



Universiteit  
Leiden  
The Netherlands

## Structural and functional models for [NiFe] hydrogenase

Angamuthu, R.

### Citation

Angamuthu, R. (2009, October 14). *Structural and functional models for [NiFe] hydrogenase*. Retrieved from <https://hdl.handle.net/1887/14052>

Version: Corrected Publisher's Version

License: [Licence agreement concerning inclusion of doctoral thesis in the Institutional Repository of the University of Leiden](#)

Downloaded from: <https://hdl.handle.net/1887/14052>

**Note:** To cite this publication please use the final published version (if applicable).

# **Structural and Functional Models for [NiFe] Hydrogenase**

**Raja Angamuthu**

**Cover-page illustration:**

Crystal structure of  $[\text{Ni}_6(\text{cpss})_{12}]$  (Chapter 6). Ni, green; S, red; Cl, yellow; C, gray.

**Printed by:**

Ridderprint BV, Ridderkerk, The Netherlands

# **Structural and Functional Models for [NiFe] Hydrogenase**

PROEFSCHRIFT

ter verkrijging van de graad van Doctor aan de Universiteit Leiden,

op gezag van Rector Magnificus prof. mr. P. F. van der Heijden,

volgens besluit van het College voor Promoties

te verdedigen op woensdag 14 oktober 2009

klokke 16.15 uur

door

**Raja Angamuthu**

geboren te Karur (Tamil Nadu), India in 1980

## **Promotiecommissie**

**Promotor** Prof. Dr. J. Reedijk

**Copromotor** Dr. E. Bouwman

**Overige leden** Prof. Dr. M. Schröder (The University of Nottingham, UK)

Prof. Dr. M. Fontecave (Université Joseph Fourier, Grenoble, France)

Dr. M. C. Feiters (Radboud Universiteit Nijmegen)

Prof. Dr. M. T. M. Koper

Prof. Dr. J. Brouwer

# Table of Contents

▪	List of Abbreviations	i
<b>1</b>	General Introduction	001
<b>2</b>	Ligand Design, Synthetic Procedures and Experimental Methods	033
<b>3</b>	[Ni(S <sub>4</sub> )Fe(C <sub>5</sub> H <sub>5</sub> )(CO)](PF <sub>6</sub> ) Complexes Containing Tetradentate S <sub>2</sub> S' <sub>2</sub> -donor Ligands: Synthesis, Characterization and Electrocatalytic Dihydrogen Production	051
<b>4</b>	Synthesis, Characterization and Electrocatalytic Properties of [Ni(S <sub>4</sub> )Fe(C <sub>5</sub> H <sub>5</sub> )(CO)](PF <sub>6</sub> ) Complexes Containing Bidentate SS'-donor Ligands	071
<b>5</b>	Heterodinuclear [NiRu] Complexes Comprising Ruthenium Bis-Bipyridine: Synthesis, Characterisation and Electrocatalytic Dihydrogen Production	087
<b>6</b>	Hexanuclear (Ni <sub>6</sub> -)Metallacrown as Functional Model of [NiFe] Hydrogenase	103
<b>7</b>	A Molecular Cage of Ni(II) and Cu(I) Resembling the Active Site of Ni-Containing Enzymes	121
<b>8</b>	Light-Induced C–S Bond Cleavage in a Nickel Thiolate Complex: Relevance to the Function of Methyl Coenzyme M Reductase (MCR)	131
<b>9</b>	Summary, Conclusions and Future Perspectives	143
▪	Samenvatting (Summary in Dutch)	151
▪	Curriculum Vitae (English)	156
▪	Curriculum Vitae (Tamil)	157
▪	Publications	158
▪	Nawoord (Afterword)	160

a.u.	Arbitrary unit
ACN	Acetonitrile
APT	Attached Proton Test
b	Broad
bpy	2,2'-Bipyridine
CoM	Coenzyme M; 2-thioethane sulfonate
COSY	Correlation Spectroscopy
CV	Cyclic voltammetry; cyclic voltammogram
Cys	Cysteine
d	Doublet
dedtc	Diethyldithiocarbamate
DMF	<i>N,N</i> -Dimethylformamide
dpa	Dipicolylamine
dppe	1,2-Bis(diphenylphosphino)ethane
$E_{pa}$	Anodic potential; oxidation potential
$E_{pc}$	Cathodic potential; reduction potential
EPPG	Edge plane pyrolytic graphite (electrode)
eq.	Equivalent
ESI-MS	Electrospray ionization mass spectrometry
Et	Ethyl
FTIR	Fourier transform infra red
GC	Glassy carbon
Glu	Glutamic acid
GSH	Glutathione
H <sub>2</sub> ase	Hydrogenase
H <sub>2</sub> bdt	Benzene-1,2-dithiol
H <sub>2</sub> bme*-daco	1,5-bis(mercaptoethyl)-1,5-diazacyclooctane
H <sub>2</sub> pdt	Propane-1,3-dithiol
H <sub>2</sub> tpdt	2-Thiopropane-1,3-dithiol
Hacac	Acetylacetone, 2,4-pentanedione
HCp	1,3-Cyclopentadiene
HER	Dihydrogen evolution reaction
HG-GSH	Hemithioacetal
His	Histidine
HS-HPT	<i>N</i> -(7-mercaptoheptanoyl)- <i>O</i> -phospho-L-threonine; coenzyme B

$J$	Coupling constant
LMCT	Ligand-to-metal charge transfer
m	Multiplet in NMR; medium in IR
$m/z$	Ratio of mass upon charge
$M_d$	Distal metal
Me	Methyl
MeCoM	Methyl-coenzyme M; 2-(methylthio)ethanesulfonate
MLCT	Metal-to-ligand charge transfer
$M_p$	Proximal metal
NMR	Nuclear magnetic resonance
NOESY	Nuclear Overhauser Effect Spectroscopy
$OTf^-$	Trifluoridomethanesulfonate
PEM	Proton exchange membrane
Ph	Phenyl
$PMe_3$	Trimethylphosphine
$PPh_3$	Triphenylphosphine
ppm	Parts per million
ROESY	Rotating-Frame NOE Spectroscopy
s	Singlet in NMR; strong in IR
SCE	Standard calomel electrode
SHE	Standard hydrogen electrode
t	Triplet
$^tBu$	<i>tertiary</i> -Butyl
TEA·HCl	Triethylamine hydrochloride
THF	Tetrahydrofuran
TMS	Tetramethylsilane
tmtu	1,1,3,3-tetramethyl-2-thiourea
TOCSY	Total Correlation Spectroscopy
tpa	Tripicolylamine
TsOH·H <sub>2</sub> O	<i>para</i> -Toluenesulfonic Acid monohydrate
UV-Vis	Ultra violet and visible spectroscopy
Val	Valine
w	Weak
$\delta$	Chemical shift
$\tau$	Trigonality index



