz, 4H), 7.00 (d, J = 8.8 Hz, 4H), 4.67 (s, 3H), 3.86 (s, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>/ MeOD): δ 160.1, 144.5, 143.7, 139.3, 133.3, 131.6, 130.4, 128.3, 127.0, 116.5, 114.5, 109.3, 80.7, 55.4, 44.9 ppm. HPLC purity: 100% (t<sub>p</sub> 9.78 min); ESI-MS: 506.33 [M].

2,6-Di([4-(4-hydroxyphenyl)phenyl]ethynyl)-1-methylpyridinium trifluoromethanesulfonate (45). 44 (27 mg, 0.04 mmol) was dissolved in 2 mL CH<sub>2</sub>Cl<sub>2</sub> and cooled to -78 °C under a nitrogen atmosphere. An 1M solution of BBr, in CH2Cl2 (0.4 mL) was added and the reaction mixture was allowed to warm up to room temperature and stirring was continued for 2 h at room temperature, after which as shown by TLC (dichloromethane:methanol 30:1) full conversion was reached. The reaction mixture was re cooled to -78 °C and quenched with H<sub>2</sub>O. After an hour the precipitate was filtered off and washed with H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> respectively and dried in vacuo. Yield (12 mg) 48%. <sup>1</sup>H NMR (400MHz, DMSO):  $\delta$  9.81 (s, 2H), 8.56 (t, J = 8.4 Hz, 1H), 8.36 (d, J = 7.6 Hz, 2H), 7.89 (d, J = 8.0 Hz, 4H), 7.83 (d, J = 8.4 Hz, 4H), 7.64 (d, J = 8.4 Hz, 4H), 6.90 (d, J = 8.4 Hz, 4Hz), 6.90 (d, J = 8.4 Hz), 6.90 (d, J = 8.4 Hz, 4Hz), 6.90 (d, J = 8.4 Hz), 6.90 (d,= 8.8 Hz, 4H), 4.58 (s, 3H); <sup>13</sup>C NMR (151 MHz, DMSO): δ 158.2, 143.6, 143.2, 133.3, 130.6, 129.1, 128.2, 126.3, 116.4, 115.9, 106.0, 81.7, 45.2 ppm. HPLC purity 99% (t<sub>p.</sub> 8.38 min); ESI-MS: 478.33 [M].

2,6-Di([4-(4-chlorophenyl)phenyl]ethynyl)-1-methylpyridinium trifluoromethanesulfonate (46). Prepared from S34e and purified by FCC (dichloromethane:methanol 30:1 to 10:1) to give the desired product. Yield (42 mg) 58%; <sup>1</sup>H NMR (400MHz, CDCl<sub>2</sub>):  $\delta$  8.43 (t, J = 8.0 Hz, 1H), 8.08 (d, J = 8.4 Hz, 2H),

7.79 (d, J = 8.4 Hz, 4H), 7.67 (d, J = 8.4 Hz, 4H), 7.56 (d, J = 8.4 Hz, 4H), 7.46(d, J = 8.4 Hz, 4H), 4.74 (s, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>2</sub>/MeOD):  $\delta$  143.8, 143.6, 139.3, 137.7, 134.8, 133.4, 130.6, 129.2, 128.4, 127.4, 117.6, 108.6, 80.9, 45.0 ppm. HPLC purity: 99% (t<sub>R</sub> 10.39 min); ESI-MS: 514.27 [M].

#### **Biology**

#### Chemical and regents

[3H]Dofetilide (specific activity 82.3 Ci·mmol-1) was purchased from Perkin-Elmer (Groningen, The Netherlands). Astemizole was purchased from Sigma Aldrich (Zwijndrecht, The Netherlands). Bovine serum albumin (BSA, fraction V) was purchased from Sigma (St. Louis, MO, USA). G418 was obtained from Stratagene (Cedar Creek, USA). All the other chemicals were of analytical grade and purchased from standard commercial sources. Human embryonic kidney cells stably expressing the K\_11.1 channel (HEK293K\_11.1) were kindly provided by Dr Eckhard Ficker (University of Cleveland, USA).

#### Cell culture and membrane preparation

HEK293K\_11.1 cells were cultured and membranes were prepared and stored as described previously.51

#### Radioligand displacement assay

The binding affinity of all tested compounds was evaluated at 25 °C in a [3H] dofetilide competitive displacement assay as reported earlier.<sup>2</sup>

#### Radioligand competition association assay

The binding kinetics of unlabeled compounds were determined at 25 °C using a [3H]dofetilide competition association assay as described before,2 and the concentrations of unlabeled compounds used in this assay were equivalent to their IC<sub>50</sub> values.

#### Patch clamp assay

HEK293K 11.1 cells were cultured on 10 mm glass coverslips and placed in a perfusion chamber (Cell Microcontrols, Norfolk, USA) at room temperature and perfused with control or test solutions. Whole cell patch clamp measurements and data acquisition were performed using an Axopatch-200B amplifier controlled by

the pClamp software package version 10 (Axon Instruments, Foster City, CA). Patch pipettes were made with a Sutter P-2000 micropipette puller (Sutter Instrument Company, Novato, USA) and had resistances of 1.5-2.5 M $\Omega$  after having been fire polished and filled with pipette solution. Control bath perfusion solution contained (mM): 140 NaCl, 4 KCl, 10 HEPES, 2 CaCl, 1 MgCl, (pH 7.2, NaOH), and the pipette filling solution consisted of (mM): 10 KCl, 125 K-Gluconate, 0.6 CaCl<sub>2</sub>, 2 MgCl<sub>2</sub>, 5 HEPES, 4 Na<sub>2</sub>ATP, 5 EGTA (pH 7.2, KOH). After establishing the whole cell configuration, capacitive transients and pipette access resistance were compensated for by 80%.

The pulse protocol for whole cell voltage clamp measurements of K\_11.1-mediated ion currents was as following. Cells were depolarized to 20 mV from a holding potential -80 mV for 4000 ms, allowing a step current due to activation and inactivation of the K<sub>2</sub>11.1 channels. This was followed by repolarization to a test potential of -50 mV for 5000 ms, during which a tail current was induced by the fast recovery from inactivation and slow deactivation of the channels. The peak tail currents were used for data analysis, and current inhibition was derived from dividing the mean tail current in the presence of selected compounds by the mean tail current under control conditions. Concentration-dependent effects of compounds 21 and 38 on K, 11.1 currents were tested at five different concentrations with an equilibrium time of 6.5 min for each concentration in a perfusion system. To determine time-dependent effects, wash-in experiments were performed with a perfusion of bath solution containing 1 µM 21 or 38. Clampfit 10.4 (Axon Instruments, Foster City, CA) and Prism v. 5.1 (GraphPad, San Diego, CA, USA) were used in analyzing, processing and plotting the data.

#### Data analysis

All experimental data were analyzed using the non-linear regression curve fitting program Prism v. 5.1 (GraphPad, San Diego, CA, USA). Apparent inhibitory binding constants (K, values) were derived from the IC, values according to the Cheng-Prusoff equation with the  $K_D$  value obtained from saturation assay.<sup>2,35</sup> Association and dissociation rates for unlabeled compounds were calculated by fitting the data into the competition association model using "kinetics of competitive binding":34

$$K_{A} = k_{1}[L] + k_{2}$$

$$K_{B} = k_{3}[I] + k_{4}$$

$$S = \sqrt{(K_{A} - K_{B})^{2} + 4k_{1}k_{3}LI10^{-18}}$$

$$K_{E} = 0.5 (K_{A} + K_{B} + S)$$

$$\begin{split} &K_{S} = 0.5 \ (K_{A} + K_{B} - S) \\ &Q = \frac{B_{max} k_{1} L 10^{-9}}{K_{F} - K_{S}} \\ &Y = Q \left( \frac{k_{4} (K_{F} - K_{S})}{K_{F} K_{S}} + \frac{k_{4} - K_{F}}{K_{F}} e^{(-K_{F}X)} - \frac{k_{4} - K_{S}}{K_{S}} e^{(-K_{S}X)} \right) \end{split}$$

Where X is the time (min), Y the specific binding of [3H]dofetilide, k, and  $k_2$  are the  $k_{on}$  (M<sup>-1</sup>·min<sup>-1</sup>) and  $k_{off}$  (min<sup>-1</sup>) of [<sup>3</sup>H]dofetilide, L the concentration of [3H]dofetilide (nM), B<sub>max</sub> the maximum specific binding (dpm) and I the concentration of the unlabelled compound (nM). Fixing these parameters allowed the following parameters to be calculated: k<sub>3</sub>, which is the k<sub>on</sub> value (M<sup>-1</sup>·min<sup>-1</sup>) of the unlabelled compound and k<sub>4</sub>, which is the k<sub>off</sub> value (min<sup>-1</sup>) of the unlabelled compound. The association and dissociation rates were used to calculate the kinetic  $K_{D}$  values from the following equation:  $K_{D} = k_{off}/k_{on}$ . The residence time (RT) was calculated according to the formula:  $RT = 1/k_{off}$ . All values in this study are means of at least three independent experiments with SEM.

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## Chapter 8

**Conclusions and perspectives** 

# About this chapter In this thesis I have explored the allosteric modulation and the kinetics of ligands at the K<sub>v</sub>11.1 channel. The research in this thesis may provide valuable information and potential approaches for relieving the K<sub>v</sub>11.1-induced cardiotoxicity of drug candidates in the future. This chapter will summarize the major findings from these studies, and propose the future directions of this research field.

#### **Conclusions of this thesis**

#### Allosteric modulations of the K<sub>2</sub>11.1 channel

Nowadays, K.11.1 blockade by potential drug candidates has received much attention since it potentially causes proarrhythmic side effects. However, at the start of this research allosteric modulation of the K\_11.1 channel as an alternative way of drug interaction had not been studied in any significant detail. The unique structural features of the K\_11.1 channel, which were thoroughly reviewed in Chapters 1 and 2, imply the presence of multiple ligand binding sites at the channel with possible allosteric interactions between them. In Chapter 3, the allosteric effects of a plethora of compounds as well as potassium ions on the K<sub>2</sub>11.1 channel were evaluated in both [3H]astemizole and [3H]dofetilide binding assays. LUF6200 was identified as a negative allosteric modulator, whereas potassium ions behaved as positive allosteric modulators. Moreover, LUF6200 and potassium ions acted on distinct allosteric sites at the K\_11.1 channel, and these binding sites showed positive cooperativity. This investigation is the first to provide direct evidence for allosteric modulation of the K\_11.1 channel using radioligand binding assays.

Subsequently, a novel negative allosteric modulator LUF7244 was characterized with several different [3H]dofetilide binding assays (Chapter 4). LUF7244 was found to decrease the affinity of the prototypical K\_11.1 blockers dofetilide, astemizole, sertindole and cisapride. Moreover, the antiarrhythmic propensity of this compound was revealed in a newly validated neonatal rat ventricular myocyte (NRVM) model. Pretreatment of the NRVM monolayers with LUF7244 relieved the arrhythmic events induced by astemizole, sertindole and cisapride. Importantly, LUF7244 per se did not significantly affect the action potential durations of NRVMs. Concomittantly, LUF7244 displayed no obvious cytotoxicity upon a long exposure to NRVMs, and exerted negligible effects on astemizole at its intentional target (the human histamine H<sub>1</sub> receptor). Taken together, this study has introduced allosteric modulation of the K\_11.1 channel as a new and effective strategy to reduce the unintended binding of drugs to the channel's central cavity and to further prevent the proarrhythmic changes associated with this binding. This opens a promising avenue for abolishment of K<sub>2</sub>11.1-induced cardiotoxicity of drug candidates via pharmacological combination therapy. Following this chapter, a series of compounds with the same scaffold as LUF7244 were synthesized and further assessed for their allosteric effects on [3H]dofetilide binding to the K<sub>v</sub>11.1 channel in **Chapter 5**. Structure-activity relationships were derived for these compounds, which offers important information for future optimization of allosteric modulators of the K\_11.1 channel. Specifically, compared to LUF7244 several more potent negative allosteric modulators like 7f and 7p were obtained, implicating improved potency in subsequent ex vivo and in vivo antiarrhythmic studies.

#### Kinetics studies of ligands at the K\_11.1 channel

Previous studies on K\_11.1 blockers have primarily focused on their equilibrium parameters such as affinity or potency at the channel. The kinetics of the interaction between a drug and the K<sub>1</sub>11.1 channel has rarely been investigated though, albeit that its importance has been reported with respect to other targets, like enzymes and G protein-coupled receptors. In Chapter 6, a novel [3H]dofetilide competition association assay was successfully developed and validated to characterize the kinetic binding parameters of unlabeled compounds at the K<sub>2</sub>11.1 channel. The association rates of fifteen prototypical, structurally diverse K\_11.1 blockers obtained in this assay were extremely divergent, with e.g., a 50,000-fold difference between moxifloxacin and clofilium, and correlated excellently to their affinity values at the channel. Meanwhile, membrane affinity determined in an immobilized artificial membrane column and other calculated physiochemical properties (molecular weight, pK<sub>3</sub>, and logD) did not correlate with the binding affinity and kinetics of these blockers at the channel. Collectively, these results lead to the conclusion that a compound's affinity at the K\_11.1 channel is mainly regulated by its association rate instead of its dissociation rate. This finding is against the general dogma that affinity values of ligands for their targets are mainly dominated by their dissociation rates with diffusion limited association rates. In this case, association rates can therefore be used to evaluate the K<sub>2</sub>11.1 liability of drug candidates, apart from the affinity. Since the dissociation rates of the selected K\_11.1 blockers in Chapter 6 only exhibited a 10-fold difference between the fastest and slowest dissociating compounds, we therefore aimed to search for and design compounds with a wide range of RTs, i.e. reciprocal of dissociation rates at the channel, in Chapter 7 in order to explore the role of the dissociation process in K<sub>2</sub>11.1 blockade.

In Chapter 7, the association rates and RTs as well as affinity values of four compound series (dofetilide, E-4031, clofilium and pyridinium derivatives) at the K<sub>2</sub>11.1 channel were determined and evaluated. Based on the results, the structure-kinetics relationships of these K<sub>2</sub>11.1 blockers were explored next to their structure-affinity relationships. Interestingly, the RTs of these compounds, in particular the newly synthesized series of pyridinium derivatives, were up to 300-fold different, in addition to dramatic differences in affinity and association

rates. For instance, compound 38 had a RT of  $105 \pm 4$  min at the K 11.1 channel, while 37 displayed a short RT that was only  $0.34 \pm 0.04$  min. Subsequently, a "k<sub>on</sub>-k<sub>off</sub>-K<sub>D</sub>" kinetic map was constructed to putatively categorize all compounds based on their potential K\_11.1 liability, together with three reference drugs (positive control astemizole, negative control ranolazine, and dofetilide). Lastly, two representative compounds were tested in patch clamp assays, and their patch clamp IC<sub>50</sub> values and current inhibition rates were in agreement with the RTs and association rates derived from radioligand binding experiments, respectively. Altogether, the importance of incorporating kinetic binding parameters (association and dissociation rates or RTs) into the current affinity-dominated paradigm is strongly suggested by our studies in order to more comprehensively and accurately evaluate the K<sub>v</sub>11.1 liability of drug candidates. Overall, this work offers a new avenue for drug researchers to efficiently design ligands with less K<sub>v</sub>11.1-induced cardiac side effects by considering ligand binding kinetics in the drug development trajectory.

#### **Future perspectives**

#### Allosteric modulator as a therapeutic agent

Although the antiarrhythmic property of a negative allosteric modulator (LUF7244) was evidenced in a NRVM model (Chapter 4), the capability of this modulator to counteract the proarrhythmic effects of unintentional K\_11.1 blockers should be further investigated in human subjects due to a species difference in K<sub>2</sub>11.1 characteristics between humans and rats. <sup>1-3</sup> Recent access to human-induced pluripotent stem cell (hiPSC) derived cardiomyocytes (hiPSC-CMs) has offered a window of opportunity to overcome this bottleneck induced by species-to-species differences. hiPSC-CMs have been demonstrated to exhibit the expected genomic, biochemical, mechanical and electrophysiological behavior of human cardiomyocytes as evidenced by RT-PCR, patch clamp, and/or immunofluorescence techniques. $^{2,4-6}$  Most importantly, the  $I_{\rm Kr}$  current mediated by the K<sub>2</sub>11.1 channel has been observed in these hiPSC-CMs.<sup>2, 5, 6</sup> To date, different techniques have been explored to assess the drug-induced electrophysiological alterations of hiPSC-CMs, including whole-cell patch clamp, impedance-based cellular and multi-electrode array (MEA) assays.<sup>7,8</sup> For example, the cardiotoxic effects of selective K 11.1 blockers E-4031 and dofetilide were successfully detected with the label-free xCELLigence RTCA Cardio system,9 and the MEA assays have been widely utilized to screen the K 11.1 liability of compounds as well.<sup>8, 10</sup> In this context, the impedance-based xCELLigence system or MEA

technique can be introduced to further investigate the antiarrhythmic effects of LUF7244 in a hiPSC-MC model. In addition, a simple and low-cost integrated platform has been recently reported to mimic the functional properties of native heart by creating large areas of aligned hiPSC-CMs.<sup>11</sup> Thus, the latter platform may produce more clinically and physiologically relevant data for drug screening, which can be applied to the follow-up studies on allosteric modulation of the K\_11.1 channel. Interestingly, K\_7.1 and K\_11.1 activators have shown compensatory effects on long QT syndromes (LQTSs) caused by dysfunctions of either channel type. 13-15 Recently, Zhang et al. established a patient- and disease-specific model with hiPSC-CMs from family members affected by the K\_7.1-mediated LQTS type 1, and found that modulation of the K\_11.1 channel by ML-T531 restored the action potential prolongation caused by this inherited dysfunction of the K<sub>2</sub>7.1 channel. <sup>16</sup> As ML-T531 was identified as a negative allosteric modulator in Chapter 4 of this thesis, K, 11.1 modulators are supposed to possess antiarrhythmic roles in congenital cardiac diseases. 16 Therefore, hiPSC-CM models with different patient-specific LQTSs can be used to evaluate the proarrhythmic propensities of LUF7244 and other more potent modulators like 7f and 7p. Taken together, negative allosteric modulators of the K<sub>2</sub>11.1 channel can enable the rescue of inherited LQTSs alone as well as prevent drug-induced arrhythmias via combination therapy, making them an attractive and promising therapy in the future.

The hiPSC-CM differentiation system mentioned above has brought unique in vitro cellular- or tissue-based assays for the future development of K\_11.1 modulators as a novel alternative to current antiarrhythmic treatments. However, a myriad of ex vivo and in vivo preclinical studies like Langendorff and animal experiments as well as clinical evaluations are still required to eventually translate our negative allosteric modulators into a new class of therapeutic agents for both acquired and inherited arrhythmias. Currently, a zebrafish model of K 11.1-mediated LQTS type 2 has been used as a high-throughput screening approach for compounds that are able to reverse the lengthened myocardial repolarization. 12 This high-throughput in vivo functional whole-organism model provides an efficient opportunity for the future preclinical investigations of K<sub>v</sub>11.1 modulators.

Of note, the current generation of negative allosteric modulators of the K<sub>2</sub>11.1 channel are not very potent (active in the micromolar range), which may raise concerns about their specificity and potential for cardiac and non-cardiac side effects. Thus, the characterization of the binding site of these modulators at the channel will be of particular interest in follow-up studies to facilitate the development of more potent and specific modulators. Here, computational methods, mutational studies and crystallization of the K<sub>2</sub>11.1 channel will bring useful information for such research in the future. 17, 18 Furthermore, more potent allosteric modulators can be synthesized according to the structure-activity relationships concluded in Chapter 5.

#### Binding kinetics of K<sub>2</sub>11.1 blockers

#### Ligand binding kinetics in patch clamp assays

Radioligand binding assays are high-throughput and useful in the early assessment of ligand binding to the K\_11.1 channel, yet they do not provide a functional endpoint. 19, 20 Conversely, voltage patch clamp assays have capabilities to produce the highest quality and most physiologically relevant data of precise and detailed activity of the channel function among all available methods, albeit with relatively low throughput. 19, 21 Historically, potency of K, 11.1 blockers has been measured using patch clamp techniques due to the mandatory requirements of the ICH S7B guideline, while the binding kinetics of drugs at the channel were not. 19, 22-24 Thus, it appears attractive and meaningful to establish patch clamp assays that can be used to directly determine the binding and unbinding kinetics of ligands at the K<sub>2</sub>11.1 channel. In **Chapter 7**, two compounds with distinguished kinetic parameters, in particular RTs, were selected to evaluate their binding kinetics in the whole-cell patch clamp assay. Although the K<sub>v</sub>11.1 current inhibition rates of tested compounds were in the same rank order as their association rates derived from the radioligand binding assay, the  $I_{\rm Kr}$  currents displayed insignificant recovery from the inhibition by both ligands during the washout experiments (data not shown). Subsequently, whole-cell radioligand binding assays, which are different from membrane radioligand binding assays used in the whole thesis, were performed in cells with or without expression of the K 11.1 channel. In this case, [3H]dofetilide exhibited comparable binding at empty HEK293 cells to that at HEK293 cells expressing the K<sub>2</sub>11.1 channel (data not shown). Collectively, we speculated that these K<sub>2</sub>11.1 blockers with an intracellular binding site at the channel could be trapped within the cell after diffusing through the cell membrane, which made the direct measurement of ligand's unbinding kinetics impossible. Therefore, a so-called inside-out patch clamp recording without drug trapping phenomena will be particularly useful to verify our speculation, as this technique will allow for accurately determining the "true" binding kinetics of ligands with an intracellular binding pocket at the K<sub>2</sub>11.1 channel.<sup>25</sup>

#### Ex vivo and in vivo investigations

The categorization of all K<sub>2</sub>11.1 blockers in **Chapter 7** is relatively arbitrary due

to the lack of arrhythmogenic data for the screened compounds. Monophasic action potentials of isolated Langendorff-perfused heart preparations can be recorded to assess ex vivo proarrhythmic liability of K 11.1 blockers, 26-28 and various animal models such as AV dogs and rabbits are widely applied to evaluate the in vivo arrhythmic propensities of drug candidates.<sup>29, 30</sup> Accordingly, the K\_11.1 blockers from Chapter 7 should be further assessed in these different ex vivo and in vivo experiments in order to define a more educated and reliable "k<sub>an</sub>-k<sub>aff</sub>-K<sub>b</sub>" map that allows for a better indication of torsadogenic risks of compounds. Notably, ex vivo and in vivo preclinical studies are commonly carried out on animal hearts or whole animals that are unlikely to mimic diseased human subjects and therefore might not be predictive of the clinical situation.<sup>3, 27, 31, 32</sup> Hence, this kinetic map could be even further improved by incorporating affinity and kinetic parameters of more reference drugs like astemizole, ranolazine and dofetilide, which are selective K\_11.1 blockers and have already been assessed for their proarrhythmic profiles in clinical trials. Ultimately, a more complete translational model based on all in vitro, ex vivo and in vivo data can hopefully be elucidated to improve the predictive index of K<sub>2</sub>11.1 liability of drug candidates in the future.

#### Rechanneling the K\_11.1 liability paradigm

The current assessment of proarrhythmic risks of K<sub>2</sub>11.1 blockers is predominantly dependent on their IC<sub>50</sub> values.<sup>3, 33</sup> However, this approach does not take into account drug binding kinetics, which can have a major impact on cardiac arrhythmias of drug candidates.<sup>24, 34, 35</sup> In 2013 a Comprehensive in vitro Proarrhythmia Assay (CiPA) was proposed as a new paradigm to replace the current regulatory strategy by the Cardiac Safety Research Consortium, and the CiPA was expected to be operational worldwide by July 2015. 34, 36, 37 In this new CiPA paradigm in silico modeling is one of the most important components. In this regard, the best simulation model is suggested to be selected before the release of the CiPA assay, and more sophisticated models that include both simple channel blockade affinity and additional drug binding kinetics are inferred to significantly improve the proarrhythmic assessment for drug candidates.<sup>36</sup> Thus, incorporating association and dissociation rates of e.g., the compounds from Chapter 6 and 7 with their binding affinity may lead to a more efficient, readily applicable computational model for the forthcoming CiPA implementation. As a branch of this extended research, a large library of K<sub>2</sub>11.1 blockers with varied affinity and kinetic parameters should also be synthesized and evaluated to provide enough data for the representative in silico model in the near future. Hopefully, the ultimate, optimized CiPA paradigm will demonstrate much better predictive value for proarrhythmic risks of pharmaceutical products, compared to the traditional strategy recommended by ICH S7A and S7B guidelines. 38, 39

#### Final note

K\_11.1-induced cardiotoxicity has emerged as an unanticipated adverse effect of many pharmacological agents and has become a major obstacle in drug development over the past decades. In this thesis, allosteric modulation of the K\_11.1 channel has been extensively explored, and negative allosteric modulators were shown to relieve the proarrhythmic effects of structurally and therapeutically diverse K<sub>2</sub>11.1 blockers. The most potent modulators can be further assessed in in vitro assays based on hiPSC-CM models, ex vivo and in vivo studies, and may be developed as a new class of antiarrhythmic medications in the future. On the other hand, kinetic binding parameters of a wide range of K<sub>v</sub>11.1 blockers at the channel have been thoroughly investigated in this thesis. Association and dissociation rates or residence times are strongly suggested to be integrated with equilibrium affinity values in the future strategies for a better and more comprehensive evaluation of K<sub>v</sub>11.1 liability of drug candidates. The "k<sub>on</sub>-k<sub>off</sub>-K<sub>D</sub>" kinetic map provides a first and promising classification of K<sub>2</sub>11.1 blockers, which could be beneficial and indicative for drug researchers to design compounds with less K<sub>2</sub>11.1-mediated cardiac side effects in the early stage of drug development. Functionally determining the kinetic parameters of K\_11.1 blockers at the channel, e.g., by using inside-out patch clamp assays, will facilitate the incorporation of ligand-channel binding kinetics into the novel CiPA paradigm. Hopefully, all findings in this thesis have brought new insights into K\_11.1-induced cardiac arrhythmias, and will offer opportunities for restoring or preventing this kind of arrhythmias in the near future.

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## Summary

The superfamily of ion channels is a major focus of drug discovery and development programs in the pharmaceutical industry due to their involvement in a number of physiologic functions. However, the binding of several drugs to ion channels is also responsible for a huge number of side effects, and pharmaceutical safety is routinely required by regulatory agencies for certain ion channels, in particular cardiotoxicity mediated by the K\_11.1 channel. All these subjects have been generally discussed in **Chapter 1** of this thesis.

Chapter 2 is a literature-based review describing research progress on the K\_11.1 channel. In this chapter, we focused on kinetic studies of the channel, including the unique gating kinetics and ligand binding kinetics. In addition, biological production and degradation, structural features, physiological functions, and on- and off-target applications of the K<sub>2</sub>11.1 channel were briefly introduced. A thorough analysis of the K<sub>v</sub>11.1 channel is beneficial for the improvement of drug design and reduction of unwanted arrhythmic side effects caused by the blockade of this channel.

In Chapter 3, a previously disclosed series of compounds together with three reference compounds, were selected and evaluated for the allosteric modulation of the K\_11.1 channel in [3H]astemizole and [3H]dofetilide binding assays. One potent negative allosteric modulator (LUF6200) was identified in radioligand dissociation assays, while potassium ions were found to be positive allosteric modulators. These two modulators interacted with the K\_11.1 channel via separate allosteric sites with positive cooperativity towards each other. Taken together, this investigation provides direct evidence for allosteric modulation of the K<sub>2</sub>11.1 channel.

As an extension of the research described in chapter 3, three compounds (ML-T531, VU0405601 and LUF7244) were extensively evaluated for their negative allosteric effects on the K 11.1 channel using different [3H]dofetilide binding assays in **Chapter 4**. The novel modulator LUF7244 was further assessed for its possible antiarrhythmic propensity in radioligand binding experiments and voltage sensitive optical mapping in a newly validated neonatal rat ventricular myocyte (NRVM) model. The channel affinity of three blockbuster drugs (astemizole, sertindole and cisapride) that have all been withdrawn from market due to their K<sub>2</sub>11.1 blockade, was diminished in the presence of LUF7244. Furthermore, the heterogeneous prolongation of the action potential duration (APD) and/or a high incidence of early afterdepolarizations (by astemizole and sertindole only) induced by these K 11.1 drugs were normalized by pretreatment with 10 µM LUF7244 in the NRVM cultures. Intriguingly, LUF7244 per se did not affect the APD and the viability, excitability and contractility of the cardiomyocytes, and exerted no obvious influence on the intentional blockade of the human histamine H, receptor by astemizole. Accordingly, these results indicate that further development of LUF7244 raises an opportunity for antiarrhythmic drugs via combination therapy.

Subsequently, in **Chapter 5** the synthesis and allosteric evaluation of 29 compounds, which share the same chemical scaffold as LUF7244, were described. Most compounds emerged as negative allosteric modulators at the K\_11.1 channel, and the structure-activity relationships of all these ligands were established. Importantly, compounds 7f and 7p were more potent than LUF7244 in negatively modulating the channel, implying promising antiarrhythmic propensities superior to LUF7244.

A new [3H]dofetilide competition association assay was successfully validated in Chapter 6, and was further utilized to determine the kinetic parameters of fifteen prototypical K 11.1 blockers. In contrast to a classical ligand-receptor interaction mechanism, the affinity of these blockers was predominantly regulated by their association rates instead of dissociation rates, implicating the role of association rates in K\_11.1 blockade. Thereafter, membrane affinity of all these ligands was measured in an immobilized artificial membrane column using HPLC. Similar to general calculated physiochemical properties, it is shown in chapter 6 that membrane interactions of K<sub>v</sub>11.1 blockers did not significantly impact their affinity and ligand binding kinetics.

Since dissociation rates or residence times (RTs) of all K\_11.1 blockers in chapter 6 only exhibited a 10-fold difference, a library of 46 compounds over a wide range of chemical scaffolds were tested for their affinity and kinetic data in Chapter 7. K 11.1 blockers were discovered with up to a 300-fold difference in RTs (0.34 min for compound 37 versus 105 min for 38), which enabled the investigation of the structure-kinetics relationships next to structure-affinity relationships. Afterwards, a "k<sub>on</sub>-k<sub>off</sub>-K<sub>D</sub>" kinetic map was constructed based on the kinetic parameters of these compounds together with those of three reference blockers (astemizole, ranolazine and dofetilide). This kinetic map offers a probable framework for a further and more precise categorization of K<sub>2</sub>11.1 blockers, inferring distinct proarrhythmic risks. Additionally, two representative compounds, 21 and 38, were measured in a whole-cell patch clamp assay. It was found that the potency (IC<sub>50</sub>) of these two ligands was influenced by their RTs, while the  $I_{Kr}$  inhibition rates were in line with their association rates. Therefore, association rates and RTs of K\_11.1 blockers are strongly suggested to be incorporated next to their affinity values into the future paradigm for proarrhythmia safety assessment.

Finally, general conclusions of all these chapters were summarized and the corresponding future research directions based on all findings in this thesis were proposed in Chapter 8. In conclusion, allosteric modulation and ligand binding kinetics at the K\_11.1 channel have been intensively discussed in this thesis. Negative allosteric modulators of the K<sub>2</sub>11.1 channel open a new avenue for mitigating unintentional, arrhythmic effects of drug candidates via pharmacological combination therapy. Likewise, incorporation of ligand binding kinetics, next to affinity for the K\_11.1 channel, into the new Comprehensive in vitro Proarrhythmia Assay (CiPA) could improve the traditional screening paradigm for assessment of K\_11.1-induced cardiotoxicity. Hopefully, considering kinetic parameters of K<sub>2</sub>11.1 blockers at the channel will lead to the design of safer drug candidates as well as to less drug attrition in the near future.

## Samenvatting

De familie van ionkanalen is vanwege zijn vele fysiologische functies al vele jaren een belangrijk aandachtsgebied voor onderzoek naar nieuwe geneesmiddelen. Helaas is gebleken dat verscheidene geneesmiddelen bijwerkingen vertonen die veroorzaakt worden door binding aan een ionkanaal, in het bijzonder door K<sub>v</sub>11.1 kanaal-gemedieerde cardiotoxiciteit. Daarom wordt er tegenwoordig door geneesmiddelautoriteiten onderzoek naar ionkanaal-interacties vereist voordat goedkeuring van een geneesmiddel kan plaatsvinden. Bovengenoemde onderwerpen worden besproken in **hoofdstuk 1** van dit proefschrift.

**Hoofdstuk 2** is een literatuurstudie die de voortgang van onderzoek naar het K<sub>v</sub>11.1 kanaal beschrijft. In dit hoofdstuk hebben we ons gericht op kinetische studies van het ionkanaal, in het bijzonder de unieke kanaalkinetiek en ligandbindingskinetiek. Daarnaast worden biosynthese, degradatie, structurele eigenschappen, fysiologische functies en on- en off-target toepassingen van het K<sub>v</sub>11.1 kanaal geïntroduceerd. Het grondig bestuderen van het K<sub>v</sub>11.1 kanaal draagt bij aan het verbeteren van geneesmiddelonderzoek en het verminderen van aritmische bijwerkingen die veroorzaakt worden door blokkering van dit ionkanaal.

In **hoofdstuk 3** is een selectie van eerder beschreven liganden, kalium ionen en drie referentie liganden onderzocht. Voor de geselecteerde stoffen is gekeken naar allostere modulatie van het K<sub>v</sub>11.1 kanaal in [³H]astemizol en [³H]dofetilide bindingsstudies. In een radioligand dissociatie studie werd gevonden dat één ligand, LUF6200, een potente negatieve allostere modulator is en dat kaliumionen zich als positieve allostere modulatoren gedragen. Deze twee modulatoren induceren hun effect op het K<sub>v</sub>11.1 kanaal via verschillende allostere bindingsplaatsen met een positieve coöperativiteit. Samengevat levert dit onderzoek direct bewijs voor allostere modulatie van het K<sub>v</sub>11.1 kanaal.

Als toevoeging op het in hoofdstuk 3 beschreven onderzoek zijn in **hoofd-stuk 4** met behulp van verschillende [³H]dofetilide experimenten drie liganden, ML-T531, VU0405601 en LUF7244, uitgebreid onderzocht op hun negatieve allostere eigenschappen ten opzichte van het K<sub>v</sub>11.1 kanaal. De nieuwe modulator LUF7244 werd daarnaast verder onderzocht op een potentieel antiaritmisch effect met behulp van radioligand bindingsstudies en een nieuw gevalideerd model, het neonatal rat ventricular myocyte (NRVM) model. De affiniteit van drie blockbuster geneesmiddelen (astemizol, sertindol en cisapride) die vanwege K<sub>v</sub>11.1 kanaal blokkering van de markt gehaald zijn, was verminderd in aanwezigheid

van LUF7244. Daarnaast werden de heterogene verlenging van de duur van de actiepotentiaal en/of een hoge incidentie van vroege napolarisaties (door enkel astemizol en sertindol) - veroorzaakt door deze blockbuster geneesmiddelen - genormaliseerd door voorbehandeling van de NRVM culturen met 10 μM LUF7244. Intrigerend genoeg heeft LUF7244 op zichzelf geen effect op de duur van de actiepotentiaal en op de levensvatbaarheid, prikkelbaarheid en contractiliteit van cardiomyocyten. Daarnaast vertoont het geen effect op de gewenste werking van astemizol, namelijk blokkering van de histamine H<sub>1</sub> receptor. Deze resultaten tonen aan dat verdere ontwikkeling van LUF7244 de mogelijkheid biedt om antiaritmische geneesmiddelen te gebruiken als combinatietherapie.

Vervolgens beschrijft **hoofdstuk 5** de synthese en allostere evaluatie van 29 liganden met een scaffold gelijk aan LUF7244. De meeste liganden bleken negatieve allostere modulatoren voor de blokkering van het K<sub>v</sub>11.1 kanaal en structuur-activiteit relaties van al deze liganden werden beschreven. Liganden **7f** en **7p** bleken potenter dan LUF7244 in het negatief moduleren van de receptor. Dit is veelbelovend omdat het laat zien dat deze stoffen mogelijk betere antiaritmische eigenschappen hebben dan LUF7244.

De ontwikkeling van een nieuw [³H]dofetilide competitie associatie experiment is beschreven in **hoofdstuk 6**. Dit experiment werd vervolgens verder gebruikt om de kinetische parameters van vijftien prototypische K<sub>v</sub>11.1 kanaal-blokkers te bepalen. In tegenstelling tot het klassieke mechanisme van ligand-receptor interactie werd de affiniteit van deze blokkers voornamelijk bepaald door hun associatie-snelheidsconstanten in plaats van hun dissociatie-snelheidsconstanten. Dit toont duidelijk de rol en het belang van associatie-snelheidsconstanten voor blokkering van het K<sub>v</sub>11.1 kanaal. Vervolgens is voor al deze liganden de affiniteit voor het celmembraan bepaald met behulp van een geïmmobiliseerde kunstmatige membraankolom in een HPLC machine. Net zoals voor gebruikelijke fysisch-chemische eigenschappen is in hoofdstuk 6 aangetoond dat de membraaninteractie van K<sub>v</sub>11.1 kanaal-blokkers geen significant effect heeft op hun affiniteit en kinetiek van ligandbinding.

Vanwege het feit dat de dissociatie-snelheidsconstanten en verblijftijden van alle in hoofdstuk 6 beschreven  $K_v11.1$  kanaal-blokkers slechts maximaal een factor 10 verschillen, werd in **hoofdstuk 7** een bibliotheek van 46 liganden getest op affiniteit en kinetiek. Er werden  $K_v11.1$  kanaal-blokkers ontdekt met een 300-voudig verschil in verblijftijd op het ionkanaal, 0.34 min voor **37** t.o.v. 105 min voor **38**. Dit maakte het voor ons mogelijk om structuur-kinetiek relaties te bestuderen naast de gebruikelijke structuur-affiniteit relaties. Vervolgens is op basis van de kinetische parameters van deze liganden samen met die van de referentie liganden, astemizol, ranolazine en dofetilide, een kinetische " $k_{on}$ - $k_{off}$ - $K_{D}$ "-

kaart gemaakt. Deze kaart biedt kader voor een verdere en nauwkeuriger indeling van  $K_v$ 11.1 kanaal-blokkers waarmee de risico's van de verschillende blokkers voorspeld kunnen worden. In aanvulling hierop zijn twee representatieve liganden, **21** en **38**, getest in een whole-cell patch clamp experiment. Het bleek dat de potentie van deze liganden ( $IC_{50}$ ) werd beïnvloed door hun verblijftijd op het ionkanaal, terwijl de  $I_{Kr}$  inhibition rates in lijn waren met de associatie-snelheidsconstanten. Om bovengenoemde redenen raden wij sterk aan om bij het bepalen van de veiligheid van potentieel pro-aritmische liganden de verblijftijd en associatie-snelheidsconstante naast de affiniteit mee te nemen ter evaluatie.

Ten slotte zijn de algemene conclusies van de hoofstukken en de bijbehorende mogelijkheden voor toekomstig onderzoek beschreven in **hoofdstuk 8**. Concluderend, allostere modulatie en ligandbindingskinetiek van het K<sub>v</sub>11.1 kanaal zijn uitvoerig beschreven en bediscussieerd in dit proefschrift. Negatieve allostere modulatoren bieden nieuwe mogelijkheden tot het verminderen van onbedoelde, aritmische effecten van kandidaat-geneesmiddelen via farmacologische combinatietherapie. Daarnaast kan het gebruikmaken van ligandbindingskinetiek, parallel aan de affiniteit, van het K<sub>v</sub>11.1 kanaal in de nieuwe FDA guideline Comprehensive *in vitro* Proarrhythmia Assay (CiPA) leiden tot verbetering van de traditionele bepaling van toxiciteit veroorzaakt door het K<sub>v</sub>11.1 kanaal. Hopelijk zal het evalueren van kinetische parameters van K<sub>v</sub>11.1 kanaal-blokkers in de toekomst leiden tot het ontwikkelen van veiliger geneesmiddelen en tot minder geneesmiddelen die van de markt gehaald moeten worden.

### Curriculum Vitae

Zhiyi Yu was born on January 5<sup>th</sup>, 1986 in Hubei, China. After graduating from high school in 2004, he moved to Sichuan to study pharmacy in Western China School of Pharmacy at Sichuan University. In 2008 he received a bachelor degree, and then he continued a master study in the same university, majoring in medicinal chemistry. During this period he performed a series of studies on chemical synthesis and biological evaluation of antimicrobial compounds under the supervision of Prof. Taiping Hou.

After obtaining his master degree in 2011, Zhiyi Yu was supported by the China Scholarship Council to conduct a Ph.D study in Division of Medicinal Chemistry at the Leiden Academic Centre for Drug Research (Leiden University, The Netherlands) under the supervision of Prof. Ad IJzerman and Dr. Laura Heitman. His doctorate dissertation consisted of two main projects, which were allosteric modulation and ligand binding kinetics at the K<sub>v</sub>11.1 (hERG) channel. During his Ph.D studies from 2011 to 2015, Zhiyi Yu together with his supervisors developed several valuable collaborations with Leiden University Medical Center and University Medical Center Utrecht, where he acquired different research skills in cellular electrophysiology. From October 2015 on, he will work as a postdoctoral fellow in the laboratory of Prof. Jonathan Cohen in Harvard Medical School at Harvard University. He will be applying protein chemistry as well as electrophysiological techniques to study allosteric modulations of neuronal AChRs and GABA<sub>a</sub> receptors.

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