

MAS NMR study of the photoreceptor phytochrome Rohmer, T.

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Appendices

Appendix A

Figure A.1: Contour plot of the ${}^{1}H^{-13}C$ HSQC (black) and ${}^{1}H^{-13}C$ HMBC (grey) NMR
dingler completion greater of BCP in mathemal d, recorded in a magnetic field of 14.1 T at dipolar correlation spectra of PCB in methanol-d⁴ recorded in a magnetic field of 14.1 T at 277 K. The HSQC assignments of correlations are indicated.

Figure A.2: Contour plot of the 2D ¹³C-¹³C RFDR ($t_{\text{mix}} = 3.9$ ms, black) and PDSD ($t_{\text{mix}} = 2.4$ ms, grey) MAS NMR spectra of $u_{-1}^{13}C$, ¹⁵N]-PCB-Cph1 Δ 2. The connectivity networks of the propionic acid side-chains are indicated by dotted lines.

Figure A.3: Contour plot of the ${}^{1}H_{-}^{13}C$ FSLG NMR dipolar correlation spectra of u-
 $I^{13}C_{-}^{15}$ NL PCP recorded at 277 K in a megnetic field of 17.6 T using a spinning frequency $[{}^{13}C, {}^{15}N]$ -PCB recorded at 277 K in a magnetic field of 17.6 T using a spinning frequency of 12 kHz and CP contact times of 512 μ s (**A**) and 1024 μ s (**B**).

Figure A.4: Contour plots of the 2D ${}^{1}H^{-15}N$ heteronuclear dipolar correlation spectrum
of ${}^{13}C {}^{15}N1$ PCP recorded at 277 K in a field of 17.6 T using a spinning frequency of 7 of u- $\binom{13}{0}$, $\binom{15}{1}$ -PCB recorded at 277 K in a field of 17.6 T using a spinning frequency of 7 kHz.

Appendix B

Figure B.1: 1D ¹³C CP/MAS NMR spectra of ¹³C5-PCB-phyA65 in the Pr (**A**) and Pfr (**P**) states recorded at 0.4 T, 10 kHz and 242 K. The actorisks indicate the ¹³C recognes of (**B**) states recorded at 9.4 T, 10 kHz and 243 K. The asterisks indicate the ¹³C response of the situal present in the buffer the glycol present in the buffer.

Figure B.2: Contour plot of $2D^{-1}H^{-13}C$ heteronuclear dipolar correlation spectra of the mathing hydrogen obtained from u $1^{13}C^{-15}$ NJ PCP Coh1A2 in the the Pr (blog) and methine bridge region obtained from u- $\left[{}^{13}C, {}^{15}N \right]$ -PCB-Cph1 $\Delta 2$ in the the Pr (black) and Pfr (grey) states at a magnetic field of 9.4 T, 10 kHz and 243 K. The asterisk indicates the position of protein backbone signals in natural abundance.

Figure B.3: Voigt deconvolution of the ¹⁵N CP/MAS NMR spectra of u -[¹³C,¹⁵N]-PCB-Cph1 Δ 2 in the Pr (**A**, 80000 scans) and Pfr (**B**, 135000 scans) recorded in a magnetic field of 17.6 T at 243 K using a spinning frequency of 8 kHz. The upper part compares the experimental NMR spectrum with the sum of the Voigt fits. The lower part gives the individual Voigt fits (dashed lines) and the residual spectra. The asterisk indicates the 15 N response of the protein backbone in natural abundance.

Figure B.4: ¹⁵N CP/MAS NMR spectra of Cph1 Δ 2 (**A**) and *phyA*65 (**B**) containing an u-[¹³C,¹⁵N]-PCB in the Pr and Pfr states recorded in a magnetic field of 17.6 T at 243 K using a spinning frequency of 8 kHz. The ^{15}N response of the protein backbone in natural abundance is indicated by an asterisk.

Figure B.5: Contour plot of the 2D ¹³C-¹³C DARR NMR spectra of u-[¹³C,¹⁵N]-PCBphyA65 in the Pr (black) and Pfr states (grey). The spectra were recorded with proton mixing times of 5 ms, at 243 K and with a spinning frequency of 9 kHz.

Appendix C

Figure C.1: Voigt deconvolution of the ¹⁵N CP/MAS NMR spectra of u-[¹³C,¹⁵N]-PCB-Cph1 Δ 2 in the Lumi-F (**A**) and Meta-F (**B**) recorded in a magnetic field of 17.6 T at 173 K and 203 K, respectively, using a spinning frequency of 8 kHz. The upper part compares the experimental NMR spectrum with the sum of the Voigt fits. The lower part gives the individual Voigt fits (dashed lines) and the residual spectra.

Chemical synthesis of ¹⁵N21-PCB-Cph1Δ**2**

 15 N21-PCB was synthesized by Dr. Bongards (Max-Planck-Institut für Bioanorganische Chemie, Mülheim an der Ruhr, Germany). The synthesis of $4E$ -4-ethylidene-3-methyl-5-thioxo(15 N)pyrrolidin-2-one (6) follows the synthesis scheme presented in Figure C.2.

Figure C.2: Synthesis scheme applied for the the chemical synthesis of $(4E)$ -4-ethylidene- 3 -methyl-5-thioxo(15 N)pyrrolidin-2-one (**6**).

 $2-(4-\text{Methoxybenzyl})({}^{15}\text{N})-1H\text{-isoindole-1,}3(2H)\text{-dione (1): A dispersion}$ of 15 N-labeled potassium phthalimide (10.00 g, 53.7 mmol), 1-(chloromethyl)-4-methoxybenzene (6.1 mL, 44.8 mmol), and 18-crown-6 (1.18 g, 4.5 mmol) in toluene (58 mL) was stirred under argon at 100 \degree C for 5 h. After cooling to ambient temperature, water (150 mL) was added to the mixture. The resulting phases were separated, the aqueous phase was extracted with dichloromethane $(4 \times 50 \text{ mL})$ and the combined organic phases were dried over $Na₂SO₄$. Evaporation of the solvent and drying of the remaining pale yellow solid under reduced pressure yielded the reaction product **1** (12.01 g, 44.8

mmol, 100 %) which could be used in the following reaction without further purification.

1-(4-Methoxyphenyl)methan(^{15}N)amine (2): A dispersion of 1 (6.41 g, 23.9 mmol) and hydrazine hydrate (51 % hydrazine, 96 mL, 1.0 mol) in methanol (1.4 L) was refluxed for 5 h. After evaporation of the solvent under reduced pressure, the remaining oil was dissolved in dichloromethane (500 mL), washed with NaOH (1 M, 5×100 mL) and dried over Na₂SO₄. Evaporation of the solvent under reduced pressure yielded the reaction product **2** (3.20 g, 23.2 mmol, 97 %) as a yellow, clear oil.

In general, the preparation of **6** follows the synthetic route of the unlabeled ring *A* compound established by Kakiuchi et al. [139]. Transformation of the ¹⁵N21-labeled ring *A* compound **6** to the full tetrapyrrole followed published procedures [140, 141].