

Fluorescence correlation spectroscopy on electron transfer reactions : probing inter- and intramolecular redox processes Sen, S.

Citation

Sen, S. (2016, June 30). Fluorescence correlation spectroscopy on electron transfer reactions : probing inter- and intramolecular redox processes. Casimir PhD Series. Retrieved from https://hdl.handle.net/1887/40761

Version:	Not Applicable (or Unknown)
License:	<u>Licence agreement concerning inclusion of doctoral thesis in the</u> <u>Institutional Repository of the University of Leiden</u>
Downloaded from:	https://hdl.handle.net/1887/40761

Note: To cite this publication please use the final published version (if applicable).

Cover Page



Universiteit Leiden



The handle <u>http://hdl.handle.net/1887/40761</u> holds various files of this Leiden University dissertation.

Author: Sen, S. Title: Fluorescence correlation spectroscopy on electron transfer reactions : probing inter- and intramolecular redox processes Issue Date: 2016-06-30

Stellingen

Behorend bij het proefschrift

"Fluorescence Correlation Spectroscopy on Electron Transfer Reactions

Probing Inter-and Intramolecular Redox Processes"

I. While studying a fluorescently labeled protein under redox conditions using fluorescence correlation spectroscopy, blinking of the fluorescent dye must be considered in analyzing the autocorrelation data.

Chapter 3 of this thesis

II. For labeled redox enzymes, the photo-induced electron transfer between dye and redoxactive centres should be taken into account together with fluorescence-resonance energytransfer for quantitative analysis of autocorrelation data.

Chapter 4 of this thesis

- III. To distinguish between the intramolecular and intermolecular electron-transfer reactions in labeled oxido-reductases using fluorescence correlation spectroscopy, the viscosity of the solution and the concentration of the reductant/oxidant must be varied. Chapter 4 of this thesis
- IV. In fluorescently labeled oxido-reductases, photo-induced intramolecular electron transfer between the label and the active center is not restricted to situations where the label is in Van-der-Waals contact with the redox center.

Chapter 4 and Chapter 5 of this thesis

V. Morishima and coworkers documented the occurrence of electron-transfer reactions in a single dye-labeled protein molecule. Besides intramolecular electron-transfer between the

dye and the redox center, intermolecular electron-transfer processes should have been considered as well.

Isao Morishima et al., J. Am. Chem. Soc., 127 2098 (2005).

- VI. Tinnefeld and coworkers introduced a reducing and oxidizing system to minimize photobleaching and blinking of a dye (ATTO647N). This method is not relevant for ATTO655 due to its different chemical structure and photophysical properties.
 Philip Tinnefeld et al., Angew. Chem. Int. Ed., 47 5465 (2008).
- VII. In contrast to the conclusions drawn by Cordes *et al.* and Grimm *et al.*, the ideal fluorescent probe still does not exist.
 Thorben Cordes et al., Nat. Methods, 9 426 (2012); Jonathan B Grimm et al., Nat. Methods, 12 244 (2015).
- VIII. The standard calibration technique proposed by Rüttinger *et al.* for the determination of effective volume by fluorescence correlation spectroscopy is not applicable in viscous solvent.

Steffen Rüttinger et al., J. Microscopy, 232 343 (2008).

- IX. Criticism may not be agreeable, but it is necessary to aid and discover new talent.
- X. To facilitate free and open exchange of new ideas for the growth of our society, the connection between academic institutions/research centers and private companies/pharmaceuticals is indispensable.

Saptaswa Sen Leiden, 30-06-2016