

Cover Page



Universiteit Leiden



The handle <http://hdl.handle.net/1887/30221> holds various files of this Leiden University dissertation

**Author:** Zope, Harshal R.

**Title:** Rationally designed peptide based functional biomaterials

**Issue Date:** 2014-12-23

# CHAPTER 2

## **Effect of the linker region on the fusogenicity of a membrane fusion model system**

*Keywords: Peptides • membrane fusion • liposomes • coiled-coils*

## Abstract

Liposome-Liposome fusion is an effective tool to study natural SNARE (soluble N-ethylmaleimide sensitive factor attachment protein receptor) mediated membrane fusion. To mimic this naturally occurring phenomenon, a model system has been developed consisting of a heterodimeric coiled-coil pair conjugated to a phospho-lipid anchor via a flexible poly(ethylene glycol) (PEG) linker. These peptide amphiphiles are tethered into the lipid bilayers of liposomes. When two batches of liposomes, modified with one of the complementary peptides are mixed, fusion occurs. In this chapter, the influence of the flexible PEG linker length on the Liposome-Liposome fusion efficiency is studied. Lipopeptides in which the PEG linker length is varied (0, 2, 4, 8 and 12 ethylene glycol units) were synthesized. The efficiency of liposome-liposome fusion was determined using dynamic light scattering (DLS), lipid mixing and content mixing assays. Our studies show that the linker has critical role in membrane fusion and PEG<sub>4</sub> provides optimum rate of fusion for the mentioned model system.

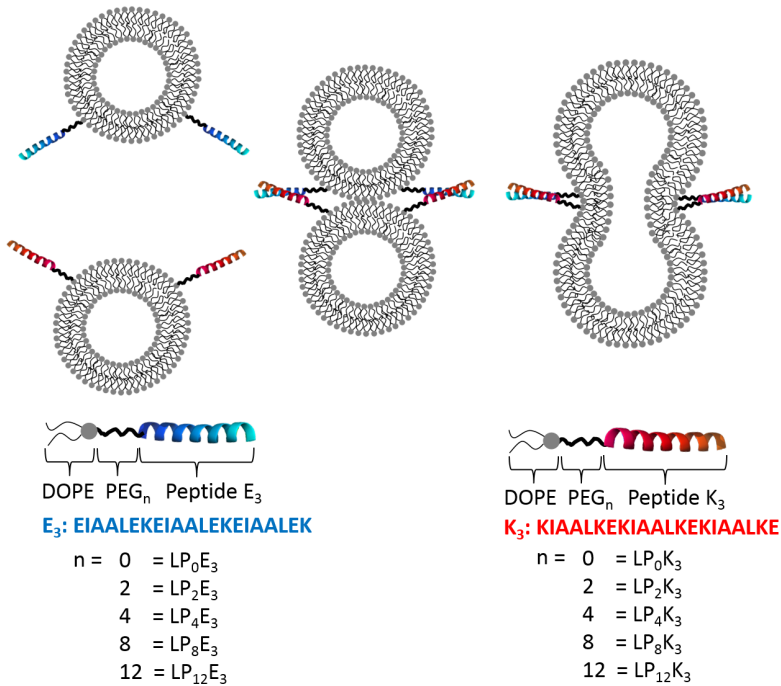
---

## Introduction

Membrane fusion is a vital physiological process as it controls the transport of chemicals in cells and is a source of inspiration for supramolecular chemists. Fusion proceeds with molecular recognition between complementary elements present on the membranes. This interaction brings the two opposing membranes in close proximity resulting in the merging of the membranes.<sup>[1]</sup> Fusion of membranes has been studied intensively using either naturally occurring proteins or supramolecular model systems. For example, in order to achieve membrane fusion in model systems, metal ion co-ordination,<sup>[2]</sup> carbohydrates,<sup>[3]</sup> zwitterion dimerization,<sup>[4]</sup> peptides,<sup>[5]</sup> [6] cyanuric acid–melamine<sup>[7]</sup> and diol-boronic acid<sup>[8]</sup> as molecular recognition units as well light responsive systems<sup>[9]</sup> have been used.

In nature, the highly conserved class of SNARE (soluble N-ethylmaleimide-sensitive factor attachment protein receptor) proteins play a crucial role in many non-viral membrane fusion events.<sup>[10]</sup> SNARES are membrane-bound proteins comprised of three functional domains; a tetrameric coiled coil motif is connected via a flexible linker region to a transmembrane domain. Even though SNARE mediated fusion has been studied intensively, the detailed mechanism of membrane fusion at the molecular level is still under debate.<sup>[11]</sup> Using the bottom-up approach, several groups have attempted to design SNARE protein mimics<sup>[12]</sup> using either pH as a stimulus,<sup>[13]</sup> DNA zipper recognition,<sup>[14]</sup> a PNA binding motif<sup>[15]</sup> or a fusogenic peptide based motif.<sup>[16]</sup> These SNARE mimicking motifs were anchored on the membrane either through lipids like stearic

acid<sup>[13a]</sup> cholesterol,<sup>[14a, 14b]</sup> phospholipids<sup>[14c, 14d, 16-17]</sup> or transmembrane domain peptides.<sup>[15, 18]</sup> All these studies predominantly focused on mimicking the binding motif or anchoring these motifs on lipid membranes, however there are only a few reports studying the role of the linker region.<sup>[14d, 19]</sup>



**Scheme 1:** Liposome-liposome fusion mediated by membrane anchored lipopeptides, along with an overview of the lipopeptides used in this study. Upon mixing of LP<sub>n</sub>K<sub>3</sub> (red) or LP<sub>n</sub>E<sub>3</sub> (blue) decorated liposomes, coiled-coil formation between peptide K<sub>3</sub> and peptide E<sub>3</sub> brings the opposite membranes in close proximity leading to a membrane fusion. The amino acid sequence of E<sub>3</sub> and K<sub>3</sub> is shown, n denotes the number of ethylene glycol units in the linker region and L denotes DOPE lipid.

## Design and Synthesis

Our group has developed a reduced SNARE model composed of three functional domains. It is based on a small heterodimeric coiled-coil motif anchored to a phospholipid on the membrane of a liposome via a flexible poly(ethylene glycol), i.e. PEG linker (**Scheme 1**).<sup>[17]</sup> This modular approach enables a systematic study of the contributions of the individual functional domains on the process of membrane fusion. Varying the number of heptad repeats of the peptides forming the coiled coil binding motif<sup>[20]</sup> as well as changing hydrophobic anchors has a significant influence on the fusion efficiency.<sup>[21]</sup> In this chapter, the influence of the PEG linker length on fusion efficiency is studied. Other studies using DNA

and natural SNAREs fragments acting as the linker region have been studied for their effect on the extent of fusion.<sup>[14d, 19]</sup> Both studies showed that the addition of a linker region between the recognition unit and the membrane anchor resulted in a decreased fusion efficiency. Here we will show that the length of the linker region has a strong influence on the fusion efficiency in our model system as well.

The coiled-coil forming peptide pair consisting of  $E_3$ , with the amino acid sequence  $(EIAALEK)_3$ , and peptide  $K_3$  with the sequence  $(KIAALKE)_3$  was chosen as the recognition unit. Coiled-coil formation between these complementary peptides results in the assembly of a heterodimeric complex is driven by hydrophobic as well as electrostatic interactions between the charged side chains of the peptides.

To anchor the coiled coil binding motif on the membrane, the phospholipid 1,2-Dioleoyl-sn-Glycero-3-Phosphoethanolamine (DOPE) was used as a trans-membrane domain. To study the influence of the length of a flexible poly(ethylene glycol) linker (PEG) on the efficiency of liposome-liposome fusion, lipopeptides  $LP_nK_3$  and  $LP_nE_3$  in which the number of ethylene glycol repeat units was varied between  $n=0$  and 12, were synthesized.

## Results and Discussion

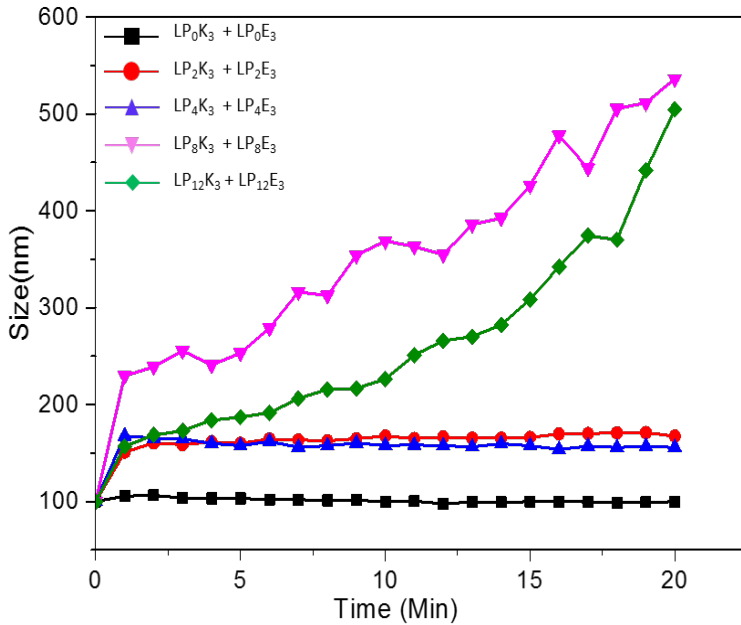
The peptides  $K_3$  and  $E_3$  were synthesized with automated standard solid phase peptide synthesis protocols using Fmoc chemistry. After coupling of the final amino acid, the resin was removed from the reaction vessel and Fmoc-NH-PEG<sub>n</sub>-COOH was coupled manually to the immobilized peptides on the resin and subsequently the Fmoc group was removed. The PEG component was comprised of 2, 8 or 12 repeating units. For the synthesis of the tetra(ethylene glycol) linker,  $N_3$ -PEG<sub>4</sub>-COOH was coupled to the resin and the azide was subsequently reduced using trimethylphosphine to obtain the free amine. Next, succinic anhydride was coupled to the n-terminus of the PEG-peptides in the presence of a base. The DOPE anchor was finally coupled to the succinate group by activation of the free acid group using HOBt/DIC in DMF:DCM (1:1) and an overnight reaction with DOPE. The lipopeptides were cleaved from the resin and subsequently purified using reversed phase chromatography yielding purities >95%. For the membrane fusion assays, lipopeptides were incorporated in liposomes composed of DOPC, DOPE and cholesterol (CH) with a molar ratio of 2:1:1. Liposomes were obtained by hydrating a lipid film containing mentioned lipid composition and 1 mol % of either  $LP_nK_3$  or  $LP_nE_3$  in a buffered solution followed by sonication at 50 °C for 5 min.<sup>[17]</sup> Dynamic light scattering (DLS) measurements revealed that the liposomes were typically 100 nm in diameter and stable for at least 24h.

Next, the effect of the PEG linker length on lipopeptide mediated membrane fusion was studied using lipid and content mixing assays as well as DLS

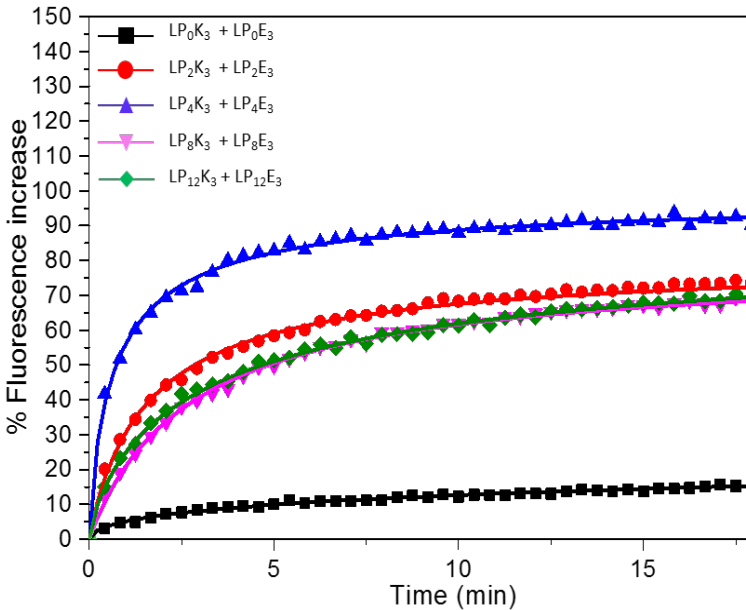
measurements. Mixing of  $LP_nK_3$  and  $LP_nE_3$  functionalized liposomes will lead to an immediate molecular recognition between the complementary peptides  $K_3$  and  $E_3$  and thereby forcing the opposing membranes in close proximity resulting in docking and subsequent fusion. As a result, the size of the resulting assemblies increases and this process was quantified using DLS measurements. When only one of the lipopeptides was present on the liposomal membranes no increase in the hydrodynamic diameter was observed, showing that coiled coil formation is critical to drive fusion. In contrast, when liposomes functionalized with  $LP_nK_3$  and  $LP_nE_3$  were mixed, a strong increase in size was observed for linkers  $PEG_8$  and  $PEG_{12}$  due to liposome docking and fusion events (**Figure 1**). Interestingly, the linker  $PEG_2$  and linker  $PEG_4$  showed a size increase up to  $\sim 160$  nm and remained stable throughout the experiment suggesting that no significant aggregation occurred. Surprisingly, with lipopeptides lacking a linker ( $PEG_0$ ), no increase in hydrodynamic diameter was observed indicating that fusion did not occur.

To confirm that the observed size increase is a result of membrane fusion and not just a result of docking events and without merging two opposing membranes, a lipid mixing assay was conducted. For this,  $LP_nK_3$  liposomes with an average size of 100 nm were decorated with the Förster resonance energy transfer (FRET) pair composed of the donor dye N-(7-nitrobenz-2-oxa-1,3-diazol-4-yl)-dioleoyl-phosphatidyl-ethanolamine (DOPE-NBD) and the acceptor dye lissamine rhodamine-phosphatidyl-ethanolamine (DOPE-LR).<sup>[22]</sup> Upon mixing with plain liposomes negligible fluorescence increase was observed (**Figure A2**). However, upon mixing with  $LP_nE$  liposomes, a significant increase in the NBD fluorescence was observed due to an increase in the average distance of the FRET pair indicative of lipid mixing between the  $LP_nK_3$  liposomes and the  $LP_nE_3$  liposomes. The lipid mixing assay shows that the  $PEG_4$  linker displays the highest efficiency in lipid mixing whereas all other PEG linkers with a longer length gave slightly weaker lipid mixing efficiency. Surprisingly, when lipopeptides were used that did not have a linker (i.e.  $LP_0K_3$  and  $LP_nE_3$ ), hardly any fusion could be detected (**Figure 2**). The observed low efficiency for the linker  $PEG_0$  could be due to reduced molecular recognition between  $K_3$  and  $E_3$  which indicates that the linker region has a significant role in achieving efficient coiled coil formation in this model system (**Table 1**).

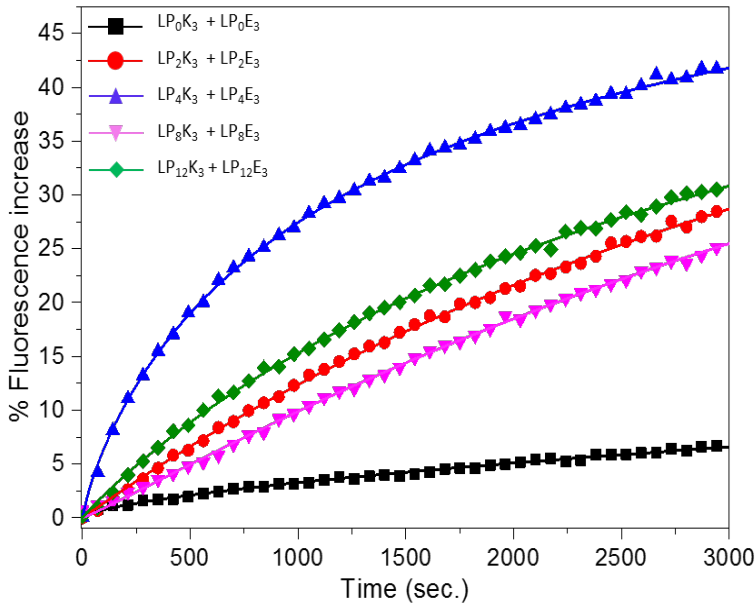
To determine that the observed lipid mixing is not due to hemi fusion which is an intermediate step in the fusion process, we performed a content mixing assay (**Figure 3**). The dye sulphorhodamine B was encapsulated at a self-quenching concentration (20 mM) in liposomes decorated with  $LP_nE_3$ . Non-encapsulated dye was separated from the liposomes using a sephadex G50 column. Upon fusion of sulphorhodamine B loaded  $LP_nE_3$  liposomes with  $LP_nK_3$  liposomes an increased fluorescence is observed due to the dilution of the fluorescent dye and concomitant relief of self-quenching.<sup>[18, 23]</sup> In the content mixing assay, the



**Figure 1:** The hydrodynamic diameter determined by DLS as a function of time and PEG-linker length. At  $t=0$  minutes,  $E_3$ -decorated liposomes were mixed with  $K_3$ -decorated liposomes in equimolar amounts. The total lipid concentration was 0.5 mM, with 1 mol% of lipopeptide, in PBS, pH 7.4.



**Figure 2:** Lipid mixing between  $E_3$  and  $K_3$  decorated liposomes as indicated by the NBD emission increase. Total lipid concentrations were 0.1 mM, with 1 mol% lipopeptide in PBS, pH 7.4.



**Figure 3:** Content mixing between  $E_3$  and  $K_3$  decorated liposomes as indicated by an increase in sulphorhodamine emission. Total lipid concentrations were 0.1 mM, with 1 mol% lipopeptide in HEPES.

same trend as for the lipid mixing assay was observed confirming that full fusion *i.e.* the mixing of the liposomal content has occurred. To ensure, that the obtained results are due to the heterodimerisation of the peptide “ $K_3$ ” and “ $E_3$ ”; control experiments were conducted. Plain liposomes were mixed with sulphorhodamine B containing  $LP_nE_3$  liposomes and little or no fluorescence increase was observed (**Figure A3**) revealing that the fusion occurs only when coiled coil interaction between  $E_3$  and  $K$  peptides is occurs.

Next, the effect of PEG linker length on the secondary structure of the membrane anchored lipopeptides was investigated to better understand the observed differences in the lipid- and content mixing assays. Circular Dichroism (CD) studies of the lipidated peptides in liposomes are shown in **Figure A4** and the main results are summarized in **Table 1**. We observed that all the lipopeptide pairs used in this study are able to form a coiled-coil complex. However, there are significant differences observed between the percentages of  $\alpha$ -helicity at post fusion state. For PEG linkers with  $n = 4, 8$  and  $12$ , similar  $\alpha$ -helicities in postfusion state (*i.e.* after mixing  $LP_nK_3$  liposomes with  $LP_nE_3$  liposomes) were observed, but the extent of fusion was different. This effect is presumably due to the fact that the distance between two membranes upon coiled coil formation is crucial in order to have efficient lipid and content mixing which is in good agreement with other previous studies performed in other model systems.<sup>[19, 24]</sup>



**Table 1:** Circular dichroism (CD) data for LP<sub>n</sub>K<sub>3</sub> and LP<sub>n</sub>E<sub>3</sub> lipopeptides bearing liposomes

PEG Linker (n)	LP <sub>n</sub> K <sub>3</sub>		LP <sub>n</sub> E <sub>3</sub>		LP <sub>n</sub> K <sub>3</sub> +LP <sub>n</sub> E <sub>3</sub>	
	$[\theta]_{224}/[\theta]_{211}$	%H	$[\theta]_{224}/[\theta]_{211}$	%H	$[\theta]_{224}/[\theta]_{211}$	%H
<b>0</b>	0.95	31.5	0.91	32	1.02	32.9
<b>2</b>	0.96	31.4	1.08	25	1.07	31.9
<b>4</b>	0.96	41.4	0.95	31.9	1.02	46.8
<b>8</b>	1.01	49.4	0.94	28.2	1.1	47.2
<b>12</b>	1.01	37.01	0.97	36.9	1.07	45.5
<b>Only Peptide</b>	0.44	20.9	0.43	22.4	0.97	65.5

*Ellipticity ratios using minima (~222/208 nm) for E<sub>3</sub>-decorated liposomes and K<sub>3</sub>-decorated liposomes in 1:1 mixtures, as measured with circular dichroism. The total lipid concentration was 0.5 mM, with 1 mol% lipopeptide, in PBS.*

Boxer and co-workers used short DNA binding motifs as the recognition unit in a membrane fusion model system to study the role of the linker region on the fusion process.<sup>[14d]</sup> It was shown that the addition of a poly-T linker to the membrane proximal region reduces the efficiency of lipid and content mixing. This study has shown that a specific length of the linker (i.e. PEG<sub>4</sub>) yields the highest efficiency in lipid and content mixing in our coiled-coil based model. It is difficult to proof what the exact role of the linker might be. It could be that the linker provides conformational flexibility for the lipidated peptides in order to bind to its complementary peptide at an opposing membrane. With longer linkers (i.e. PEG<sub>8</sub> or PEG<sub>12</sub>) the distance between the two liposome membranes increases and/or the steric hindrance increases due to favorable peptide-PEG interactions results in a lower membrane fusion efficiency.<sup>[25]</sup> Future efforts will be dedicated to investigating and formulating a detailed understanding of the linker region by comparing with natural SNARE linkers.

## Conclusion

In our membrane fusion model system, the length of the flexible PEG-linker is critical and significantly influences the kinetics and the final yield of liposomal fusion as shown by the differences in the DLS measurements, lipid mixing and content mixing assays. From this study it is clear that the tetra(ethylene glycol) linker is the optimal length in combination with the DOPE anchor to obtain efficient fusion between liposomes. A longer linker reduces the fusion efficiency,

while when the linker is too short or absent the ability of coiled coil formation is lowered resulting in low rates of membrane fusion. PEG<sub>4</sub> linker aids an optimal membrane fusion between liposomes due to an efficient coiled coil formation and providing a most favorable distance between two opposing membranes, which give rise to less aggregation and better rate of membrane fusion.

## Appendix

### Abbreviations

CD, circular dichroism; CH, cholesterol; DIPEA, N,N-diisopropylethylamine; DMF, N,N-dimethylformamide; NMP, N-Methyl-2-Pyrrolidone; DOPC, 1,2-dioleoyl-sn-glycero-3-phosphatidylcholine; DOPE, 1,2-dioleoyl-sn-glycero-3-phosphatidylethanol-amine; DOPE-LR, 1,2-dioleoyl-sn-glycero-3-phosphatidylethanolamine-lissamine-rhodamine B; DOPE-NBD, 1,2-dioleoyl-sn-glycero-3-phosphoethanolamine-N-(7-nitro-2-1,3-benzoxadiazol-4-yl) (ammonium salt); Fmoc, fluorenylmethoxycarbonyl; DIC, N,N'-Diisopropylcarbodiimide; HCTU, 1H-Benzotriazolium 1-[bis(dimethylamino)methylene]-5chloro-hexafluorophosphate (1-),3-oxide; MALDI-TOF, matrix-assisted laser desorption-ionization time-of-flight; RP-HPLC, reversed-phase high-pressure liquid chromatography; TEA, triethyl amine; TFA, trifluoroacetic acid; TIS, triisopropylsilane; PEG, polyethylene glycol.

### Materials

Amino acids Fmoc-Lys(Boc)-OH, Fmoc-Glu(OtBu)-OH, Fmoc-Ile-OH, Fmoc-Leu-OH and Fmoc-Ala-OH as well as DMF, DCM and NMP (peptide synthesis grade) were purchased from Biosolve. trimethylphosphine and the lipids DOPE, DOPC and Cholesterol were obtained from Sigma-Aldrich while the fluorescent lipids, DOPE-NBD and DOPE-LR were purchased from Avanti Polar Lipids. Fmoc-NH-(PEG)<sub>2</sub>-COOH (mw 385.42 g/mol), Fmoc-NH-d(PEG)<sub>8</sub>-COOH (mw 663.75 g/mol) and Fmoc-NH-(PEG)<sub>12</sub>-COOH (mw 839.98 g/mol) linkers were purchased from Iris biotech. N<sub>3</sub>-PEG<sub>4</sub>-COOH (mw 279 g/mol) was synthesized as described previously.<sup>[26]</sup> Milli-Q water with a resistance of more than 18.2 mΩ/cm was provided by a Millipore Milli-Q filter system with filtration through 0.22 μM millipak filter. Phosphated buffered saline, PBS: 5 mM KH<sub>2</sub>PO<sub>4</sub>, 15 mM K<sub>2</sub>HPO<sub>4</sub>, 150 mM NaCl, pH 7.4

### Lipopeptide synthesis

Peptide synthesis was performed with a Biotage Syro I on a 100 μM scale using Rink amide resin with a degree of loading of 0.62 mmol/g. In each coupling reaction, 4 eq. of amino acid, 4 eq. of HCTU and 8 eq. of DIPEA were dissolved in 2 mL of DMF and added to the resin for 45 min. Fmoc deprotection was carried out using 2 cycles consisting 40 vol% piperidine and 20 vol% piperidine in DMF for 2 and 15 minutes respectively. The N-terminal free amine was used to couple the PEG linker through standard solid phase chemistry using 1.25 eq of Fmoc-PEG<sub>n</sub>-COOH, 3 eq. of DIC and 3 eq. of HOBt in 3 ml DMF.

N<sub>3</sub>-PEG<sub>4</sub>-COOH was synthesized and utilized using procedure described earlier.<sup>[26c]</sup> The N-terminal free amine of resin was used to couple N<sub>3</sub>-PEG<sub>4</sub>-COOH,

by adding 2 eq. of  $N_3$ -PEG<sub>4</sub>-COOH, 3 eq. of HOBt and 3 eq. of DIC in 3 ml DMF. Reaction was left over the shaker for 18 h. After several resin washes with DMF and DCM, the azide functionality was reduced to an amine with the aid of 10 eq of trimethylphosphine in 7 ml dioxane/water (6:1 v/v). After 2 h the reactants were removed from the reaction vessel and the procedure was repeated one more time to obtain N-terminal free amine. Afterwards, several washes with dioxane/water and DMF were carried out.

Except for  $N_3$ -PEG<sub>4</sub>-COOH containing lipopeptides, Fmoc deprotection of the PEG component was achieved using 20 vol% piperidine in DMF for 20 min to obtain an N-terminal free amine. Further, succinic anhydride was coupled to the N-terminal free amine using 10 eq. of succinic anhydride and 10 eq. TEA in 7 ml NMP. The reaction was left overnight on a shaker. The resin was washed thoroughly with NMP, DCM, and NMP to remove excess of reagents. Next, the resin was activated using 4 eq. of HOBt and 4 eq. of DIC in 2 ml of DCM:DMF (1:1, v/v) for 1hr. In separate vial, 2 eq. of DOPE was dissolved in 2 ml of DCM:DMF (1:1, v/v) containing 2 eq. of TEA. After 1 hr, the DOPE solution was added to the resin and gently heated to ~55 °C for 10 min to achieve efficient coupling and then the reaction was left shaking overnight at room temperature. Next, the resin was washed thoroughly with DMF and DCM to remove excess reactants. The lipopeptide was cleaved from the resin and concomitantly deprotected for 3 hours in 7 ml of a TFA:TIS:Water (95%:2.5%:2.5%) mixture and resulting mixture was co-evaporated with toluene on the rotary evaporator to ensure complete removal of TFA to obtain crude lipopeptides. Amount of crude lipopeptide was determined and dissolved in water: acetonitrile: tert-butanol (1:1:1, v/v/v) to obtain 20 mg/ml lipopeptide solution which was further injected in HPLC for purification.

RP-HPLC was performed with a Shimadzu HPLC system with two LC-8A pumps, and an SPD-10AVP UV-VIS detector. Sample elution was monitored by UV detection at 214 nm and 256 nm. Samples were eluted with a linear gradient from A to B, A being 10 % (v/v) acetonitrile, 0.1 % (v/v) TFA in water, and B being 90 % (v/v) acetonitrile, 0.1 % (v/v) TFA in water. Purification of the lipopeptides was performed on a Vydac C4 reversed phase column (214TP1022, 22 mm diameter, 250 mm length, 10.00 μm particle size) with a flow rate of 20 mL/min. Collected fractions were tested for >95% purity using LC-MS with a Gemini C18 column, lyophilized and to store at -20 °C.

### LC-MS analysis

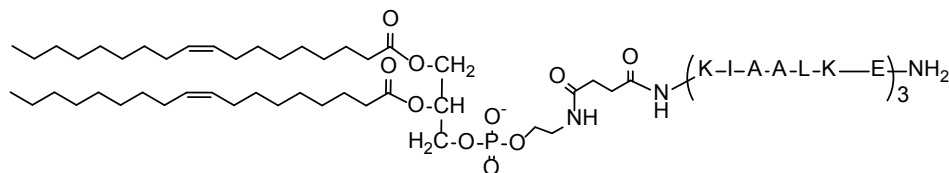
Purity was confirmed using LC-MS analysis equipped with Gemini 3μ C18 column coupled with Finnigan LCQ advantage max (Thermo) ESI-MS analyzer. 0.1%TFA

containing acetonitrile/water was used as a mobile phase. The following gradient (Table A1) was used with a flow rate of 1ml/min.

**Table A1:** Gradient used in LCMS analysis of peptides and lipopeptides.

Time (min)	Gradient
0-2	10% (v/v) acetonitrile in water (0.1% TFA)
2-12	10%-90%(v/v) acetonitrile in water (0.1% TFA)
12-13	90%(v/v) acetonitrile in water (0.1% TFA)
13-15	10%(v/v) acetonitrile in water (0.1% TFA)

1. LP<sub>0</sub>K

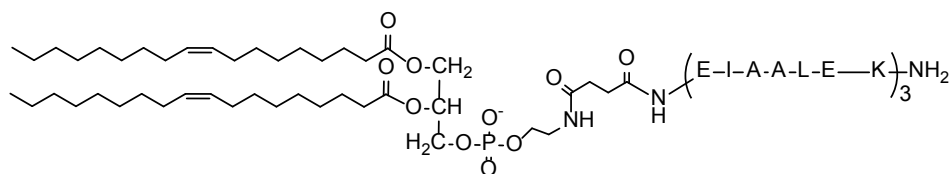


Chemical Formula: C<sub>150</sub>H<sub>271</sub>N<sub>29</sub>O<sub>37</sub>P<sup>-</sup> [3103.96]

Retention time: 10 min

LC-MS: Calcd. [1552.9, M+2H]<sup>+2</sup>, Found 1553.01

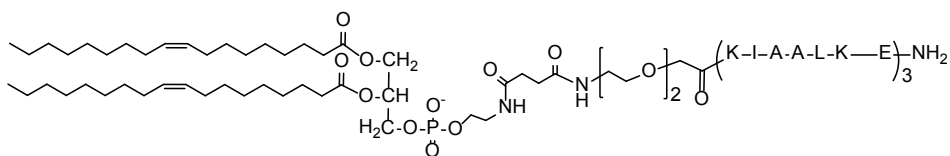
2. LP<sub>0</sub>E



Chemical Formula: C<sub>147</sub>H<sub>256</sub>N<sub>26</sub>O<sub>43</sub>P<sup>-</sup> [3106.78]

Retention time: 12 min

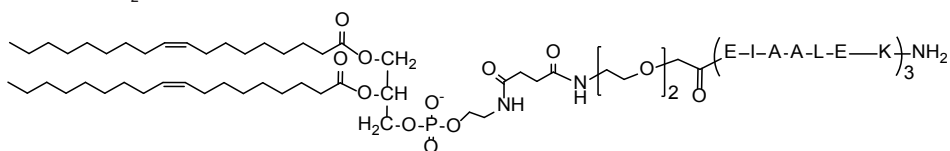
LC-MS: Calcd. [1554.39, M+2H]<sup>+2</sup>, Found 1563.44, [H+NH<sub>4</sub>]<sup>+2</sup>

3. LP<sub>2</sub>K

Chemical Formula: C<sub>156</sub>H<sub>282</sub>N<sub>30</sub>O<sub>40</sub>P<sup>-</sup> [3249.12]

Retention time: 10min

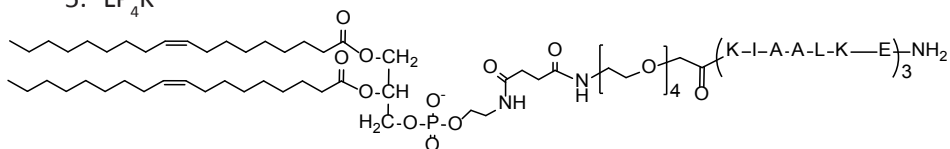
LC-MS: Calcd. [1084.04, M+2H]<sup>+</sup>, Found 1084.03

4. LP<sub>2</sub>E

Chemical Formula: C<sub>153</sub>H<sub>267</sub>N<sub>27</sub>O<sub>46</sub>P<sup>-</sup> [3251.94]

Retention time: 12 min

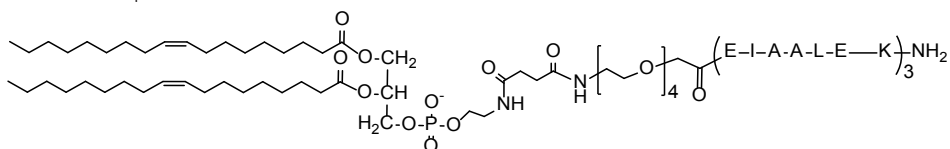
LC-MS: Calcd. [1084.98, M+2H]<sup>+</sup>, Found 1084.98

5. LP<sub>4</sub>K

Chemical Formula: C<sub>160</sub>H<sub>290</sub>N<sub>30</sub>O<sub>42</sub>P<sup>-</sup> [3337.22]

Retention time: 10 min

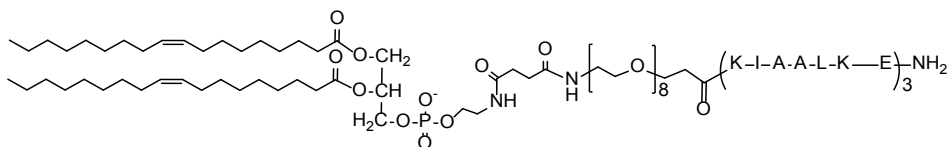
LC-MS: Calcd. [1113.41, M+3H]<sup>+</sup>, Found 1113.72

6. LP<sub>4</sub>E

Chemical Formula: C<sub>157</sub>H<sub>275</sub>N<sub>27</sub>O<sub>48</sub>P<sup>-</sup> [3340.04]

Retention time: 12 min

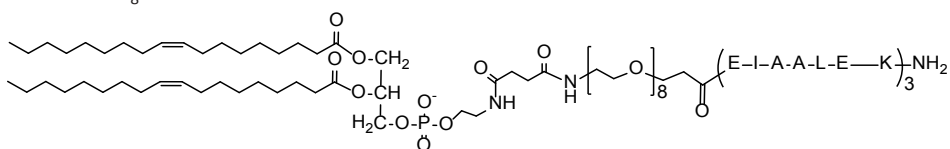
LC-MS: Calcd. [1114.34, M+3H]<sup>+</sup>, Found 1114.33

7. LP<sub>8</sub>K

Chemical Formula: C<sub>169</sub>H<sub>308</sub>N<sub>30</sub>O<sub>46</sub>P [3527.46]

Retention time: 10 min

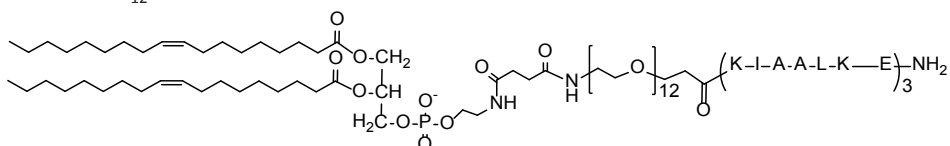
LC-MS: Calcd. [1176.82, M+3H]<sup>+</sup>, Found 1177.09

8. LP<sub>8</sub>E

Chemical Formula: C<sub>166</sub>H<sub>293</sub>N<sub>27</sub>O<sub>52</sub>P [3530.28]

Retention time: 12min

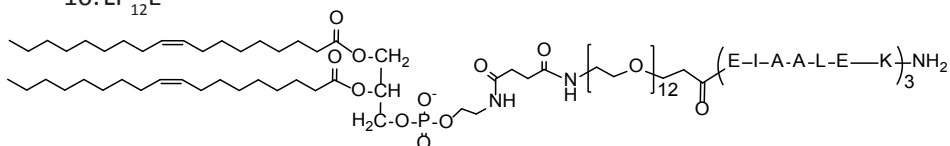
LC-MS: Calcd. [1177.76, M+3H]<sup>+</sup>, Found 1178.04

9. LP<sub>12</sub>K

Chemical Formula: C<sub>177</sub>H<sub>324</sub>N<sub>30</sub>O<sub>50</sub>P [3703.67]

Retention time: 10 min

LC-MS: Calcd. [1235.55, M+3H]<sup>+</sup>, Found 1235.79

10. LP<sub>12</sub>E

Chemical Formula: C<sub>174</sub>H<sub>309</sub>N<sub>27</sub>O<sub>56</sub>P [3706.49]

Retention time: 12 min

LC-MS: Calcd. [1236.49, M+3H]<sup>+</sup>, Found 1236.74

## Liposome Preparation

A 1 mM stock solution of DOPC:DOPE:CH (50:25:25 mol%) was prepared in dry chloroform and stored at -20 °C. Lipopeptide stock solutions (10 μM) were prepared in chloroform:methanol (1:1) and also stored at -20 °C. Liposomes (0.1 mM) in PBS buffer were prepared containing 1 mol% lipopeptide as reported before.<sup>[17]</sup> In 20 ml glass vial, lipids (100 μl from stock solution) and lipopeptide (100 μl from stock solution) were mixed and the mixture was dried under continuous air flow to obtain lipid/lipopeptide film. This film was hydrated with 1 ml PBS buffer (pH 7.2) and sonicated for 3-5 min at 50 °C using a Branson bath sonicator to obtain 0.1 mM liposomes with an average diameter of 100nm.

## Lipid mixing assay

LP<sub>n</sub>K<sub>3</sub> decorated liposomes (DOPC/DOPE/CHOL/NBD-DOPE/RHD-DOPE; 50/24.75/24.75/0.5/0.5 mol %) were mixed with LP<sub>n</sub>E<sub>3</sub> (1 mol%) decorated liposomes (DOPC/DOPE/CHOL; 50/25/25 mol %). Fluorescence measurements for lipid mixing were performed using a luminescence spectrometer LS50B (Perkin Elmer). All spectra were obtained at room temperature using a cuvette with a 1 cm path length. The NBD fluorescence was used to calculate the lipid mixing percentage with time. Fluorescence measurements were started immediately after mixing 750 μL of the fluorescent-labelled LP<sub>n</sub>K<sub>3</sub> liposome suspension with 750 μL of unlabelled LP<sub>n</sub>E<sub>3</sub> liposome suspension in the cuvette. The NBD fluorescence intensity at 530 nm was monitored in a continuous fashion. To calculate the percentage of fusion the following equation (1) was used:

$$F_{(\%) } = [(F_{(t)} - F_{(0)}) / [(F_{(max)} - F_{(0)})] \times 100 \quad (1)$$

where F(t) is the fluorescence intensity measured at time t, F<sub>0</sub> is the 0% fluorescence and F<sub>max</sub> is the fluorescence intensity measured using 1:1 lipid mixing.

## Content Mixing assay

Content mixing experiments were carried out as reported earlier.<sup>[27]</sup> A dried film containing DOPC/DOPE/CH (50:25:25 mol%) and the corresponding LP<sub>n</sub>E<sub>3</sub>-peptides (1 mol%) were hydrated and sonicated (5 min at 50°C) with a sulforhodamine B (20 mM) containing HEPES buffer solution (20 mM HEPES, 90 mM NaCl, pH 7.2). The final lipid concentration was 1 mM. To remove non-encapsulated dye the liposomal solution was subjected to Sephadex column (G50, Superfine, 15 cm in length) using HEPES (20 mM HEPES-Na, 90 mM NaCl) buffer as eluent. The fraction containing the liposomes was collected and diluted to a final liposome concentration of 0.1 mM. 600 μL of the LP<sub>n</sub>E<sub>3</sub> containing liposomes with encapsulated sulforhodamine B were added to a small volume disposable



cuvette. Fluorescence measurements for content mixing were performed using a luminescence spectrometer LS50B (Perkin Elmer). All spectra were obtained at room temperature using a cuvette with a 1 cm path length. The fluorescence signal of the Sulforhodamine ( $\lambda_{em} = 580$  nm) was detected and 600  $\mu$ L of the corresponding LP<sub>n</sub>K<sub>3</sub> containing liposomes (0.1 mM) in HEPES-buffer were added and the increase of sulforhodamine B fluorescence, due to a relief of self-quenching, was detected. After 30 min, 120  $\mu$ L of 10% (v/v) solution of Triton X was added to lyse the liposomes and reach maximum fluorescence. To calculate the percentage of fusion, equation (2) was used:

$$F_{(\%)} = [(F_{(t)} - F_{(0)}) / (F_{(max)} - F_{(0)})] \times 100 \quad (2)$$

where  $F(t)$  is the fluorescence at a certain time,  $F_{(max)}$  is the fluorescence after lysis of the liposomes with Triton X and  $F(0)$  is the starting fluorescence after addition of the K-peptide containing liposomes.

### DLS Measurements

Particle size distributions were obtained with the aid of a Malvern Zetasizer Nano ZS which was equipped with a peltier controlled thermostatic holder. The laser wavelength was 633 nm and the scattering angle was 173°. To obtain an estimation of the hydrodynamic radius,  $D_h$ , the Stokes-Einstein relation (3) was used:

$$D = \frac{k_b T}{3\pi\eta D_h} \quad (3)$$

Here,  $k_b$  is the Boltzmann constant and  $\eta$  is the viscosity of the solvent. Measurements were carried out at room temperature.

For the DLS measurements, 0.5 mM liposomes bearing 1 mol% of LP<sub>n</sub>K<sub>3</sub> or LP<sub>n</sub>E<sub>3</sub> were prepared. LP<sub>n</sub>K<sub>3</sub> liposomes were mixed with LP<sub>n</sub>E<sub>3</sub> liposomes and size was measured for 30 min at 25 °C.

### CD measurements

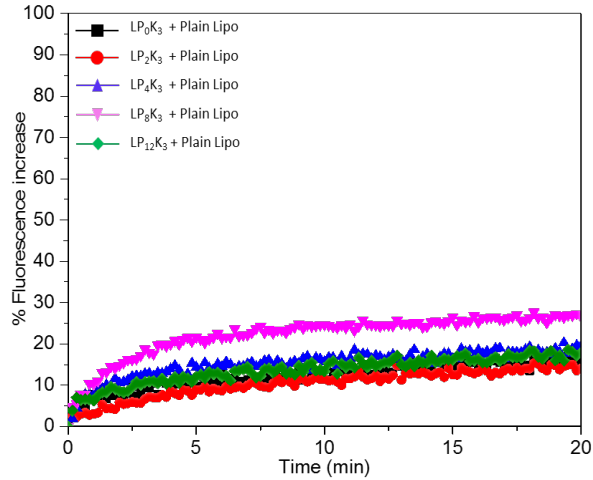
CD spectra were obtained using a Jasco J-815 spectropolarimeter equipped with a peltier-controlled thermostatic cell holder (Jasco PTC-423S). 0.5 mM liposomes of 100 nm size were decorated with 1 mol% LP<sub>n</sub>K<sub>3</sub> or LP<sub>n</sub>E<sub>3</sub> lipopeptides. Spectra were recorded from 260 nm to 200 nm in a 5.0 mm quartz cuvette at 25 °C. Data was collected at 0.5 nm intervals with a 1 nm bandwidth and 1 s readings. Each spectrum was the average of 5 scans. Measurements were started immediately after mixing. All spectra were recorded in 20 mM phosphate buffer at pH 7.4. For analysis each spectrum had the appropriate background spectrum (buffer or liposomes of 100 nm in size) subtracted. The ellipticity is given as mean residue molar ellipticity,  $[\theta]$  ( $10^3$  deg cm<sup>2</sup> dmol<sup>-1</sup>), calculated using equation (4)

$$[\vartheta] = (\vartheta_{obs} MRW) / (10lc) \quad (4)$$

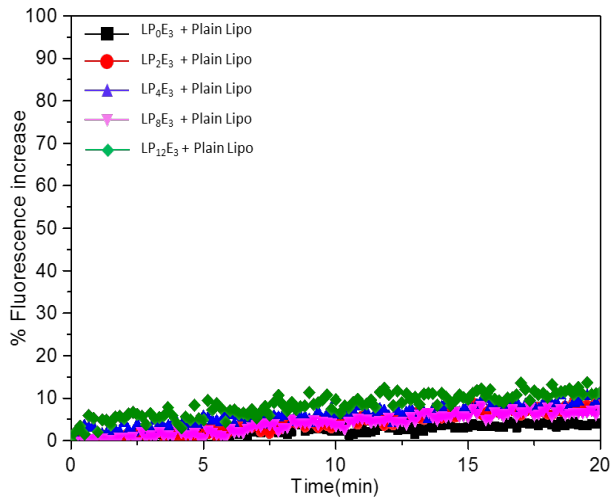
Where  $\vartheta_{obs}$  is the ellipticity in millidegrees, MRW is the mean residue molecular weight (i.e. the molecular weight of the peptide divided by the number of amino acids residues),  $l$  is the path length of the cuvette in cm and  $c$  is the peptide concentration in mg/mL. From this mean residue molar ellipticity, the ellipticity ratio for the peptide could be calculated, using equation (5)

$$Ellipticity\ Ratio = [\vartheta]_{224} / [\vartheta]_{211} \quad (5)$$

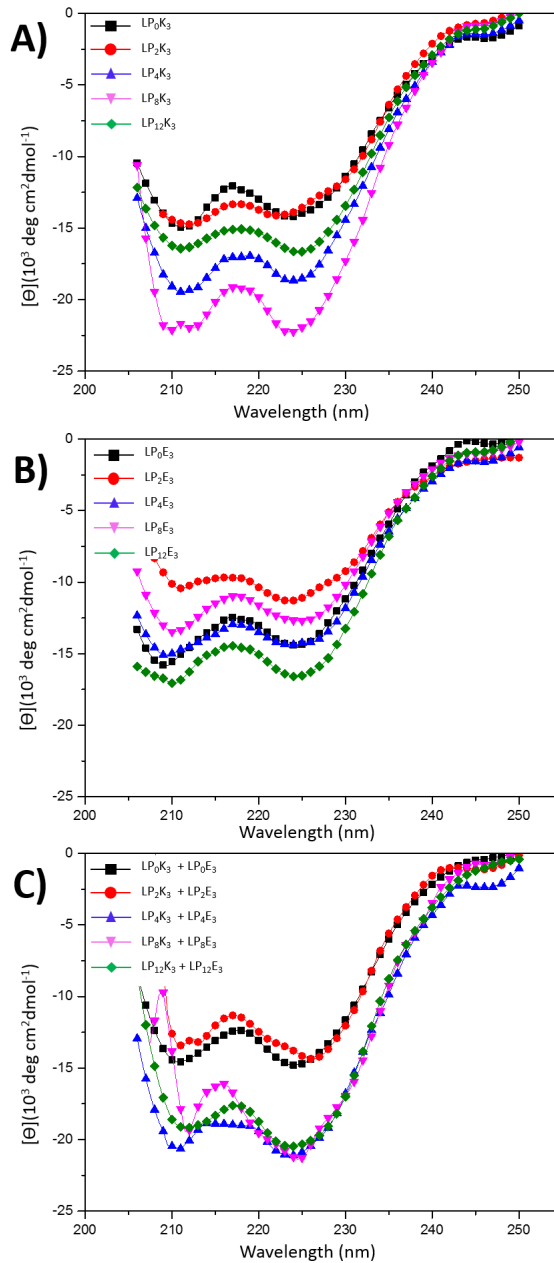
## Control Experiment for Lipid mixing:



**Figure A2.** Lipid mixing between  $K$  decorated liposomes and plain liposomes, as indicated by an increase in NBD fluorescence. Non-fluorescent liposomes ( $0.1 \text{ mM}$ ) were added to fluorescent  $K$  liposomes ( $0.1 \text{ mM}$ ,  $1\% \text{ LP}_n K_3$ ).



**Figure A3:** Content mixing between  $\text{LP}_n E_3$  and  $\text{LP}_n K_3$  modified liposomes ( $0.1 \text{ mM}$  total lipid concentration,  $1 \text{ mol}\% \text{ LP}_n E_3$  peptides). To test no specific leakage during fusion  $\text{LP}_n E_3$  decorated liposomes ( $0.1 \text{ mM}$  total lipid concentration,  $20 \text{ mM}$  sulphorhodamine B) were added to the plain liposomes ( $0.1 \text{ mM}$  total lipid concentration).



**Figure A4:** Circular Dichroism spectra of A)  $LP_nK_3$  liposomes B)  $LP_nE_3$  liposomes and C) (1:1) mixture of  $LP_nK_3$  liposomes and  $LP_nE_3$  liposomes where the baseline was corrected with plain liposomes of 100nm in size. 0.5 mM liposomes in PBS were decorated with either 1 mol%  $LP_nK_3$  or  $LP_nE_3$ .

## References

- [1] a) L. V. Chernomordik, M. M. Kozlov, *Nat Struct Mol Biol* **2008**, *15*, 675-683; b) P. I. Hanson, J. E. Heuser, R. Jahn, *Curr Opin Neurobiol* **1997**, *7*, 310-315.
- [2] A. Richard, V. Marchi-Artzner, M. N. Lalloz, M. J. Brienne, F. Artzner, T. Gulik-Krzywicki, M. A. Guedeau-Boudeville, J. M. Lehn, *Proc Natl Acad Sci U S A* **2004**, *101*, 15279-15284.
- [3] a) U. Kauscher, B. J. Ravoo, *Beilstein J Org Chem* **2012**, *8*, 1543-1551; b) J. Voskuhl, M. C. A. Stuart, B. J. Ravoo, *Chem-Eur J* **2010**, *16*, 2790-2796.
- [4] J. Voskuhl, T. Fenske, M. C. Stuart, B. Wibbeling, C. Schmuck, B. J. Ravoo, *Chemistry* **2010**, *16*, 8300-8306.
- [5] M. Rauschenberg, S. Bomke, U. Karst, B. J. Ravoo, *Angew Chem Int Edit* **2010**, *49*, 7340-7345.
- [6] F. Nomura, T. Inaba, S. Ishikawa, M. Nagata, S. Takahashi, H. Hotani, K. Takiguchi, *Proc Natl Acad Sci U S A* **2004**, *101*, 3420-3425.
- [7] M. M. Ma, A. Paredes, D. Bong, *J Am Chem Soc* **2008**, *130*, 14456-+.
- [8] a) A. Kashiwada, M. Tsuboi, K. Matsuda, *Chem Commun (Camb)* **2009**, 695-697; b) A. Kashiwada, M. Tsuboi, T. Mizuno, T. Nagasaki, K. Matsuda, *Soft Matter* **2009**, *5*, 4719-4725.
- [9] a) S. K. M. Nalluri, B. J. Ravoo, *Angew Chem Int Edit* **2010**, *49*, 5371-5374; b) S. K. M. Nalluri, J. B. Bultema, E. J. Boekema, B. J. Ravoo, *Chem-Eur J* **2011**, *17*, 10297-10303.
- [10] a) T. C. Sudhof, J. E. Rothman, *Science* **2009**, *323*, 474-477; b) G. van den Bogaart, R. Jahn, *Proceedings of the National Academy of Sciences* **2011**, *108*, 11729-11730.
- [11] S. Martens, H. T. McMahon, *Nat Rev Mol Cell Biol* **2008**, *9*, 543-556.
- [12] H. R. Marsden, I. Tomatsu, A. Kros, *Chem Soc Rev* **2011**, *40*, 1572-1585.
- [13] a) A. Kashiwada, K. Matsuda, T. Mizuno, T. Tanaka, *Chem-Eur J* **2008**, *14*, 7343-7350; b) R. A. Parente, S. Nir, F. C. Szoka, *J Biol Chem* **1988**, *263*, 4724-4730.
- [14] a) G. Stengel, R. Zahn, F. Hook, *J Am Chem Soc* **2007**, *129*, 9584-9585; b) G. Stengel, L. Simonsson, R. A. Campbell, F. Hook, *J Phys Chem B* **2008**, *112*, 8264-8274; c) Y. H. Chan, B. van Lengerich, S. G. Boxer, *Biointerphases* **2008**, *3*, FA17; d) Y. H. Chan, B. van Lengerich, S. G. Boxer, *Proc Natl Acad Sci U S A* **2009**, *106*, 979-984.
- [15] A. S. Lygina, K. Meyenberg, R. Jahn, U. Diederichsen, *Angew Chem Int Edit* **2011**, *50*, 8597-8601.
- [16] a) Y. Gong, Y. Luo, D. Bong, *J Am Chem Soc* **2006**, *128*, 14430-14431; b) Y. Gong, M. Ma, Y. Luo, D. Bong, *J Am Chem Soc* **2008**, *130*, 6196-6205.
- [17] H. Robson Marsden, N. A. Elbers, P. H. Bomans, N. A. Sommerdijk, A. Kros, *Angew Chem Int Ed Engl* **2009**, *48*, 2330-2333.
- [18] K. Meyenberg, A. S. Lygina, G. van den Bogaart, R. Jahn, U. Diederichsen, *Chem Commun (Camb)* **2011**, *47*, 9405-9407.
- [19] J. A. McNew, T. Weber, D. M. Engelman, T. H. Sollner, J. E. Rothman, *Mol Cell* **1999**, *4*, 415-421.
- [20] J. V. TingTing Zheng, Frank Versluis, Harshal Zope, Itsuro Tomatsu, Hana Robson Marsden, Alexander Kros *Chem Commun (Camb)* **2013**.
- [21] F. Versluis, J. Voskuhl, B. van Kolck, H. Zope, M. Bremmer, T. Albrecht, A. Kros, *J Am Chem Soc* **2013**, *135*, 8057-8062.
- [22] D. K. Struck, D. Hoekstra, R. E. Pagano, *Biochemistry* **1981**, *20*, 4093-4099.
- [23] M. L. Rachida El Jastimi, *Biospectroscopy* **1999**, *5*, 133-140.
- [24] a. T. W. James A. McNew, a Francesco Parlati, a Robert J. Johnston, a Thomas J. Melia, a Thomas H. Söllner, a and James E. Rothman, *J Cell Biol.* **2000**, *150*, 105-118.
- [25] G. W. Vandermeulen, D. Hinderberger, H. Xu, S. S. Sheiko, G. Jeschke, H. A. Klok, *Chemphyschem* **2004**, *5*, 488-494.
- [26] a) P. S. Shirude, V. A. Kumar, K. N. Ganesh, *Eur J Org Chem* **2005**, 5207-5215; b) J. T. t. Lundquist, J. C. Pelletier, *Org Lett* **2001**, *3*, 781-783; c) J. Voskuhl, C. Wendeln, F. Versluis, E. C. Fritz, O. Roling, H. Zope, C. Schulz, S. Rinnen, H. F. Arlinghaus, B. J. Ravoo, A. Kros, *Angew Chem Int Ed Engl* **2012**, *51*, 12616-12620.
- [27] a) F. Versluis, J. Dominguez, J. Voskuhl, A. Kros, *Faraday Discuss* **2013**, *166*, 349-359; b) H. Robson Marsden, A. V. Korobko, T. Zheng, J. Voskuhl, A. Kros, *Biomaterials Science* **2013**, *1*, 1046-1054.



