

Laboratory studies of water ice in space : optical and photochemical properties

Kofman, V.

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Laboratory Studies of Water Ice in Space – Optical and Photochemical Properties –

Vincent Kofman

- 1. Simultaneously studying astronomical ice analogues in the UV-vis and IR holds great potential in characterizing the photochemical processing of these ices. *Chapters 2 and 5*
- 2. The availability of temperature and wavelength dependent values of the refractive index of water ice will help in characterizing interstellar environments. *Chapter 3*
- 3. Triphenylene's D_{3h} symmetry and its fully benzenoid character makes this PAH aesthetically very attractive, as well as a molecule of astrophysical interest. *Chapter 4*
- 4. Studying photochemical reactions both in a rare gas matrix and a water environment allows one to separate primary and secondary photochemical reactions. *Chapters 2 and 5*
- 5. When only relying on infrared spectroscopy to study astronomical ices, the sensitivity limit constrains the depth of understanding one will be able to reach.
- 6. Understanding of ice in space requires a balanced contribution from both observational and laboratory studies, and the latter should aim to work within the limits of the first.
- 7. Using UV-vis spectroscopy and the newly derived optical constants to quantify ice thicknesses will result in more accurate infrared band strengths.
- 8. Despite its deceptive familiarity, the physical and chemical characteristics of water are exceedingly complex.
- 9. Both in science and in life, one should focus on what one *has*, not on what is missing.
- 10. One very important thing our educational system fails to teach is selfconfidence.
- 11. Yoshikawa: One should not confuse education with intelligence.