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## Materials and energy : a story of linkages

Kleijn, E.G.M.

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**Author:** Kleijn, René

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## Chapter 3

# Resource constraints in a hydrogen economy based on renewable energy sources: an exploration.

### Abstract

In order to tackle climate change, a transition to a renewable based energy system is crucial. A renewable based hydrogen economy is one of the possible implementations of such a system. The world receives ample energy from the sun that can be harvested by PV solar cells and, indirectly, by wind turbines. In order to use the most optimal locations for collecting and concentrating energy from these diffuse sources a long distance transmission network is needed. Mature and semi-mature technologies are available for all parts of the system: from collection to transmission to end-use. In an early stage the development, when new technologies have to win market share from the existing energy system, their development is driven almost exclusively by the reduction of costs per J delivered. However, if a technology should be able to deliver tens to hundreds of EJ, resource constraints can become show-stoppers. Many of the newest, most cost-efficient, energy technologies make use scarce resources and, although they may play an important role in the transition process, they cannot be scaled up the level we need for a complete transition. In most cases however other technologies are available that use more abundant materials, be it often at a cost of efficiency. The issue is not only with scarce resources. The sheer size of the energy transition will also challenge the industrial capacity for the mining and production of bulk materials like steel and copper.

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### 3.1 Introduction

The world is preparing for a transition from a fossil fuels based energy system to a system of renewable energy. One important option being discussed is the so-called hydrogen economy (European Commission 2003; The White House 2004; U.S. Department of Energy 2004; Okano 2003): an energy system relying on hydrogen as its main energy carrier, instead of fossil fuels. This envisaged transition is partly driven by concerns about the environmental impact, especially global warming, connected to the current energy system, and partly by arguments related to security of energy supply.

Large-scale transitions cause many changes in society. It is a good idea, therefore, to assess beforehand what the consequences might be of such a transition. These consequences can be various. A first question is: does it actually solve the problem it is supposed to solve? Although a hydrogen economy is generally propagated as a "clean"

alternative for the current fossil fuel based economy, it is the question how "clean" this hydrogen economy truly may be. Even if renewable energy sources would be used to produce hydrogen, a number of environmental issues would remain whilst additional new problems might occur. In a hydrogen economy, hydrogen and water are emitted. The emission of molecular hydrogen has been the topic of a debate on the possible effects of these emissions, specifically on stratospheric ozone and on climate change (Tromp et al. 2003; Schultz et al. 2003; Prather 2003).

A second question relates to the feasibility of such a new system. Can it actually be built and implemented? When starting to envisage what a hydrogen economy would look like, a first step is to conceive a technological description of such a system. The production, transport and use of hydrogen as a fuel requires a whole series of new processes, materials, products and infrastructure which all will require resources and will have an environmental impact of their own. The same is true for the actual generation of energy, which in a renewable energy system will be done by other means than combusting fossil fuels, for example by using solar cells, wind power, biomass etc. There is a high risk of running into resource constraints of one sort or another, simply because of the stunningly large scale of such a transition.

In this article, we will address this last issue, as a very important one. The central question in this paper is the following: Will a transition to a hydrogen economy based on renewable energy sources run into resource constraints? In order to answer this question we try to create a picture of a world with hydrogen and electricity as the main energy carriers and renewables as the only energy source. We will, as well as possible, quantify the main parts of such a system, and we will assess whether or not resource constraints can be expected. Although it is clear to us that a world without fossil fuels is an extreme case (Odell 2004). and there will be many hurdles on the way towards such a world, the dynamics of the transition process itself will not be addressed in this exploratory work. Also, we will not address economic and social consequences or the difficulties of actually realizing the transition politically or legally. It will be a technical description, targeted at the issue of potential "show stoppers" in a technical sense.

### **3.2 Core assumptions in the case study**

For the technical specification of a renewables based hydrogen economy, we have defined the following starting points:

- the case describes a hypothetical energy system, roughly in the year 2050; the path towards this situation is not a part of the study;
- basic data like population, GDP and energy demand in 2050 are taken from the UNEP Markets First Scenario, see Table 3-1 (UNEP 2002);
- although in the Markets First Scenario the bulk of the primary energy still comes from fossil fuels, we describe a purely hypothetical situation in which all energy comes from renewable sources;
- we assume that the role of biomass based energy will be limited;

- we assume that the renewables mix for primary energy will be 80% PV solar, 15% wind and 5% other (renewable) sources;
- electricity and hydrogen are considered the only two energy carriers.

These starting points are elaborated below.

*Table 3-1: Key variables of the IPCC SRES Markets First Scenario*

	1990	2050	unit	multiplier
population	5302	9386	million	1.8
GDP	25293	191767	billion US \$ (1995)	7.6
primary energy use	356	1278	EJ	3.6

*Source: (UNEP 2002).*

### 3.2.1 Base year & scenario

The choice for the year 2050 is rather arbitrary as a year which is sufficiently far in the future to envisage a novel energy system while not too distant to still have reasonable estimates of basic data like world population and GDP. Furthermore, if we seriously want to address climate change the energy system needs to change dramatically within this time frame. The Markets First Scenario was chosen because (1) it is most in line with current world economic and demographic developments and (2) it results in a relatively high future energy demand which makes it a worst case. The Markets First scenario envisages a world in which market-driven developments converge on the values and expectations that prevail in industrialized countries (UNEP 2002). World integration, globalization, free trade and reliance on market-based approaches are the main directions for development. GDP/capita and energy consumption/capita both grow rapidly in this scenario between now and 2050.

### 3.2.2 Renewables only

In the Markets First Scenario, the total primary energy production in 2050 will be about 1300 EJ, more than 2.5 times the primary energy use in 2005 (about 514 EJ) (see Table 3-1). In almost all current projections of future energy use fossil fuels will still be the dominant source of energy in the coming decades (IEA 2007; Shell 2008). However, in our extreme case study we assume that all energy is generated from renewable sources. In the Markets First scenario, where globalization and free trade rule, the primary energy mix will be dependent on local availability and world market prices. If one looks at the potential availability of the different renewable sources (Table 3-2) it is clear that the bulk of the primary energy in the Markets First scenario will have to come from wind and solar power.

## Chapter 3

*Table 3-2: Availability of renewable energy (excluding biomass and extraterrestrial collection)*

	W	J/a	multiplier for 1990 energy use ( $3.56 \cdot 10^{20}$ J)	multiplier for 2050 energy use in MF scenario ( $1.28 \cdot 10^{21}$ J)	source
<b>Solar</b>					
solar energy reaching Earth surface	$2.6 \cdot 10^{16}$	$8.2 \cdot 10^{23}$	2303	642	(Sørensen 2000)
collectable in Sahara desert with current PV efficiency (10%)	$2.1 \cdot 10^{14}$	$6.5 \cdot 10^{21}$	18	5	Calculated
<b>Wind</b>					
total kinetic energy in Earth atmosphere	$3.5 \cdot 10^{15}$	$1.1 \cdot 10^{23}$	310	86	(Leijendeckers et al. 2008)
total in bottom 100 meter	$1.3 \cdot 10^{15}$	$4.0 \cdot 10^{22}$	113	31	(Leijendeckers et al. 2008)
maximum practically realizable	$5.7 \cdot 10^{12}$	$1.8 \cdot 10^{20}$	0.51	0.14	(Williams 2001)
technical potential	$1.5 \cdot 10^{13}$	$4.8 \cdot 10^{20}$	1.3	0.37	(Hoogwijk 2004)
<b>Hydro</b>					
on basis of precipitation on land	$1.0 \cdot 10^{13}$	$3.0 \cdot 10^{20}$	0.8	0.2	(Sørensen 2000)
principally available	$1.2 \cdot 10^{12}$	$3.8 \cdot 10^{19}$	0.11	0.030	(Sørensen 2000)
realistic reserve hydro	$6.3 \cdot 10^{11}$	$2.0 \cdot 10^{19}$	0.055	0.015	(Sørensen 2000)
<b>Tidal</b>					
total tidal energy	$3.0 \cdot 10^{12}$	$9.5 \cdot 10^{19}$	0.27	0.074	(Sørensen 2000)
tidal world potential at best sites	$1.2 \cdot 10^{11}$	$3.8 \cdot 10^{18}$	0.011	0.0030	(Sørensen 2000)
<b>Wave</b>					
total wave energy	$1.1 \cdot 10^{11}$	$3.6 \cdot 10^{18}$	0.010	0.0028	(Sørensen 2000)
<b>Geothermal</b>					
total stored on Earth		$4.0 \cdot 10^{30}$	$1.1 \cdot 10^{10}$	$3.1 \cdot 10^9$	(Sørensen 2000)
total heat outflow	$3.0 \cdot 10^{13}$	$9.5 \cdot 10^{20}$	2.7	0.7	(Sørensen 2000)

### 3.2.3 Large scale biomass is not included

Next to wind and solar sources, biomass is the only source that could in theory provide a substantial contribution to the world's energy supply in 2050. The estimates of the potential contribution of biomass to the total world energy supply vary from 0 to over 1000 EJ/a of which over 90% would have to come from high quality agricultural land (Hoogwijk et al. 2003). In a detailed and regionalized study Hoogwijk

concludes that the contribution of biomass to the fuel production in 2050 could reach a level equal to the present oil demand (180 EJ/a) (Hoogwijk 2004). The largest part of the biomass would have to come from the former USSR, East Asia, Africa and South America, assuming a large increase in agricultural productivity in these areas. Others (Doornbosch and Steenblik 2007) dispute this and state that the required increase in biomass production of an order of magnitude is impossible in view of land and water constraints. By-and-by it is becoming clear that biomass can provide only a limited amount of the world's energy. We consider competition with nature and food production for the limited land area that can be used for the production of biomass<sup>1</sup> such a problem that we exclude biomass as a significant source of renewable energy. This in contrast to solar energy which can be collected in areas where biodiversity is minimal and agriculture is not possible, and to wind turbines which can either be used off shore or combined with agriculture. Moreover, biomass production dedicated for energy purposes is less efficient from an energy perspective than PV solar cells<sup>2</sup>. The use of waste biomass and residues as a source of energy is less problematic. However, the contribution to the total primary energy production will probably be relatively small: 30 -108 EJ/a (Hoogwijk 2004).

### **3.2.4 Wind and solar provide 95% of energy**

Wind may provide about 180 EJ /a (Williams 2001) or about 15% of the 2050 total energy demand, hydropower up to 3% (Sørensen 2000). All other renewable energy sources will be used in addition in places where they are readily available but will probably not provide more than 1% of the world's total primary energy use<sup>3</sup>. In order to simplify the analysis at hand we will focus on the 95% that can be provided via wind power (15%) and solar radiation (80%) leaving out all other sources.

### **3.2.5 Hydrogen and electricity as carriers**

We will focus on two main carriers for the future energy system: electricity and hydrogen. Current renewable energy sources (except biomass) directly produce electricity. Since any energy conversion is associated with a loss of exergy, conversions are limited as much as possible in an optimal energy system. Electricity is an excellent

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<sup>1</sup> The production algae and kelp does not require arable land but is related to problems of its own. For example in order to have substantial production levels of kelp nutrient rich water would have to be pumped towards the kelp fields which would require energy and lower the overall efficiency dramatically (Bungay 2004).

<sup>2</sup> The energy need for the production of Rape Methyl Ester is for example 16 GJ ha<sup>-1</sup> a<sup>-1</sup> while it delivers an equivalent of 43 GJ of fossil energy (Kaltschmitt and Reinhardt 1997). With poplar clones a net energy collection of 260 GJ ha<sup>-1</sup> a<sup>-1</sup> has been reported (European Commission 2005). This is equivalent to an efficiency of respectively 0.08 % and 0.8 % (biomass out (HHV)/ solar in) while current PV systems have an efficiency between 6-15 % (Green et al. 2006).

<sup>3</sup> At the moment the way in which society uses energy is quite wasteful. High exergy energy sources like fossil fuels are often used in applications that actually only require low-exergy energy. An example is heating of buildings. Although some heating is done by using waste heat from industry, low temperature geothermal sources and passive solar energy, the contribution of these sources is still very limited, mainly because the use of high-exergy energy is relatively cheap and requires little investments because existing infrastructure can be used. Based on the current trends and the Markets First scenario, we explore the scenario in which low-exergy energy sources will remain of marginal importance.

energy carrier for stationary applications but it is difficult to store in large quantities. This is problematic in a renewable based energy system because the availability of renewable sources is highly variable through weather conditions, seasonal changes and day and night rhythms. It also implies that electricity is not very useful in mobile applications that require more than a few Watt, like transportation. Only recently battery technology has improved in such a way that the first full electric vehicles are being brought to consumer market. It is not clear which technology, hydrogen fuel cells or full electric cars will be dominant. Hydrogen is still seen by many as the preferable energy carrier for the use in mobile applications. In our scenario for 2050 we assume hydrogen will be used as the energy carrier in high power mobile applications (mainly in transport).

### **3.3 System specification**

The key figures from the Markets First scenario are combined with data from literature on the availability of renewable energy sources and data on relevant energy technologies. We have used this information to generate a rough but more or less complete and quantitative overview of what a world based on renewable energy sources and hydrogen as an energy carrier could look like in 2050. The entire energy "life-cycle" will be taken into account: from the energy collection, via the transmission and distribution of this energy, to the utilization of this energy by the end-users. The infrastructure of the renewable based hydrogen economy is described and the material flows connected to this infrastructure are quantified over the next sections. On the basis of this quantified system description, the bottlenecks with regard to resource availability will be identified.

A general overview of our hydrogen energy system based on renewable energy sources is given in Figures 3-1 and 3-2. Renewable energy is collected and concentrated, preferably in locations where conditions are optimal for a given source, transmitted to the centres of economic activity and distributed to the end users. Both electricity (Figure 1) and hydrogen (Figure 2) can be used as a carrier for long range transmission.



## Resource constraints H<sub>2</sub>-economy

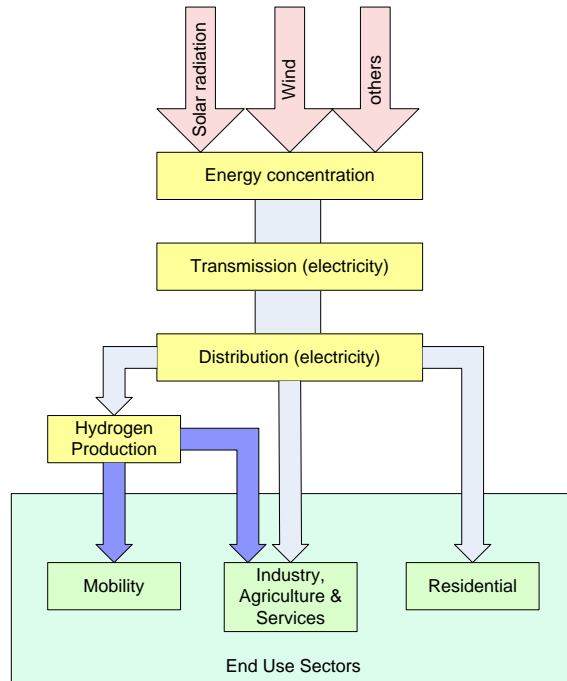


Figure 3-1: Main structure of a renewable based hydrogen energy system with electricity as the carrier in transmission.

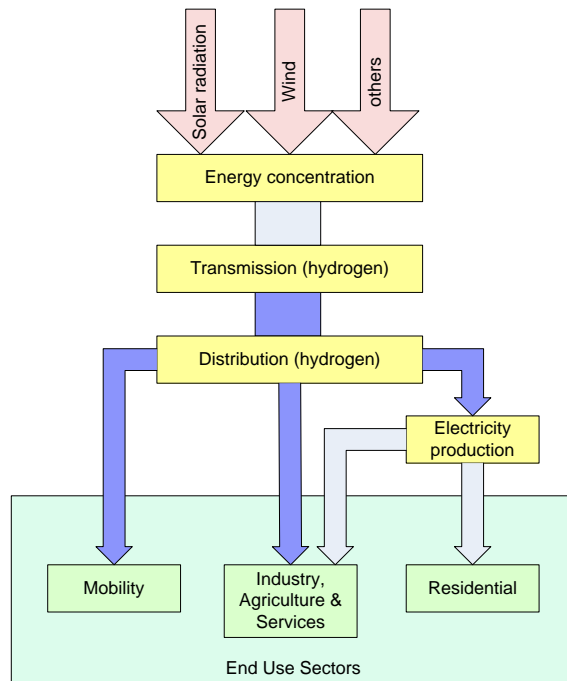


Figure 3-2: Main structure of a renewable based hydrogen energy system with hydrogen as the carrier in transmission.

### 3.3.1 Collection and concentration

In order to collect substantial amounts of useful energy from wind and solar radiation, large surface areas are needed, preferably in places with optimal conditions for collection. These optimal locations are often not the same as the locations where the energy is used. For example, the best places to collect solar energy, deserts, are often located far from the economic and industrial centers of the world (Knies and Bennouna 2002; Czisch 2004). Therefore, an extensive long distance transmission and distribution infrastructure is needed to get the energy to the end-users. Next to that, solar and wind energy are intermittent by nature which increases the importance of a long distance transmission network in order to buffer the system for temporal local shortages (Seboldt 2004).

### 3.3.2 Hydrogen production

Hydrogen can be produced via two routes:

1. via electricity in a two step process: solar and wind energy are used to produce electricity, which in a next step is used for the electrolysis of water to form hydrogen.
2. directly from water and solar energy via photolysis.

The second route has two main advantages: an energy transformation step would be eliminated, thereby increasing the efficiency of energy system as a whole, and two expensive and complex pieces of equipment, PV cells and electrolysis cells, would be replaced by one piece of equipment, the photolysis cell. However, the technology still suffers from severe practical problems and needs to be developed further<sup>4</sup>.

Next to this photochemical photolysis, a biological route exists. Algae or bacteria or their enzymes (hydrogenases) are used in light-driven bioreactors to split water (Melis and Happe 2001). However, this route is also still in the phase of early laboratory scale experiments. Because both the direct photolytic route and the biological route are still in the early stages of development, it is hard to say anything about their practical potential or link them to an estimate of resource requirements. We will therefore focus on PV coupled to electrolysis for hydrogen production.

Electrolysis of water can be done with different processes:

- Alkaline electrolyzers (commercial large scale units)
- Polymer Electrolyte Membrane (PEM) electrolyzers (prototypes and commercial units)
- Solid Oxide electrolyzers (lab-stage and commercial units)

All three of these options will be analysed although the main focus will be on alkaline electrolyzers.

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<sup>4</sup> This technique was first demonstrated by (Fujishima and Honda 1972). Problems like the lack of efficient light absorption and the corrosion of the semiconductor have been identified by (Khaselev and Turner 1998). In order to avoid these problems, tandem cells are now being developed in which a photo-electrolysis cell is put in series with a solar cell which allows the use of semiconductors that are more stable in water like tungsten oxide. However, the published efficiencies of these tandem cells are at the moment as low as 4.5% (Grätzel 2001).

### 3.3.3 Transmission

Both hydrogen and electricity can be used for long range transmission of energy, and both options will be explored in this study. The technology to transport electricity over long distance via high voltage AC power grid is already operational today. Transmission losses depend on the transport distances. In the 1990s the UK the average losses, excluding transformation and distribution, were 1.5 to 2% (Oxera 2003). In Table 3-3 an overview is given of the losses in different types of transmission: overhead AC, HVDC and hydrogen pipelines. High Voltage DC (HVDC) lines are today used to transport about 2% of all electricity in around 90 projects all over the world (International Energy Agency 2008). Trans-ocean HVDC cables that have been used to transport electricity from Scandinavia to other parts of Europe have losses of under 0.4% per 100 km. However, a 1-2% loss should be added for the AC/DC/AC conversion (International Energy Agency 2008; Sørensen 2007) so, from an energy efficiency perspective, this technique is only preferred for distances over 300 km. From a viewpoint of investment costs, DC transmission becomes favourable over AC lines over 50 km<sup>5</sup>. In our case study we therefore assume HVDC for long distance transmission. HVDC cables are either bipolar or monopolar. A bipolar cable has a better performance (Basslink 2002) and will therefore be assumed in our case study. Conductors in HVDC lines are traditionally made from copper although aluminium conductors are nowadays sometimes used in light weight, low capacity cables that are applied for shorter distances.

Table 3-3: A comparison of transmission losses between AC, HVDC and H<sub>2</sub> pipelines

	overhead AC	HVDC	H <sub>2</sub> pipeline
energy losses / 100 km(%)	1	0.4	0.8
AC/DC conversion losses	0	1-2%	NA
% losses for 1500 km	15	8	12
% losses for 3000 km	30	14	24
<i>Source</i>	<i>(Sorensen 2000)</i>	<i>(International Energy Agency 2008)</i>	<i>(Weindorf et al. 2003)</i>

Hydrogen can be transported as a gas via pipelines or as a liquid in cryogenic dewar vessels on ships, trucks or trains. Since hydrogen is widely used in oil refining and chemical industry, a commercial system of hydrogen distribution is already in place for the use of hydrogen in oil refining and in the chemical industry (e.g. (Zittel and Altmann 1996)). When hydrogen is transported via pipelines specifically designed for hydrogen transport, losses can be very low. Overall losses from the pipelines in Germany are about 0.1% per year (Zittel and Altmann 1996). The pumping of a gas

<sup>5</sup> Other advantages of DC lines are (Wright et al. 2002): (1) it is possible to link two asynchronous power grids (i.e. the frequency at either end can differ); (2) DC lines, unlike AC lines, do not have the drawback of so-called reactive current which can substantially reduce the power throughput. This problem does not occur in overhead lines; (3) the direction and the magnitude of the power flow can be controlled; and (4) a given cable can carry more power in DC than in AC.

through a pipeline requires energy. In natural gas pipelines about 0.2% of the transported amount of gas is used for compressors every 100 km at a rate of 10 m/s (Bossel et al. 2003). Because hydrogen contains roughly a factor three less energy per unit of volume than natural gas, more energy is needed for compression when transporting hydrogen through pipelines: about 0.8 % every 100 km (Bossel et al. 2003; International Energy Agency 2008). In Table 3-3 transmission via electricity is compared with hydrogen pipelines. For a transmission distance of 1500 km losses from H<sub>2</sub> pipelines are 1.5 times as high as those from HVDC cables. If the distances increase the relative difference increases as well. However, by optimizing the hydrogen pipelines (larger diameter, improved gas turbines for compression etc.) losses could be reduced to less than 10% for 3000 km (Weindorf et al. 2003; Ball et al. 2009).

Because hydrogen molecules are very small hydrogen can easily diffuse through most materials and seals and it can cause severe degradation of steels. Therefore, pipelines used for transporting hydrogen must be made of high quality non-porous materials such as stainless steel (Ball et al. 2009). Hydrogen can be added to steel natural gas pipelines in mixtures up to 30 vol % (10% in energy terms) (Ball et al. 2009; Blazek et al. 1992). A problem for transport through steel pipes is the so-called hydrogen embrittlement. This phenomenon is not completely understood, but it refers to cracks and blisters in the metal due to hydrogen permeation in the steel, which result in leakages. To avoid embrittlement pipes can be coated, small quantities of CO, SO<sub>2</sub> or O<sub>2</sub> can be added or embrittlement-resistant steels can be used. Transport of hydrogen through plastic pipes is also problematic because hydrogen permeates through plastic four to six times as fast as natural gas (Ball et al. 2009; Ogden 1999) .

Liquefaction of hydrogen can be applied by cooling hydrogen to extremely low temperatures (below -253 °C). The energy content of one litre of liquefied hydrogen is much higher than the energy content of a litre of compressed gaseous hydrogen. The main disadvantages are that the energy costs are high and that the losses during handling and transport are relatively high too. One of the reasons for these losses is that boil-off, evaporation of hydrogen due to temperature changes, is inevitable. However, the hydrogen that escapes in this way can, in some cases, be captured again and used for the propulsion of the vehicle in which the transport takes place. Although the transport of liquid hydrogen is technically feasible, the costs, both in terms of energy as in terms of money, will be so high that it will probably only be used under very specific circumstances and it will not compete with pipeline transport. Pipelines are the preferred transport option for large quantities and long distances while liquid hydrogen is preferred for smaller volumes over long distances (Ball et al. 2009). Therefore, in our case we assume pipelines for long distance bulk transport of hydrogen.

### 3.3.4 End Use

Although electricity and hydrogen are interchangeable as energy carriers in distribution, this is not the case in end-use. As transport fuel, hydrogen can either be combusted like gasoline and kerosene, or used in fuel cells. Fuel cells convert

hydrogen into electricity which is used to power the electric motors that are used to make the wheels turn and propel the vehicles. Even fuel cell cars that are filled up with hydrogen will thus ultimately be powered by electricity. In cars, the dominant hydrogen based technology is fuel cell cars. Prototypes of internal combustion engine (ICE) based hydrogen cars exist (e.g. BMW), but this should be seen purely as an intermediate technology that will help the transition to a hydrogen fuel cell vehicles (Wietschel et al. 2009). In aviation, the dominating propulsion technology is jet engines. Since these make direct use of the exhaust gases to propel the plane, combustion of hydrogen would be the preferred option here (Faaß 2001). Consumer electronics that need a lot of power, like refrigerators, washing machines, TV's and stereos, will remain grid connected and powered by electricity. Smaller electronics like laptops and mobile phones will either be battery powered or powered by small fuel cells, that in turn produce the electricity that powers these devices.

### 3.4 Identification of possible resource constraints

To identify possible resource constraints of the hydrogen economy, a quantification will be attempted with regard to the most important elements of the system described above:

- energy collection via PV cells and wind turbines
- hydrogen production via electrolysis
- long range electricity and hydrogen transmission
- end-use: fuel cells & electric motors

#### 3.4.1 Resource constraints related to energy collection

##### *PV Solar Cells*

If 80% of the primary energy in 2050 comes from solar, this means that around 1000 EJ will be produced via solar energy. Part of this could come from built-up area. The IEA (IEA 2002) concludes that in selected developed countries between 30 and 120 % of the electricity use in that year could come rooftops. Here we assume a theoretical maximum of about 15% of primary energy in 2050 (around 200 EJ) that could come from built-up area<sup>6</sup>. Even with this maximized amount of PV from built-up area this would mean that 65% of the primary energy would have to come from other surfaces. This implies that land will be used that is currently used for agriculture, or that is currently not used by humans. In (sub)tropical desert areas on average about 7.2 GJ/m<sup>2</sup> reaches Earth surface every year (Leijendeckers et al. 2008). Assuming a 10% efficiency, an area of 1.0 million km<sup>2</sup> would be needed to produce 845 EJ/a (65% of 1300 EJ/a, see earlier). This area is equivalent to about 10% of the Sahara desert area, divided over the different continents. At a latitude of 50° (Vancouver , Paris) the average annual incoming solar energy on ground level is around 3.6 GJ/m<sup>2</sup>

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<sup>6</sup> This assumption is based on data on built-up area in different countries (WRI online database land use), the assumption that 20% of this area would be suited for PV collection, an average incoming solar radiation of about 110 W/m<sup>2</sup> and a PV efficiency of 10%.

(Leijendeckers et al. 2008). Based on this irradiation an area of 2.0 million km<sup>2</sup> would be needed to supply the remaining 65% of the total energy use, equivalent to about four times area of France. If, for example, OECD Europe would want to be self-sufficient in energy terms, it should be able to collect 100 EJ/a in 2050 (IMAGE team 2001). In order to achieve this, 4% of the total land area (or 10% of the agricultural area) would have to be re-allocated to collect solar energy, in addition to a 15% use of wind energy and a 15% use of solar cells on built-up land, and the assumed 5% other renewable sources.

At the moment, different solar cells are available. There are different types of silicon cells: monocrystalline, polycrystalline and amorphous. All three forms of silicon are produced from sand of which supply is abundant. The only other materials in the cells themselves are dopants, mostly phosphorus and boron. For both materials the application in electronics is only a very small fraction of the total production so no constraints are in sight (USGS 2009b). Amorphous Si cells are cheap but have low efficiency, while crystalline Si cells have high efficiencies but they are also expensive and energy intensive to produce (Wadia et al. 2009). Therefore new thin-film PV cells have been developed that can be produced at lower costs (Wadia et al. 2009; Fthenakis 2009). At the moment the main thin-film technologies are cadmium telluride (CdTe) and copper indium gallium diselenide (CIGS) (Fthenakis 2009). Typical efficiencies of different cells are given in Table 3-4. Since these thin film cells use a combination of less common materials, the supply of these materials can be an issue of concern once production reaches a substantial level.

*Table 3-4: Efficiency of PV cells*

	<b>record cell efficiency (%)</b>	<b>record module efficiency (%)</b>
Si (crystalline)	24.7	22.7
Si (multicrystalline)	20.3	15.3
Si (thin film)	16.6	8.2
Si (amorphous)	9.5	N/A
CIGS (cell)	18.4	13.4
CdTe	16.5	10.7
concentrator Si (parabolic disc)	26.8	N/A
concentrator GaInP/GaAs/Ge (2-terminal)	34.7	N/A

*Source: (Green et al. 2006)*

During the last decade several studies have addressed the issue of material constraints of thin-film PV (Wadia et al. 2009; Fthenakis 2009; Andersson et al. 1998; Andersson 2001; Andersson and Jacobsson 2000). The material needs of the four main non-silicon solar cells have been calculated by (Andersson and Jacobsson 2000). On the basis of these numbers and our scenario we calculated resources constraints for these technologies which are presented in Table 3-5. It is apparent that current reserves would not allow a substantial contribution of these cells to the solar energy collection we need in our scenario for 2050. All four technologies will be limited to a maximum

of less than 1% (10 EJ/a) of the total required energy collection. In more recent and more detailed work the limits for current thin film technologies (CdTe and CIGS) are estimated at around 10-100 EJ/a. However there are large differences in the material constraints for different technologies. Some, more unconventional, thin film PV technologies (especially FeS<sub>2</sub>) look very promising from the perspectives of material constraints and can be scaled up to the level of over thousand EJ/a (Wadia et al. 2009; Fthenakis 2009). Considering these material constraints<sup>7</sup> it seems that the use of current thin film cells based on rare materials will be limited. In a world based largely based on solar energy, the bulk of the solar cells would thus have to be based on new thin film materials or amorphous silicon. Next to the PV cells themselves the production of PV modules requires additional materials, these are not quantified in this study.

Table 3-5: Material constraints of different elements for four thin film solar cell technologies

cell type	element	require- ment g/m <sup>2</sup>	reserve 1998 (Gg)	reserve 2012 (Gg)	production 1997 (Gg/a)	produc- tion 2010 (Gg/a)	needed for 1.7 10 <sup>6</sup> km <sup>2</sup> (Gg)
CdTe							
	Cd	6.3	600	640	20	21	10000
	Te	6.5	20	24	0.29	NA	11000
CIGS							
	Se	4.8	70	93	2.2	2.1	7900
	Ga	0.53	110	NA	0.054	0.18	880
	In	2.9	2.6	NA	0.2	0.61	4800
aSiGe							
	Ge	0.44	2	NA	0.063	0.12	730
dye-sensitised							
	Ru	0.1	6	NA	0.011	NA	170

Sources: (Andersson et al. 1998; Andersson 2001; Andersson and Jacobsson 2000). The 1.7 10<sup>6</sup> km<sup>2</sup> is based on 15% of the total energy demand from roofs on higher latitudes (50°) (0.7 10<sup>6</sup> km<sup>2</sup>) and 65% PV cells in desert areas (1.0 10<sup>6</sup> km<sup>2</sup>).

### Wind

As stated before wind may provide about 180 EJ or about 15% of the 2050 total energy demand. This would mean that about 195 EJ/a will be produced via wind

<sup>7</sup> Another issue with technologies using rare materials is, that the energy pay-back time as well as environmental impacts are relatively high because of the energy-intensive mining and refining processes. It was calculated, based on current technologies, that the full life cycle energy payback time is 6.3 years for an indium gallium phosphide (InGaP) module and only 3.5 years for a mono-crystalline Si module (Meijer et al. 2003). They also estimated that the environmental impact of the InGaP module was 80% higher than the impact of a mono-crystalline Si module. This reinforces the conclusion that these types of solar cells cannot contribute more than a small fraction to worldwide energy generation.

turbines. In January 2008, 94 GWe of wind power was installed, delivering about 1.4% of global electricity demand (International Energy Agency 2008). Thus in order to produce 190 EJ in 2050 the installed capacity has to increase by a factor of 250<sup>8</sup>, resulting in a capacity of 24 TWe. Present turbines have a capacity of around 1-2 MWe on land and up to 3-5 MWe for offshore sites. However, the size of wind turbines has been growing rapidly: from 0.5 MW in 1992 to 4.5 MW in 2002 (Kuik 2001). If we assume an average capacity of 2 MWe, 12 million of these large turbines would have to be installed.

An indication of the material use in wind turbines is given in Table 3-6. The amount of copper needed for these turbines is between 2 tons/MWe (onshore) and 3 tons/MWe (offshore) per turbine. This means that for the total 24 TWe, more than 50 million ton copper is needed. This is equivalent to about 7% of current reserve base<sup>9</sup> and more than 4 times the worldwide mine production in 2008. The use of iron and steel is around 140 ton/MWe and thus more than 3 billion ton for the total 24 TWe. This is equivalent to around 2% of the base reserve of iron ore and, in terms of iron content of the ore, about 3 times the current iron ore production.

Table 3-6: Material use wind turbines

	2MW offshore	800 kW onshore	2MW offshore	800 kW onshore	% reserve base 24 TWe with 2 MW offshore	% reserve base 24 TWe with 800kW onshore	multiplier of world annual mine production 2008	
	ton/MWe	ton/MWe	Mton / 24 TWe	Mton / 24 TWe			2MW offshore	800kW onshore
steel	148	135	3551	3244	2.2	2.0	3.0	2.7
aluminium	0.42	0.26	10	6				
copper	2.75	1.83	66	44	6.6	4.4	4.2	2.8
lead	3.79	0	91	0				
plastics	2.09	1.92	50	46				
concrete	1050	306	25100	7340			1.3	0.38
gravel	150	0	3600	0				
neodymium	0.15		3.6				180	

Source: (Frischknecht R. 2007); Neodymium (Polinder et al. 2006; USGS 2009a), reserves & production USGS Reserve Base is defined by the USGS as "That part of an identified resource that meets specified minimum physical and chemical criteria related to current mining and production practices, including those for grade, quality, thickness, and depth. The reserve base is the in place demonstrated (measured plus indicated) resource from which reserves are estimated." (USGS 2009a)

Next to the base metals which are used in wind turbines more exotic materials are sometimes used as well. Neodymium is a rare earth metal which is used in permanent

<sup>8</sup> This may seem like a very high number but at an annual growth rate of 14% would be enough to achieve this in 2050. The last ten years, 1999 to 2008, the annual growth rate in installed wind capacity has been around 25% (International Energy Agency 2008).

<sup>9</sup> As defined by the USGS: base reserve: "That part of an identified resource that meets specified minimum physical and chemical criteria related to current mining and production practices, including those for grade, quality, thickness, and depth."



magnets. Neodymium magnets are preferred above other permanent magnets because they can provide the same magnetic strength with smaller and lighter magnets. Using permanent Nd magnets in wind turbines can slightly increase their efficiency, reduce maintenance costs and reduce the weight of the generator, all because in a they can make gearboxes redundant. In state of the art wind turbines with direct-drive permanent-magnet generators about 550 kg permanent magnets or 150 kg of neodymium is used per MWe (Polinder et al. 2006; Hatch 2008). This would mean that around 3.6 million ton Nd would be needed for the total 24 TWe in our scenario. This is about 180 times the current annual mine production. At the moment very few wind turbines actually use direct-drive with permanent magnets and most use electromagnets in geared generators which are made of copper and iron. Neodymium constraints will therefore not be a show stopper for large scale application of wind turbines in general but it will limit the market share of certain new types of wind turbines. Even more so because the turbine producers would have to compete with car manufacturers (see under 4.4) and with other renewable energy sources in which permanent magnets are used (Hatch 2008).

The amount of concrete needed to build a wind turbine is between 300 ton/MWe (onshore) and 1000 ton/MWe (offshore) which means that for 24 TWe between 7 and 25 billion ton of concrete needs to be produced. If the average cement content of concrete is set at 15% the amount of cement needed to produce this concrete is between 40% and 130% of the annual world cement production in 2008 (USGS 2009a). For cement, the base reserve is not very relevant: although there can be regional shortages of the raw materials, like limestone and sand, are geologically abundant and widespread. However, there may be environmental constraints: cement production is an important source of CO<sub>2</sub> (Kleijn et al. 2008).

The conclusion that can be drawn is that if wind energy is to contribute 15% to the global primary energy use in 2050, this would mean that a substantial amount of the base reserve of copper and iron ore needs to be invested in this sector. The worldwide production of copper, iron ore and cement would have to increase dramatically. If wind turbines are to be scaled up to the level of TWhs it does not seem likely that neodymium containing direct-drive permanent-magnet generators will play a significant role because of resource constraints related to Neodymium.

### **3.4.2 Resource constraints related to hydrogen production**

Electrolysers based on the alkaline process have been in commercial use for decades. PEM electrolysis cells and solid oxide based electrolysers have recently been commercially introduced. In this analysis we will therefore focus on alkaline electrolysis. The electrolyte is an alkaline solution, usually potassium or sodium hydroxide, and the electrodes are made of different coated metals. The anode is often made of nickel and copper and coated with oxides of metals and the cathode is made of platinum coated nickel. A theoretical maximum efficiency of 83% is given by

(Sørensen 2000). In practice the efficiency based on LHV<sup>10</sup> is between 50 and 75 % (Ogden 1999; IEA 2005; Koroneos et al. 2004). Here we assume a future efficiency for the production of hydrogen from water of 75%. This means that if hydrogen is used as the energy carrier in transmission, the energy collection and the connected material flows and land use will have to be 25% higher than calculated above.

In electrolysis, hydrogen is produced from fresh water. This is problematic in many places that would be ideal for PV solar power plants. In desert-like areas that are close to oceans, like the Western part of the Sahara, one could overcome this problem by combining PV plants with desalination installations. Although desalination has become much more efficient over recent years it still requires a significant energy input which would decrease the overall efficiency of hydrogen production.

Current electrolyzers are made from stainless steel, plastics and the materials needed for membranes and electrodes and the electrolyte. There is little quantitative information on the material use in the membranes and electrodes in the public domain. The electrolyte is normally potassium hydroxide or sodium hydroxide. Both of them are bulk chemicals and the use in this application would still be small compared the total annual use. The membranes are made of organic polymers, such as nafion and polybenzimidazole, and are not related to material constraints either. The electrodes are made of metals: the anode is usually a combination of nickel and copper coated with a layer of metal oxides. The anode is made of nickel coated with a layer of platinum (Koroneos et al. 2004). Although we do not have much specific data on the quantity of the metals in electrolyzers we assume here that they will be in the same range as that of fuel cells. This would imply potential resource constraints with regard to Platinum (see further in this text, under “Resource constraints related to hydrogen use: Fuel Cells”). Technological innovations have a potential to solve these material issues. Recently a study was published on a promising new technique that uses a solution of much more abundant cobalt and phosphate as the main active components<sup>11</sup> (Kanan and Nocera 2008).

It can be concluded that material constraints of hydrogen production via electrolysis using alkalines can be expected related to platinum. PEM electrolyzers also use platinum as a catalysts and these would run into similar constraints (see 4.4). For solid oxide this is presently unclear, as they are not developed fully yet, but resource constraints are not necessarily absent.

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<sup>10</sup> The LHV assumes that the latent heat of vaporization of the water that is produced during the electrolysis is not recovered. For hydrogen the Higher Heating Value is about 18% higher than the lower heating value.

<sup>11</sup> The authors use a solution with cobalt and phosphate ions in aqueous solution at neutral pH. In the experiments a newly formed catalyst is deposited on the electrode consisting of both cobalt and phosphate. This catalyst is highly active and will regenerate spontaneously. In the experiment an indium tin oxide (ITO) catalyst was used which by including indium has its own resource scarcity issues. However, this material was itself did not seem to be taking part the electrolysis mechanism.

### 3.4.3 Resource constraints related to transmission

In May 2008 the longest HVDC cable became operational and connects the electricity grids of Norway and the Netherlands. Specifications of this cable are given in Table 3-7. The capacity of the cable about 600 MW and it has a copper conductor of 760 mm<sup>2</sup>. Total transmission losses are estimated at 4% with full load (Tennet 2007; Statnett 2002). In a scenario where solar energy is collected in remote desert areas power cables would be needed that are about a factor 10 longer than the longest operational cables today. The main materials used in these cables are copper, steel, lead and polyethylene. In Table 3-8 the material needs for transmission of 65% of the primary energy in 2050 through HVDV cables is calculated. The copper use in the 600 MW NorNed cable is about 7.5 kg/m. With increased conductor size the maximum power through this types of cables today can be several 1000 MW (ABB 2009). If it is assumed that the conductor size would have to double when doubling the power throughput. In a 1200 MW cable the copper use would also be doubled to 14 kg/m, in a bipolar cable this would again be doubled to 28 kg/m. If 65 % of the primary energy use in 2050 would be transmitted over an average distance of 1500 km, 22 thousand 1200 MW (bipolar) cables would be needed. In order to produce these cables 900 million ton copper would be needed. This is equivalent to about 90 % of current reserve base and twice the cumulative production between 1900 and 2001 (Porter and Edelstein 2003), 24% of the total copper resource and almost 60 times the current annual world production. The long term availability of copper has been discussed in literature (Gordon et al. 2006; Tilton and Lagos 2007). Although copper is not a rare element, still resource constraints are in sight if electricity transmission has to take place in these amounts over these large distances.

*Table 3-7: Details of the HVDC power cable between Norway and the Netherlands*

conductor size	2 x 760 mm <sup>2</sup> (bipolar)
conductor material	Copper
weight	85 kg/m (bipolar)
outer diameter	120 mm
rated power	600 MW
rated voltage	450 kV
length	580 km
maximum water depth	410 m

*Source: (Tennet 2007; Statnett 2002)*

If 65% of the primary energy use in 2050 would be delivered by hydrogen through stainless steel pipelines (current technology) this would require substantial amounts of nickel and chromium. The exact specifications of pipelines may vary substantially. An estimate is given in Table 3-9. The starting point of the calculations was that 600 MW<sub>H<sub>2</sub></sub> stainless steel (API 303/304/316) pipelines would be used. The number of pipelines would then be identical to the number of HVDC cables in the case of electricity transmission. Based on typical wall thickness and diameter from (Ball et al. 2009) and typical chromium and nickel content in stainless steel this result a demand for Ni of about 45 times the current annual mine production and for Cr of about 5500 times the current annual mine production. The amount of nickel needed would be

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about halve the current reserve base. The USGS does not give global reserves for chromium but the amount is about a third of the base reserve of countries that are included in their data.

*Table 3-8: Calculation copper use in transmission if 65% of the world primary energy supply in 2050 (845 EJ) would come from PV solar plants in the deserts, assuming transmission with HVDC power cables..*

<b>Data on HVDC cables</b>			
	<b>unit</b>	<b>amount</b>	<b>source</b>
power through existing cable	W	6.00 10 <sup>8</sup>	(Tennet 2007; Statnett 2002)
conductor cross section for 600 MW	mm <sup>2</sup>	760	(Statnett 2002)
maximum power per cable	W	1.20 10 <sup>9</sup>	(ABB 2009)
conductor cross section for 1200 MW	mm <sup>2</sup>	1520	assumption: double power, double conductor cross section
volume copper per m cable	cm <sup>3</sup>	3200	calculated
density conductor (copper)	g/cm <sup>3</sup>	8.9	(CRC 2009)
mass copper	g/m cable	13500	calculated
<b>Calculation cables needed</b>			
total energy use 2050	J/a	1.3 10 <sup>21</sup>	assumption in this study
total energy use 2050	W	4.1 10 <sup>13</sup>	calculated
65% from deserts	W	2.6 10 <sup>13</sup>	calculated
number of cables needed	-	21951	calculated
average distance to end-users	km	1500	assumption in this study
km cable needed	km	3.3 10 <sup>7</sup>	calculated
<b>Calculation total copper use for transmission</b>			
total copper use monopolar	kg	4.5 10 <sup>11</sup>	calculated
total copper use if bipolar	kg	8.9 10 <sup>11</sup>	calculated
<b>Comparison to copper production &amp; base reserve</b>			
base reserve (Cu)	Kg	1 10 <sup>12</sup>	(USGS 2009a)
resources (land and ocean)	Kg	3.7 10 <sup>12</sup>	(USGS 2009a)
mine production	Kg/a	1.6 10 <sup>10</sup>	(USGS 2009a)
cumulative world production 1900-2001	Kg	4.2 10 <sup>11</sup>	(Porter and Edelman 2003)
multiplier copper use transmission vs base reserve	-	0.89	calculated
multiplier copper use transmission vs resources	-	0.24	calculated
multiplier copper use transmission vs mine production	-	57	calculated
multiplier copper use transmission vs cumulative mine production (1900-2001)	-	2.1	calculated

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*Table 3-9: Calculation nickel, chromium and iron use in transmission if 65% of the world primary energy supply in 2050 (845 EJ) would come from PV solar plants in the deserts, assuming transmission with hydrogen pipelines.*

<b>Data on hydrogen pipelines</b>			
	<b>unit</b>	<b>amount</b>	<b>source</b>
power through pipeline (600 MW_H2)	W	6.0 10 <sup>8</sup>	(Ball et al. 2009)
pipe diameter	cm	25	(Ball et al. 2009)
wall thickness	cm	1.0	average API 10 pipe
volume steel per m pipe	cm <sup>3</sup>	2500	calculated (The Physics Factbook 2009)
density stainless steel	g/cm <sup>3</sup>	8	
mass steel	kg/m pipe	20	calculated
fraction Ni in stainless steel	-	0.11	average 304,306,316
fraction Cr in stainless steel	-	0.18	average 304,306,316
fraction Fe in stainless steel	-	0.71	remaining fraction
mass Ni	kg/km pipe	2200	calculated
mass Cr	kg/km pipe	3600	calculated
mass Fe	kg/km pipe	14200	calculated
<b>Calculation pipelines needed</b>			
total energy use 2050	J/a	1.3 10 <sup>21</sup>	assumption in this study
total energy use 2050	W	4.1 10 <sup>13</sup>	calculated
65% in centralised scenario from deserts	W	2.6 10 <sup>13</sup>	calculated
number of pipelines needed	-	22000	calculated
average distance to end-users	km	1500	assumption in this study
km pipeline needed	km	3.3 10 <sup>7</sup>	calculated
<b>Calculation total Ni and Cr use for transmission</b>			
total Ni use	kg	7.2 10 <sup>10</sup>	calculated
total Cr use	kg	1.2 10 <sup>11</sup>	calculated
total Fe use	kg	4.7 10 <sup>11</sup>	calculated
<b>Comparison to production &amp; base reserves</b>			
base reserve (Ni)	kg	1.50 10 <sup>11</sup>	(USGS 2009a)
mine production (Ni)	kg/a	1.61 10 <sup>9</sup>	(USGS 2009a)
			(USGS 2009a) total based on main producers
base reserve (Cr)	kg	3.74 10 <sup>11</sup>	
mine production (Cr)	kg/a	2.15 10 <sup>7</sup>	(USGS 2009a)
base reserve (Fe)	kg	1.6 10 <sup>14</sup>	(USGS 2009a)
mine production (Fe)	kg/a	1.2 10 <sup>12</sup>	(USGS 2009a)
multiplier Ni use pipelines / base reserve	-	0.48	calculated
multiplier Ni use pipelines / mine production	-	45	calculated
multiplier Cr use pipelines / base reserve	-	0.32	calculated
multiplier Cr use pipelines / mine production	-	5500	calculated
multiplier Fe use pipelines / base reserve	-	0.0029	calculated
multiplier Fe use pipelines / mine production	-	0.39	calculated

From the analysis above it is clear that, using current technologies, the use of copper in HVDC lines would be prohibit the use of HVDC transmission in our extreme scenario. The use of the much more abundant aluminium conductors might solve this issue but it would also decrease the overall efficiency of the transmission. Also the use of current technology stainless steel for the transport of hydrogen cannot be scaled up to the level that we would need in our scenario due to material constraints of chromium and nickel. Other steels or coated steels might be the best alternative here or possibly steel could be avoided completely by the use of new composite materials.

#### **3.4.4 Resource constraints related to end use: fuel cells & electric motors**

##### *Fuel cells in cars*

In 2001 there were 555 million cars in the world (Worldwatch Institute 2002). In the Markets First scenario, GDP per capita will grow by a factor of 3.7 from 2001 to 2050. It is assumed that the number of cars will grow with the same factor. This would mean that there will be roughly 2 billion cars on the road by 2050. We assume that all these cars will have a PEM fuel cell on board. Rade & Andersson (Råde and Andersson 2001) use in their baseline scenario, based on current technologies, that in PEM fuel cells on average 0.39 g of Pt is needed per kW. With 2 billion cars on the road in 2050 this would mean that 39,000 ton of Pt would be needed. Since the platinum catalyst is an expensive part of the fuel cell, R&D will be aimed at reducing the amount of platinum needed in fuel cell vehicles. The UK Department for Transport reports an ultimate goal of about 3 g per car which would mean a demand for Platinum of about 6,000 ton in 2050 (UK Department for Transport 2003). Currently platinum group metals (PGM) are used as catalysts. The USGS estimates the reserve base at 80,000 ton and world resources at over 100,000 ton (USGS 2009a). World production in 2008 was 200 ton Platinum and 206 ton Palladium (USGS 2009a). The cumulative production of PGM from 1900 to 2000 is 9120 ton (Kelly and Hilliard 2003). In Table 3-10 the amount of platinum in fuel cells is calculated. If the goals for the reduction of Pt in fuels cells are met less than 10% of the base reserve is needed. However, using current technologies the amount of Pt in cars in 2050 would exceed the world resources. In any case the production rates would have to grow dramatically to fulfil the demand.

The problem of Pt use might be solved in future by using alternative catalysts which do not contain rare metals. The two main alternatives are the use of base metals and the use of enzymes. However, for hydrogen fuelled fuel cells there are no known (combinations of) base metals that can be practical substitutes for PGM metals yet. The enzyme route is more promising although a practical enzyme based biofuel cell is (Armstrong et al. 2009) also still in the early stages of fundamental research. Currently technologies are being developed to adsorb enzymes like hydrogenase on electrodes which brings the biofuel cell a bit closer to practical application (Lamle et al. 2003).

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Table 3-10: Calculation of Pt and Nd use in fuel cell cars in 2050.

		unit	source
number of cars on the road in 2050	2.0 10 <sup>9</sup>	-	Assumption based on GDP growth
Pt per car (high)	60	g/car	(UK Department for Transport 2003)
Pt per car (low)	20	g/car	(Råde and Andersson 2001)
Pt per car (minimum)	3	g/car	(UK Department for Transport 2003)
Pt for all cars in 2050 (high)	120000	ton	calculated
Pt for all cars in 2050 (low)	47000	ton	calculated
Pt for all cars 2050 (minimum)	6000	ton	calculated
world production Pt 2007	220	ton	(USGS 2009b)
multiplier (high)	530	-	calculated
multiplier (low)	220	-	calculated
multiplier (minimum)	28	-	calculated
Nd per car (low)	1	kg/car	(Gorman 2009) uncertain !
Nd per car (high)	2	kg/car	(Gorman 2009) uncertain !
Nd for all cars in 2050 (low)	2000	kton	calculated
Nd for all cars in 2050 (high)	4000	kton	calculated
world production Nd 2008	20	kton	(Gorman 2009)
multiplier (high)	100	-	calculated
multiplier (low)	200	-	calculated

### *Stationary fuel cells for power generation*

If hydrogen is used for the transmission of energy a substantial part of this should be transformed again to electricity. This can be done by combustion of hydrogen which in power plants that are comparable to current natural gas fired power plants. This option would thus not be confronted with additional material constraints. However, it would be more efficient to use fuel cells instead. This would mean that stationary fuel cells would be used to convert hydrogen into electricity. Large stationary fuel cells are small compared to current power plants. This means that power generation will be distributed in which case the power grid would only be used for local distribution. There are different types of stationary fuel cells: Phosphoric-acid fuel cells (PAFC), Molten-carbonate fuel cells (MCFC) and solid oxide fuel cell (SOFC). Although we did not quantify this it seems likely that both PAFCs and SOFCs would be confronted with resource constraints: in PAFCs platinum is used and in SOFCs lanthanum is

used. In MCFCs less scarce materials like Ni, Cr and Al can be used as electrode materials. The material constraints with in stationary fuel cells will be less severe than for the use of fuel cells in cars.

#### *Electric motors in cars*

Fuel cells in cars are used to produce electricity which is used to power the electromotors that make the wheels turn. The use of neodymium in electromotors make these motors smaller, lighter and more efficient. Reliable sources on the amount of Nd in cars have not been found. However, figures of 1 to 2 kg of neodymium for a Toyota Prius can be found in the media (e.g. (Gorman 2009)). For the 2 billion cars on the road in 2050, 2 to 4 million ton of neodymium would be needed, about 200 times the current annual production. For full electric cars using permanent magnet motors these number would be higher. However, like in wind turbines alternatives are possible. The Tesla roadster for example, uses an induction motor without permanent magnets.

Resource constraints will prohibit the scale up of the use of fuel cells in cars due to constraints in the availability of platinum. An extreme reduction of the amount of platinum per fuel cell might solve this issue. The use of other, less scarce, electrode materials might also be a way out. Stationary fuel cells needed to produce electricity in a scenario where hydrogen will be used for the transmission will run into similar, though less severe because the installed capacity can be much lower because they are used almost 24/7. Neodymium magnets are an issue in electric vehicles if permanent magnet motors will be used. However, alternatives like the induction motor that do not contain permanent motors are available.

### **3.5 Conclusions and Discussion**

The limiting factor for a renewable based hydrogen energy system does not lie in the amount of renewable energy that is available. There is plenty of wind and especially solar energy to provide for an exponentially growing world economy. However, since wind and solar energy are diffusive energy sources the infrastructure needed to harvest substantial amounts is huge. In order to limit the amount of infrastructure needed for harvesting it is important to choose locations where the circumstances for harvesting are optimal. Especially for solar energy these locations are quite often far from the locations where this energy is used. Therefore an extensive transmission network is needed in addition, to get the energy in the right place.

In this study we have shown that the choice of technology has a strong influence on the material resources that are needed in such a system. In three crucial parts of the system we describe the use of relatively scarce materials as *show stoppers* for specific technologies: rare elements in current thin film solar cells, neodymium in direct-drive permanent magnet wind turbines and electro motors, and platinum in electrolysis and fuel cells. Resource constraints prevent these technologies from upscaling to a level



that would be needed for a substantial transition to a renewable based hydrogen economy. For a worldwide conversion to renewable energy, therefore, different technologies must be chosen and/or developed that do not rely on these scarce materials. Many of these technologies are available be it sometimes at the cost of efficiency and/or economics.

Furthermore, the amount of steel that would be needed for both wind turbines and hydrogen pipelines is very high. For wind turbines, this would be 6 times the current annual production of iron for hydrogen pipelines 40%. However, if current technology stainless steel is used for the pipelines about 45 times the annual production of nickel would be needed and 5500 times the annual production of chromium. If electricity is chosen as a carrier, the amount of copper that would be needed in the system is equivalent to 70 times the current annual world production. For the wind turbines an additional 4 times the current annual world copper production would be needed. Although resource constraints for iron ore are not in sight, it would imply a major increase in worldwide energy and, in the current system, fossil fuel use. The same is true for copper. While scarcity is not an immediate issue, the energy required to mine increasingly lower ore grades could become very large, thus creating a negative spiral.

This paper is a first step in the analysis of changes in material needs as a result of energy transitions. There are some important issues that we have not addressed yet. This study focuses on the direct material needs only. The life cycle perspective is not applied. If it would be, the material needs would probably be much higher. Moreover, the energy system that we use as a starting point is an extreme case that is not predicted in any of the major energy scenarios. However, it was not our intention to sketch a realistic energy scenario for 2050 but to explore the material needs of a complete transition. A partial transition would of course decrease the material needs, as would a limitation of primary energy needs. However, we also made some major simplifications that underestimate material needs. For example the intermittent nature of solar and wind as energy sources will make some sort of buffering an essential part of the energy system. In the large scale energy system perspective that we have used here, this could be done for example by high capacity east-west power lines that would compensate for the day-night rhythm of PV. These cables would have to be much longer than the ones we have assumed here. which would decrease the efficiency of the system as a whole. If hydrogen would be the energy carrier of choice for transmission, intermittency might be buffered by allowing changes in pressures in the pipelines.

Further in-depth analysis is needed to assess the material needs of a transition to a renewable based energy systems. Including the life cycle perspective could be a means to detect mechanisms that have a potential to make the system spiralling out of control. The increasing use of energy to produce the materials required to generate energy could be such a mechanism. Another large research area is introducing the

dynamics of actual transformations. This will lead to more detailed and realistic estimations of material needs through time in different scenarios.

Another issue is the environmental aspects of these energy systems. The systems are designed to reduce greenhouse gas emissions. A first relevant question would be, if this is indeed the case. If material requirements explode, so might the use of fossil fuels in build-up phase of the system. But even if a substantial reduction of greenhouse gas emissions would be reached, still there may be side-effects in other areas of environmental impacts, such as land and water use, emissions of toxic substances etc.. These side-effects need to be investigated as well, in order to be able to have a balanced assessment of alternative energy systems.

Finally, A sustainability approach is not limited to scarcity and environmental impacts, but also includes economic and social aspects. These have not been addressed in this paper, but should be addressed in a more comprehensive sustainability assessment.

## References

- ABB. 2009. HVDC delivering success. Accessed November 12 2009 2009.
- Andersson, B. A. 2001. Material Constraints on Technology Evolution: The Case of Scarce Metals and Emerging Energy Technologies thesis, Department of Physical Resource Theory, Chalmers University of Technology and Göteborg University, Göteborg.
- Andersson, B. A. and S. Jacobsson. 2000. Monitoring and assessing technology choice: the case of solar cells. *Energy Policy* 28(14): 1037-1049.
- Andersson, B. A., C. Azar, J. Holmberg, and S. Karlsson. 1998. Material constraints for thin-film solar cells. *Energy* 23(5): 407-411.
- Armstrong, F. A., N. A. Belsey, J. A. Cracknell, G. Goldet, A. Parkin, E. Reisner, K. A. Vincent, and A. F. Wait. 2009. Dynamic electrochemical investigations of hydrogen oxidation and production by enzymes and implications for future technology. *Chemical Society Reviews* 38(1).
- Ball, M., W. Weindorf, and U. Bünger. 2009. Hydrogen Distribution In *The Hydrogen Economy, Opportunities and Challenges* edited by M. Ball and M. Wietschel. New York: Cambridge University Press.
- Basslink. 2002. *Response by Basslink Pty Ltd to Draft Report of the Basslink Joint Advisory Panel. Submission 2 of 3, Proposed Metallic Return*. Melbourne: Basslink.
- Blazek, C. F., R. T. Biederman, F. S. E, and W. Jasionowski. 1992. Underground Storage and Transmission of Hydrogen. In *Third Annual US Hydrogen Meeting*. Washington, DC.
- Bossel, U., B. Eliasson, and G. Taylor. 2003. The future of the hydrogen economy Bright or Bleak. Energy and the Hydrogen Economy. In *2003 Fuel Cell Seminar*. European Fuel Cell Forum.
- Bungay, H. R. 2004. Confessions of a bioenergy advocate. *Trends in Biotechnology* 22(2): 67-71.
- CRC. 2009. *Handbook of Physics and Chemistry*: Taylor and Francis Group, LLC.
- Czisch, G. 2004. *Least-Cost European/Trans-European Electricity Supply Entirely with Renewable Energies*. Kassel, Germany: ISET, Universität Kassel.
- Doornbosch, R. and R. Steenblik. 2007. *Biofuels: is the cure worse than the disease?* . SG/SD/RT(2007)3. Paris: OECD.
- European Commission. 2003. Hydrogen Energy and Fuel Cells; A vision of our future. Brussels: European Commission.

- European Commission. 2005. *Biomass, green energy for Europe*. EUR 21350. Directorate-General for Research, Directorate J – Energy, Unit 3 – New and Renewable Energy Sources.
- Faaß, R. 2001. *Flugzeuge mit wasserstoffantrieb (Aircraft with hydrogen propulsion)*. Hamburg: Airbus Deutschland.
- Frischknecht R., A. H.-J., Doka, G., Dones, R., Heck T., Hellweg, S., Hirschier, R., Jungbluth N., Nemecek, T., Rebitzer, G., Spielmann, M. 2007. *Overview and Methodology. Final report ecoinvent v2.0 No. 1*. Duebendorf, CH.: Swiss Centre for Life Cycle Inventories.
- Fthenakis, V. 2009. Sustainability of photovoltaics: The case for thin-film solar cells. *Renewable and sustainable energy reviews* 13(9): 2746-2750.
- Fujishima, A. and K. Honda. 1972. Electrochemical Photolysis of Water at a Semiconductor Electrode. *Nature* 238(5358): 37-38.
- Gordon, R. B., M. Bertram, and T. E. Graedel. 2006. Metal stocks and sustainability. *Proceedings of the National Academy of Sciences of the United States of America* 103(5): 1209-1214.
- Gorman, S. 2009. As hybrid cars gobble rare metals, shortage looms. *Reuters* [Los Angeles], August 31, 2009.
- Grätzel, M. 2001. Photoelectrochemical cells. *Nature* 414(6861): 338-344.
- Green, M. A., K. Emery, D. L. King, Y. Hishikawa, and W. Warta. 2006. Solar cell efficiency tables (version 28). *Progress in Photovoltaics: Research and Applications* 14(5): 455-461.
- Hatch, G. P. 2008. Going Green: The Growing Role of Permanent Magnets in Renewable Energy Production and Environmental Protection. In *Magnetics 2008* Denver (CO), USA.
- Hoogwijk, M., A. Faaij, R. van den Broek, G. Berndes, D. Gielen, and W. Turkenburg. 2003. Exploration of the ranges of the global potential of biomass for energy. *Biomass and Bioenergy* 25(2): 119-133.
- Hoogwijk, M. M. 2004. On the global and regional potential of renewable energy sources. thesis, Universiteit Utrecht, Utrecht, The Netherlands.
- IEA. 2002. *Potential for Building Integrated Photovoltaics* Paris: IEA.
- IEA. 2005. *Prospects for hydrogen and fuel cells*. Paris, France: OECD/IEA.
- IEA. 2007. *World Energy Outlook 2007. China and India Insights*. Paris, France.: OECD/IEA
- IMAGE team. 2001. *The IMAGE 2.2 implementation of the SRES scenarios: A comprehensive analysis of emissions, climate impacts in the 21st century*. RIVM CD-ROM publication 481508018. Bilthoven, The Netherlands: National Institute for Public Health and the Environment.
- International Energy Agency. 2008. *Energy Technology Perspectives*. Paris: EIA.
- Kaltschmitt, M. and G. A. Reinhardt. 1997. *Nachwachsende Energieträger. Grundlagen, Verfahren, ökologische Bilanzierung*. : Vieweg Verlagsgesellschaft.
- Kanan, M. W. and D. G. Nocera. 2008. In Situ Formation of an Oxygen-Evolving Catalyst in Neutral Water Containing Phosphate and Co<sup>2+</sup>. *Science* 321(5892): 1072-1075.
- Kelly, T. D. and H. E. Hilliard. 2003. *Platinum-Group Metals Statistics*. USGS.
- Khaselev, O. and J. A. Turner. 1998. A Monolithic Photovoltaic-Photoelectrochemical Device for Hydrogen Production via Water Splitting. *Science* 280(5362): 425-427.
- Kleijn, R., E. van der Voet, and H. A. Udo de Haes. 2008. The need for combining IEA and IE tools: The potential effects of a global ban on PVC on climate change. *Ecological Economics* 65(2): 266-281.
- Knies, G. and A. Bennouna. 2002. Potential of Moroccan renewable energies for the defence against climate change. . In *FIER 2002*. Tetuoan, Morocco.
- Koroneos, C., A. Dompros, G. Roumbas, and N. Moussiopoulos. 2004. Life cycle assessment of hydrogen fuel production processes. *International Journal of Hydrogen Energy* 29(14): 1443-1450.

- Kuik, G. A. M. v. 2001. Are wind turbines growing too fast? In *European Wind Energy Conference*. Copenhagen.
- Lamle, S. E., K. A. Vincent, L. M. Halliwell, S. P. J. Albracht, and F. A. Armstrong. 2003. Hydrogenase on an electrode: a remarkable heterogeneous catalyst. *Dalton Transactions*(21).
- Leijendeckers, P. H. H., J. L. A. Jansen, and D. C. E. M. Keukeleere. 2008. *Energie zakboek (Energy Handbook)*. Doetinchem, The Netherlands: Reed Business.
- Meijer, A., M. A. J. Huijbregts, J. J. Schermer, and L. Reijnders. 2003. Life-cycle assessment of photovoltaic modules: Comparison of mc-Si, InGaP and InGaP/mc-Si solar modules. *Progress in Photovoltaics: Research and Applications* 11(4): 275-287.
- Melis, A. and T. Happe. 2001. Hydrogen Production. Green Algae as a Source of Energy. *Plant Physiology* 127(3): 740-748.
- Odell, P. R. 2004. *Why carbon fuels will dominate the 21st century's global energy economy*. Brentwood Essex, UK: Multi-Science Publishing Co. Ltd.
- Ogden, J. M. 1999. Prospects for building a hydrogen energy infrastructure. *Annual Review of Energy and the Environment* 24(1): 227-279.
- Okano, K. 2003. Hydrogen and fuel cell activities in Japan. In *PATH hydrogen workshop*. Mexico.
- Oxera. 2003. *The impact of average zonal transmission losses applied throughout Great Britain*. Oxford: Department of Trade and Industry.
- Polinder, H., F. A. van der Pijl, G.-J. de Vilder, and P. J. Tavner. 2006. Comparison of Direct-Drive and Geared Generator Concepts for Wind Turbines. *IEEE Transactions on energy conversion* 21(3): 8.
- Porter, K. E. and D. I. L. Edelman. 2003. *Copper Statistics*. USGS.
- Prather, M. J. 2003. An Environmental Experiment with H<sub>2</sub>? *Science* 302(5645): 581-582.
- Råde, I. and B. A. Andersson. 2001. *Platinum group metal resource constraints for fuel-cell vehicles*. PRT Report 2001:01. Göteborg, Sweden: Chalmers University of Technology and Göteborg University.
- Schultz, M. G., T. Diehl, G. P. Brasseur, and W. Zittel. 2003. Air Pollution and Climate-Forcing Impacts of a Global Hydrogen Economy. *Science* 302(5645): 624-627.
- Seboldt, W. 2004. Space- and Earth-based solar power for the growing energy needs of future generations. *Acta Astronautica* 55(3-9): 389-399.
- Shell. 2008. *Shell energy scenarios to 2050*. Shell International BV. .
- Sørensen, B. 2000. *Renewable Energy, Its physics, engineering, use, environmental impacts, economy and planning aspects*. . London: Academic Press.
- Sørensen, B. 2007. *Renewable Energy, conversion, transmission and storage*. . London: Academic Press.
- Statnett. 2002. *NorNed cable, HVDC project*. Oslo, Norway: Statnett.
- Tennet. 2007. *NorNed, Europe's link to the future*. Arnhem, The Netherlands: Tennet.
- The Physics Factbook. 2009. The Physics Factbook. <http://hypertextbook.com/facts/>. Accessed.
- The White House. 2004. A new generation of American innovation, edited by T. W. House: The White House.
- Tilton, J. E. and G. Lagos. 2007. Assessing the long-run availability of copper. *Resources Policy* 32(1-2): 19-23.
- Tromp, T. K., R.-L. Shia, M. Allen, J. M. Eiler, and Y. L. Yung. 2003. Potential Environmental Impact of a Hydrogen Economy on the Stratosphere. *Science* 300(5626): 1740-1742.
- U.S. Department of Energy. 2004. *Hydrogen posture plan. An integrated research, development and demonstration plan*. U.S. Department of Energy,.
- UK Department for Transport. 2003. *Platinum and hydrogen for fuel cell vehicles*. UK Department for Transport.
- UNEP. 2002. *Global Environment Outlook 3*. London: Earthscan Publications Ltd.

- USGS. 2009a. *Mineral Commodity Summaries 2009*. USGS.
- USGS. 2009b. *Minerals Yearbook 2007*. USGS.
- Wadia, C., A. P. Alivisatos, and D. M. Kammen. 2009. Materials Availability Expands the Opportunity for Large-Scale Photovoltaics Deployment. *Environmental Science & Technology* 43(6): 2072-2077.
- Weindorf, W., U. Bünger, and J. Schindler. 2003. *Comments on the paper by Baldur Eliasson and Ulf Bossel "The Future of the Hydrogen Economy: Bright or Bleak"*. Ottobrun, Germany L-B-Systemtechnik GmbH.
- Wietschel, M., M. Ball, and P. Seydel. 2009. Hydrogen Today. In *The Hydrogen Economy, Opportunities and Challenges*, edited by M. Ball and M. Wietschel. New York: Cambridge University Press.
- Williams, R. H. 2001. Nuclear and Alternative Energy Supply Options for an Environmentally Constrained World; A Long-Term Perspective. In *Nuclear Control Institutes 20th Anniversary Conference Nuclear Power and the Spread of Nuclear Weapons: Can We Have One Without the Other?*. Center for Energy and Environmental Studies, Princeton University
- Worldwatch Institute. 2002. *Vital Signs 2002, Database CD-ROM*: Worldwatch Institute.
- Wright, S. D., A. L. Rogers, J. F. Manwell, and A. Ellis. 2002. *Transmission options for offshore wind farms in the United States*. Amherst, USA: University of Massachusetts.
- Zittel, W. and M. Altmann. 1996. Molecular hydrogen and water vapour emissions in a global hydrogen energy economy. Paper presented at 11th World Hydrogen Energy Conference, June 1996, Stuttgart, Germany.

