

Oxidation catalysis on Pt and Au : complexity of simple chemistry Spronsen, M.A. van

Citation

Spronsen, M. A. van. (2016, June 29). Oxidation catalysis on Pt and Au : complexity of simple chemistry. Casimir PhD Series. Retrieved from https://hdl.handle.net/1887/41415

Version: Not Applicable (or Unknown)

License: License agreement concerning inclusion of doctoral thesis in the

Institutional Repository of the University of Leiden

Downloaded from: https://hdl.handle.net/1887/41415

Note: To cite this publication please use the final published version (if applicable).

Cover Page



Universiteit Leiden



The handle http://hdl.handle.net/1887/41415 holds various files of this Leiden University dissertation

Author: Spronsen, Matthijs A. van

Title: Oxidation catalysis on Pt and Au : complexity of simple chemistry

Issue Date: 2016-06-29

Propositions

Accompanying the PhD thesis "Oxidation catalysis on Pt and Au: Complexity of simple chemistry"

- I. It is not appropriate to classify the structures formed on Pt(111) under oxygenrich reaction conditions as either chemisorption structures or oxides, as they are structurally, electronically, and most importantly, chemically different from both. Chapters 2 & 3 of this thesis
- II. The surface oxides discovered on Pt(111) and Pt(110) contain the most reactive oxygen atoms and, therefore, are most relevant for catalysis.
 Chapters 3 & 4 of this thesis
- III. Stepped surfaces are usually described by the facets forming the terraces and the steps. However, for the anisotropic (110) step, the direction of the step within this (110) facet could be more significant than the structure of the terrace, in binding of interacting adsorbates.
 Chapter 4 of this thesis
- IV. In the water-promoted oxidation of carbon monoxide over gold catalysts, hydroxides play a minor role.
 Chapter 5 of this thesis
- V. Seriani and Mittendorfer overestimate the significance of the (110) facet in their predictions of the shape of a platinum nanoparticle under oxygen-rich conditions, because they do not consider the proper surface oxides.
 N. Seriani and F. Mittendorfer: "Platinum-group and noble metals under oxidizing conditions", J. Phys.: Condens. Matter. 20, 184023 (2008)
- VI. The large difference in conditions between surface science experiments and catalytic reactions has been identified as the "pressure gap". However, this is misleading as all its implications are more related to time than to pressure. P. Stoltze and J.K. Nørskov: "Bridging the "Pressure Gap" between Ultrahigh-Vacuum Surface Physics and High-Pressure Catalysis", Phys. Rev. Lett. 55, 2502–2505 (1985)
- VII. It is unclear how well the steps of a vicinal surface can serve as a model for the facet edges of a nanoparticle catalyst.
 B. Lang, R.W. Joyner, and G.A. Somorjai: "Low energy electron diffraction studies of chemisorbed gases on stepped surfaces of platinum", Surf. Sci. 30, 454–474 (1972)

- VIII. Recently developed in situ and operando instrumentation has led to highly specialized equipment, which, in turn, determines the frame of reference of the conclusions obtained with this equipment.
 A. Ziegler, H. Graafsma, X.F. Zhang, and J.W.M. Frenken (Eds.): "In-situ Materials Characterization", Springer Series in Materials Science (2014)
- IX. X-ray photoelecton spectroscopy is like playing guitar; it is easy to start doing, but incredibly hard to master.
- X. The shift of focus in scientific publishing from communication of results and reasoning towards ranking scientists, has significantly degraded the quality of the published work.

Matthijs A. van Spronsen Leiden, 28-06-2016