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On the geometry of demixing: A study of lipid phase separation on curved surfaces

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Appendix

No.	Name	Sequence
1	10 nm Base	Cholesterol-TEG-3'-GTT-AGC-CCG-ATT-ACA-GAG-CGT- -TCT-TT-3'
2	10 nm Inert	Cholesterol-TEG-5'-TTT-GAA-CGC-TCT-GTA-ATC-GGG- -CTA-AC-3'
3	20 nm Base	Cholesterol-TEG-3'-TTT-TAG-CGA-TGG-GAA-GCG-TGT- -CAG-TTA-GAT-CTC-TCG-GGA-CGG-AAT-GC-5'
4	20 nm Inert	Cholesterol-TEG-5'-TTT-ATC-GCT-ACC-CTT-CGC-ACA- -GTC-AAT-CTA-GAG-AGC-CCT-GCC-TTA-CGA- 3'
5	20 nm Single Linker A	Cholesterol-TEG-5'-TTT-ATC-GCT-ACC-CTT-CGC-ACA- -GTC-AAT-CTA-GAG-AGC-CCT-GCC-TTA-CGA- <i>GTA-GAA-</i> <i>GTA-GG-3'-6FAM</i>
6	20 nm Single Linker A'	Cholesterol-TEG-5'-TTT-ATC-GCT-ACC-CTT-CGC-ACA- -GTC-AAT-CTA-GAG-AGC-CCT-GCC-TTA-CGA- <i>CCT-ACT-</i> <i>TCT-AC-3'-Cy3</i>
7	30 nm Base	5'-TCG-TAA-GGC-AGG-GCT-CTC-TAG-ACA-GGG-CTC-TCT- -GAA-TGT-GAC-TGT-GCG-AAG-GTG-ACT-GTG-CGA-AGG- -GTA-GCG-ATT-TT-3'
8	30 nm Single Linker A	Double Stearyl-HEG-5'-TT-TAT-CGC-TAC-CCT-TCG- -CAC-AGT-CAC-CTT-CGC-ACA-GTC-ACA-TTC-AGA-GAG- -CCC-TGT-CTA-GAG-AGC-CCT-GCC-TTA-CGA- <i>GTA-GAA-</i> <i>GTA-GG-3'-6FAM</i>
9	30 nm Single Linker A'	Double Stearyl-HEG-5'-TT-TAT-CGC-TAC-CCT-TCG- -CAC-AGT-CAC-CTT-CGC-ACA-GTC-ACA-TTC-AGA-GAG- -CCC-TGT-CTA-GAG-AGC-CCT-GCC-TTA-CGA- <i>CCT-ACT-</i> <i>TCT-AC-3'-Cy3</i>

Table A.1: Summary of all DNA strand sequences and their names. Sticky ends are marked in cursive.

Sphere	Cube	Symm. dumbbell	Asymm. dumbbell
18 $\pm$ 3	11 $\pm$ 1	13 $\pm$ 1	13 $\pm$ 2
26 $\pm$ 2	22 $\pm$ 2	23 $\pm$ 5	32 $\pm$ 2
35 $\pm$ 3	28 $\pm$ 3	50 $\pm$ 2	
45 $\pm$ 3	41 $\pm$ 2	57 $\pm$ 4	
52 $\pm$ 5	52 $\pm$ 2	67 $\pm$ 2	
58 $\pm$ 2	60 $\pm$ 2	80 $\pm$ 2	
69 $\pm$ 2	72 $\pm$ 2	89 $\pm$ 1	
80 $\pm$ 8	80 $\pm$ 1		

Table A.2: Area fractions in percentage of liquid disordered phase of the CSLBs shown in Chapter 3, Figure 3.6.

BSM [%]	POPC [%]	chol [%]
30	50	20
35	45	20
40	35	25
50	20	30
50	25	25

Table A.3: Lipid mixtures allowing for phase separation.

Lipid composition. In supported lipid bilayers, the substrate can affect the physical properties of phase separation, such as the temperature at which phase separation occurs⁷¹. Therefore, we varied the lipid composition on spherical supports to identify the conditions under which phase separation occurs. We found that phase separation can be obtained with mixtures of porcine brain sphingomyelin (BSM), 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine (POPC) and cholesterol (chol) in mole ratios equal to the ones reported on Table A.3.

No phase separation was observed for 0 to 10% of cholesterol on spherical CSLBs. We used the mixture 2:1:1 BSM:POPC:chol. This ratio has also been shown to phase separate in free-standing bilayers both at 23°C and 37°C, albeit with a different type of sphingomyelin, the palmitoylsphingomyelin PSM¹⁴⁰. We confirmed that for our mixture phase separation also occurs in free standing bilayers (giant unilamellar vesicles) made via electro-swelling (see Figure A.1).

Surface properties analysis. To exclude that surface properties of the supporting colloidal particles, such as the roughness or the type of silica, affect the phase separation landscape, we compared dumbbell shaped CSLBs made on two different types of dumbbell-shaped substrates: silica dumbbells made via destabilisation of colloidal silica spheres and silica-coated dumbbells made from PS-TPM-Si particles. We studied the likelihood of phase separation and of number of domains. We observed that these quantities are similar

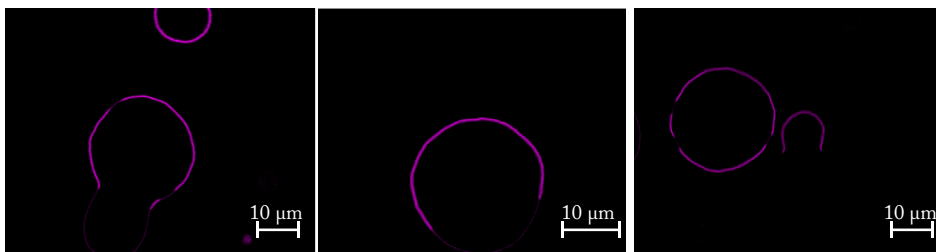


Figure A.1: Fluorescence images of phase-separated giant unilamellar vesicles. Lipid composition 2:1:1 of SM:POPC:chol. The liquid disordered phase is labelled in magenta with DOPE-Rhodamine.

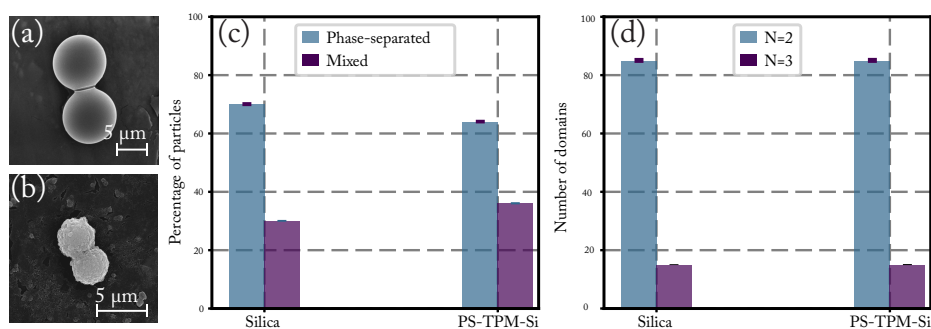


Figure A.2: Analysis of the surface effect. (a) Scanning electron microscopy (SEM) image of a silica dumbbell made via destabilisation of silica spheres. (b) SEM image of a PS-TPM-Si dumbbell. (c) Percentage of phase separation of dumbbell shaped CSLBs made of aggregated silica spheres (left) and PS-TPM-Si dumbbells (right). (d) Percentage of number of coexistent liquid ordered and disordered domains of the same CSLBs as before.

for the two types of particles, indicating that the properties of the surfaces of the supports that we use do not significantly influence phase separation properties (Figure A.2).

Details on the numerical simulations of Chapter 3. Here, we explain how we constructed the 3D geometries corresponding to the experimental CSLBs and how we implemented the numerical solver to find equilibrium configurations relative to Equation 3.4 and 3.5, which were essential to produce Figure 3.8 and 3.9.

Gradient Flows

The general strategy to find the equilibria of phase-field models is to implement an evolution equation that flows an arbitrary configuration smoothly towards energy minima. Extending previous work¹⁶⁸ we choose to implement a gradient flow with conserved global order parameter:

$$\partial_t \phi = \epsilon \nabla^2 \phi - \frac{1}{\epsilon} f'(\phi) - k'(\phi) H^2 - \bar{k}'(\phi) K + \lambda, \quad (\text{A.1})$$

where ∇^2 is the Laplace-Beltrami operator on S and $\phi = \phi(x^i, t)$ has been promoted to a function of both space and flow parameter t . The right hand side consists just of (minus) the first functional derivative of the total free energy with a term involving the conservation of the total concentration. When minimising Equation 3.5 one has to replace $k(\phi) \rightarrow k(\phi)/\epsilon$ and $\bar{k}(\phi) \rightarrow \bar{k}(\phi)/\epsilon$ in the dynamical equation.

The flow generated by Equation A.1 is fictitious and does not reflect the actual coarsening dynamics of the binary fluid. This approach offers nonetheless a practical way to generate stable equilibrium configurations for arbitrary geometries. Note that t has dimension of a length.

The numerical integration of Equation A.1 is done via a local Runge-Kutta-Fehlberg adaptive time-step algorithm²²⁰, where the criterion for acceptance of a given time-step is averaged over S .

Metastable equilibrium is defined as the fixed point of the right hand side of Equation A.1, *i.e.* when $\partial_t \phi$ vanishes everywhere. Numerically, we assume that the simulation has reached equilibrium when the surface-averaged squared time-derivative

$$\langle (\partial_t \phi)^2 \rangle_S = \frac{1}{A_S} \int_S dA (\partial_t \phi)^2, \quad (\text{A.2})$$

decreases below a given threshold (usually of the order of 10^{-6} for surfaces of unit area).

The initial conditions, at $t = 0$, are either random ϕ values distributed around a specific Φ value (for Equation 3.5) or a random set of small domains with constant $\phi = \phi_*$ surrounded by a $1 - \phi_*$ background (for Equation 3.4 and to generate the phase diagram of Figure 3.8a).

Mesh construction

We solve Equation A.1 numerically using a finite difference scheme on unstructured triangular meshes. Meshes are constructed using the software package *Gmsh*¹⁹⁰. As in the case of planar droplets on the plane, the rotational symmetry of the substrate is not necessarily inherited by the minimisers of the Gibbs free energy, thus it is often necessary to solve the full two-dimensional problem. For the case of dumbbells and snowmen, the

radial profile has been obtained by joining two circular arcs by an interpolating polynomial of degree eight, chosen such that the neck interpolation and the circular arc match smoothly up to the fourth derivative at each of the two gluing points.

Due to the arbitrariness of the geometry of S , our implementation of the solver is entirely coordinate-independent. Computation of discrete Laplace-Beltrami weights and of the mean curvature H on each node is done by using the cotangent method. Since the ideal surface is smooth and differentiable at least twice, we use the numerically computed H as a criterion for the quality of the triangulation: if H appears to be reasonably smooth on every portion of S we accept the mesh, otherwise we refine it.

For every triangulation, the length scale in a simulation is fixed by the requirement that the area of S has to match the dimensions in μm^2 of the observed CSLBs. Furthermore, the value of ϵ is automatically computed for each triangulation and is fixed by the requirement that the interface should be resolved anywhere on the membrane by at least six grid points.

Equilibrium equation in non-homogeneous potentials

If the free energy depends explicitly on the position through the non-constancy of curvatures, one can divide S in regions - each labelled by an index i - where both H and K can be regarded as approximately constant so that

$$F = \sum_i \varphi_i F_i(\phi_i), \quad (\text{A.3})$$

where the sum runs over the partition of S , φ_i is the area fraction occupied by the subregion i (so that $\sum_i \varphi_i = 1$). The variable ϕ_i is the restriction of the field ϕ to the region i . If S is a smooth surface, it is always possible to find partitions that satisfy the above requirement to an arbitrary degree of accuracy. The condition for equilibrium is obtained by differentiating G in each subdomain:

$$\frac{dF_i}{d\phi_i} = \mu, \quad (\text{A.4})$$

provided the average concentration

$$\sum_i \varphi_i \phi_i = \Phi, \quad (\text{A.5})$$

is kept fixed. Equation A.4 implies that, at equilibrium, the derivative of each free energy has to take the same value over all of S . This should not be confused with the usual requirement that, at equilibrium, two coexisting phases always have the same value of the chemical potential: here the free energies are position-dependent functions and Equation A.4 refers to the equilibrium condition of a single, mixed phase in different locations in space. Note that now, for generic curvature dependence, the mixed phase $\phi_i = \Phi$ is not a solution of Equation A.4. If some, or all, F_i happen to be concave, then one should check whether the spinodally decomposed solutions can become locally favoured. We found the solutions of Equation A.4 when S consists of the disjoint union of two spheres to produce the equilibrium phase diagrams of Figure 3.9b-d.

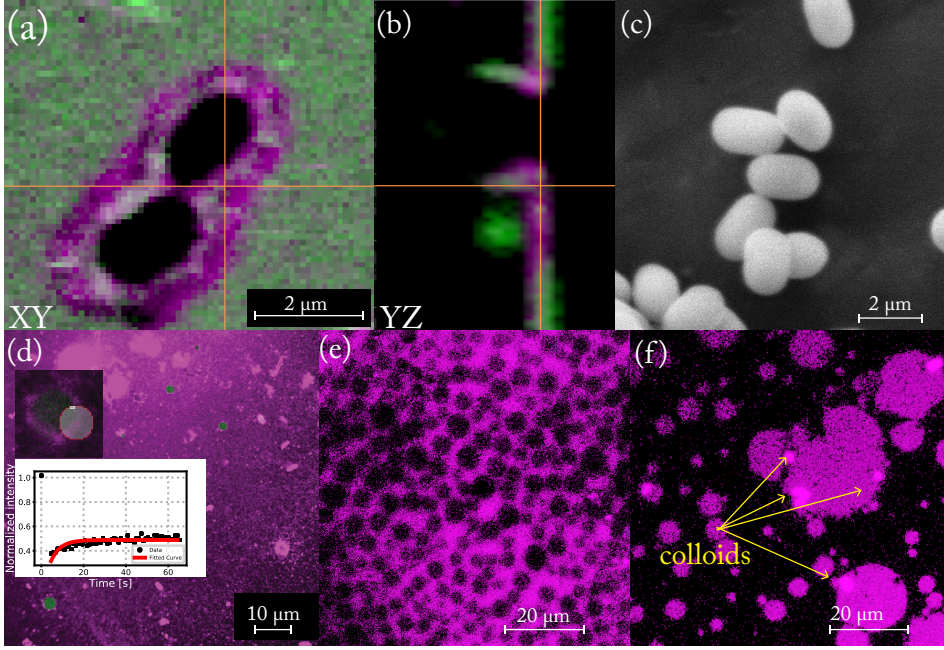


Figure A.3: Phase separation on a flat substrate. Fluorescence microscopy image of a lipid bilayer on a cluster of two colloidal spherocylinders silica shells obtained from a synthesis procedure analogous to the one of the colloidal cubic shells. The projections of the bilayer on the (a) XY and (b) YZ planes are reported to show the localisation of the LD phase at the contact area between the colloids and the flat substrate. (c) SEM image of the spherocylinders. (d) Fluorescence microscopy image of a lipid bilayer on a flat substrate without colloids taken five days after lipid coating. Circular LO domains are surrounded by LD domains. In the inset, we report the result of a FRAP experiment on the bleached area indicated with a red circle. We can observe that the fraction of mobile lipids decreased over the five days. (e) Fluorescence image of a phase-separated bilayer on MICA substrate, where only the LD phase was labelled in magenta. We can observe circular domains. (f) Fluorescence image of a phase-separated bilayer on a MICA substrate with colloidal particles attached. The particles are indicated in yellow. By comparing the two figures, we can hypothesise that the colloids may alter the regular pattern of circular domains shown in (e).

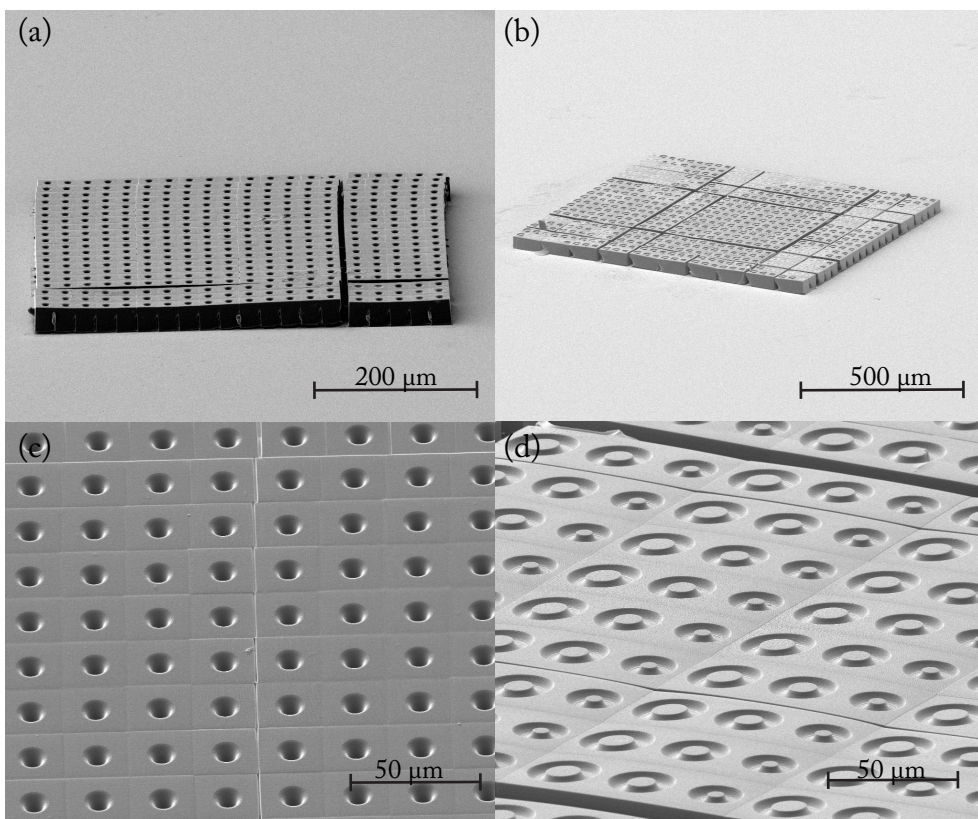


Figure A.4: SEM images of micro-printed molds. (a)-(c) Two views of molds of inverted Gaussian bumps. **(b)-(d)** Two views of molds of inverted torii.

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